



Nuclear magnetic resonance study of anion and cation dynamics in CsSiH₃

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ABSTRACT

In order to study the dynamical properties of cesium silanide CsSiH₃, we have measured the ¹H and ¹³³Cs nuclear magnetic resonance (NMR) spectra and spin-lattice relaxation rates in this compound over the temperature range of 5–354 K. The results of the ¹H NMR measurements indicate that [SiH₃][−] anions in CsSiH₃ retain unusually high reorientational mobility down to low temperatures. In particular, a significant narrowing of the proton NMR spectrum due to [SiH₃][−] reorientations is observed in the range ~10–14 K. The order-disorder (β→α) phase transition accompanied by the strong acceleration of [SiH₃][−] reorientations and by the change in the reorientational mechanism is observed above 200 K; according to the ¹H spin-lattice relaxation data, this transition appears to be gradual. For the high-temperature disordered α-phase, the activation energy for [SiH₃][−] reorientations is found to be 41(4) meV. The ¹³³Cs NMR results are consistent with the onset of diffusive motion of Cs⁺ cations at the frequency scale of ~10⁴ s^{−1} above 300 K.

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1. Introduction

Alkali-metal silanides MSiH₃ (M = K, Rb, Cs) consisting of M⁺ cations and pyramidal [SiH₃][−] anions have received significant recent attention as potential hydrogen-storage materials [1–5]. These compounds can be prepared by reversible hydrogenation of Zintl phases MSi with the equilibrium hydrogen gas pressure of 0.1 MPa near 410 K [3]. Above 300 K, the crystal structure of all three MSiH₃ compounds is cubic (α-phase, space group *Fm* $\bar{3}$ *m*) with NaCl-type arrangement of M⁺ cations and orientationally disordered [SiH₃][−] anions. At lower temperatures, these disordered α-phases are transformed into the ordered β-phases with orthorhombic (β-KSiH₃, space group *Pnma*) or monoclinic (β-RbSiH₃ and β-CsSiH₃, space group *P2*₁/*m*) structures [3,6,7]. The corresponding entropy-driven α↔β phase transitions are found to be reversible, but hysteretic [7,8].

It is known that in many complex hydrides, the reorientational motion of complex anions (such as [BH₄][−], [AlH₄][−], [NH₂][−], [B₁₂H₁₂]^{2−}) [9–13] strongly contributes to the balance of energies determining the thermodynamic stability of these compounds; therefore, information on the anion reorientational dynamics is important for understanding the fundamental properties of complex hydrides. The first evidence of the reorientational motion of [SiH₃][−] anions in alkali-metal silanides was obtained in the wide-line nuclear magnetic resonance (NMR) study [14] that revealed the narrowing of the ¹H NMR spectrum of KSiH₃ above 170 K. The motional narrowing of the ²D NMR spectrum was also observed for deuterium-substituted KSiD₃ and RbSiD₃ above 200 K [8]. Recent quasielastic neutron scattering (QENS) measurements [7,8,15,16] have shown that phase transitions from the ordered β-phases of MSiH₃ to the disordered α-phases are accompanied by the strong acceleration of anion reorientations. Typical reorientational jump rates for α-phases of all MSiH₃ compounds just above room temperature are of the order of 10¹² s^{−1}. It is interesting to note that similar order-disorder phase transitions accompanied by an acceleration of anion reorientations have been found to occur in a

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number of complex hydrides, including the family of alkali-metal *closo*-hydroborates with $[B_{12}H_{12}]^{2-}$ and $[B_{10}H_{10}]^{2-}$ anions [13,17–19]. The high-temperature disordered phases of lithium and sodium *closo*-hydroborates and their derivatives exhibit extremely high Li^+ and Na^+ ionic conductivities [19–21], which are believed to be related [22–24] to fast anion reorientations. For alkali-metal silanides, a possible relation between anion reorientations and cation diffusion has not been studied so far.

In the present work, we use nuclear magnetic resonance measurements of 1H and ^{133}Cs spectra and spin-lattice relaxation rates to investigate both the reorientational motion of $[SiH_3]^-$ anions and the diffusive motion of Cs^+ cations in $CsSiH_3$ over wide temperature ranges. The results of this first NMR study of $CsSiH_3$ indicate that $[SiH_3]^-$ reorientations in β - $CsSiH_3$ are unusually fast, and on the frequency scale of $10^5\ s^{-1}$ they are not “frozen out” down to very low temperatures. The diffusive motion of Cs^+ cations becomes observable at the NMR frequency scale in α - $CsSiH_3$ above 300 K.

2. Experimental details

$CsSiH_3$ sample was prepared by the direct solid – gas reaction between $CsSi$ and H_2 , as described in Ref. [3]. The Zintl phase $CsSi$ was synthesized by high-temperature annealing of a 1:1.03 mixture of cesium (99.8%, Alfa Aesar [25]) and silicon (powder, –325 mesh, Sigma Aldrich). All materials were handled in an argon-filled glovebox. The mixture was arc-weld-sealed within a stainless-steel ampoule inside the glovebox. The ampoule was heated to 873 K and kept at this temperature for 48 h. Then it was cooled with the rate of 0.2 K/min down to room temperature. A slight excess of silicon was added to ensure a complete reaction of the starting cesium. The resulting $CsSi$ compound was first activated via evacuation at 473 K for 12 h and then hydrogenated at 373 K with ~5 MPa of hydrogen for 24 h. For NMR experiments, the powdered $CsSiH_3$ sample was flame-sealed in a glass tube under ~0.5 bar of nitrogen gas.

Low-field 1H NMR measurements were performed on a pulse spectrometer with quadrature phase detection at the frequencies $\omega/2\pi = 28$ and 90 MHz. The magnetic field was provided by a 2.1 T iron-core Bruker magnet. A home-built multinuclear continuous-wave NMR magnetometer working in the range 0.32–2.15 T was used for field stabilization. For rf pulse generation, we used a home-built computer-controlled pulse programmer, a PTS frequency synthesizer (Programmed Test Sources, Inc.) and a 1 kW Kalmus wideband pulse amplifier. A probehead with the sample was placed into an Oxford Instruments CF1200 continuous-flow cryostat using nitrogen or helium as a cooling agent. The sample temperature monitored by a chromel – (Au-Fe) thermocouple was stable to ± 0.1 K. High-field ^{133}Cs NMR measurements were performed on a Bruker AVANCE III 500 spectrometer at the frequency $\omega/2\pi = 65.62$ MHz. Typical values of the $\pi/2$ pulse length were 3–4 μs for both 1H and ^{133}Cs . All the NMR measurements were performed with increasing temperature and with 15 min stabilization time at each temperature prior to a measurement. Nuclear spin-lattice relaxation rates were measured using the saturation – recovery method. NMR spectra were recorded by Fourier transforming the solid echo signals (pulse sequence $\pi/2_x - t - \pi/2_y$ with $t = 20$ –22 μs).

3. Results and discussion

3.1. 1H NMR spectra and spin-lattice relaxation rates

The evolution of the proton NMR spectra for $CsSiH_3$ with temperature is shown in Fig. 1. As can be seen from this figure, the spectra exhibit a considerable narrowing with increasing temperature. Fig. 2 shows the temperature dependence of the line width

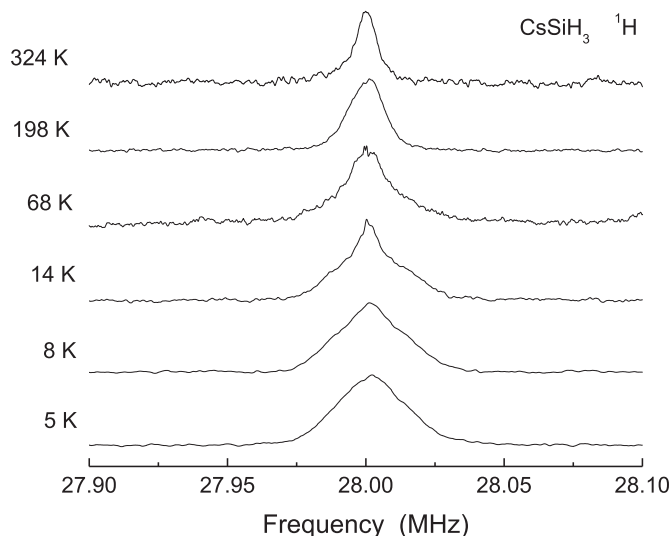


Fig. 1. Evolution of the proton NMR spectrum for $CsSiH_3$ with temperature.

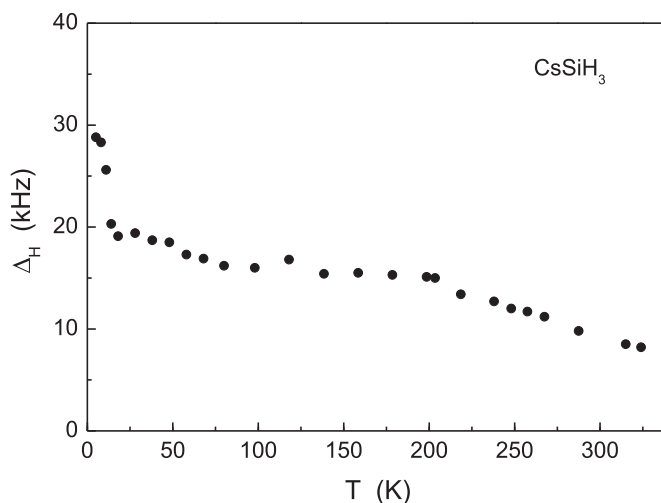


Fig. 2. Temperature dependence of the width (full width at half-maximum) of the 1H NMR spectrum measured for $CsSiH_3$ at 28 MHz.

Δ_H (full width at half-maximum). At low temperatures, the width of the 1H NMR spectrum is determined by dipole-dipole interactions between static nuclear spins; this is the ‘rigid-lattice’ line width Δ_{HR} . A rough theoretical estimate of Δ_{HR} can be obtained from the second moment calculations using the structural data for β - $CsSiH_3$ [3], see Fig. 3. The calculated ‘rigid-lattice’ second moment of the 1H NMR line for β - $CsSiH_3$ is $5.2 \times 10^9\ s^{-2}$. Assuming a Gaussian shape of the spectrum, this second moment should correspond to the Δ_{HR} value of 27.1 kHz. The experimental Δ_H value at $T = 5$ K (28.8 kHz) is close to this estimate.

With increasing temperature, the line width starts to decrease, since the onset of H jump motion leads to a partial averaging of dipole-dipole interactions. Such motional narrowing is expected to be substantial at the temperature at which the H jump rate τ^{-1} becomes nearly equal to $2\pi\Delta_{HR}$ [26]; for β - $CsSiH_3$, this characteristic frequency is about $1.7 \times 10^5\ s^{-1}$. As can be seen from Fig. 2, a considerable line narrowing in β - $CsSiH_3$ is observed already at 14 K. This means that SiH_3 groups in the cesium silanide retain high reorientational mobility down to very low temperatures. For comparison, the onset of the proton NMR line narrowing in β - $KSiH_3$

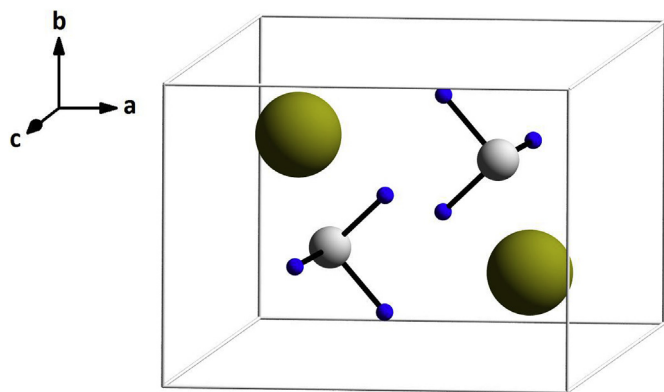


Fig. 3. Schematic view of the structure of β -CsSiH₃. Olive spheres: Cs atoms; gray spheres: Si atoms; blue spheres: H atoms. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

occurs near 170 K [14]. Recent QENS studies of MSiH₃ compounds [16] have shown that the reorientational SiH₃ motion in β -CsSiH₃ is much faster than in β -KSiH₃ and β -RbSiH₃. However, because of the limited energy resolution of QENS measurements, they can only probe the jump rates down to $\sim 4 \times 10^8 \text{ s}^{-1}$ (for β -CsSiH₃, this low limit corresponds to the temperature of about 140 K [16]). Since NMR measurements are sensitive to much slower jump rates, they allow us to trace the reorientational motion down to lower temperatures. An Arrhenius extrapolation of the QENS results for β -CsSiH₃ [16] to the low- T region indicates that the H jump rate of 10^5 s^{-1} is expected to occur near 73 K. Thus, SiH₃ reorientations are not “frozen out” at the frequency scale of 10^5 s^{-1} down to much lower temperatures than those expected on the basis of extrapolation of the QENS data. It should be noted that, although the low- T reorientational SiH₃ motion in CsSiH₃ is unusually fast, it cannot be treated in terms of the rotational tunneling [27], since the value of Δ_H at the lowest temperature corresponds to the calculated ‘rigid-lattice’ limit. In the known cases of the rotational tunneling [27], Δ_H at the lowest temperatures is considerably smaller than this ‘rigid-lattice’ value.

As can be seen from Fig. 2, in the temperature range of 100–200 K, the value of Δ_H remains nearly constant. This plateau regime is related to the fact that even when τ^{-1} is much higher than $2\pi\Delta_H$, a localized H motion (such as reorientational motion) leads to only partial averaging of the dipole-dipole interactions. The plateau value of Δ_H should depend on the mechanism of reorientations. For β -CsSiH₃, the second moment of the proton NMR line is dominated by $^1\text{H} - ^1\text{H}$ interactions within the same SiH₃ group (the ‘intramolecular’ contribution). A fast uniaxial rotation of SiH₃ group around the 3-fold symmetry axis is expected to result in a 75% drop of this contribution to the second moment with respect to its ‘rigid-lattice’ value [14,28]. Assuming that the line shape is Gaussian, this corresponds to a 50% drop of the line width. The experimental plateau value of Δ_H for β -CsSiH₃ ($\sim 15 \text{ kHz}$) is indeed close to one-half of the low-temperature value; this is consistent with the model of uniaxial reorientations, in agreement with the QENS results for β -phases of MSiH₃ [16]. The additional decrease in the line width above 220 K (Fig. 2) can be attributed to the change in the reorientational mechanism related to the $\beta \rightarrow \alpha$ phase transition [16]. QENS data for the orientationally-disordered α -phases of MSiH₃ [7,15,16] are consistent with the reorientational mechanism in which each H atom traverses many sites on the surface of a sphere roughly centered on the Si atom. The effects of such nearly isotropic reorientational motion on Δ_H can be estimated in the following way. Since fast isotropic rotations should

average out the dipole-dipole interactions within SiH₃ groups, we have to calculate only the ‘intermolecular’ contribution to the second moment (between different SiH₃ groups and between SiH₃ groups and ^{133}Cs spins). This contribution can be roughly estimated by placing all protons in each SiH₃ group to its center (Si atom) and taking into account only the distances between centers of different groups [10]. The second moment for α -CsSiH₃ calculated using this approach is $4.1 \times 10^8 \text{ s}^{-2}$; for a Gaussian shape of the proton NMR line, this value corresponds to the line width of 7.6 kHz. The experimental value of Δ_H at the high-temperature end of our measurements (8.2 kHz) is close to this estimated level.

The behavior of the proton spin-lattice relaxation rates R_1^H measured at two resonance frequencies ($\omega/2\pi = 28$ and 90 MHz) is shown in Fig. 4. We have not found any significant deviations from a single-exponential recovery of the nuclear magnetization over the entire temperature range studied. General features of the observed temperature dependence of R_1^H are typical of complex hydrides with reorienting anions [9], where the dominant relaxation mechanism is related to motional modulation of the dipole-dipole interaction between nuclear spins. For this mechanism, the proton spin-lattice relaxation rate exhibits a peak at the temperature at which the reorientational jump rate τ^{-1} becomes nearly equal to the nuclear magnetic resonance frequency ω . According to the standard theory [26], in the limit of slow motion ($\omega\tau \gg 1$), R_1^H is proportional to $\omega^{-2}\tau^{-1}$, and in the limit of fast motion ($\omega\tau \ll 1$), R_1^H is proportional to τ being frequency-independent. If the temperature dependence of the jump rate τ^{-1} follows the Arrhenius law with the activation energy E_a ,

$$\tau^{-1} = \tau_0^{-1} \exp(-E_a/k_B T), \quad (1)$$

a plot of $\ln R_1^H$ versus T^{-1} is expected to be linear in the limits of both slow and fast motion with the slopes of $-E_a/k_B$ and E_a/k_B , respectively. We have verified that there are no additional R_1^H peaks below 80 K; the relaxation rate continues to decrease with decreasing temperature. As can be seen from Fig. 4, the observed behavior of R_1^H for CsSiH₃ strongly deviates from that predicted by the standard theory. First, the high-temperature slope of the $\ln R_1^H$ vs. T^{-1} peak appears to be much steeper than the low-temperature

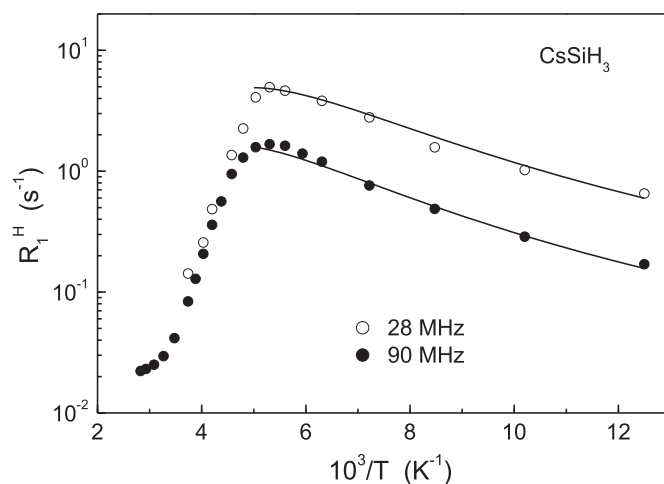


Fig. 4. Proton spin-lattice relaxation rates measured at 28 and 90 MHz for CsSiH₃ as functions of the inverse temperature. The solid lines show the simultaneous fit of the model with a Gaussian distribution of the activation energies to the data in the range 80–200 K.

one. Second, the observed frequency dependence of R_1^H at the low-temperature slope is considerably weaker than the predicted ω^{-2} dependence. Furthermore, there are certain deviations from the linear behavior of the $\ln R_1^H$ vs. T^{-1} plot both at high and low temperatures. The most evident reason for the unusual behavior of the proton spin-lattice relaxation rate in CsSiH₃ is the $\beta \rightarrow \alpha$ phase transition accompanied by the strong acceleration of the reorientational motion of SiH₃ groups. This phase transition in CsSiH₃ was studied both by neutron diffraction and quasielastic neutron scattering [16]. According to the QENS data [16], in the heating regime (corresponding to our present NMR measurements), the most pronounced changes in τ^{-1} occur between 200 K and 300 K. Because of this phase transition, the steep high-temperature slope of the $R_1^H(T)$ peak cannot be described in terms of regular thermally-activated changes in τ^{-1} ; this slope should reflect rapid changes in the H jump rate due to the phase change. It is interesting to note that in some complex hydrides the order-disorder phase transitions lead to abrupt changes in R_1^H ; for example, the order-disorder transition in sodium *closo*-hydroborate Na₂B₁₂H₁₂ near 520 K is accompanied by the two-orders-of-magnitude drop of R_1^H over the temperature interval of 5 K [13]. For CsSiH₃, the relaxation rate drop is not so sharp; this may be related to specific features of the $\beta - \alpha$ phase transitions in MSiH₃ occurring via intermediate phases [8,16]. The presence of disordered α -like nanodomains at temperatures much lower than the phase transition points was first reported for KSiH₃ and RbSiH₃ [8]. Because of their small size (2–4 nm [8]), these nanodomains may be invisible in diffraction experiments. The comprehensive QENS results for CsSiH₃ [16] are consistent with the presence of a nanostructured intermediate phase (*i*-phase) with the distinct reorientational dynamics being much faster than that for the β -phase, but slower than that for the α -phase. According to the QENS data [16], the α -, *i*-, and β -phases in CsSiH₃ can coexist over the temperature range of at least 170–250 K.

The complex behavior of CsSiH₃ related to the order-disorder phase transition prevents us from any attempts to parametrize the $R_1^H(T)$ data over the entire temperature range studied. Therefore, we have only tried to parametrize the $R_1^H(T)$ data at low temperatures ($T < 200$ K), where the β -phase is expected to be the dominant one. As noted above, the observed frequency dependence of the proton spin-lattice relaxation rate at the low- T slope of the $R_1^H(T)$ peak is considerably weaker than the ω^{-2} dependence. This feature is consistent with the presence of a certain distribution of H jump rates [29]. For parametrization of our low-temperature $R_1^H(T)$ data, we have used the simplest model based on a Gaussian distribution of the activation energies [29]; for this model, the observed relaxation rate is expressed as

$$R_1^H = \int R_1^H(E_a) G(E_a, \bar{E}_a, \Delta E_a) dE_a \quad (2)$$

Here $G(E_a, \bar{E}_a, \Delta E_a)$ is a Gaussian distribution function centered on \bar{E}_a with the dispersion ΔE_a , and $R_1^H(E_a)$ is determined by combining the standard expression that relates the relaxation rate and τ ,

$$R_1^H(E_a) = \frac{2\Delta M_{HH}\tau}{3} \left[\frac{1}{1 + \omega^2\tau^2} + \frac{4}{1 + 4\omega^2\tau^2} \right], \quad (3)$$

and the Arrhenius law (Eq. (1)). It should be noted that all the known expressions for the proton relaxation rate due to the reorientational motion (including a number of exact solutions for some simple types of reorientations [26]) show the same functional form

of dependence on τ and ω , as Eq. (3). The parameters of the model are the average activation energy \bar{E}_a , the dispersion ΔE_a , the pre-exponential factor τ_0 and the amplitude factor ΔM_{HH} determined by the strength of the fluctuating part of dipole-dipole interaction between protons. The contribution of dipole-dipole ^1H – ^{133}Cs interactions to the proton relaxation rate in CsSiH₃ can be neglected, since these interactions give only 2.6% of the total calculated ‘rigid-lattice’ second moment of the ^1H NMR line. The model parameters have been varied to find the best fit between the model $R_1^H(T)$ and the experimental data for $T < 200$ K at two resonance frequencies simultaneously. The results of such a simultaneous fit are shown by the solid curves in Fig. 4; the corresponding fit parameters: $\bar{E}_a = 226$ (8) meV, $\Delta E_a = 78$ (4) meV, $\tau_0 = 3.7(5) \times 10^{-15}$ s, and $\Delta M_{HH} = 3.4(1) \times 10^9$ s⁻². Note that the resulting fluctuating part of the second moment, ΔM_{HH} , corresponds to ~65% of the calculated ‘rigid-lattice’ second moment (see above); this is close to the expected ‘intramolecular’ value (75%) [14,28] for the case of uniaxial rotation of SiH₃ groups around the 3-fold symmetry axis.

The distribution width ΔE_a resulting from the fit appears to be rather large (about one-third of \bar{E}_a). This feature may indicate the presence of an additional jump process not resolved in the spin-lattice relaxation measurements. Indeed, if the difference between two frequency scales of H jump motion is relatively small (about one order of magnitude), an additional jump process may contribute only to the low-temperature slope of the single-peak $R_1^H(T)$ dependence that mimics the typical behavior for the model with a continuous jump rate distribution. Similar behavior of $R_1^H(T)$ was reported for a number of systems with two frequency scales of H jump motion, including some Laves-phase hydrides [30] and the ordered *closo*-hydroborate salt Rb₂B₁₀H₁₀ [31]. The presence of an additional jump process in the temperature range, where the β -phase is expected to be the dominant one, is also consistent with recent QENS results for CsSiH₃ [16]. These results indicate that the description of the elastic incoherent structure factor measured on the backscattering neutron spectrometer at 200 K in terms of the model of 3-fold SiH₃ reorientations (the only physically reasonable model for the ordered β -phase) requires an extra contribution (43%) to the elastic peak intensity. Such a large extra contribution to the elastic peak intensity is consistent [16] with the presence of H atoms moving much faster than those in the β -phase. However, we have not tried to use a two-peak distribution of the activation energies for parametrization of our low- T relaxation rate data, since this would require an introduction of a number of additional fit parameters.

The observed change in the high-temperature slope of the $\ln R_1^H$ vs. T^{-1} plot near 300 K (see Fig. 4) can be attributed to the upper boundary of the phase transition region. This interpretation implies the existence of a single α -phase of CsSiH₃ above 300 K, so that the behavior of R_1^H at $T > 300$ K should be governed by the regular Arrhenius temperature dependence of τ . The activation energy for H jump motion in this temperature range can be directly estimated from the proton spin-lattice relaxation data, since in the limit of fast motion (see above) R_1^H is proportional to τ . The estimate based on an Arrhenius approximation of our $R_1^H(T)$ data in the range 324–354 K gives the activation energy of 41(4) meV. This value is close to the activation energy for α -phase of CsSiH₃ (48(2) meV) found from QENS experiments [16].

3.2. ^{133}Cs NMR spectra and spin-lattice relaxation rates

The evolution of the ^{133}Cs NMR spectra for CsSiH₃ with temperature is shown in Fig. 5. As in the case of the proton NMR spectra, with increasing temperature we observe a considerable

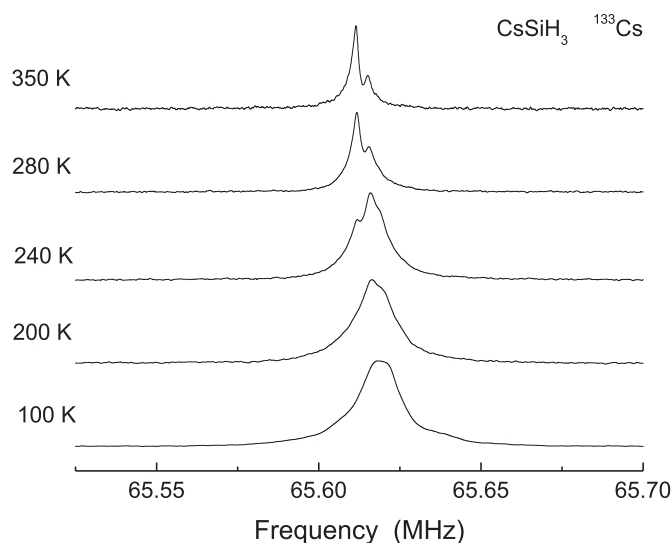


Fig. 5. Evolution of the ^{133}Cs NMR spectrum for CsSiH_3 with temperature.

narrows of the ^{133}Cs spectra accompanied by changes in their shape. The temperature dependence of the ^{133}Cs NMR line width Δ_{Cs} (full width at half-maximum) is shown in Fig. 6. Note that the onset of the order-disorder phase transition (in the temperature range 200–260 K) only slightly affects the value of Δ_{Cs} ; this is consistent with the localized nature of H motion. In the studied temperature range of 100–350 K, the most pronounced drop of Δ_{Cs} is observed between 260 K and 300 K, and at $T > 300$ K, the line width reaches a plateau. Such a behavior is typical of the case of diffusing cations in ionic conductors (see, for example, the “steps” in ^7Li NMR line widths in Refs. [32–34]). It should be noted, however, that the presence of a certain structure of the ^{133}Cs NMR spectrum for CsSiH_3 at 350 K suggests that the ^{133}Cs electric quadrupole interaction is not fully averaged out at this temperature. It is extremely difficult to simulate NMR line shapes in situations when a quadrupole interaction is partially averaged by atomic motion. Furthermore, we cannot exclude a possibility of multiple Cs sites characterized by different shifts. Similar results were reported for ^{23}Na NMR spectra in $\text{Na}_2(\text{BH}_4)(\text{NH}_2)$ showing fast Na^+ diffusion [35]. For CsSiH_3 , we have to distinguish between

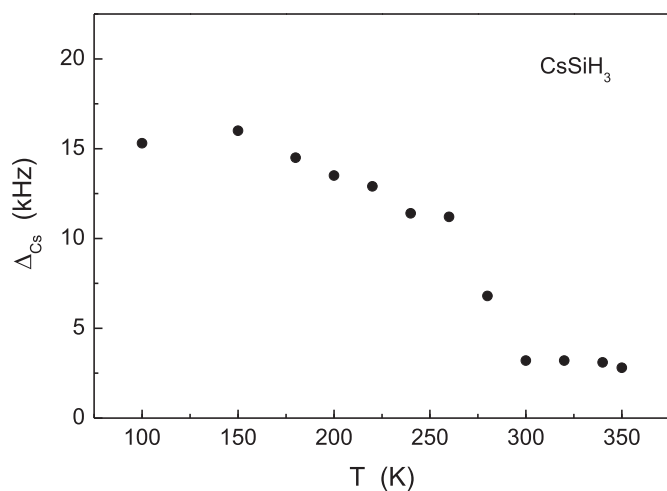


Fig. 6. Temperature dependence of the width (full width at half-maximum) of the ^{133}Cs NMR spectrum measured for CsSiH_3 at 65.62 MHz.

the effects due to the reorientational H motion and the diffusive motion of Cs^+ cations. For this purpose, we can use the results of the ^{133}Cs spin-lattice relaxation rate measurements shown in Fig. 7.

Comparison of Figs. 7 and 4 indicates that below 300 K, the behavior of the ^{133}Cs spin-lattice relaxation rate R_1^{Cs} resembles that of R_1^{H} . In particular, the $R_1^{\text{Cs}}(T)$ peak is observed at nearly the same temperature as the $R_1^{\text{H}}(T)$ peak. Furthermore, the low-temperature slope of the $R_1^{\text{Cs}}(T)$ peak is close to that of the $R_1^{\text{H}}(T)$ peak. Thus, the observed ^{133}Cs relaxation rate peak can be attributed to the reorientational motion of SiH_3 groups that modulates the relatively weak $^{133}\text{Cs} - ^1\text{H}$ dipole-dipole interaction. This is supported by the small amplitude of the $R_1^{\text{Cs}}(T)$ peak (0.36 s^{-1}) which is two orders of magnitude lower than the amplitude of the $R_1^{\text{H}}(T)$ peak. Such a difference in the maximum values of the spin-lattice relaxation rates is in general agreement with the estimated contributions of $^{133}\text{Cs} - ^1\text{H}$ and $^1\text{H} - ^1\text{H}$ dipole-dipole interactions to the corresponding second moments. It should be noted that reorientational jumps of SiH_3 groups are not expected to add any electric quadrupole contribution to the ^{133}Cs relaxation rate, since each reorientational jump does not change the charge configuration around Cs sites. In contrast to the case of the electric quadrupole interaction, the dipole-dipole interaction is pairwise, i.e., any changes in the radius vector connecting a pair of interacting nuclear spins should contribute to the nuclear spin relaxation. It can be concluded that the behavior of the ^{133}Cs spin-lattice relaxation rate below 300 K is governed by the reorientational motion of SiH_3 groups via $^{133}\text{Cs} - ^1\text{H}$ dipole-dipole interactions.

Above 300 K, the ^{133}Cs relaxation rate starts to increase with increasing temperature (Fig. 7). This behavior of R_1^{Cs} differs qualitatively from that of R_1^{H} in the same temperature range (Fig. 4); it is consistent with the onset of an additional low-frequency jump process, such as diffusive jumps of Cs^+ ions. It should be noted that R_1^{Cs} starts to increase at temperatures just above the characteristic “step” in the line width Δ_{Cs} (Fig. 6). This feature is typical of the onset of cation diffusive motion on the NMR frequency scale [32–34]; it corresponds to the case when the cation jump rate is higher than $\sim 10^4 \text{ s}^{-1}$ (sufficient to cause the NMR line narrowing), but is still considerably lower than $\sim 10^8 \text{ s}^{-1}$ (not reaching a new relaxation rate maximum).

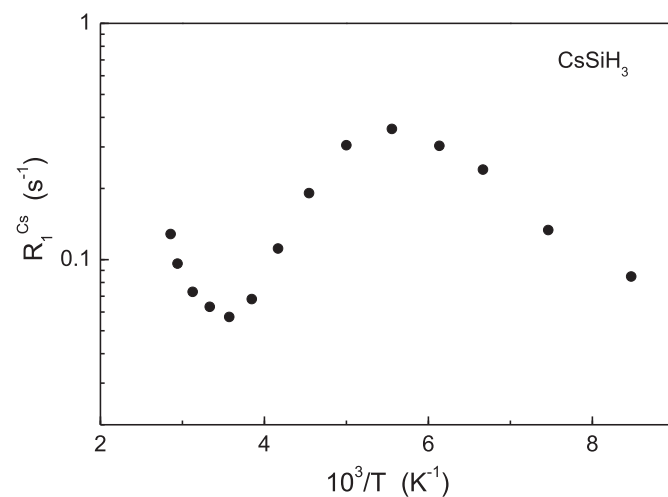


Fig. 7. ^{133}Cs spin-lattice relaxation rate measured at 65.62 MHz for CsSiH_3 as a function of the inverse temperature.

4. Conclusions

The results of our proton NMR measurements indicate that $[\text{SiH}_3]^-$ groups in CsSiH_3 retain unusually high reorientational mobility down to low temperatures. According to the ^1H NMR line width measurements, even at $T = 14\text{ K}$ the jump rate of SiH_3 reorientations in CsSiH_3 is about 10^5 s^{-1} . In agreement with recent QENS results for CsSiH_3 [16], our proton spin-lattice relaxation data are consistent with a coexistence of at least two reorientational jump processes below 200 K. The order-disorder ($\beta \rightarrow \alpha$) phase transition accompanied by the strong acceleration of SiH_3 reorientations and by the change in the reorientational mechanism leads to the considerable drop of the measured ^1H spin-lattice relaxation rate R_1^H ; however, this drop is not as abrupt as in the case of the order-disorder transitions in borohydrides [36] and closo-hydroborates [13]. The smearing of the R_1^H “step” in the phase transition region for CsSiH_3 suggests that the order-disorder transition in this compound occurs via the formation of an intermediate phase [16]. For the high-temperature disordered α -phase of CsSiH_3 , the activation energy for SiH_3 reorientations estimated from our $R_1^H(T)$ data is 41(4) meV. The ^{133}Cs NMR results are consistent with the onset of diffusive motion of Cs^+ ions at the frequency scale of $\sim 10^4\text{ s}^{-1}$ above 300 K.

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