



# Solid microparticles based on chitosan or methyl- $\beta$ -cyclodextrin: A first formulative approach to increase the nose-to-brain transport of deferoxamine mesylate



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## ARTICLE INFO

### Article history:

Received 10 December 2014

Received in revised form 20 January 2015

Accepted 21 January 2015

Available online 22 January 2015

### Keywords:

Deferoxamine mesylate

Chitosan chloride

Methyl- $\beta$ -cyclodextrin

Nasal formulations

Nose-to-brain transport

Pharmacokinetic studies

## ABSTRACT

We propose the formulation and characterization of solid microparticles as nasal drug delivery systems able to increase the nose-to-brain transport of deferoxamine mesylate (DFO), a neuroprotector unable to cross the blood brain barrier and inducing negative peripheral impacts. Spherical chitosan chloride and methyl- $\beta$ -cyclodextrin microparticles loaded with DFO (DCH and MCD, respectively) were obtained by spray drying. Their volume-surface diameters ranged from  $1.77 \pm 0.06 \mu\text{m}$  (DCH) to  $3.47 \pm 0.05 \mu\text{m}$  (MCD); the aerodynamic diameters were about  $1.1 \mu\text{m}$  and their drug content was about 30%. In comparison with DCH, MCD enhanced the in vitro DFO permeation across lipophilic membranes, similarly as shown by ex vivo permeation studies across porcine nasal mucosa. Moreover, MCD were able to promote the DFO permeation across monolayers of PC 12 cells (neuron-like), but like DCH, it did not modify the DFO permeation pattern across Caco-2 monolayers (epithelial-like). Nasal administration to rats of 200  $\mu\text{g}$  DFO encapsulated in the microparticles resulted in its uptake into the cerebrospinal fluid (CSF) with peak values ranging from  $3.83 \pm 0.68 \mu\text{g/mL}$  (DCH) to  $14.37 \pm 1.69 \mu\text{g/mL}$  (MCD) 30 min after insufflation of microparticles. No drug CSF uptake was detected after nasal administration of a DFO water solution. The DFO systemic absolute bioavailabilities obtained by DCH and MCD nasal administration were 6% and 15%, respectively. Chitosan chloride and methyl- $\beta$ -cyclodextrins appear therefore suitable to formulate solid microparticles able to promote the nose to brain uptake of DFO and to limit its systemic exposure.

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## 1. Introduction

Deferoxamine (DFO) is a hexadentate chelator binding ferric iron at a 1:1 M ratio. The iron complex of DFO, ferrioxamine, is highly stable thereby preventing iron ions from catalyzing redox reactions that lead to free radical formation (alternately “iron ions from participating in free radical reactions”) [1,2]. DFO as mesylate salt is the treatment of choice for acute iron intoxication and chronic iron overload due to transfusion-dependent anemia [2,3]. Due to its poor oral bioavailability and short plasma half-life, DFO is only used parenterally, often having a negative impact on the patients' quality of life due to secondary effects [2]. A clinical trial of iv DFO for intracerebral hemorrhage was halted due to an increased incidence of acute respiratory distress syndrome [4]. This should provide more rationale for an intranasal approach.

Direct brain targeting of DFO could be beneficial for the treatment of some diseases characterized by central nervous system (CNS) iron dysregulation. Data concerning animal and human studies show that DFO has diverse neuroprotective effects and its use is beneficial in Huntington's disease [5,6], Alzheimer's disease [7,8], Parkinson's [9–11] disease, amyotrophic lateral sclerosis [12], prionopathies [13], neuroferritinopathy [14], intracerebral hemorrhage [3,15], ischemic stroke [16,17], multiple sclerosis [18], and spinal cord injury [19]. Unfortunately, most of these studies used forms of systemic administration or intracranial injections that are not suitable for human use [9]. Whereas some studies indicate that DFO can rapidly penetrate the blood–brain barrier (BBB) and accumulate in the brain tissue in significant concentrations after systemic administration [3,20,21], other authors claim that the BBB is relatively impermeable to DFO [6,18,22]. Many therapeutic agents, including DFO, are rapidly delivered to the CNS by intranasal delivery and have exhibited therapeutic effects in both animals and humans [8,17,23–25]. The advantages of intranasal administration

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of DFO include targeting of the brain, reducing systemic exposure with possible side effects, and a noninvasive rapid method of administration [26], which rapidly bypasses the BBB reducing systemic exposure. Compared with intravenous administration, intranasal treatment with DFO increased targeting to the frontal cortex by 271-fold [17]. In this study, DFO was delivered in the nose as solution in saline or phosphate buffer solution. This approach, however, fails to take account of several factors limiting the nasal absorption of drugs, such as the (1) low membrane permeability, especially for the larger molecular weight and polar drugs; and (2) a rapid clearance of the drug formulation from the nasal cavity as a result of the mucociliary clearance mechanism [27]. It is possible to greatly improve the nasal absorption of polar drugs by administering them in a drug delivery system and in particular in solid formulations. Powders tend to stick to the moist surface of the nasal mucosa before being dissolved and cleared [28]. Considering the above-mentioned aspects, our aim was to formulate nasal drug delivery systems able to increase the nose to brain transport of DFO. The effect of two different penetration enhancers, methyl- $\beta$ -cyclodextrin and chitosan, was studied.

Numerous studies have demonstrated that cyclodextrins are efficient absorption enhancers in nasal drug delivery [29,30]. In the case of water-soluble drugs, CDs enhance drug absorption and/or bioavailability because of their ability to form inclusion complexes with specific membrane lipids, such as cholesterol, phosphatidylcholine and sphingomyelin, which cause perturbation of bilayer integrity and increase membrane fluidity [31]. Chitosan has the additional effect of reversibly opening tight junctions, with the potential to increase extracellular transport along olfactory and trigeminal nerve pathways into the CNS [32]. Chitosan, moreover, forms electrostatic interactions with the negatively charged surface of epithelial cells to reduce clearance from the nasal epithelium. This biodegradable polymer absorbs water from the mucus layer in the nasal cavity, swells and forms a gel-like layer prolonging drug residence time at the site of absorption and improving its bioavailability [33].

## 2. Materials and methods

### 2.1. Materials

Deferoxamine mesylate (aprox. 95% TLC) and 2-Propanol were purchased from Sigma-Aldrich (St. Louis, USA). Chitosan chloride, Protasan UP CL 113 (Mw: 160,000 g/mol; deacetylation degree, 86%) were purchased from NovaMatrix/FMC Biopolymer (Sandvika, Norway). Methyl- $\beta$ -cyclodextrin, Cavasol® W7 M Pharma (Mw: 1300 g/mol; molecular substitution: 1.7) was purchased from Wacker-Chemie GmbH (München, Germany). Phenyl Trimethicone, Dow Corning 556 cosmetic grade fluid was obtained from Dow Corning (Brussels, Belgium). Ultrapure grade water was obtained by Millipore filtration system (Billerica, MA, USA). All used solvents were of analytical grade. Male Sprague-Dawley rats were purchased from Harlan Italy S.r.l., Udine, Italy. Caco-2 and PC-12 cell lines were purchased from Sigma-Aldrich (St. Louis, USA).

### 2.2. Preparation of microspheres

Two different loaded formulations were prepared: one containing chitosan chloride (DCH) and the other one based on methyl- $\beta$ -cyclodextrin (MCD). The instrument used was Büchi Mini Spray Dryer B-191 (Büchi Laboratoriums-Technik AG, Flawil, Switzerland). A precisely weighed quantity of chitosan chloride or methyl- $\beta$ -cyclodextrin was dissolved in MilliQ water; DFO was added after the complete dissolution of the polymer. The resultant feed solution contains a total amount of solid (DFO and polymer) in solution equal to 1% (w/v) with a drug-polymer ratio of 1–2 (w/w). The solution was aspirated by a peristaltic pump, set up at 8%, and atomized through the 0.7 mm nozzle within the nebulization chamber. Here, a hot air flow (100 °C),

generated by 90% aspirator, forced the rapid evaporation of the solvent and, consequently, the formation of the solid microparticles inside the high-performance cyclone. At the end of the process, the solid particles were collected and placed in a desiccator until the moment of their use. The yield of production was calculated. In order to verify the different properties and behaviors between microspheres and simple mixtures, feed solutions were also freeze-dried for approximately 8 h, using a Lio 5P Cinquepascal (Trezzano sul Naviglio, Italy). These formulations were denominated DCHL and MCDL. Also a water solution of DFO was freeze-dried and coded as DFOL. Unloaded formulations were also prepared, as comparison and coded as CH and MC.

### 2.3. Evaluation of drug content and encapsulation efficiency

The drug content, which corresponds to the real amount of drug loaded in the microspheres, was determined by dissolving 10 mg of each formulation in 10 mL of MilliQ water and then adding pH 6.5 phosphate buffer up to 100 mL. A sample (1 mL) was collected and analyzed by HPLC, according to the method described below. The result obtained is the average of three determinations ( $n = 3$ ;  $\pm$  standard deviation, SD). Encapsulation efficiency was calculated [34].

### 2.4. HPLC analysis of deferoxamine mesylate

For the quantification of DFO in buffer samples, a Varian ProStar 210 with AutoSampler 410 and a PDA photodiode array detector (Varian Inc Scientific Instruments, Walnut Creek, CA USA) were employed. The chromatographic separation was performed on a C18 column with polar endcapping (Phenomenex Synergi Hydro-RP 80A, 150  $\times$  4.6 mm I.D. and 4  $\mu$ m of particle size). The HPLC analysis was conducted at room temperature with a flow rate of 1.2 mL/min and UV detection at 210 nm. The volume of the injected samples was 20  $\mu$ L and the analysis time was 5 min per sample. The binary mobile phase consisted of acetonitrile (pH 4.6 with 0.5 M  $\text{H}_3\text{PO}_4$ ) and a 0.1 M  $\text{KH}_2\text{PO}_4$  (3.4 g/250 mL), 120 mg/L (30 mg/250 mL) 1-eptane sulphonic acid sodium salt solution ratio 17:83 (v/v) adjusted to pH 4.60 by 0.5 M phosphoric acid. All data were collected in triplicate and processed using Varian Star Chromatography Workstation, system control, version 6.20 (Varian, Inc. Cary, NC, USA). Concentrations of analytes were calculated by interpolation of their respective standard curves.

### 2.5. Particle size analysis

Particle size was quantified via Coulter Laser Diffraction (Coulter LS 100Q Laser sizer, Beckman Coulter, Miami, USA). 2-Propanol was used as dispersion fluid for DCH formulation, while Dow Corning 556 was employed in case of MCD and MCDL. Microspheres (2 mg) were dispersed in the medium, sonicated for 30 s and analyzed under gentle magnetic stirring. Three dispersions were made for each formulation and the values reported are the averages of three determinations for each dispersion ( $n = 9$ ). The average particle size of the microspheres was reported as the mean volume-surface diameter,  $d_{vs}$  ( $\mu$ m) and the particle size distribution was graphically expressed by reporting the distribution curves obtained by plotting the volume of particles in percentage versus size ( $\mu$ m) shown in logarithmic scale. The coefficient of uniformity (CU) was also calculated by applying the following formula:  $\text{CU} = d_{10}/d_{90}$ . The values of  $d_{10}$  and  $d_{90}$  indicate that, respectively, 10% and 90% of the particles of the sample have a diameter less than the number indicated. A CU value approaching 1 indicates a dimensionally uniform sample.

### 2.6. Morphological analysis

The morphological attributes of the microparticles were studied by Scanning Electron Microscopy (SEM). A small amount of powder was

spread on an aluminium stub, which was placed after gold sputtering in an SEM chamber (Zeiss EVO LS10, Zeiss, Germany).

## 2.7. Water uptake

The ability of the formulations to absorb water when placed in contact with an aqueous fluid was measured using a modified apparatus of Enslin. A sample of each formulation (2 mg) was uniformly dispersed on a disk of paper filter ( $d = 1$  cm,  $A = 0.78$  cm<sup>2</sup>), saturated with phosphate buffer pH 6.5, which was placed on a porous septum. The volume of buffer absorbed by the formulation within 60 min was measured. The result obtained is the average of three determinations ( $n = 3$ ;  $\pm$  SD).

## 2.8. True density and aerodynamic diameter

True density of loaded microspheres was measured by helium pycnometry (Micromeritics Accupyc II 1340 Analysis system, Peschiera Borromeo, Italy) at 21 °C. Microparticles of known mass were transferred into the cell of the instrument (1.03 cm<sup>3</sup> volume). The density ( $\rho$ ) of the powder was determined in triplicate for each batch as follows:  $\rho = M/V$ , where  $M$  is the mass per each batch of microspheres and  $V$  is the average volume occupied with powder, calculated from the 5 cycles of emptying and puffiness with helium. The aerodynamic diameter was calculated. Assuming the particle shape to be spherical, the relation between mass mean aerodynamic diameter,  $d_{ae}$ , and geometrical diameter,  $d_g$ , is given by:  $d_{ae} = d_g \sqrt{\rho}$ , with  $\rho$  being the true density of particles.

## 2.9. Solid state characterization

Drug, polymers and formulations were characterized by X-ray diffraction (XRD). XRD analyses were performed at room temperature with a Siemens D5000 diffractometer equipped with a copper tube and a graphite monochromator on the diffracted beam. The acquisition parameters were 40 kV, 30 mA,  $2\theta$  range of 2–50°, step size of 0.020° ( $2\theta$ ), and time per step of 2 s.

## 2.10. In vitro permeation test

Experiments were performed using a modified Franz diffusion system incorporating three in-line flow-through diffusion cells [34]. Cellulose acetate membranes (pore size 0.45  $\mu$ m) were employed as hydrophilic layer; regenerated cellulose membranes (pore size 0.45  $\mu$ m) saturated with octanol were chosen as lipophilic layer. An amount of microspheres equivalent to about 2.5 mg of DFO was uniformly distributed above each membrane. Then, 1 mL of acceptor fluid was taken at predetermined time intervals (0–120 min) and analyzed by HPLC. The withdrawn volume was restored with fresh buffer at  $37 \pm 0.5$  °C. The results reported are the mean of three determinations and are expressed as cumulative amount of DFO permeated per unit of time. The effective permeability coefficient,  $P_{eff}$ , under steady state conditions across the synthetic membranes has been mathematically expressed, as follows:  $P_{eff} = (dc/dt)_{ss}V/(ACD)$ , where  $(dc/dt)_{ss}$  was determined by the slope of the plot of the permeated amount versus time in the steady state,  $A$  is the permeation area,  $V$  is the volume of the receiver compartment and  $CD$  is the initial concentration of DFO in donor compartment [35]. The Transport Enhancement Ratio (TER) of formulation compared to the drug was calculated from  $P_{eff}$  values:  $TER = P_{eff}(\text{formulation}) / P_{eff}(\text{drug})$ .

## 2.11. Ex vivo drug permeation study

This procedure was similar to the in vitro permeation method, replacing the synthetic membrane with fragments of porcine nasal mucosa, obtained from the local slaughterhouse [36]. As acceptor medium, phosphate buffer pH 6.5 has been used. At predetermined time

intervals, samples (1 mL) of acceptor medium were taken and the volume was replaced with fresh buffer at  $37 \pm 5$  °C. Samples were passed through a 0.45  $\mu$ m cellulose acetate filter before HPLC analysis. The results reported are the mean of three determinations and are expressed as cumulative amount of DFO permeated per unit of time ( $n = 3$ ;  $\pm$  SD). The effective permeability coefficient under steady state conditions across the nasal mucosa and the Transport Enhancement Ratio have been calculated as described above.

## 2.12. Cellular uptake studies

Cell uptake of DFO released from microspheres was assessed using Caco-2 and PC-12 cells as model of epithelial and neuron-like phenotypes, respectively. Caco-2 cells (human colorectal adenocarcinoma), were cultured in flasks in Dulbecco's Modified Eagle's Medium (DMEM, Sigma-Aldrich) supplemented with 1% (v/v) MEM nonessential amino acids (Invitrogen), 10% (v/v) fetal bovine serum (Sigma-Aldrich), 1% (v/v) penicillin and streptomycin solution (Lonza) and were grown in a humidified atmosphere of 5% CO<sub>2</sub> at 37 °C. Cells were subcultured at 80% confluence. PC-12 cells (rat pheochromocytoma, ECACC) were cultured in flasks in differentiation medium (RPMI-1640 medium, Sigma-Aldrich) supplemented with 10% (v/v) horse serum, 5% (v/v) fetal bovine serum (not of USA origin, Sigma-Aldrich) and L-glutamine 2 mM (Sigma-Aldrich) and were grown in a humidified atmosphere of 5% CO<sub>2</sub> at 37 °C. For differentiation to a neuron-like phenotype, cells were suspended in RPMI-1640 medium, supplemented with 1% horse serum, 100 ng/mL of nerve growth factor (NGF) (Sigma-Aldrich). The medium was replaced every 2–3 days. Six-day differentiated PC-12 cells were used as an in vitro model for neuron drug permeation studies. Caco-2 cells were harvested from flasks with trypsin–EDTA (BiocromAG) and seeded in Transwell PET inserts (12 well, 0.4  $\mu$ m pore size, Corning, USA) at a concentration of  $1 \times 10^5$  cells/membrane (number passage 50). PC-12 cells, previously grown in suspension, were seeded in Transwell-COL PTFE insert (12 well, 0.4  $\mu$ m pore size, collagen coated, Corning, USA) at a concentration of  $3 \times 10^4$  cells/membrane and number passage of 5. Cell monolayers were washed twice with warm phosphate buffered saline (PBS) and basolateral and apical compartments were filled with 1.5 and 0.5 mL of DMEM (Caco-2 cells) or RPMI-1640 (PC-12 cells). Microspheres containing 2.6 mg DFO and the same quantity of non-formulated drug were suspended or solubilized in 0.15 mL medium and added to the apical side. Cells were incubated in a humidified atmosphere of 5% CO<sub>2</sub> at 37 °C. At fixed intervals (7.5, 15, 30, 45, 60 min), aliquots (20  $\mu$ L) from basolateral side were collected and replaced with an equal volume of fresh medium. Monolayer integrity was verified before and during the experiments by transepithelial electric resistance (TEER). DFO was quantified by HPLC. Drug apical concentrations at 7.5 min and 60 min were also evaluated. All tests were performed in duplicate.

## 2.13. Liquid chromatographic–mass (LC–MS) spectrometric analysis

LC–MS analyses for in vivo pharmacokinetic studies were undertaken with a microHPLC Surveyor (Thermo Scientific) coupled to a LTQ XL Linear Ion trap MS detector (Thermo Scientific). Chromatographic conditions were applied to a Luna HST C18 100  $\times$  2.1 mm column (Phenomenex) packed with 2.5  $\mu$ m particles. Mobile phases were mixtures of formic acid 0.1% v/v in water (channel A) and formic acid 0.1% v/v in methanol (channel B); the gradient profile ranged from 50/50 water/methanol to 10/90 in 5 min, then it continued isocratically for 3 min. Column flow rate was 150  $\mu$ L/min at 25 °C. The following MS conditions were applied to the electrospray ion source (ESI): capillary temperature 275 °C; spray voltage 4 kV; capillary voltage 50 V; and tube lens voltage 100 V. The molecular ions for DFO, aluminoxamine (AO) and ferrioxamine (FO) were detected at 561.4, 585.3 and 614.3 m/z, respectively. The internal standard CPA was detected at 336 m/z. All ions were in the form  $[M + H]^+$ . Target analytes in samples



were DFO, AO and FO and the sum of these three forms was considered as the total amount. Quantitative determination is undertaken by internal standard calibration with the use of N<sup>6</sup>-cyclopentyladenosine (CPA). DFO, AO and FO were co-eluted from the chromatographic column with a retention time of 4.5 min, while CPA has a retention time of 6 min. Calibration range was 0.05–50  $\mu\text{M}$  for both determinations in plasma and cerebrospinal fluid (CSF) matrices. For CSF simulation, standard aliquots of balanced solution (PBS Dulbecco's without calcium and magnesium) in the presence of 0.45 mg/mL bovine serum albumin (BSA) were employed [37,38]. Calibration data were fitted to straight lines in both cases with  $R^2$  values of 0.9910 and 0.9929 for plasma and CSF, respectively ( $P < 0.001$ ). Limit of detection (LOD) and limit of quantification (LOQ) were estimated with a confidence level of 95% by using the Huabaux and Vos method. Calibration in plasma matrix was characterized by LOD = 0.15  $\mu\text{M}$  (0.08  $\mu\text{g/mL}$ ) and LOQ = 0.5  $\mu\text{M}$  (0.28  $\mu\text{g/mL}$ ), while for CSF LOD = 0.24  $\mu\text{M}$  (0.13  $\mu\text{g/mL}$ ) and LOQ = 0.8  $\mu\text{M}$  (0.45  $\mu\text{g/mL}$ ). The average recovery of total analytes was estimated as 85%. Recovery for CPA was already tested in a previous work [39]. Chromatograms and related MS/MS spectra referred to a DFO standard solution and to DFO detected in CSF are reported, as representative (Supplemental Fig. S1a–b). The positive ion mode MS/MS spectra are those expected for DFO by fragmentation of parent ion 561 m/z. Daughter ions correspond to those reported in literature (201, 243, 319.2, 361.2, 401.2 and 443.2 m/z) [40].

#### 2.14. In vivo deferoxamine administration and quantification

Male Wistar rats (200–250 g) anesthetized by a continuous inhalation of isoflurane/air (1.5% mixture) during the experimental period, received a femoral intravenous infusion of 0.2 mg/mL deferoxamine dissolved in a medium constituted by 20% (v/v) DMSO and 80% (v/v) physiologic saline, with a rate of 0.2 mL/min for 5 min. At the end of infusion and at fixed time points, blood samples (300  $\mu\text{L}$ ) were collected and CSF samples (50  $\mu\text{L}$ ) were withdrawn by cisternal puncture method described by van den Berg et al. [41]. CSF samples (10  $\mu\text{L}$ ) were immediately injected into HPLC system for LC–MS analysis. Blood samples were centrifuged at 4000  $\times g$  for 15 min, then 100  $\mu\text{L}$  of plasma was quenched in 300  $\mu\text{L}$  of ethanol (4  $^{\circ}\text{C}$ ); 100  $\mu\text{L}$  of internal standard (1  $\mu\text{M}$  CPA) was added. After centrifugation at 13,000  $\times g$  for 10 min, 300  $\mu\text{L}$  aliquots were reduced to dryness under a nitrogen stream, dissolved in 200  $\mu\text{L}$  of mobile phase and, after filtration, 10  $\mu\text{L}$  was injected into the HPLC system for LC–MS analysis. The in vivo half-life of deferoxamine in the blood was calculated by nonlinear regression (exponential decay) of concentration values in the time range within 2 h after infusion and confirmed by linear regression of the log concentration values versus time. Nasal administration of DFO was performed on anesthetized rats laid on their backs, following two procedures. The first one consisted of the introduction of 40  $\mu\text{L}$  of an aqueous solution of DFO (5 mg/mL) in one nostril of rats using a semiautomatic pipette which was attached to a short polyethylene tubing. The tubing was inserted approximately 0.6–0.7 cm into each nostril. After the administration, blood (300  $\mu\text{L}$ ) and CSF samples (50  $\mu\text{L}$ ) were collected at fixed time points, and they were analyzed with the same procedures described above. Four rats were employed for nasal administration of DFO solution. The second procedure was based on the insufflations of DFO-loaded microparticles (DCH and MCD) to one nostril of anesthetized rats by single dose Monopowder P<sup>®</sup> insufflators (Valois Dispray SA, Mezzovico, Switzerland) [42] loaded with about 0.7 mg of DFO microparticles (corresponding to about 200  $\mu\text{g}$  of DFO). Blood (300  $\mu\text{L}$ ) and CSF samples (50  $\mu\text{L}$ ) were collected at fixed time points, and they were analyzed with the same procedures described above. Four rats were employed for nasal administration of each preparation. All in vivo experiments were performed in accordance with the guidelines issued by the Italian Ministry of Health (D.L. 116/92 and D.L. 111/94-B), the Declaration of Helsinki, and the Guide for the Care and Use of Laboratory Animals as adopted and promulgated by the National Institute of Health (Bethesda, Maryland). The protocol

of all the in vivo experiments has been approved by Local Ethics Committee (University of Ferrara, Ferrara, Italy). Any effort has been done to reduce the number of the animals and their suffering. The area under concentration curves of DFO in the blood and CSF (AUC,  $\mu\text{g/mL min}$ ) was calculated by the trapezoidal method. All the calculations were performed by using the computer program Graph Pad Prism (GraphPad Software Incorporated, La Jolla, CA, USA).

#### 2.15. Statistical analysis

Data were analyzed using the nonparametric Kruskal–Wallis test; individual differences were evaluated using a post hoc Dunn's multiple comparison test (GraphPad Prism, version 6.02; GraphPad Software Incorporated). When suitable, analysis of variance (ANOVA) followed by a Tukey test was done.

### 3. Results

#### 3.1. Evaluation of yield of production, drug content and encapsulation efficiency

Microspheres prepared by spray-drying resulted in production yields exceeding 50%. As expected, yields were higher for formulations produced by lyophilization (Table 1). Real drug content was very close to theoretical values and the encapsulation efficiency ranged from 93 to 98% (Table 1). These parameters are not affected by the excipient used ( $P > 0.05$ ).

#### 3.2. Particle size analysis

Cyclodextrin microparticles had a larger volume-surface diameter ( $d_{vs}$ ) compared with chitosan particles (Table 1,  $P < 0.01$ ); no significant differences were observed between loaded and unloaded microspheres ( $P > 0.05$ ). Distribution curves of loaded microspheres were almost superimposed with those of the drug-free formulations (Supplemental Fig. S2). All formulations had a wide size distribution, as indicated by CU, that decreased in the range order DCH > CH > MCD > MC ( $P < 0.05$ ). It was not possible to analyze the size of the DCHL particles because chitosan formed a sponge-like film as seen on SEM. MCDL particles had dimensions substantially greater and more heterogeneous than its counterpart produced by spray-drying:  $d_{vs}$  was  $9.29 \pm 0.50 \mu\text{m}$  and CU was  $0.07 \pm 0.01$  ( $P < 0.05$ ).

#### 3.3. Morphological analysis

Chitosan and cyclodextrin microspheres had distinct morphologies. DCH had a spherical shape and smooth surface (Fig. 1 left), whereas multiple invaginations were present in MCD particles (Fig. 1 right).

No significant morphological changes were observed with microsphere loading. SEM pictures showed that freeze-dried formulations had a very different morphology compared with the corresponding microspheres. DCHL had crumpled surfaces resembling dry leaves. MCDL was less homogeneous: with particles of variable dimensions having both curved and smooth surfaces. The free drug also changed its morphology when freeze-dried; DFO as received showed aggregates with irregular surfaces, heterogeneous for dimension and shape, whereas the lyophilized drug formed drapes with smoother surfaces, sometimes having filaments on the rims (Supplemental Fig. S3).

#### 3.4. Water uptake

Formulations containing cyclodextrin absorbed very little water and then rapidly solubilized (Fig. 2). Chitosan-based formulations absorbed significantly more water ( $P < 0.01$ ), regardless of preparation method.

**Table 1**

Results of the yield of production, drug content and encapsulation efficiency (mean  $\pm$  SD). Volume-surface diameter ( $d_{vs}$ ) and coefficient of uniformity (CU) of loaded and unloaded formulations (value  $\pm$  SD).

Formulation	Yield of production (%)	Drug content (%)	Encapsulation efficiency (%)	$d_{vs}$ ( $\mu$ m)	CU
DCH	63.18 $\pm$ 6.76	30.77 $\pm$ 0.25	92.72 $\pm$ 1.24	1.77 $\pm$ 0.06	0.26 $\pm$ 0.01
MCD	55.62 $\pm$ 3.45	31.71 $\pm$ 3.39	95.07 $\pm$ 10.04	3.47 $\pm$ 0.05	0.17 $\pm$ 0.00
DCHL	95.90 $\pm$ 5.80	32.92 $\pm$ 0.82	98.02 $\pm$ 2.80	1.92 $\pm$ 0.01	0.20 $\pm$ 0.00
MCDL	72.47 $\pm$ 8.20	32.45 $\pm$ 1.05	97.54 $\pm$ 3.49	3.81 $\pm$ 0.09	0.13 $\pm$ 0.01

### 3.5. True density and aerodynamic diameter

Absolute densities of DCH and MCD microspheres were very similar ( $1.414 \pm 0.003$  and  $1.414 \pm 0.001$  g/cm<sup>3</sup>, respectively) ( $P > 0.05$ ), and did not differ from that of drug as received ( $1.371 \pm 0.001$  g/cm<sup>3</sup>,  $P > 0.05$ ). The freeze drying process increased the density of DFOL ( $2.196 \pm 0.058$  g/cm<sup>3</sup>) and DCHL ( $1.663 \pm 0.045$  g/cm<sup>3</sup>), but decreased that of MCDL ( $1.355 \pm 0.005$  g/cm<sup>3</sup>) compared with that of the corresponding spray dried preparation ( $P < 0.05$ ).

Aerodynamic diameter ( $d_{ae}$ ) ranged from 1.10 to 1.16  $\mu$ m for MCD and DCH, respectively ( $P > 0.05$ ).

### 3.6. Solid state characterization

The XRD pattern of DFO (Fig. 3) shows that it has a crystalline structure, with sharp peaks (i.e. the high ratio signal/noise) indicating a high degree of crystallinity. The patterns of MC and CH demonstrate that these polymers are amorphous (Fig. 3). In particular, MC shows two main broad reflections at about 11° and 18° of 2 $\theta$ , and a third, weaker, at about 24°, whereas the XRD pattern of CH is characterized by two slight humps at about 12° and 24° of 2 $\theta$ . MCD, the MC-DFO formulation, has an amorphous structure. The DFO peaks are not present in the XRD pattern, that is superimposable to that of MC except for the weaker intensity of the first broad reflection (Fig. 3). The formulation DCH is also amorphous; the XRD pattern is perfectly superimposable to that of CH. The freeze-drying process led to the amorphization of the pure drug (DFOL in Fig. 3). Conversely, lyophilization did not produce significant variations in the XRD patterns of the pure polymers, or of the formulations MCDL and DCHL with respect to their spray-dried counterparts (XRD patterns not reported).

### 3.7. In vitro permeation test

The in vitro permeation profiles of the drug and formulations are illustrated in Fig. 4. DFO as received permeated the hydrophilic membrane more rapidly than membranes saturated with octanol ( $P < 0.01$ ) in the range between 15–120 min. After freeze-drying, this difference was not observed ( $P > 0.05$ ).

Encapsulation in microspheres affected the permeation behavior of DFO. Both materials reduced the cumulative amount of drug passing

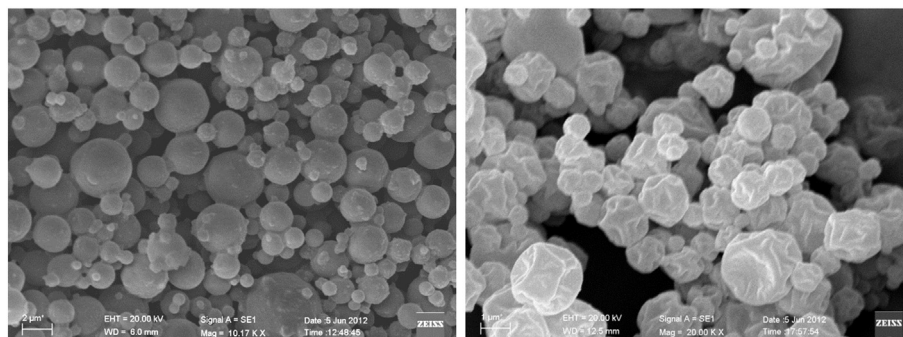
through the hydrophilic membrane ( $P < 0.01$ ). Chitosan reduced the permeation of loaded drug through lipophilic membranes, whereas methyl- $\beta$ -cyclodextrin increases it ( $P < 0.05$ ). This was evident by observing the  $P_{eff}$  under steady state conditions (Fig. 5). MCD demonstrated higher flux through lipophilic membranes when compared with DCH, as also deductible by TER values which are 1.63 and 0.12, respectively. The permeation profiles of the freeze-dried formulations did not differ from the corresponding spray dried formulations in the applied conditions ( $P > 0.05$ ). The presence of DFO and excipient at the same time in the freeze-dried powders modified the  $P_{eff}$  of drug alone through both membrane types, increasing the TER values from 0.22 to 1.27 for DCHL and MCDL, respectively ( $P < 0.05$ ), indicating that CD was the most effective in facilitating the permeation of DFO through the two membranes.

### 3.8. Ex vivo drug permeation study

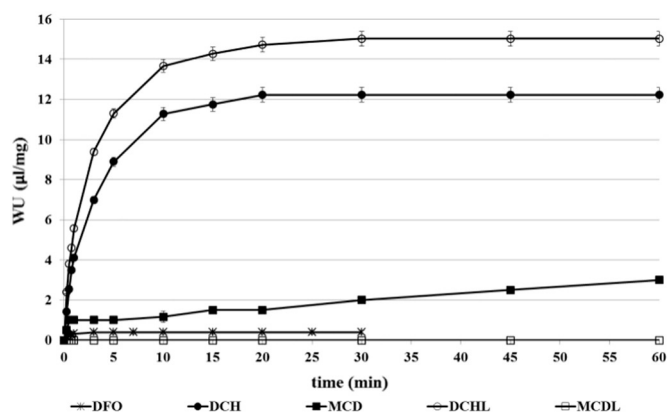
All profiles were characterized by a lag time that increased from 15 to 30 and 60 min in case of DFO, MCD and DCH, respectively (Fig. 6) which is compatible with the water behavior of the carrier and consistent with the water uptake and in vitro permeation results. Afterwards, MCD and DFO show the same permeation rates ( $P > 0.05$ ), that are higher than that of DCH. At the end of the test, almost 50% of DFO as received and loaded in MCD was recovered in the acceptor medium. On the contrary, only 30% of drug encapsulated in DCH microspheres permeated nasal mucosa at the same time. The  $P_{eff}$  under steady state conditions are  $3.28 \pm 0.20$  and  $1.67 \pm 0.16$  mg/cm<sup>2</sup> min for MCD and DCH, respectively ( $P < 0.01$ ).

### 3.9. Cellular uptake studies

Permeation capacity of DFO released from microspheres through Caco-2 and PC-12 cells was evaluated (Fig. 7). Cell monolayers maintained their integrity during the experiments: TEER remained on a high level (Supplemental Fig. S4). All formulations showed the same permeation profiles through epithelial cells, superimposable to drug ( $P > 0.05$ ). On the contrary, MCD permitted a faster release and passage of drug through PC-12 cell monolayers ( $P < 0.05$ ), in the range of 15 to 45 min. The permeation rate of MCD and DFO was the same, regardless the cell phenotypes ( $P > 0.05$ ).



**Fig. 1.** Images obtained by SEM of DCH (left) (10.17 KX) and MCD (right) (20.00 KX).

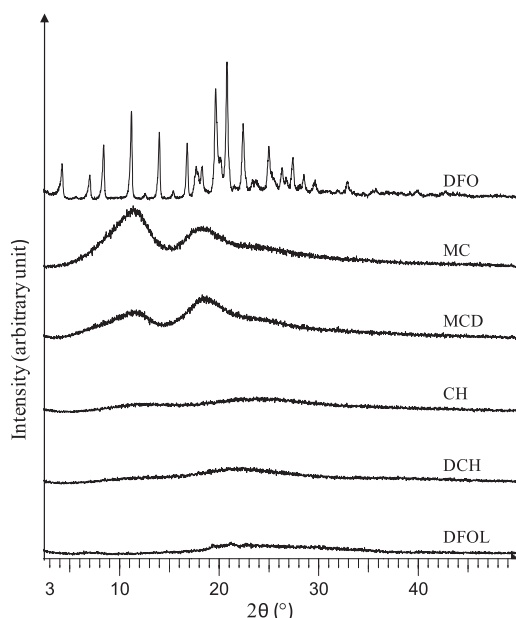


**Fig. 2.** The water uptake capacity ( $\mu\text{L}/\text{mg}$ ) of the microspheres as well as freeze-dried formulation compared to the free drug as received. Data are reported as the mean value  $\pm$  SD ( $n = 3$ ).

Furthermore, after 7.5 min and 60 min, the amount of drug in the donor compartment was measured (Fig. 8). From the difference between the amount of DFO permeated and recovered, the percentage of drug inside cells was calculated. Chitosan microspheres rapidly released about 44% of drug that remained in the cells and then slowly diffused; the leftover amount persisted in the microspheres until 60 min, regardless of the cells used. On the contrary, the drug released from MCD remained for the longest time in the donor compartment and only 15% and 30% of DFO were in Caco-2 and PC-12 cells, respectively, after 7.5 min. DFO as received had the same behavior of MCD.

### 3.10. In vivo DFO administration

DCH and MCD were tested for nasal administration of deferoxamine in order to verify its potential uptake into the CNS. Nasal administration of an aqueous solution of DFO was also tested as control. The analysis of rat blood samples following the intravenous infusion of 0.2 mg of deferoxamine indicated that the drug concentration in the blood stream was  $6.40 \pm 0.38 \mu\text{g}/\text{mL}$  and decreased over time with an apparent first order kinetic ( $n = 6$ ,  $r = 0.980$ ,  $P < 0.01$ ) and a half-life of  $15.4 \pm 1.8$  min (Fig. 9). No DFO was detected in CSF within 120 min of intravenous administration.



**Fig. 3.** XRD pattern of: DFO, MC, MCD, CH, DCH and DFOL.

### 3.11. Nasal DFO administration

Nasal administration of pure DFO as powder was not performed, due to the very low dose required (about 200  $\mu\text{g}$  in rats). Therefore, as control, we employed a water solution of the raw drug. No significant amounts of DFO were observed in the blood or CSF within 120 min of treatment (data not shown). On the contrary, nasal administration of the powder constituted by the loaded DCH and MCD microparticles (0.7 mg, about 200  $\mu\text{g}$  of DFO) produced detectable DFO levels in both blood and CSF. Following the nasal administration, the peak plasma concentration was detected at 30 min, with values just above the LOD of the analytical method ( $0.19 \pm 0.04 \mu\text{g}/\text{mL}$  by DCH powder;  $0.36 \pm 0.06 \mu\text{g}/\text{mL}$  by MCD powder), indicating poor permeation of the drug from the nose to the bloodstream. Indeed, a comparison of the AUC values obtained after intravenous administration ( $145.1 \pm 3.6 \mu\text{g}/\text{mL} \cdot \text{min}$ ) and nasal administration ( $8.92 \pm 1.11 \mu\text{g}/\text{mL} \cdot \text{min}$  for DCH powder;  $14.89 \pm 2.05 \mu\text{g}/\text{mL} \cdot \text{min}$  for MCD powder) indicated absolute bioavailability values of 6.15% and 10.26% for the microparticulate powders based on chitosan and cyclodextrins, respectively.

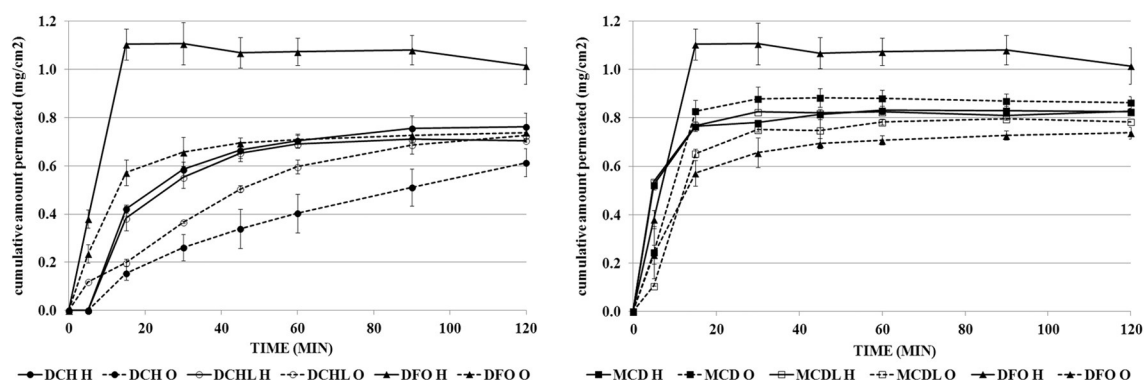
Conversely, significant quantities of DFO were detected in the CSF following the nasal administration of the microparticulate systems. In particular, the CSF peak concentrations of the drug were detected at 30 min, showing values of  $3.83 \pm 0.68 \mu\text{g}/\text{mL}$  for DCH powder and  $14.37 \pm 1.69 \mu\text{g}/\text{mL}$  for MCD powder. DFO concentrations slowly decreased within 60 min of administration ( $2.77 \pm 0.50 \mu\text{g}/\text{mL}$  for DCH powder;  $11.68 \pm 11.43 \mu\text{g}/\text{mL}$  for MCD powder), then an abrupt decrease was registered within 90 min (Fig. 10).

The AUC values obtained for DFO in CSF following nasal administration of the DCH and MCD powders were  $198.2 \pm 17.4 \mu\text{g}/\text{mL} \cdot \text{min}$  and  $815.7 \pm 47.0 \mu\text{g}/\text{mL} \cdot \text{min}$ , respectively. The ratio between MCD and DCH powders was 4.12, indicating that the nasal formulation containing cyclodextrins allowed approximately four times more DFO uptake into the CSF than the chitosan-based formulation.

## 4. Discussion

Freeze-drying remains the gold standard for the drying technology used in the pharmaceutical industry; spray drying represents the most mature alternative drying technology to lyophilization. The spray drying process provides an opportunity to control particle properties such as crystallinity, particle size, residual water content, bulk density, and morphology [43]. In this research work the properties of freeze-dried and spray-dried powders were compared in order to remark that spray drying is the most useful technique for preparing nasal formulations. In fact, freeze-drying is a technique not applicable for all drug-polymer combinations. DCHL form a sponge-like film with crumpled surfaces whereas MCDL production is characterized by low yield probably due to the aspiration of the smallest solid particles from the vacuum system of the freeze dryer. Moreover, by spray drying, particles with small diameter and narrow size distribution are obtained. Size together with density and morphology are the properties which can affect the aerodynamic features of microparticles and therefore the achievement of targeted deposition sites [44]. MCD and DCH show similar density and size and therefore they do not differ in aerodynamic features among themselves, even if they differ in morphology. Inthavong and co-workers (2009) demonstrated that near-spherical particles with a rough surface show lower deposition in comparison with spherical particles. As a consequence, particles can avoid impaction in the anterior part all the nasal cavity. On the basis of these considerations and of the low  $d_{ae}$  values, the particles produced have great possibility to reach the olfactory region in the roof of the nasal cavity. After inertial impaction, which is the dominant mechanism of deposition for micron ranged particles [44], because of their rapid water absorption, microspheres can adhere or dissolve at the deposition site. Obviously, in vivo deposition of particles will be influenced also by the dispenser used for the administration and the mucoadhesive properties of

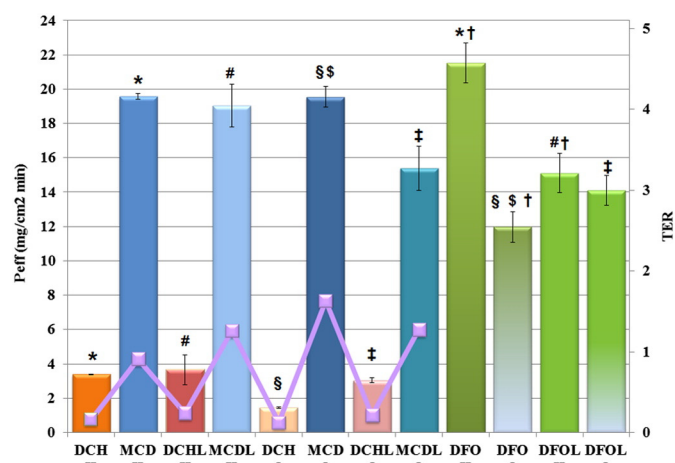




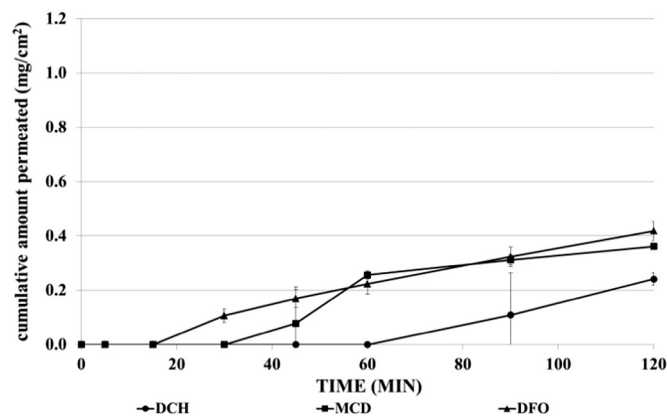
**Fig. 4.** The permeation profiles of DFO through hydrophilic (solid line) and lipophilic (dotted line) membranes from spray dried (filled symbol) and freeze-dried (empty symbol) formulations of the same concentration. Profiles are compared with those of drug as received and after freeze-drying. H and O letters have been used to indicate the permeation through hydrophilic and lipophilic membranes, respectively. Data are reported as the mean  $\pm$  SD of three independent experiments.

microparticles. Both drying techniques generate amorphous state of DFO as shown in XRD patterns; in spite of literature data [45], the higher molecular mobility of this form does not lead to enhanced permeation rate probably due to the high solubility of DFO. On the contrary, the in vitro and in vivo permeation behavior of DFO is affected by materials used as carriers for microparticulate drug delivery systems. In order to study the DFO pharmacokinetics, we have developed a liquid phase extraction method of this drug from plasma, suitable for LC–MS analysis. We have then performed an intravenous administration of the drug (200  $\mu$ g) to rats and calculated its terminal half-life ( $15.4 \pm 1.8$  min) evidencing a value similar to that found in dogs [46] and confirming that after intravenous administration DFO is rapidly eliminated from the bloodstream [17,47]. Moreover, following the intravenous administration of the same dose, DFO was not detected in the CSF of rats at any considered time point. This result appears in agreement with the poor aptitude of the drug to reach the CNS from the bloodstream [17]. It has been indeed reported that the nasal administration of DFO solutions constitutes a viable noninvasive alternative to systemic injections for the treatment and prevention of stroke. In particular, intranasal administration of 6 mg DFO to rats resulted in significantly higher CNS concentrations in the brain (at least two orders of magnitude) and significantly reduced systemic exposure compared with intravenous administration of the same dose [17]. We have previously demonstrated that nasal microparticulate formulations based on

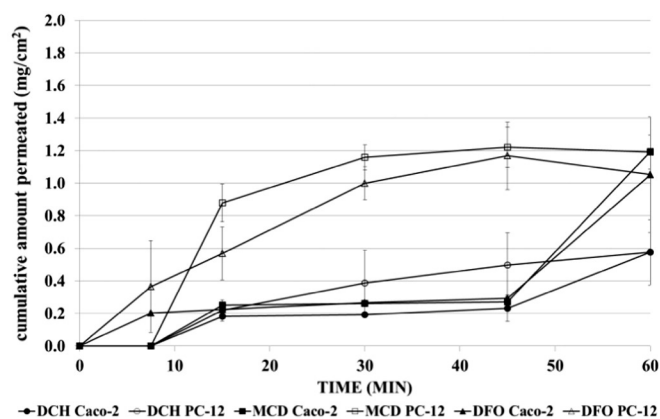
chitosan salts (chloride or glutamate) can be very useful for brain targeting of neuroactive drugs [36,48,49]. We have then formulated the DCH formulation for nasal administration. Very recently,  $\beta$ -cyclodextrins were identified as absorption enhancers for transmucosal delivery of drugs [29,34]; the methylated form is more efficient cholesterol chelator than non-methylated  $\beta$ -cyclodextrin [50]. As a consequence, the MCD microparticulate system was prepared as solid nasal formulation for DFO. A dosage of 200  $\mu$ g DFO was chosen for nasal formulations, in order to minimize the potential peripheral adverse effects of DFO and taking into account that similar nasal dosages allowed us to obtain the CSF uptake of significant amounts of neuroactive drugs, formulated as chitosan based microparticles [36,48,49]. The nasal administration of the aqueous solution of DFO (200  $\mu$ g) did not allow the drug to reach neither the bloodstream or the CSF of rats, as previously found by us after the nasal administration similar doses of poorly permeable drugs formulated as water suspensions [48,49]. This can be explained considering that the liquid formulation could be rapidly removed from the nasal cavity by the mucociliary clearance, the main protection mechanism of the nose. The nasal administration of 200  $\mu$ g DFO contained in the DCH and MCD formulations allowed us to detect significant amounts of the drug both in the bloodstream and the CSF of rats. In particular, the systemic absolute bioavailability values of DFO ranged around 8%. These data indicate the ability of the solid formulations to induce the permeation of DFO across the respiratory mucosa, where systemic absorption occurs. The effect induced by DCH preparation can be attributed to its mucoadhesive properties and to the ability of chitosan to transiently open the tight junctions of the



**Fig. 5.**  $P_{eff}$  (bars) and TER values (line) of formulations compared with drug. Error bars indicate standard deviation of three independent measurements. H and O letters have been used to indicate the permeation through hydrophilic and lipophilic membranes, respectively.  $P < 0.05$ : \*DCH H versus (vs.) MCD H, DCH H vs. DFO H; #DCHL H vs. MCDL H, DCHL H vs. DFO L H; §DCH O vs. MCD O, DCH O vs. DFO O; ‡DCHL O vs. MCDL O, DCHL O vs. DFO O; \*†DFO H vs. DFO O, and DFO H vs. DFO L.

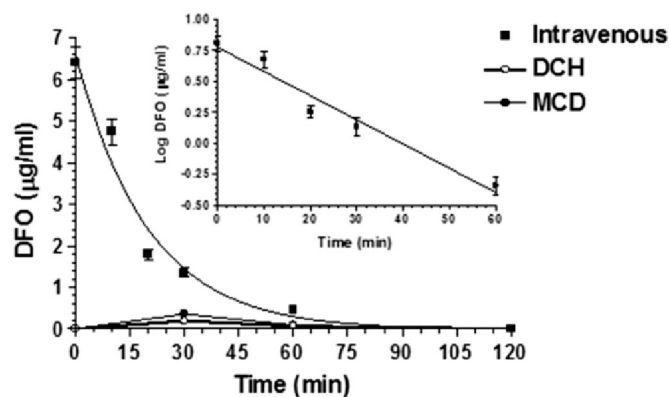


**Fig. 6.** Ex vivo permeation profiles of DFO from microspheres. Profiles are compared with that of DFO as received. Data are reported as the mean  $\pm$  SD of three independent experiments.



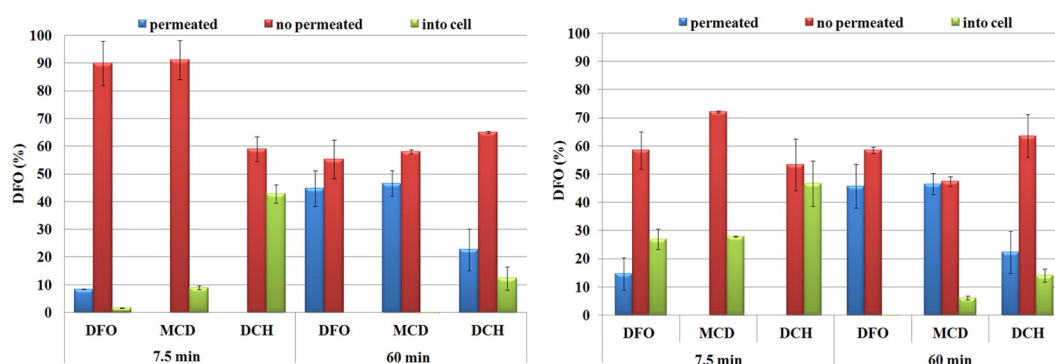
**Fig. 7.** Permeation profiles of DFO from microspheres through Caco-2 (filled symbol) and PC-12 cells (empty symbol). Profiles are compared with that of DFO as received. Data are reported as the mean  $\pm$  SD of two independent experiments.

respiratory mucosa [48,49], whereas the results obtained by the MCD preparation seem to confirm the ability of  $\beta$ -cyclodextrins to act as absorption enhancers for transmucosal delivery of drugs [29,30,34,41]. In the case of water-soluble drugs, CDs enhance drug absorption and/or bioavailability because of their ability to form inclusion complex with specific membrane lipids, such as cholesterol, phosphatidylcholine and sphingomyelin, which cause perturbation of bilayer integrity and increase membrane fluidity [31]. The DFO concentrations detected in CSF of rats after nasal administration of the DCH and MCD preparations indicate their ability to induce also a direct nose to brain pathway for DFO. Indeed, its appearance in CSF after insufflations of the microparticulate powders cannot be imputable to a systemic absorption and a successive crossing of BBB, since this drug was undetectable in CSF after IV administration of the same dose (200  $\mu$ g). DCH preparation confirms therefore the ability of chitosan salt microparticles to induce the CNS uptake from the nose of their encapsulated drug, as previously found by us with other neuroactive drugs [36, 48,49]. This phenomenon has been attributed to the activation of a paracellular pathway across the olfactory epithelium after transient opening of mucosal tight junctions [51,52]. The hypothesis appears supported by the relatively fast appearance of DFO in CSF after nasal administration of DCH preparation. Indeed, 30 min after insufflations, about 3  $\mu$ g/mL of DFO (about 5  $\mu$ M) was detected in the CSF. As a consequence, intracellular axonal transport mechanisms through olfactory neuronal cells seem to be excluded, requiring a long time (up to 24 h) compared to transcellular or paracellular mechanisms that enable the transport of drugs to the CNS within minutes [53]. Interestingly, the MCD preparation appeared more efficient than the DCH formulation in inducing the DFO uptake into the CNS, producing a



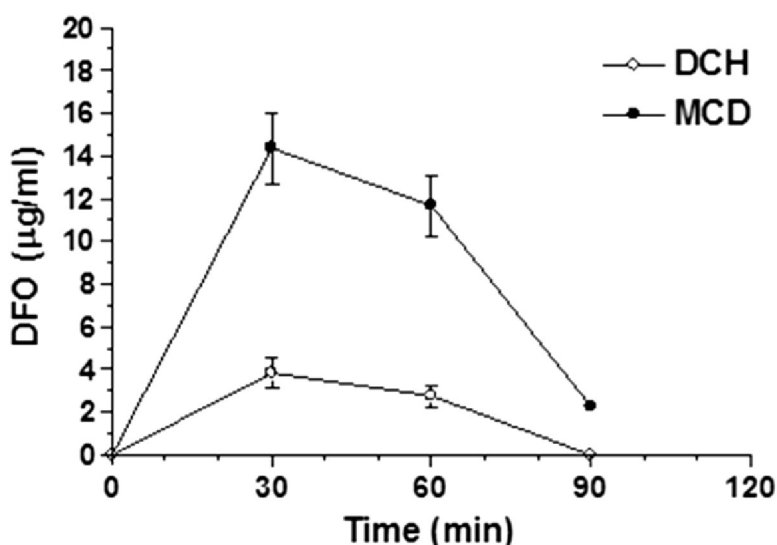
**Fig. 9.** Elimination profile of DFO after 0.2 mg infusion to rats. Data are expressed as the mean  $\pm$  SD of four independent experiments. The elimination followed an apparent first order kinetic, confirmed by the semilogarithmic plot reported in the inset ( $n = 6$ ,  $r = 0.980$ ,  $P < 0.01$ ). The half-life was calculated to be  $15.4 \pm 1.8$  min. The figure reports also the profile of plasma concentrations of DFO following nasal administration of the same dose of the drug encapsulated in DCH and MCD microparticles.

concentration of about 15  $\mu$ g/mL (about 25  $\mu$ M) within 30 min. Accordingly, the CSF bioavailability of DFO in MCD microparticles was about four times higher than the bioavailability obtained by the DCH formulation. It is important to underline that other authors performed a nasal administration of a solution containing 6 mg of DFO to rats; 30 min after the administration they found that the drug concentration in the brain of rats reached values up to 15  $\mu$ M [17]. Taking into account that the nasal administration to rats of a MCD sample containing 200  $\mu$ g of DFO allowed us to detect its concentration up to 25  $\mu$ M in CSF, we can conclude that the  $\beta$ -cyclodextrin microparticles showed great efficacy in inducing the DFO uptake in the CNS. This phenomenon appears due to the ability of this type of microparticle to differently interact with the respiratory and olfactory epithelium, due to the different cell composition [54]. As evidenced by our permeation studies across the monolayers obtained with PC-12 cells and confirmed by in vivo studies, methyl- $\beta$ -cyclodextrin enhances DFO permeation across the neuronal component of the olfactory mucosa. The affinity of the MCD microparticles to a lipophilic substrate was firstly seen in our in vitro permeation studies where unlike the chitosan, cyclodextrin increases the flow of drug through the membranes saturated with octanol. Considering the relatively fast appearance of DFO in the CSF, the intracellular axonal transport mechanisms through olfactory neuronal cells seem to be excluded. For this type of behavior an extracellular pathway along neurons to the CNS has been hypothesized [53]. Intranasal administration, thus, increases drug concentration in the brain and decreases drug systemic circulation avoiding secondary effects. Nose-to-brain delivery of DFO by microspheres is a promising approach also by



**Fig. 8.** Percentage of drug permeated in the acceptor medium, recovered in the donor compartment (no permeated) and calculated into the cells after 7.5 min and 60 min through Caco-2 (left) and PC-12 cell lines (right). Error bars indicate standard deviation of two independent measurements.





**Fig. 10.** DFO concentrations ( $\mu\text{g}/\text{mL}$ ) detected in the CSF after nasal administration of DCH and MCD microparticles. Each dose contained  $200 \mu\text{g}$  of the drug. Data are expressed as the mean  $\pm$  SD of four independent experiments.

considering the recent limitation to the IV administration [4]. Further investigations are required in order to detect the in vivo potential cellular toxicity or therapeutic effects obtained by the nasal administration of the microparticles.

## 5. Conclusions

Microparticles based on chitosan chloride or methyl- $\beta$ -cyclodextrins can be obtained by spray drying as carriers of DFO, in order to promote its nose to brain permeation. In particular, the microparticles based on methyl- $\beta$ -cyclodextrins appear to be of great efficacy in inducing the drug absorption at CSF level, probably across the neuronal component of the olfactory mucosa. Taking into account the poor systemic bioavailability obtained by the nasal administration of the microparticles, the MCD system appears promising for a noninvasive clinical application of DFO as a neuroactive drug in various brain diseases, while minimizing systemic drug exposure.

## Acknowledgment

Authors thank Dott.ssa Claudia Juliano for helping in cell studies.

## Appendix A. Supplementary material

Supplementary data to this article can be found online at <http://dx.doi.org/10.1016/j.jconrel.2015.01.025>.

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