



# Non-Native Structure Appears in Microseconds during the Folding of *E. coli* RNase H

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## Abstract

The folding pathway of *Escherichia coli* RNase H is one of the best experimentally characterized for any protein. In spite of this, spectroscopic studies have never captured the earliest events. Using continuous-flow microfluidic mixing, we have now observed the first several milliseconds of folding by monitoring the tryptophan fluorescence lifetime (60  $\mu$ s dead time). Two folding intermediates are observed, the second of which is the previously characterized  $I_{\text{core}}$  millisecond intermediate. The new earlier intermediate is likely on-pathway and appears to have long-range non-native structure, providing a rare example of such non-native structure formation in a folding pathway. The tryptophan fluorescence lifetimes also suggest a deviation from native packing in the second intermediate,  $I_{\text{core}}$ . Similar results from a fragment of RNase H demonstrate that only half of the protein is significantly involved in this early structure formation. These studies give us a view of the formation of tertiary structure on the folding pathway, which complements previous hydrogen-exchange studies that monitored only secondary structure and observed sequential native structure formation. Our results provide detailed folding information on both a timescale and a size-scale accessible to all-atom molecular dynamics simulations of protein folding.

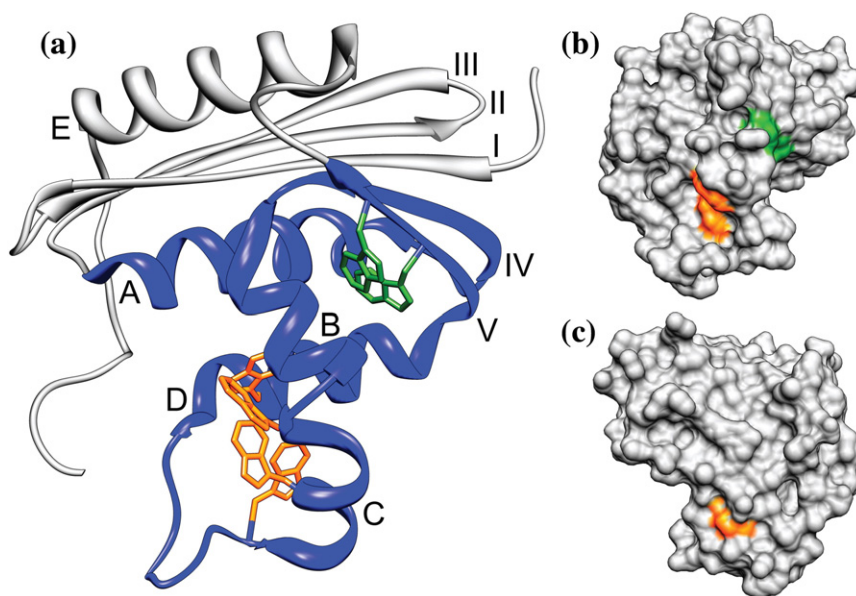
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## Introduction

What does a protein look like as it folds? What partially folded intermediates are populated on the way to reaching the native state? It is particularly difficult to answer these questions for the earliest stages of folding. Protein folding spans a wide range of timescales—some single-domain proteins fold to their native state in microseconds while others need several seconds—but for even the slower folders, many reach partially structured intermediate states within microseconds or milliseconds. These early structural events are important determinants of the overall reaction and are critical for understanding

protein-folding pathways, but are the hardest to observe directly.

While events on the microsecond-to-millisecond timescale are the hardest to observe experimentally, they are the events most accessible to detailed simulations of folding [1]. The improvement of folding simulations—important for our ability to predict protein dynamics and behavior from sequence—requires detailed experimental information for comparison and validation. Here, we use a microfluidic continuous-flow (CF) mixer to observe the early folding of an important protein folding model system, *Escherichia coli* RNase H, at a timescale—and a size scale—amenable to simulation.



**Fig. 1.** Structure of *E. coli* RNase H. (a) Ribbon diagram. Helices are labeled with letters and  $\beta$ -strands are labeled with Roman numerals. The region that is structured in the  $I_{\text{core}}$  intermediate is colored blue. Tryptophan residues are shown in stick (in the 4Trp variant, the two green tryptophans are mutated to phenylalanine, leaving only the four orange tryptophans). (b) Surface contour of the RNase H crystal structure in a very similar orientation as in (a). Only the tryptophan side chains are colored to highlight the solvent exposure of these side chains in the native state. (c) Structure in (b) rotated  $180^\circ$  about the vertical axis. W104 on helix D is the only completely buried tryptophan.

The folding of *E. coli* RNase H has been studied extensively; the protein is known to populate an obligate, on-pathway, partially folded intermediate within several milliseconds of folding, with subsequent folding to the native state occurring in seconds [2,3]. (All work on *E. coli* RNase H discussed here refers to a cysteine-free variant [4,5].) The intermediate, termed  $I_{\text{core}}$ , was initially characterized using pulse-labeling hydrogen exchange (HX) monitored by NMR [2] and mutational analysis [6] and was found to contain native-like secondary structure in approximately half of the protein (Fig. 1a). Although very well characterized, until recently, the folding to this intermediate had never been observed directly, as it occurs within the dead time of a standard stopped-flow or quench-flow instrument.

Recently, using pulse-labeling HX and a novel mass spectrometry (MS) technique, we identified two new early folding intermediates in addition to  $I_{\text{core}}$  [7]. (The experiment was conducted at  $10^\circ\text{C}$  instead of the  $25^\circ\text{C}$  conditions of previous experiments, slowing early folding events so that they were accessible in a quench-flow instrument.) While this work provides detailed structural characterization of early folding events, it gives only a rough sense of the rates associated with these early steps, and it provides no information about tertiary structure formation and its role during the early folding of RNase H.

In the present work, we use ultra-rapid CF mixing to monitor RNase H folding spectroscopically from

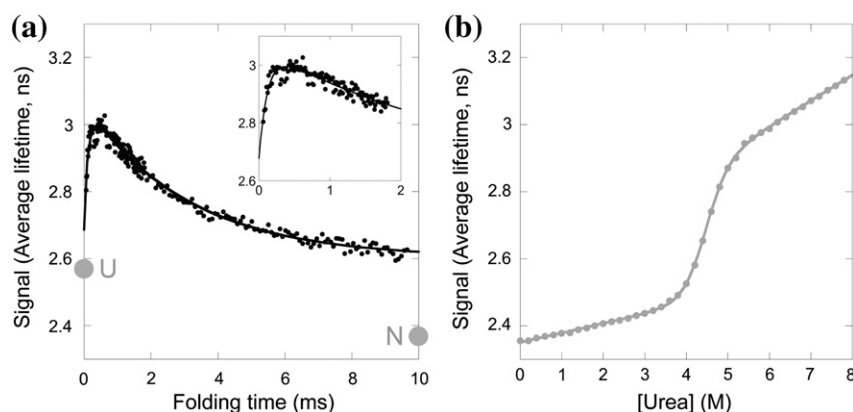
60  $\mu\text{s}$  to 9 ms [8], characterizing early folding kinetics with high temporal resolution. We use intrinsic tryptophan fluorescence to monitor the progress of the folding reaction, providing a window into tertiary structure formation. RNase H has six tryptophans, all within the structured portion of  $I_{\text{core}}$  (Fig. 1).

We observed two kinetic steps in the first few milliseconds of RNase H folding, revealing the formation of a new early intermediate ( $I_{\text{early}}$ ) in addition to the formation of  $I_{\text{core}}$ . Kinetic modeling, mutational analysis, and comparison with the HX-MS data [7] suggest that  $I_{\text{early}}$  is an on-pathway intermediate containing some non-native structure. Using a fragment of RNase H [9], we confirm that only half the protein is significantly involved in these early folding steps. These results, together with the previous HX-MS data [7], provide a detailed model for the early folding of RNase H on both a timescale and size-scale amenable to comparison with atomistic folding simulations.

## Results

### Direct observation of two kinetic phases in the first 9 ms of folding

Folding of *E. coli* RNase H was initiated using a urea concentration jump (6 M to 0.6 M) in a microsecond-resolved CF mixing device with a 60- $\mu\text{s}$  dead time.

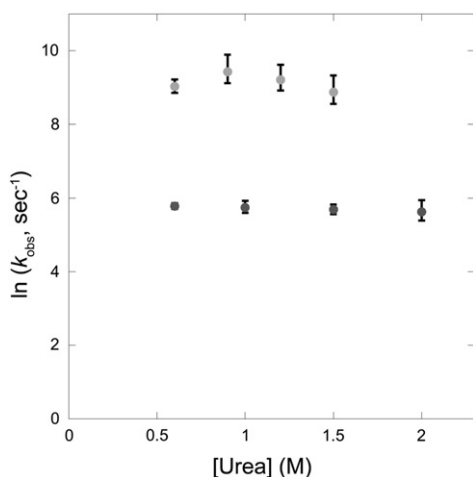


**Fig. 2.** WT *E. coli* RNase H. (a) Refolding into 0.6 M urea monitored by average fluorescence lifetime. For comparison, the extrapolated signal of the unfolded state (U) and the signal of the final equilibrium state (N) at this condition are indicated. Inset: Refolding monitored in just the fastest mixing device. (b) Equilibrium urea denaturation monitored by average fluorescence lifetime. Data are fit with a two-state model.

Folding was monitored by the change in average fluorescence lifetime of the tryptophans, determined using time-correlated single photon counting (TCSPC). Plotting the average fluorescence lifetime *versus* folding time reveals two kinetic phases, clearly distinguishable by the opposite directions of their change in amplitude (Fig. 2a, inset). The time constant of the second kinetic phase, however, is poorly determined under these experimental conditions where folding is only monitored out to ~2 ms.

In order to better resolve the second kinetic phase, we repeated the experiment using a larger CF mixing device that has an approximately 1 ms mixing time,

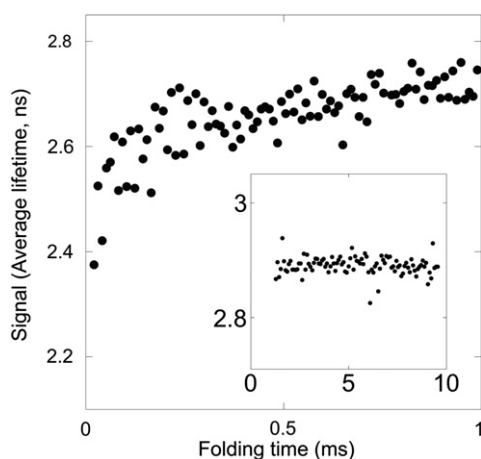
with folding observed out to ~9 ms. The second kinetic phase is well resolved in this time range. The data from both experiments were combined; a fit of the change in average fluorescence lifetime as a function of folding time (Fig. 2a) yields time constants of  $120 \pm 20 \mu\text{s}$  and  $3.1 \pm 0.2 \text{ ms}$  for the first and second kinetic phases, respectively. We are using the average fluorescence lifetime to evaluate the progression of the reaction (see [Materials and Methods](#)) but it should be noted that the opposite trend in amplitudes is observed when monitoring the total fluorescence intensity: the intensity decreases in the first few hundred microseconds and then rises over milliseconds (Fig. S1).



**Fig. 3.** Urea dependence of WT *E. coli* RNase H early folding kinetics. Refolding was monitored by average fluorescence lifetime at multiple final urea concentrations spanning the accessible range of the CF mixing experiments. Error shown is the 95% confidence interval calculated using the standard deviation of the fits, which likely underestimates the true error.

### Equilibrium unfolding monitored by average fluorescence lifetime

To determine the expected signal for the unfolded and native states (starting and end points), we carried out an equilibrium urea titration monitored by TCSPC (Fig. 2b). The resulting cooperative unfolding transition, fit to a two-state model [10], agrees with previous work on RNase H [2], with a  $\Delta G_{\text{unf}}$  of  $9.5 \pm 0.4 \text{ kcal/mol}$  and an  $m$ -value of  $2.13 \pm 0.09 \text{ kcal/mol/M}_{\text{urea}}$ . The average fluorescence lifetime for the native state at 0.6 M urea is less than the final value observed at the end of our millisecond kinetics (2.36 ns *versus* 2.60 ns). Extrapolation of the unfolded baseline back to 0.6 M urea results in an expected average fluorescence lifetime of 2.58 ns (the signal assumed for the unfolded state at the start of the kinetic experiment), while the amplitude of the observed kinetics extrapolated to zero time yields a fluorescence lifetime of 2.69 ns. These values are just outside of our error (error is approximately  $\pm 0.05 \text{ ns}$ , based on an error of  $15 \mu\text{s}$ —one time step—for determining the zero time). The difference could be due to the inaccuracy of such a long



**Fig. 4.** Early folding kinetics of the I53D variant. Refolding of I53D was performed in both mixing devices. A fast kinetic phase is observed in the fastest mixer, analogous to the fastest kinetic phase in the refolding of the WT protein. Inset: No millisecond refolding kinetics are observed. There is no kinetic process overlapping between the two data sets so they cannot be combined.

extrapolation of the unfolded baseline or could indicate that there is a very early small amplitude kinetic phase occurring within our 60  $\mu$ s mixing time.

#### Urea dependence of the observed kinetics

In an effort to obtain insights into the solvent-accessible surface area of the transition state for the two early kinetic steps and to aid in the development of a kinetic model, we repeated the ultra-rapid mixing experiments as a function of denaturant concentration. The results are plotted in Fig. 3 (raw data shown in Fig. S2). The early folding steps are characterized

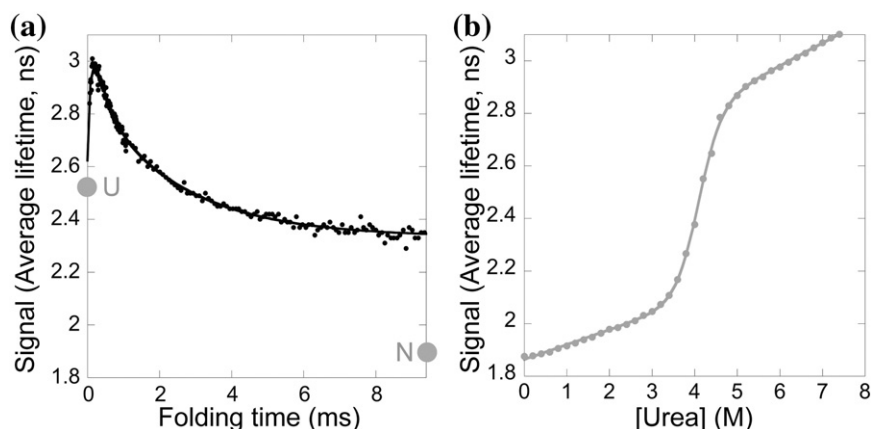
by very low refolding  $m$ -values (denaturant dependence of the barriers), indicative of a transition state that is very unfolded like [11]. The unfolding leg of the chevron for these early steps, expected at  $>2$  M final urea concentration, was not resolved in these experiments because higher urea concentrations decrease the amplitude of both phases and also increases the pressure beyond the physical limits of the mixing devices.

#### The effect of the I53D mutation on early folding

The single amino acid substitution I53D is known to destabilize the  $I_{\text{core}}$  intermediate such that folding monitored by stopped-flow CD appears two state (the  $I_{\text{core}}$  intermediate is never detectably populated) [12,13]. To test whether our CF mixing experiments monitor the formation of the  $I_{\text{core}}$  intermediate, we repeated the same rapid-mixing experiments with the I53D variant. No kinetics were observed in the slower mixer (Fig. 4, inset); however, we still observe a fast kinetic phase in the faster mixer (Fig. 4). (Data from the two mixers are displayed separately since we cannot combine them without an overlapping kinetic process.) Overall, the I53D mutation abolishes the second kinetic phase observed in the wild-type (WT) experiments, indicating that the second phase is indeed formation of  $I_{\text{core}}$ .

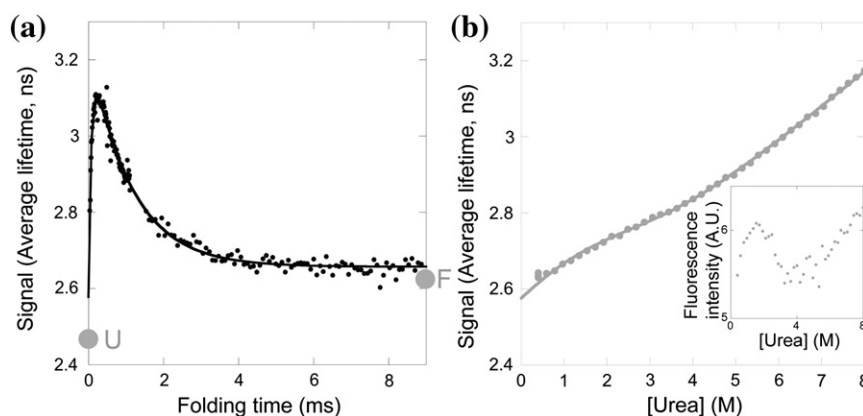
#### Structural information from monitoring select tryptophans

To obtain further structural information about the two kinetic phases, we reduced the number of tryptophan reporters. We made conservative substitutions for two of the six tryptophans, W118F and W120F, creating a variant with just four tryptophans,



**Fig. 5.** The 4Trp variant. (a) Refolding into 0.6 M urea was monitored by average fluorescence lifetime. Two kinetic phases, similar to WT, are observed. The extrapolated signal of the unfolded state (U) and the signal of the final equilibrium state (N) at this condition are indicated. (b) Equilibrium urea denaturation monitored by average fluorescence lifetime. Data are fit with a two-state model.





**Fig. 6.** The  $I_{\text{core}}$  fragment. (a) The fragment refolding into 0.5 M urea was monitored by average fluorescence lifetime. The extrapolated signal of the unfolded state (U) and the final equilibrium state (F) at these conditions are indicated. (b) Equilibrium urea denaturation of the fragment monitored by average fluorescence lifetime, fit to a two-state model by performing a global fit with the total fluorescence intensity data (inset).

all in one region of the protein (herein referred to as the 4Trp variant) (Fig. 1).

Refolding of the 4Trp variant was monitored as described previously (Fig. 5a). The observed kinetics look qualitatively very similar to WT, indicating that the remaining region of tryptophans (Trp81, Trp85, Trp90, and Trp104) is the primary reporter on both kinetic phases observed in the WT experiments. Quantitatively, a global fit of the 4Trp data fits best to three phases rather than to two phases (with time constants of  $91 \pm 2 \mu\text{s}$ ,  $300 \pm 40 \mu\text{s}$ , and  $2.3 \pm 0.2 \text{ ms}$ , all with amplitudes of similar magnitude). The observation of three kinetic phases may indicate that the mutations have altered the early folding pathway or may indicate that the WT protein also folds in three steps but one step is obscured by signal from the additional tryptophans.

An equilibrium urea titration with the 4Trp variant monitored by average fluorescence lifetime (Fig. 5b) can be fit to a two-state model, with a  $\Delta G_{\text{unf}}$  of  $8.5 \pm 0.3 \text{ kcal/mol}$  and an  $m$ -value of  $2.19 \pm 0.08 \text{ kcal/mol/M}_{\text{urea}}$ , indicating that these mutations destabilize the native state by only 1 kcal/mol. Comparing fluorescence lifetimes between the equilibrium titration and the kinetic experiment indicates that the starting point for refolding and the extrapolated equilibrium unfolded state are just outside of error from each other, as seen for WT (2.63 ns *versus* 2.53 ns). The comparison also suggests that the region containing the remaining four tryptophans is not natively folded in the  $I_{\text{core}}$  intermediate: the final fluorescence lifetime of the millisecond kinetics is 2.34 ns while the fluorescence lifetime of the 4Trp native state in 0.6 M urea is 1.90 ns.

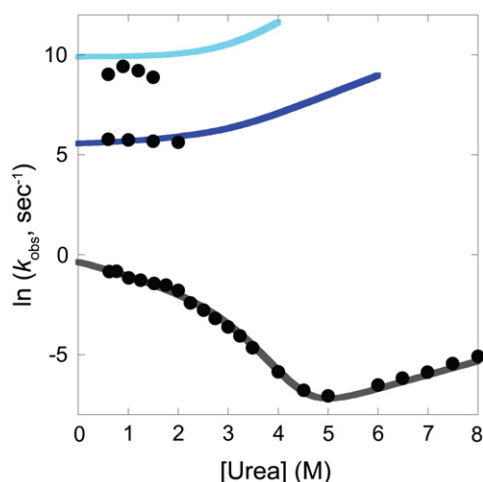
### Structural information from a truncation mutant

The current model of the  $I_{\text{core}}$  folding intermediate does not involve any structure in the region encom-

passing native-state strands I–III and helix E (Fig. 1a) [2,6,7,14]. Since there are no tryptophans in this region of the protein, our kinetic experiments do not report on its role in the early folding pathway. However, we recently made a fragment of RNase H with the I–III/E regions removed [9]. This fragment folds and serves as a mimic of  $I_{\text{core}}$ . Therefore, we carried out the CF folding experiment with this fragment and compared the results to WT to determine whether the I–III/E regions impact the early folding pathway.

The first 9 ms of the folding of the fragment was monitored using a urea concentration jump (5 M to 0.5 M) in both mixers (Fig. 6a). Qualitatively, the kinetics are very similar to that observed in the WT protein: two kinetic phases with similar amplitude changes. When the data are fit together to a two-exponential model, the time constants for the two phases are  $80 \pm 10 \mu\text{s}$  and  $1.11 \pm 0.06 \text{ ms}$ , slightly faster than that observed for WT (even when an extrapolation to 0.5 M is considered). These observations fit with a model where the I–III/E regions of the protein play only a minor role in slowing the early events in the refolding of full-length protein.

The equilibrium urea denaturation of the fragment (Fig. 6b) reveals a steep dependence on urea concentration in the folded and unfolded baselines, making it difficult to fit these data to a model. Globally fitting the average fluorescence lifetime together with total fluorescence intensity (Fig. 6b, inset) to a two-state model yields a stability of  $1.0 \pm 1.0 \text{ kcal/mol}$  and an  $m$ -value of  $0.5 \pm 0.2 \text{ kcal/mol/M}_{\text{urea}}$ ; however, improvements in  $\chi^2$  using a three-state model hint that the fragment may populate an intermediate during equilibrium unfolding. Three-state equilibrium unfolding by fluorescence would explain why the stability and  $m$ -value determined here using a two-state model are lower than previously determined using CD data fit to a



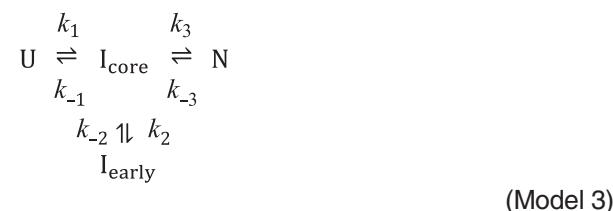
**Fig. 7.** The on-pathway kinetic model. Comparison between results from the global fit and the observed rate constants from the present CF mixing experiments and previous folding and unfolding experiments to/from the native state [2].

two-state model (stability of  $3.0 \pm 0.3$  kcal/mol and  $m$ -value of  $1.16 \pm 0.05$  kcal/mol/ $M_{\text{urea}}$ ) [9]. Nevertheless, the equilibrium average lifetime data are useful to compare to the kinetic amplitudes. The final amplitude of the fluorescence lifetime in the kinetic experiment (2.66 ns) is very similar to the equilibrium fluorescence lifetime of the fragment at 0.5 M urea (2.62 ns), suggesting that the fragment folds to its equilibrium conformation within the time window of our experiment. The kinetic starting point and the extrapolated equilibrium unfolded state are again just outside of error from each other (2.58 ns versus 2.47 ns).

### Fitting a kinetic model

The observation of two kinetic phases indicates folding to an early intermediate (which we now term  $I_{\text{early}}$ ) in addition to the formation of  $I_{\text{core}}$ .  $I_{\text{core}}$  is

known to be an obligate and on-pathway intermediate [3]. There are three ways to incorporate  $I_{\text{early}}$  into this folding pathway without invoking any unjustified complexity:  $I_{\text{early}}$  is on-pathway to  $I_{\text{core}}$  (Model 1),  $I_{\text{early}}$  is off-pathway from the unfolded state (Model 2), or  $I_{\text{early}}$  is off-pathway from  $I_{\text{core}}$  (Model 3). These three models are depicted below.



Results for the I53D variant allow us to discount Model 3. The I53D variant populates the intermediate with the long fluorescence lifetime ( $I_{\text{early}}$ ) on a 100- $\mu$ s timescale without build-up of  $I_{\text{core}}$  intermediate (Fig. 4). This result is only compatible with Model 3 if I53D  $I_{\text{core}}$  forms on this 100- $\mu$ s timescale and  $I_{\text{early}}$  formation occurs even faster. By comparison to WT, this would require that one early folding step or both early folding steps were significantly sped up by the destabilizing I53D mutation—a very unlikely scenario. In contrast, both Models 1 and 2 allow for the I53D mutation to significantly slow down the formation of  $I_{\text{core}}$ , which is much more probable.

We performed global fits of the data based on Models 1 and 2 to determine how consistent these

**Table 1.** Parameters from the global fit to Model 1.

	Global fit parameters <sup>a</sup>	Previously published [2]
$k_1$ ( $s^{-1}$ )	$20,000 \pm 3000$	n.a.
$k_{-1}$ ( $s^{-1}$ )	$130 \pm 150$	n.a.
$k_2$ ( $s^{-1}$ )	$240 \pm 80$	n.a.
$k_{-2}$ ( $s^{-1}$ )	$24 \pm 20$	n.a.
$k_3$ ( $s^{-1}$ )	$0.78 \pm 0.11$	$0.74 \pm 0.02$
$k_{-3}$ ( $s^{-1}$ )	$1.9 \times 10^{-5} \pm 0.1 \times 10^{-4}$	$1.69 \times 10^{-5} \pm 0.04 \times 10^{-5}$
$m_1$ (kcal/mol/M)	0.0	n.a.
$m_{-1}$ (kcal/mol/M)	-0.97	n.a.
$m_2$ (kcal/mol/M)	0.0	n.a.
$m_{-2}$ (kcal/mol/M)	-0.57	n.a.
$m_3$ (kcal/mol/M)	0.34	$0.454 \pm 0.005$
$m_{-3}$ (kcal/mol/M)	-0.41	$-0.42 \pm 0.03$
$\Delta G_{U \rightarrow \text{early}}$ (kcal/mol)	$-3.0 \pm 0.7$	n.a.
$\Delta G_{U \rightarrow \text{core}}$ (kcal/mol)	$-4.3 \pm 1.2$	-3.5
$\Delta G_{U \rightarrow N}$ (kcal/mol)	$-10.6 \pm 1.3$	-9.9

<sup>a</sup> Error propagated from the 68% confidence interval from a rigorous error analysis.

models are with our measurements, as well as to determine possible values for the microscopic rate constants at each kinetic step. Models were fit simultaneously to the raw data from the CF and equilibrium experiments (Fig. S2 and Fig. 2b). To constrain the fits to be consistent with previous stopped-flow and manual mixing results, we simulated kinetic traces for refolding and unfolding to/from the native state according to the parameters in Raschke and Marqusee [2] and included them with appropriate weighting in the global chevron analysis (see Materials and Methods).

The on-pathway model (Model 1) reasonably fits the raw kinetic data, rate constants, and amplitudes from the kinetic experiments and reproduces the equilibrium thermodynamics (Fig. 7, Figs. S2 and S3, and Table 1). The deviations between the fit and the data are of a reasonable magnitude considering the number of simultaneous data sets being considered and the small difference in temperature between previous and present experiments (the CF experiments were performed at 21 °C while the experiments in Raschke and Marqusee [2] were performed at 25 °C).

In contrast, it is not possible to reasonably fit the off-pathway model (Model 2) (Fig. S3). If forced, the off-pathway model can reproduce the experimentally observed rate constants (eigenvalues) reasonably well, but this results in unphysical amplitudes (e.g., negative average fluorescence lifetimes) and statistically poorer fits ( $\chi^2_{\text{off}}/\chi^2_{\text{on}} = 1.24$ ,  $P < 10^{-6}$ ). This strongly rules out Model 2, suggesting that  $I_{\text{early}}$  is likely an on-pathway intermediate.

Because the global kinetic analysis is constrained by the equilibrium unfolding curve and the stopped-flow/manual mixing results, thermodynamic information can be obtained about the two intermediates,  $I_{\text{early}}$  and  $I_{\text{core}}$ , without direct unfolding kinetic data. The global analysis suggests that  $I_{\text{early}}$  has a stability of  $3.0 \pm 0.7$  kcal/mol (with an  $m$ -value of 0.97 kcal/mol/ $M_{\text{urea}}$ ) and that  $I_{\text{core}}$  is slightly more stable:  $4.3 \pm 1.2$  kcal/mol (with an  $m$ -value of 1.5 kcal/mol/ $M_{\text{urea}}$ ) (errors were propagated from the 68% confidence interval from a rigorous error analysis, see Materials and Methods). The equilibrium  $m$ -value for both intermediates is dominated by the unfolding leg of the chevron, consistent with an unfolded-like transition state for formation of  $I_{\text{early}}$  and an  $I_{\text{early}}$ -like transition state leading to  $I_{\text{core}}$ .

## Discussion

Using a CF rapid mixing device, we can now directly observe the first several milliseconds of *E. coli* RNase H folding (with a 60- $\mu$ s dead time) by monitoring the change in tryptophan fluorescence. Two kinetic phases are clearly observed: one on the hundred-microsecond timescale and one, an order of magni-

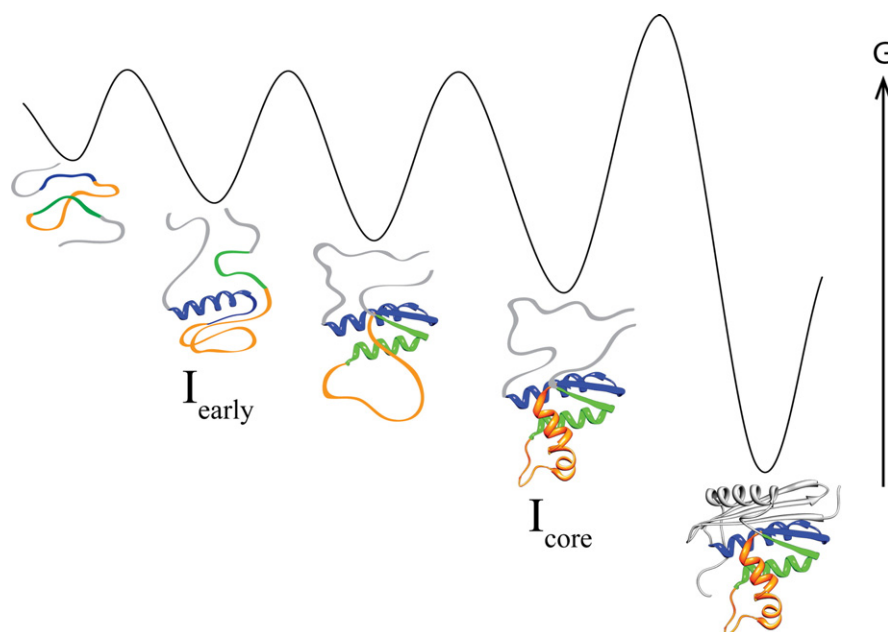
tude slower, on the millisecond timescale. These kinetics represent folding to intermediates, with final folding to the native state occurring on a slower timescale (seconds), beyond the range of these experiments. Experiments with the I53D variant demonstrate that the millisecond-timescale kinetic phase is the formation of the previously characterized  $I_{\text{core}}$  intermediate. Therefore, our data indicate that we are observing the transient population of an early intermediate ( $I_{\text{early}}$ ) in addition to formation of  $I_{\text{core}}$ . Kinetic modeling demonstrates that  $I_{\text{early}}$  likely forms on-pathway to  $I_{\text{core}}$ .

## Structural interpretation

How does the change in tryptophan environment give rise to the observed fluorescence lifetime for each state? Fluorescence quenching (often characterized by a shorter lifetime) is typically associated with solvent exposure but may also arise from close proximity of quenching-competent groups such as other tryptophans, phenylalanines, protonated histidines, charged amino groups, and protonated carboxyl groups [15–18]. Quenching by other tryptophans is a strong possibility in RNase H because four tryptophans (green in Fig. 1) are within 5–7 Å of each other in the native structure and the two additional tryptophans (118 and 120, orange in Fig. 1) are sequence local and also stacked cofacially in the native structure [19].

$I_{\text{early}}$  folds within the first few hundred microseconds and has a longer fluorescence lifetime relative to all other states. This is true for all constructs, including the 4Trp variant, indicating that the tryptophan region in helices C and D (orange in Fig. 1) undergoes a dramatic change in environment from U to  $I_{\text{early}}$  and from  $I_{\text{early}}$  to  $I_{\text{core}}$ . The long lifetime suggests that these tryptophans are significantly protected from solvent due to burial in a collapsed structure. It is noteworthy that this presumably collapsed region overlaps with the most extensive network of isoleucine, leucine, and valine (ILV) residues in the native protein (Fig. S4)—ILV residues have been shown to be important for guiding early structure formation [20]. Interestingly, the average fluorescence lifetime is much longer in  $I_{\text{early}}$  than in the native state, suggesting the presence of non-native structure (consistent with a model incorporating the previous HX data, described below).

While a longer fluorescence lifetime would be predicted to correlate with an increase in total fluorescence intensity,  $I_{\text{early}}$  exhibits slightly decreased fluorescence intensity relative to the unfolded state and  $I_{\text{core}}$  (Fig. S1). This is likely because we are monitoring fluorescence only at the peak emission for solvent-exposed tryptophans (335–379 nm) and the tryptophan emission spectrum shifts in hydrophobic environments. (Though this observation could also arise in the event of



**Fig. 8.** A model for the early folding pathway of RNase H. This model incorporates the results presented here for the early folding monitored by fluorescence and the HX-MS results from Hu *et al.* [7].

preferential quenching of Trp residues or rotamers with shorter lifetimes.)

The transition from  $I_{\text{early}}$  to  $I_{\text{core}}$  is accompanied by a reduction in the fluorescence lifetime followed by an even further reduction when  $I_{\text{core}}$  folds to the native state. In the native state, most of the tryptophans have some solvent exposure (Fig. 1b and c) and all are packed against other tryptophans, both of which are undoubtedly the basis of the observed quenching. Previous studies using HX [2,7], mutagenesis [6], and equilibrium mimics [9] all suggest a native-like topology for  $I_{\text{core}}$ . However, since all of the tryptophans reside within the structured region of  $I_{\text{core}}$ , the non-native fluorescence signal of  $I_{\text{core}}$  suggests that it contains some non-native features. Importantly, this is also true for the 4Trp variant where the tryptophans adjacent to the interface with the unstructured region of  $I_{\text{core}}$  have been removed.

Previously,  $I_{\text{core}}$  was thought to be a molten globule [2,21], which could account for the observed difference in fluorescence between  $I_{\text{core}}$  and N. However, we recently determined that  $I_{\text{core}}$  likely has closely packed side chains [9]. If this is the case, there must be, at a minimum, a small reorientation of tryptophan side chains as the protein folds from  $I_{\text{core}}$  to N, and therefore,  $I_{\text{core}}$  must have non-native packing. This reorientation must occur in a way that makes the tryptophans in the native state more solvent exposed or else packed slightly differently so as to be more efficient at quenching each other or more efficiently quenched by neighboring side chains. In the future, monitoring anisotropy [22] and/or iodide quenching could be used

to directly address the mobility and solvent exposure of the tryptophans in  $I_{\text{core}}$ , as well as in  $I_{\text{early}}$ .

### A detailed model of the folding pathway

Any model for early folding must be reconciled with previous experiments. We recently explored the folding pathway of RNase H using a high-resolution pulse-labeled HX experiment monitored by MS [7] and found that folding occurred in three steps by sequential addition of native units of secondary structure: helix A and strand IV, then helix D and strand V, then finally helices B and C, forming the previously characterized  $I_{\text{core}}$  intermediate. In view of this work, our present results are most readily explained by the model depicted in Fig. 8.

The initial burial of the tryptophans on helices C and D must precede or accompany the protection observed in helix A and strand IV. According to the scheme in Fig. 8,  $I_{\text{early}}$  may be the same species as the on-pathway, HX-MS A/4 intermediate, with non-native structure (orange) stabilizing the first native secondary structure elements (blue). This interaction would explain the mystery of why helix A is the first helix to show protection, when it is not the helix with the highest intrinsic helicity (which is helix E). This model makes a case for non-native structure formation that is on-pathway, supporting the folding of native units of secondary structure.

In subsequent folding steps, the non-native interaction is replaced by native-like packing between helices A and D, and then finally helices B and C assemble



to form  $I_{\text{core}}$ . This model is consistent with our observation of three-state early folding in the 4Trp variant, suggesting that, in the WT protein, the strand V tryptophans (green in Fig. 1) obscure observation of the latter two folding events as discrete steps. Additionally, while Fig. 8 depicts the  $I_{\text{core}}$  intermediate as containing secondary structure elements in a native-like orientation, native-like packing has not been achieved, as discussed above. This intermediate may be best described as a subset of native secondary structure and topology but with some non-native tertiary interactions. Such intermediates have been observed for other protein folding model systems [23–25]. Perhaps resolution of closely packed non-native structure in  $I_{\text{core}}$  is what creates the subsequent rate-limiting barrier, as we have previously suggested [9].

The scheme in Fig. 8 also predicts that the I–III/E regions (gray) are not involved in the early folding pathway and only become structured in the rate-limiting step. This is supported by our observation that a fragment of the protein encompassing the contiguous sequence from helix A to strand V folds with kinetics and amplitudes very similar to the WT protein. Because the fragment folds [9] and appears to mirror the early folding of WT RNase H, the fragment could be a good target for atomistic folding simulations, which are generally limited to small proteins. (Proteins studied by detailed folding simulations are typically no larger than ~80 amino acids in size [1]; the fragment is 81 amino acids.)

Overall, we have presented a detailed model for the early folding pathway of RNase H at a timescale that is currently accessible by atomistic simulations of folding. Moreover, we show that the region of the protein gaining structure on this timescale is on a size-scale amenable to atomistic simulations. Our model makes the prediction that non-native structure is present in multiple on-pathway intermediates. This underscores the importance of using methods that monitor both secondary and tertiary structure when analyzing protein-folding pathways.

## Materials and Methods

### Materials

Cysteine-free *E. coli* RNase H was expressed and purified as described previously [5], with the following modifications. Heparin column fractions containing RNase H were diluted with water 1:1 and adjusted to a pH of 5.5. Precipitated protein was removed by centrifugation and the supernatant was further purified using cation-exchange chromatography (HiTrap Capto S column). Purified protein was dialyzed against 50 mM ammonium bicarbonate, lyophilized, and resuspended into the designated buffer before use.

The I53D variant [12] and the fragment variant [9] were constructed previously. The 4Trp variant was generated using

the QuikChange mutagenesis protocol, using pSM101 as a template, and then subcloned into a pET27 vector.

Expressions were performed in the Rosetta2(DE3) pLysS strain. The 4Trp variant was purified similar to the WT. The fragment expresses insolubly in inclusion bodies and was purified according to the protocol from Rosen *et al.* [9]. The I53D variant was purified the same way, but without the final size chromatography step.

The purity and molecular mass of all the variants were verified by SDS-PAGE and MS (data not shown).

### Experimental conditions

For all experiments, the buffer conditions were 20 mM sodium acetate (pH 5.5) and 50 mM potassium chloride. All experiments were performed at room temperature (21 °C). Final protein concentration after mixing was 1.2  $\mu\text{M}$  for WT, fragment, and I53D and was 1.5  $\mu\text{M}$  for the 4Trp variant.

### Time-resolved tryptophan fluorescence

Details of the TCSPC apparatus equipped with a very similar microsecond CF mixer have been previously described [8,22]. In this work, we use mixers that are single-piece quartz microfluidic devices from Translume Inc. (Ann Arbor, MI). The faster one has custom, arrow-shaped geometry with a 75  $\mu\text{m} \times 100 \mu\text{m}$  cross-section and the large one has a T-shaped geometry with a 300  $\mu\text{m} \times 300 \mu\text{m}$  cross section. Both are 3 cm in length after the point of mixing. Prior to use, both mixers were covalently modified using silanated polyethylene glycol in order to passivate the surfaces and prevent protein adsorption on the channel walls. This was performed by slowly flowing 2-[methoxy(polyethyleneoxy)propyl]trimethoxysilane, tech-90, through the mixer for over 3 h followed by rinsing with methanol.

Flow to the microchannel mixer was provided by two syringe pumps (500D; Isco Inc., Lincoln, NE) operating at a combined flow rate of 8 mL/min in the smaller mixer, corresponding to an ~60- $\mu\text{s}$  dead time, and 15 mL/min in the larger mixer, corresponding to an ~2-ms dead time (the dead time is the mixing time). The exception is that the 0.6 M urea WT experiment in the smaller mixer was performed with a 7 mL/min combined flow rate. Excitation at 293 nm with a repetition rate of 3.8 MHz was provided by the vertically polarized third harmonic of a Ti:sapphire laser. Typical excitation power was 500  $\mu\text{W}$ . Tryptophan emission was measured using a 357-nm filter with 44 nm bandwidth from Semrock Inc. (Rochester, NY). A Glan-Taylor polarizer at magic angle (54.7° from vertical) orientation was used for all measurements. The variation in excitation intensity along the flow channel was corrected using a standard (*N*-acetyl-L-tryptophanamide), as described previously [8]. Separate instrument responses were recorded for each channel by recording of scattered light signal or by numerical deconvolution from the *N*-acetyl-L-tryptophanamide decay curve. The photon count rate was typically  $7 \times 10^4$  to  $1 \times 10^5$  cps. Each scan of the channel consisted of approximately 120 points, with 0.5 s/point. Typically, 6–8 scans were summed.

The excited-state lifetime,  $\tau_{\text{ave}}$ , was determined by calculating the first moment (center of mass) for the time-resolved

fluorescence decay, based on the following equality:

$$\tau_{\text{ave}} = \frac{\sum_i \alpha_i \tau_i^2}{\sum_i \alpha_i \tau_i} = \frac{\sum_j I_j t_j}{\sum_j I_j}$$

where  $\alpha_i$  and  $\tau_i$  are the amplitude and lifetime of the  $i$ th phase in a Trp fluorescence decay curve, and  $I_j$  is the intensity at the  $t_j$  point in decay time. The use of the first moment was adopted because it provided a model-independent approach for monitoring the kinetics based on a metric that was insensitive to protein concentration and relatively independent of signal to noise. The latter feature allowed more robust comparisons between the high signal-to-noise equilibrium experiments and kinetic data. For all data sets, the first moment was calculated over the  $-0.16$  to  $14.7$  ns time range relative to the zero time defined by the excitation pulse arrival.

### Fits to the kinetic data

The kinetic data were fit to one exponential or two exponentials according to

$$\text{Signal} = A \exp(-k_{\text{obs}} t) + C,$$

or

$$\text{Signal} = A_1 \exp(-k_{\text{obs},1} t) + A_2 \exp(-k_{\text{obs},2} t) + C,$$

to determine the rate constants for the observed kinetic phases (in the text, we report time constants of the kinetic phases: the time constant is the reciprocal of the observed rate constant). The data from experiments performed in both mixing devices (WT 0.6 M and 1.5 M urea, 4Trp, and fragment experiments) were combined and fit to two exponentials, and both observed rate constants were considered. Experiments performed in only the fastest mixing device (WT 0.9 M and 1.2 M urea) were fit to two exponentials but only the fastest observed rate constant is considered. Experiments performed only in the slower mixing device (WT 1 M and 2 M urea) were fit to one exponential. Where shown, the 95% confidence intervals were calculated using the standard deviation of the fits and likely underestimate the true error. Fits were performed using Savuka. To validate the use of the first moment, we also carried out fits of the kinetic data using a global analysis that incorporated the first moment weighted by the total intensity. Because the total intensity (used as a proxy for quantum yield) change throughout the microsecond-to-second time range was less than 20%, fits of the first moment did not depend on weighting by quantum yield.

### Equilibrium titrations

Equilibrium unfolding experiments were set up using a titrator and measured using a homemade autosampler assembled from linear stepper motor stages (Zaber Technologies Inc., Vancouver, British Columbia, Canada) and a Hamilton Microlab 500 syringe pump. Signal was measured using the same TCSPC apparatus as used in the kinetic experiments. For comparison to kinetic data, signal from the equilibrium titration was adjusted to match equilibrium end-points measured within the mixing device.

### Fitting kinetic models to the chevrons

Global fits of the kinetic data at various final denaturant concentrations were accomplished by the “cumbersome” [26] approach of fitting the raw data. The amplitudes and time constants were calculated from the eigenvalues and eigenvectors of a general kinetic master equation as detailed previously [27,28]. The baselines for the optical properties of the intermediates were assumed to follow the Z-approximation, where  $Z = (Y_{\text{int}} - Y_{\text{native}})/(Y_{\text{unfolded}} - Y_{\text{native}})$  and  $Y_{\text{unfolded}}$ ,  $Y_{\text{native}}$ , and  $Y_{\text{int}}$  are the optical properties of the unfolded, native, and intermediate species (e.g.,  $I_{\text{early}}$  or  $I_{\text{core}}$ ), respectively, in the absence of denaturant. For the fits utilizing Model 1 and Model 2, the refolding  $m$ -values were constrained to be positive and unfolding  $m$ -values are constrained to be negative. Equilibrium data were included in the global fits with the thermodynamic parameters calculated from the microscopic rate constants and associated  $m$ -values.

Ideally, the global fit would utilize raw kinetic data from the previous stopped-flow CD work with RNase H in Raschke and Marqusee [2]. We attempted this approach, but unfortunately, the signal to noise of these data on the millisecond timescale was insufficient to constrain the global fits so that the slowest kinetic step agreed with the published work. Therefore, in order to be able to use the longer timescale data published previously [2], we created a synthetic data set based on the published chevron. The noise level of the synthetic data set was adjusted to the maximum amount that produced eigenvalues statistically within the 68% confidence interval of those in Raschke and Marqusee as determined by a Z-statistic comparing the residuals of the two chevrons. When the synthetic data set is included in the global fit with our CF data, the resulting microscopic rate constants for the final step of the mechanism,  $I_{\text{core}}$  to N in this manuscript and I to N in Raschke and Marqusee, are nearly identical.

Errors for the microscopic kinetic rates were determined using sensitivity (rigorous) analysis. This was accomplished by sampling of the parameter over parameter space (typically 20–40 points) and keeping the parameter fixed at each value while allowing all other parameters to vary as before. The chi-square obtained for each point along the parameter space was used to obtain a one-dimensional chi-square surface. This was converted to an  $F$ -statistic (using the minimum chi-square value for the ratio) and converted into a one-dimensional probability curve to obtain the confidence interval for each parameter.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <http://dx.doi.org/10.1016/j.jmb.2014.10.003>.

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### Abbreviations used:

HX, hydrogen exchange; MS, mass spectrometry; CF, continuous flow; TCSPC, time-correlated single photon counting; WT, wild type.

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