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Simple and fast evaluation of relaxation parameters of magnetic nanoparticles

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Abstract

The efficacy of magnetic hyperthermia treatment depends on the optimal available magnetic nanoparticles (MNPs) that are excited in a given alternating magnetic field and viscosity of the region of interest. In this regard, assessing the relevant relaxation parameters is of utmost importance and could improve the speed of development of efficient applications. Here, we demonstrate how to deduce all relevant magnetic parameters from fast, independent, and simple experimental measurements such as dynamic light scattering, vibrating sample magnetometer, and lock-in thermography. We study the thermal behavior of two MNPs with different forms, i.e. spherical and cubical, synthesized in-house by thermal decomposition and coated with 4 different surface agents. By determination of specific absorption rate (SAR) values, hydrodynamic diameters and $M-H$ curves it is possible to compute the magnetic particle volume, dominant relaxation time and magnetocrystalline anisotropy constant. The calculated SAR values derived from these parameters, show good agreement with the experimental determined SAR data, demonstrating the applicability of the reported procedure. Additionally, our results indicate that surface coatings can have minor impacts on the thermal dissipation of Néel relaxation dominated MNPs.

Introduction

Magnetic nanoparticles (MNPs) gained much attention in recent years, due to their biocompatibility,[1] biodegradability,[2,3] superparamagnetic properties,[4] and ability to convert magnetic energy into heat.[1,5,6] The conversion of magnetic energy, generated by an alternating magnetic field (AMF), into thermal energy is based on relaxation mechanisms, which can be categorized into Brownian and Néel relaxations.[7] In short, Brownian relaxation describes the complete rotation of the entire particle towards the direction of the applied external magnetic field and generates heat *via* rotational friction,[8] while in Néel relaxation only the magnetic moment inside the particles follows the magnetic field direction and the magnetic energy is dissipated as heat.[7] The efficiency of this conversion is dictated by external parameters including magnetic field frequency [9] and amplitude,[10] as well as by several intrinsic particle properties, such as crystallinity,[11] surface coating,[6] polydispersity index,[12] colloidal stability,[13] and size.[14] In addition, the latter factor, particle size, determines to a large extent the effective relaxation mechanism.[7,14] Even though the effective relaxation is a combination of both, it is widely accepted that in small MNPs Néel relaxation is the dominant mechanism, while Brownian relaxation dominates for particles with a size above a certain threshold.[14] This value can vary from batch to batch depending upon the magnetocrystalline anisotropy constant, which can be described as the energy needed to change the direction of the dipole moment from the easy axis of magnetization.[15] In short, the higher the magnetocrystalline anisotropy constant for a given particle, the smaller is the nanoparticle size for which the transition from Néel- to Brownian-dominated relaxation occurs.[7] To achieve efficient magnetic hyperthermia treatment, it is of utmost importance to know which is the dominant relaxation mechanism, since Brownian relaxation is hindered at high viscosities and will not dissipate sufficient heat to alter the cells.[16] Hence, the intended area of application and the used magnetic field frequency dictates which relaxation particle type is favored to yield maximum heat dissipation.[7] However, assessing the dominant relaxation mechanism and magneto-crystalline anisotropy constant are challenging and require a combination of several characterization techniques. Brownian relaxation is related to the viscosity coefficient η of the surrounding fluid, the absolute temperature T , and the hydrodynamic volume of the nanoparticle V_H . [14] While viscosity and temperature are easy to determine, the hydrodynamic particle volume requires more elaborated measurement techniques, e.g. dynamic light scattering (DLS) or Taylor dispersion analysis (TDA).[17,18] In contrast, the characteristic Néel relaxation time is given by the characteristic time τ_0 and the ratio of the magnetocrystalline anisotropy constant K_V and the magnetic particle volume V_p to the absolute temperature T . [19] The value of τ_0 is often approximated as $\sim 10^{-9}$ s,[20] the magnetic particle volume can be estimated from the magnetization (M - H) curve,[21] while K_V requires more complex experiments and/or calculations based on susceptibility measurements or M - H curves of MNP samples, since its values for nanoparticles often differs substantially from the corresponding bulk material values.[22,23]

While it was just recently reported how to evaluate K_V , V_p , and τ_0 using the M - H curve and AC susceptibility of immobilized and suspended MNPs,[24] we present here a

straightforward way to assess the same parameters using simple measurement techniques. In detail, by determining the hydrodynamic diameters *via* DLS, the *M-H* curves *via* vibrating sample magnetometer (VSM), and the specific absorbance rate (SAR) values *via* lock-in thermography (LIT), we were able to deduce the above-mentioned properties. For that purpose, we synthesized 2 different MNPs batches by thermal decomposition method, transferred them to water, coated each particle batch with 4 different surface agents, and subsequently analyzed the most common particle characteristics using simple and fast measurement techniques. The validity of this approach is shown by the good agreement between experimental assessed and modelled SAR values. Furthermore, our results implicate that surface coatings have little impact on the heating efficiency of Néel relaxation dominated MNPs.

2. Materials and methods

2.1 SPIONs synthesis and coating

Cubic and spherical superparamagnetic iron oxide nanoparticles (SPIONs) were synthesized by thermal decomposition method described elsewhere.[11,25,26] Cubic morphology was achieved by introducing sodium oleate into the reaction.[11,27] The obtained particles were transferred into an aqueous environment by a ligand exchange with citric acid (CA).[28] Catechol-conjugated (Cat-) polymer (Polyethylene glycol, PEG and Polyvinyl alcohol, PVA) was synthesized following a published procedure [29] and dissolved in Milli-Q water. Catechol-PEG was synthesized by a modified procedure from ref. 29. In brief, 6 mL of dry dimethylformamide (DMF) and 3 mL of dichloromethane (DCM) were introduced into a balloon. The liquids were degassed with argon and 500 mg PEG-NHS was added to the solution. Subsequently, 95 mg dopamine hydrochloride and trimethylamine were added and the reaction was stirred overnight at RT. CA coated SPIONs were added (10 molecules polymer/nm² nanoparticle surface) and the mixtures were sonicated at room temperature overnight. No further purification was done thus keeping the iron concentration constant at 4 mg/mL for all applied coatings. This procedure was performed for both morphologies and resulted in Cat-PEG 5kDa, Cat-PEG 2kDa, and Cat-PVA 5kDa coated SPIONs.

2.2 Particle analysis

The iron content of CA coated SPIONs was analyzed according to a procedure published elsewhere [12] using a PerkinElmer Avio 200 inductively coupled plasma optical emission spectroscopy (ICP-OES). Diluted samples of CA-coated cubic and spherical SPIONs were prepared for transmission electron microscopy (TEM) by a reported procedure.[30] In brief, by mixing MNPs with a defined concentration of bovine serum albumin (BSA), the individual particles get stabilized against aggregation and drying artefacts are avoided. TEM images were collected with a side-

mounted Olympus Veleta CCD camera using a FEI Tecnai Spirit, operating at a voltage of 120 kV. X-ray diffractograms (XRD) were collected with an Ultima IV X-Ray diffractometer from Rigaku. The dried samples were measured between 20° and 80° and the resulting diffractograms were normalized to show the same maximum peak height. Zeta potential measurements were estimated via the Henry equation using the Smoluchowski approximation and measured at 25°C in a Milli-Q water/PBS mixture using a Brookhaven ZetaPALS. Dynamic light scattering (DLS) experiments were carried out on a 3D LS spectrometer of LS Instruments at 25 °C as described elsewhere.[11] To assess the magnetic properties, coated nanoparticle suspensions were deposited onto hydrophobic cotton pellets and dried. Vibrating sample magnetometer (VSM) measurements were carried out on a Model 3900 VSM of Princeton Measurements Corporation. Magnetization curves were measured at room temperature and normalized by the dried sample mass of iron. All thermal measurements of nanoparticle dispersions were conducted with an in-house developed lock-in thermal (LIT) imaging set-up.[31,32] MNP heating was induced in an alternating magnetic field of 523 kHz – 10.45 mT, 331 kHz – 7.62 mT, 173 kHz – 6.04 mT, and 110 kHz – 5.97 mT, respectively, and recorded with an infrared camera. VSM and LIT data were analysed by the Shapiro-Wilk normality test and their significance level determined using the two-sample t-test.

2.3 Modelling of heat dissipation

Linear response theory (LRT) was used to interpret the heat generated by SPIONs exposed to an alternating magnetic field. According to this theory, the power P generated per unit mass of particles exposed to a magnetic field with an amplitude H_0 , frequency f :

$$\begin{aligned}
 P &= \pi \mu_0 H_0^2 f \chi_0 \frac{2\pi f \tau}{1 + (2\pi f \tau)^2} \\
 \chi_0 &= \frac{\mu_0 M_s^2 V_p}{3\rho kT} \frac{3}{\xi} \left(\coth(\xi) - \frac{1}{\xi} \right) \\
 \xi &= \frac{\mu_0 M_s V_p H_0}{kT} \\
 \frac{1}{\tau} &= \frac{1}{\tau_B} + \frac{1}{\tau_N} \Rightarrow \begin{cases} \tau_B = \frac{3\eta V_H}{kT} \\ \tau_N = \frac{\sqrt{\pi}}{2} \tau_0 \frac{\exp(\sigma)}{\sqrt{\sigma}} \end{cases} \\
 \sigma &= \frac{K_V V_p}{kT}
 \end{aligned}$$

In the above equation, M_s is the saturation magnetization of particles, which is obtained from magnetization data, V_p is the particle magnetic volume, k is Boltzmann constant, μ_0 is the vacuum magnetic permeability, T the absolute temperature, τ is the overall relaxation time (combination of Brownian and Néel relaxation times), τ_0 is the characteristic internal relaxation time of the material, which is often assumed to be $\sim 10^{-9}$ s, V_H is the hydrodynamic volume of particles, η is the viscosity of the solution,

K_V is the magnetocrystalline anisotropy constant and ρ is the particle density. While the hydrodynamic volume of the particles is determined from DLS data, the magnetic volume can only be determined by looking at magnetization curves. As already suggested in the literature, we have assumed that, in spite of the narrow size distribution of particles as seen from TEM images, the prepared SPIONs can still have a broader distribution of magnetic size, and consequently of magnetic volumes, due to defects in the crystalline structure and to the existence of magnetic dead layer. Therefore, we have assumed that it is possible to describe the magnetic size distribution by means of the lognormal function. We have further assumed that each particle volume has a magnetization behaviour that can be described by means of a Langevin function. Therefore, the magnetization curve of the particles can be described as follow:

$$M(B) = M_s \frac{\int g(r) r^3 \left(\coth(x) - \frac{1}{x} \right) dr}{\int g(r) r^3 dr}$$

$$x = \frac{4\pi r^3 M_s B}{3k_b T}$$

$$g(r) = \frac{1}{\sqrt{2\pi} r \cdot s} \exp \left(-\frac{(\log(r) - m)^2}{2s^2} \right)$$

In the above equation, B is the intensity of the applied magnetic field, $g(r)$ is the lognormal magnetic particle size distribution, with s and m being the mean and standard deviation of the natural logarithm of the size r . A fitting of the magnetization curves, allowing us to retrieve the two parameters s and m for the each SPIONs sample, is shown in the supporting information (Figure S3).

3. Results and discussion

Cubic and spherical SPIONs were synthesized in house following the well-known thermal decomposition method [11,25,26] and subsequently transferred to water by ligand exchange with CA.[28] Figure 1 shows that the particles are well dispersed after the ligand exchange and are entirely composed of magnetite/maghemite. The diameter was found to be 14.3 ± 2.0 nm for the cubes and 18.7 ± 2.4 nm for the spheres (determined using ImageJ particle sizing software, $n > 1000$ nanoparticles).

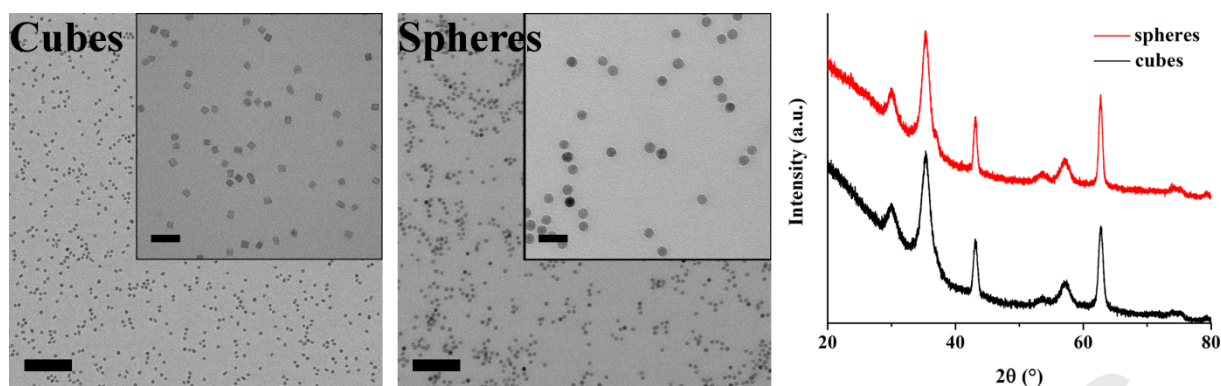


Figure 1. TEM micrographs of CA coated SPIONs and corresponding x-ray diffractograms. Scale bar = 200 nm, insets = 50 nm.

Furthermore, zeta potential and DLS measurements confirm the colloidal stability of the CA coated particles (Table 1) and hence, were the starting point for all further functionalization. For that purpose, catechol-conjugated polymers were dissolved in Milli-Q water and cubic and spherical SPIONs, respectively, were added to the polymer solutions and sonicated at room temperature overnight. To ensure comparable results, we choose the same particle concentration in all dispersions, performed all coatings at the same time using the exact same parameters and also included the CA coated SPIONs into the sonication process. All data presented in this work, including the CA coated SPIONs, represent the results obtained after overnight sonication. The success of each individual coating process is displayed in Table 1. As can be expected, the respective hydrodynamic diameters increase upon coating the particles with the 5kDa polymers, whereas the shorter 2kDa PEG resulted in hydrodynamic sizes comparable to CA SPIONs. However, the increased zeta potential proves successful coating since the strongly negative citric acid is partially replaced by the grafted (neutral) polymer. Additionally, DLS shows that no aggregation occurred upon polymer coating (Figure S1, supplementary information), which allowed us to study the effect of different coatings of single MNPs on heating.

Table 1. Zeta potential and hydrodynamic diameter determined by DLS as a function of particle type and applied coatings. Characteristic time and magnetocrystalline anisotropy constant calculated from experimental data as a function of particle type and applied coating.

Particle type	Coating	Zeta potential (mV)	$d_{H, DLS}$ (nm)	τ_0 (s)	K_V (J/m ³)
Cubes	Citric Acid	- 26.4 ± 0.6	19.2 ± 0.2	1·10 ⁻⁹	4587
	Catechol-PEG 2kDa	- 4.5 ± 1.1	19.0 ± 0.2	1·10 ⁻⁹	3595
	Catechol-PEG 5kDa	- 6.6 ± 0.6	32.4 ± 0.8	1·10 ⁻⁹	3919
	Catechol-PVA 5kDa	- 4.7 ± 0.6	27.8 ± 0.4	1·10 ⁻⁹	3527
Spheres	Citric Acid	- 27.2 ± 0.8	23.0 ± 0.2	6·10 ⁻⁹	6205
	Catechol-PEG 2kDa	- 4.4 ± 0.9	23.2 ± 0.2	6·10 ⁻⁹	5007
	Catechol-PEG 5kDa	- 5.2 ± 0.5	34.0 ± 2.4	6·10 ⁻⁹	6137
	Catechol-PVA 5kDa	- 4.3 ± 1.4	31.2 ± 0.2	6·10 ⁻⁹	5749

In a second step, we analysed the MNPs' magnetic and thermal properties (Figure 2). All particles display superparamagnetic behaviour and no hysteresis could be detected in the $M-H$ curve (Figure S2). Figure 2 shows that the normalized (by iron mass) magnetization of spherical particles is lower than the magnetization of cubic MNPs. This can be explained with shape effects, resulting in an increased magnetic core volume and a lowered surface to volume ratio for quasi-cubic particles,[33] as well as the synthesis process, which leads to paramagnetic impurities in larger particles.[34] Nonetheless, regarding the effect of coating, no significant difference in magnetization can be observed, neither between the 4 different coated cubic particles nor between the spherical particles coatings which is in good agreement with previous reported results.[35] In contrast, the SAR values for all three polymer coated spheres were found to be statistically significant lower than the SAR value of the CA spheres. This behaviour was only partially observable for the cubes, where the Catechol-PEG 5kDa show a significant decrease in SAR to CA coated cubes, while Catechol-PEG 2kDa and Catechol-PVA 5kDa are not significantly different. These small changes might be attributed to the overall lower SAR values of the cubes due to their smaller size.[12]

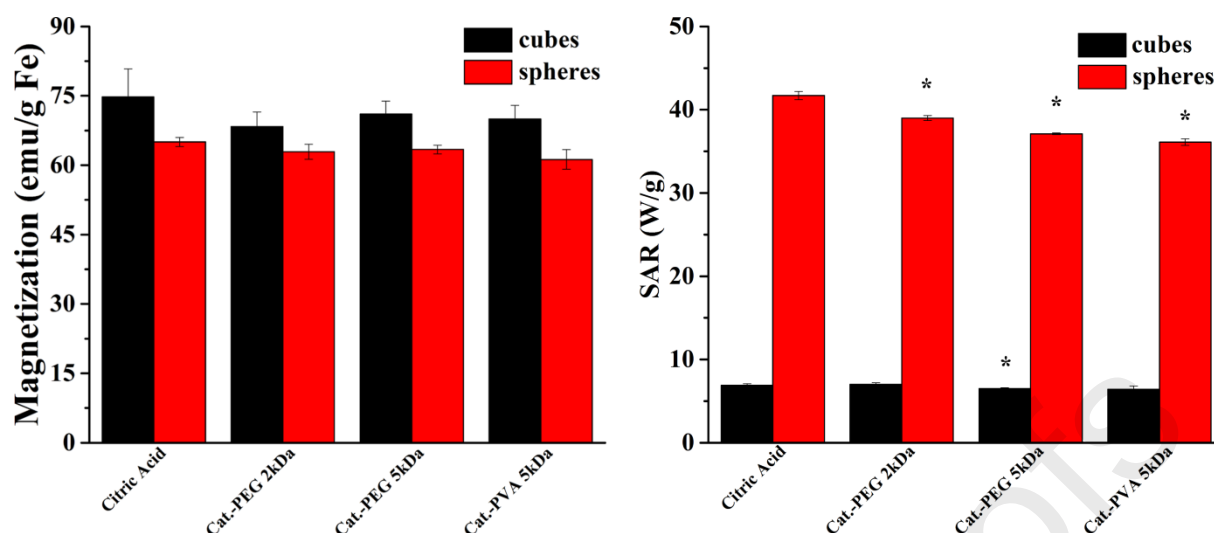


Figure 2. Saturation magnetization (left) and SAR values (right) as a function of particle morphology and applied coating. Saturation magnetization values show no significant differences upon coating for both particle types. Spheres exhibit a significant difference* in SAR values between polymer coated and CA coated SPIONs whereas cubes show only a significant difference* for Catechol-PEG 5kDa coating. Values were considered significant compared with their respective CA coated SPIONs when $p < 0.05$.

In order to obtain a better understanding of the heating behaviour of cubic and spherical MNPs, we decided to investigate the heat dissipation mechanism in more detail. We used linear response theory [14] in order to compute the heat generated by the MNPs. It is well known that the magnetic particle size distribution plays a central role in the heat generated. Therefore, we first used magnetization data to estimate the magnetic particle size distribution, which is different from the crystallite size distribution determined via TEM. The calculation has been made under the assumption that the magnetization curve of particles with a given magnetic size follows the Langevin equation. By assuming that the magnetic size distribution can be described effectively by a lognormal distribution, the two parameters of the distribution have been obtained by fitting the magnetization data. With the knowledge of the magnetic size distribution, we have performed calculation using LRT, and used the magnetocrystalline anisotropy constant and the relaxation time constant τ_0 as adjustable parameters to fit the experimental SAR data. In order to minimize the uncertainty in these parameters and precisely determine which the dominant relaxation mechanism is, we have performed for all the spherical samples SAR measurements at four different field frequency values. We have found that the small dependence of the SAR value to the change in surface coating shown in Figure 2 can only be explained by assuming that Brownian relaxation is not the dominant mechanism of heat dissipation in our system. However, Néel relaxation is the prevailing mechanism, which is supported by the low value of the magnetocrystalline anisotropy constant (Table 1) that can be used to model the experimental SAR data. The corresponding values of the relaxation time τ_0 are for all samples of spherical particles about $6 \cdot 10^{-9}$ s. With these values, the heating behaviour of all spherical SPIONs can be quantitatively accounted for, as shown in Figure 3.

In the case of cubes, a similar analysis can be carried out. Because their size regime is in the area of Néel dominated relaxation time, we determined SAR values at only one frequency to show the simplicity of this approach. It is found that using relaxation time

value of 10^{-9} s, values of the magnetocrystalline anisotropy constant are ca. 3500-4500 J/m³, depending on the type of coating, lead to results in good agreement with the experimental data.

In conclusions, the results of our calculations indicate that the low magnetocrystalline anisotropy constant values found for all nanoparticles cause Néel relaxation to be the dominant heat dissipation mechanism, which explains the small differences in SAR values observed for all particles. In addition, the small differences in SAR values appear to be related to the little differences in the magnetocrystalline anisotropy values for the particles bearing different coatings.

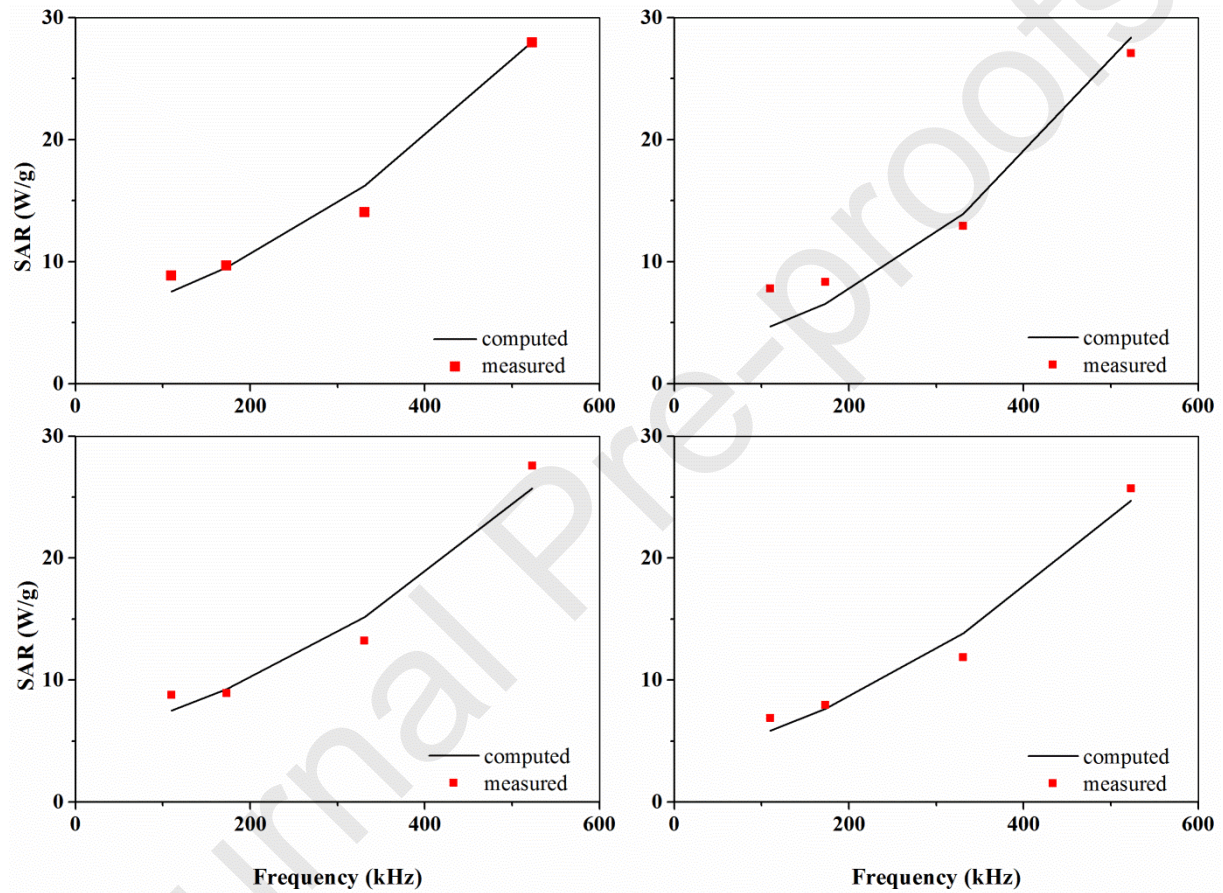


Figure 3. Comparison between experimental and computed SAR data of spheres as a function of applied coating and the magnetic field frequency.

One important aspect should be noted, however. There is a significant difference between the SAR values of particles measured a day after preparation at a frequency of 523kHz, reported in Figure 2, and the same particles measured a few weeks after preparation at various frequencies (including 523kHz), reported in Figure 3. This suggests that the SAR values decrease in time after preparation, and the calculations indicate that the magnetocrystalline anisotropy constant is lower directly after preparation ($K_v=1700-2100$ J/m³) than after a few weeks ($K_v=5000-6200$ J/m³). Such change might be attributed to aging of MNPs in the presence of oxygen, resulting in oxidation of magnetite to maghemite even inside the nanoparticle core.[36,37] The conclusion of dominant Néel relaxation mechanism applies anyway to all datasets.

4. Conclusions

We have presented the derivation of the magnetocrystalline anisotropy constant and effective relaxation mechanism of MNPs from simple, independent, and fast measurement techniques. We demonstrated the applicability of this approach using two SPIONs batches, synthesized in-house by thermal decomposition method, and applied four different surface coatings. By acquiring their $M-H$ curves, SAR values, and hydrodynamic diameters it is possible to calculate all relevant parameters needed to assess the magnetic properties which are imperative for magnetic hyperthermia. Hence, a simple and reliable quantification of the dominant relaxation time and magnetocrystalline anisotropy constant can be useful for a better understanding of the areas of application and could facilitate the development of more efficient magnetic hyperthermia treatments. Additionally, our results demonstrate the influence of surface coatings on the heating efficiency of Néel relaxation dominated MNPs. Minor changes in SAR values attributed to the different coatings can be observed, based on changed magnetocrystalline anisotropy constants.

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Highlights

- Tune efficacy of magnetic hyperthermia by efficiently assessing all relevant parameters
- DLS, VSM and lock-in thermography can be used to determine all relevant magnetic parameters
- Easy determination of magnetocrystalline anisotropy constant
- Surface coatings have minor impacts on the thermal dissipation of Néel relaxation dominated magnetic nanoparticles