



Total synthesis of sparstolonin B via a palladium-catalyzed aldehyde α -arylation



Dalton Kim ^a, Aaron Nash ^b, Jef De Brabander Jr. ^b, Uttam K. Tambar ^{b,*}

^a Department of Chemistry, Northwestern University, 2145 Sheridan Road, Evanston, IL 60208-3113, United States

^b Department of Biochemistry, The University of Texas Southwestern Medical Center, 5323 Harry Hines Boulevard, Dallas, TX 75390-9038, United States

ARTICLE INFO

Article history:

Received 3 April 2018

Accepted 3 May 2018

Available online 7 May 2018

Keywords:

Palladium catalysis

Arylation

Total synthesis

Autoredox

Cross-coupling

ABSTRACT

A concise and convergent total synthesis of sparstolonin B was developed. A palladium-catalyzed aldehyde α -arylation was utilized to construct the carbon skeleton of the natural product. A subsequent simple one-pot procedure effected global deprotection and closure of the final two rings via an unusual autoredox mechanism for the conversion of a bis-hydroquinone intermediate to the natural product. The 6 step synthetic sequence was realized in 18% overall yield.

© 2018 Elsevier Ltd. All rights reserved.

1. Introduction

Transition metal catalysis has fundamentally transformed the total synthesis of biologically active natural products.¹ Palladium-catalyzed cross-coupling chemistry is especially useful for the formation of carbon–carbon bonds in complex molecular settings.² We became interested in the synthesis of the natural product sparstolonin B because of its unique biological profile and the potential to assemble its carbon skeleton via palladium-catalyzed carbon–carbon bond forming chemistry.

Since the isolation of sparstolonin B in 2011 from the Chinese herb *Sparganium stoloniferum*,³ the polyphenolic isocoumarin has been shown to demonstrate a diverse array of biological activities.^{3,4} The natural product's earliest recognized biological activity was selective antagonism of toll-like receptors (TLRs) 2 and 4.³ Given the crucial role of TLRs in innate immunity,^{4a} antagonists of these proteins are attractive candidates for the treatment of chronic inflammatory diseases,^{4b} as well as severe sepsis.^{4c} Protective effects of sparstolonin B have already been demonstrated in septic mice.^{4d} Sparstolonin B has also been shown to attenuate early liver inflammation in a mouse model of non-alcoholic steatohepatitis.^{4e} More recently, sparstolonin B was shown to inhibit HIV-1 transcription by a novel mechanism^{4f} and induce apoptosis in

neuroblastoma cells.^{4g} These diverse and valuable biological activities have rendered sparstolonin B an attractive synthetic target.^{5,6}

The first total synthesis of sparstolonin B was reported by Wang and co-workers in 2015.⁵ Tang et al. reported an alternative synthesis in 2017 featuring milder reaction conditions.⁶ We were interested in devising a concise and convergent total synthesis of sparstolonin B that would facilitate the preparation of analogs for medicinal chemistry studies to optimize the various biological activities of the natural product. We envisioned the use of a palladium-catalyzed carbon–carbon bond forming reaction to facilitate the efficient assembly of the core structure of the natural product.

Herein, we describe an efficient and convergent total synthesis of sparstolonin B that features a palladium-catalyzed aldehyde α -arylation to construct the carbon skeleton of the natural product. A one-pot procedure was devised to effect global deprotection and dual cyclization to yield sparstolonin B. Total synthesis of the natural product was achieved in 18% overall yield in 6 steps from inexpensive and commercially available 2,5-dimethoxybenzoic acid.

2. Results

Our retrosynthetic analysis of sparstolonin B (**1**) commenced with a disconnection of the diaryl ether to furnish hydroquinone **2**, which could readily be prepared by global demethylation of **3**

* Corresponding author.

E-mail address: uttam.tambar@utsouthwestern.edu (U.K. Tambar).

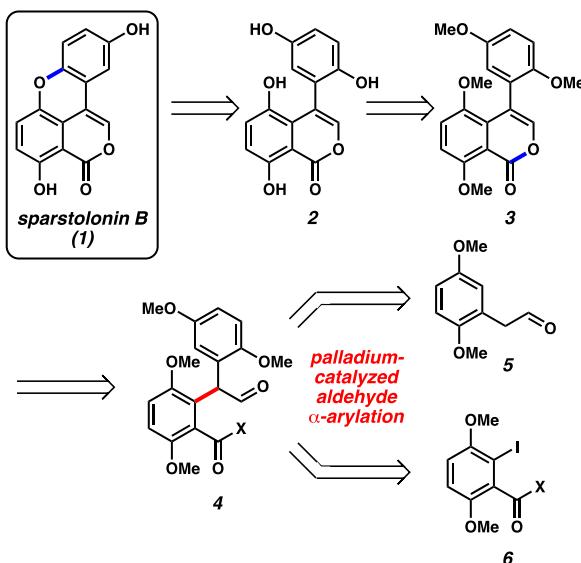


Fig. 1. Retrosynthetic analysis of sparstolonin B (1).

(Fig. 1). We hypothesized that 4-arylisocoumarin **3** could be derived from the key α -arylation of aldehyde **5** with aryl iodide **6**. Fragments **5** and **6** could in turn be accessed from commercially available starting materials.

The synthesis of aldehyde **5** is depicted in Fig. 2. Commercial (2,5-dimethoxy)phenylacetic acid (**7**) was quantitatively reduced to aryl alcohol **8** by an excess of lithium aluminum hydride in THF at 0 °C. Treatment of the resultant alcohol with Dess-Martin periodinane⁷ in CH₂Cl₂ yielded the desired aldehyde **5** in 83% yield.

The proposed palladium-catalyzed α -arylation of aldehyde **5** with aryl iodide **6** is the key transformation in our synthesis of sparstolonin B, effectively assembling the carbon backbone of the natural product. Traditionally, the α -arylation of aldehydes has been a challenging transformation due to competing aldol reactions under basic conditions. An efficient protocol for the palladium-catalyzed α -arylation of aldehydes was reported by Terao in 2002 (Fig. 3).⁸ Various aldehydes **9** were coupled with bromobenzene **10** to furnish α -arylated aldehydes **11**. Soon thereafter, a similar procedure was developed by Lessi for the direct synthesis of 4-arylisocoumarins **16** from 2-arylacetraldehydes such as **12** and methyl 2-halobenzoates such as **13**.⁹ This transformation presumably proceeds through a tandem α -arylation/enol lactonization (**12 + 13 → 14 → 15 → 16**).

Before attempting the palladium-catalyzed α -arylation of aldehyde **5** with aryl iodide **6**, we devised model substrate 2-halobenzoate **17** to test the viability of the proposed key reaction

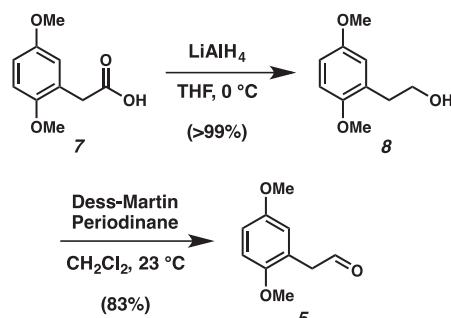


Fig. 2. Syntheses of aldehyde **5**.

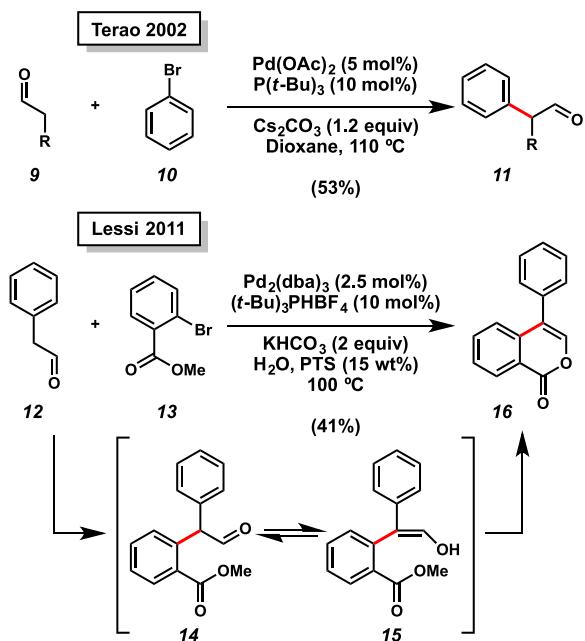


Fig. 3. Previous protocols for α -arylation of aldehydes.

(Table 1). We examined a series of reaction conditions, with various solvents, bases, palladium sources, and ligands. To our delight, we obtained the desired 4-arylisocoumarin **20** in 65% yield in the presence of Cs₂CO₃ as the base, Pd(OAc)₂ as the palladium source, t-Bu₃PHBF₄ as the ligand precursor, and dioxane as the reaction medium. This reaction proceeds through initial formation of α -arylation product **18**, followed by enolization to ester **19** and subsequent cyclization to 4-arylisocoumarin **20**.

With the generation of model 4-arylisocoumarin **20** in hand, we pursued the synthesis of 2-iodobenzoate **6** for the total synthesis of sparstolonin B (Fig. 4). Commercially available 2,5-dimethoxybenzoic acid (**21**) was converted to benzamide **22** to facilitate installation of the iodide by directed *ortho* metalation.¹⁰ Refluxing benzoic acid **21** in thionyl chloride yielded the corresponding acid chloride, which upon exposure to diethylamine rendered benzamide **22** in 88% overall yield. Directed *ortho* lithiation followed by exposure to 1,2-diidoethane yielded iodo-benzamide **6** in 70% yield.

Next, we attempted the key palladium-catalyzed aldehyde α -arylation and subsequent cyclization. Gratifyingly, the coupling of aldehyde **5** and 2-iodobenzamide **6** yielded an equilibrating mixture of α -arylated aldehyde **23** and enol **24**. While these intermediates did not undergo spontaneous cyclization under the palladium-catalyzed conditions, treatment with refluxing glacial acetic acid generated the desired 4-arylisocoumarin **3** in 78% yield.

We attempted to perform a global removal of the four methyl protecting groups of tricycle **3** with refluxing HBr and glacial acetic acid under an atmosphere of oxygen. Surprisingly, these acidic conditions resulted in the formation of the desired natural product sparstolonin B (**1**).

To account for the conversion of 4-arylisocoumarin **3** to sparstolonin B (**1**), we propose the following mechanism (Fig. 5). Initial global deprotection of the four methoxy groups results in the formation of bis-hydroquinone **2**. Mono-oxidation in the presence of oxygen yields the mixed hydroquinone-quinone intermediate **25**. Nucleophilic attack of the hydroquinone to the electrophilic quinone forms tetracycle **26**, and dehydration furnishes oxocarbenium ion **27**. Reduction of intermediate **27** with concomitant

Table 1

Optimization of tandem palladium-catalyzed aldehyde α -arylation/coumarin formation. PTS = Polyoxyethylene- α -Tocopheryl Sebacate.

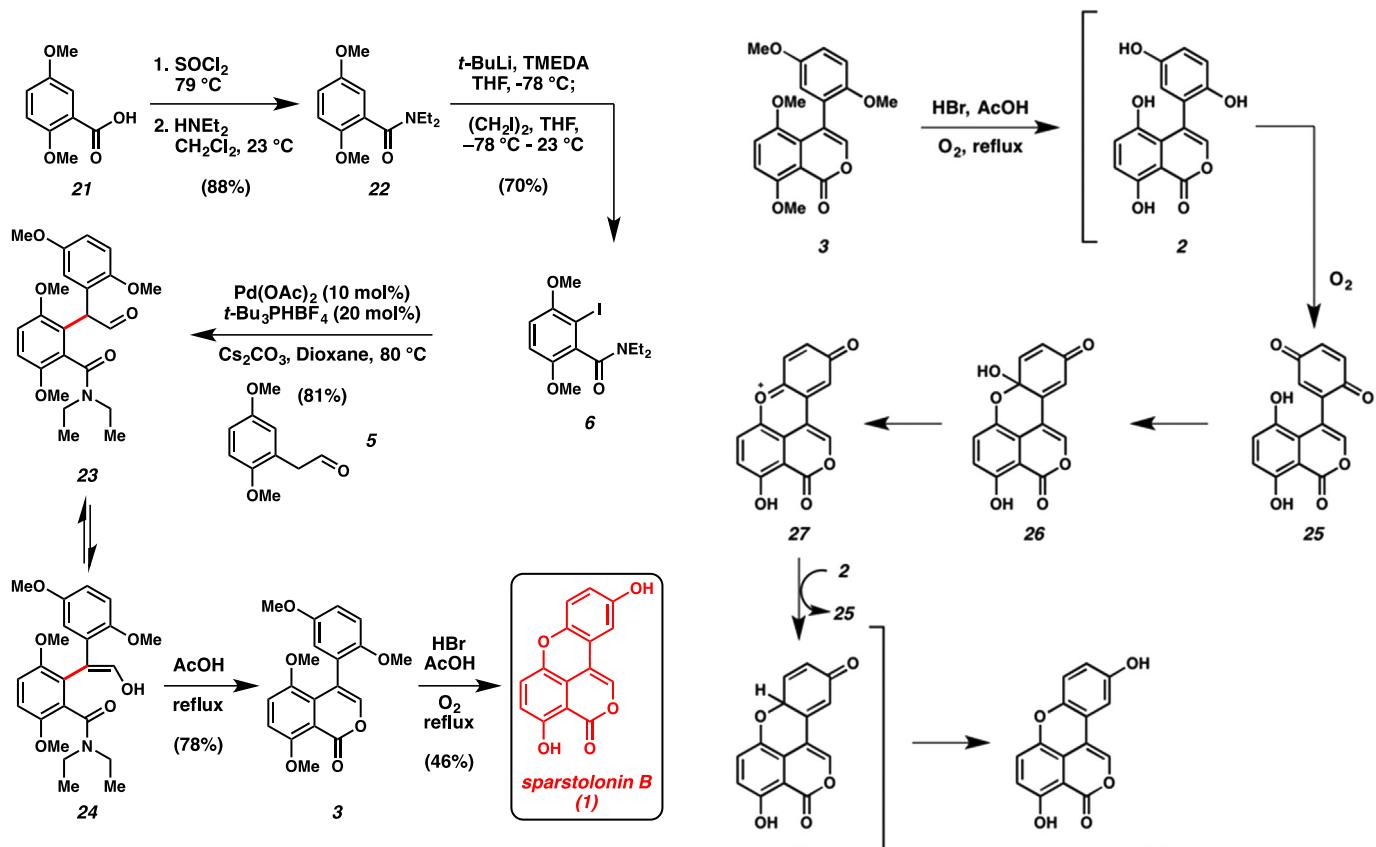
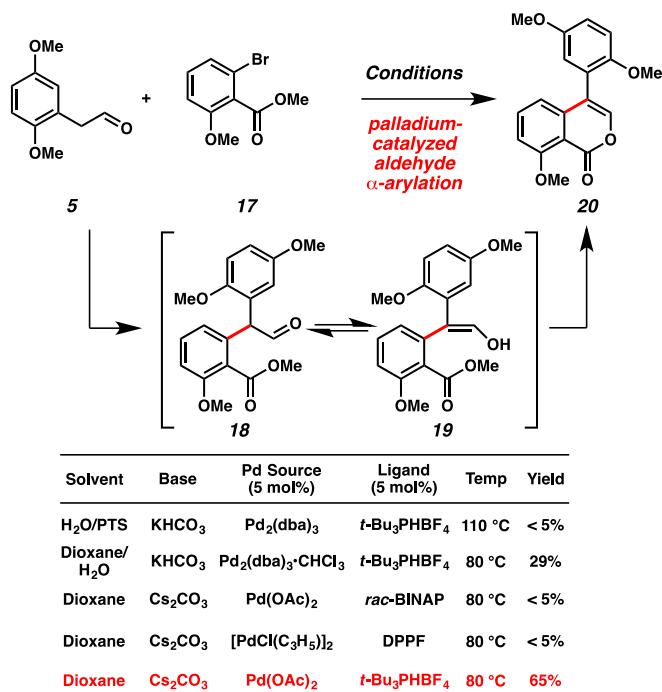


Fig. 5. Proposed mechanism for the conversion of tricycle **3** to sparstololin B (**1**).

oxidation of another equivalent of **2** to **25** generates tetracycle **28** in an autoredox process.¹¹ Subsequent tautomerization of intermediate **28** leads to the formation of sparstolonin B (**1**).

In conclusion, we have developed a 6 step synthesis of sparstolonin B from commercially available 2,5-dimethoxybenzoic acid (**21**) in 18% overall yield. Key steps in our approach include a palladium-catalyzed aldehyde α -arylation to assemble the core structure **3** and an unusual autoredox mechanism for the conversion of bis-hydroquinone **2** to the natural product. The application of this flexible synthetic strategy to the synthesis of unnatural analogs of sparstolonin B and the exploration of biological activity for these structures are on-going interests in our group.

Acknowledgments

Financial support was provided by W. W. Caruth, Jr. Endowed Scholarship, Welch Foundation (I-1748), National Institutes of Health (R01GM102604), National Science Foundation (1150875), Sloan Research Fellowship, and UTSW Quantitative and Physical Sciences Summer Undergraduate Research Fellowship (QP-SURF).

Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.tet.2018.05.012>.

References

- (a) Hudlicky T, Kutchan TM, Naqvi SM. *Org React*. 1985;33:247–335;
 (b) Brunner H. *Pure Appl Chem*. 1994;66:2033–2036;
 (c) Trost BM. *Pure Appl Chem*. 1994;66:2007–2014;
 (d) Fuerstner A. *Synlett*. 1999;1523–1533;
 (e) Knoelker H-J. *Curr Org Synth*. 2004;1:309–331;
 (f) Ulaczyk-Lesanko A, Hall DG. *Curr Opin Chem Biol*. 2005;9:266–276;
 (g) Bouyssi D, Monteiro N, Balme G. *Curr Org Chem*. 2008;12:1570–1587;
- (h) Toure BB, Hall DG. *Chem Rev*. 2009;109:4439–4486;
 (i) Denmark SE, Liu JHC. *Angew Chem Int Ed*. 2010;49:2978–2986;
 (j) Cramer N. *Chimia*. 2011;65:656–658;
 (k) Trost BM, Crawley ML. *Top Organomet Chem*. 2011;38:321–340;
 (l) Chen DYK, Youn SW. *Chem Eur J*. 2012;18:9452–9474;
 (m) Hamada Y. *Chem Pharm Bull*. 2012;60:1–20;
 (n) Anjum S, Soriano E, Marco-Contelles JL. *Stud Nat Prod Chem*. 2013;40:51–69;
 (o) Scarso A, Strukul G. In: *Transition-metal-catalyzed Stereoselective Oxidations in Drug and Natural Product Synthesis*. vol. 2. John Wiley & Sons, Inc.; 2013: 1043–1069;
 (p) Wender PA. *Tetrahedron*. 2013;69:7529–7550;
 (q) Duefert S-C, Hierold J, Tietze LF. In: *Domino Reactions in the Total Synthesis of Natural Products*. Wiley-VCH Verlag GmbH & Co. KGaA; 2014:523–578;
 (r) Xu P-F, Wei H. In: *Use of Transition Metal-catalyzed cascade Reactions in Natural Product Synthesis and Drug Discovery*. John Wiley & Sons, Inc.; 2014: 283–331;
 (s) Hall DG, Rybak T, Verdelet T. *Acc Chem Res*. 2016;49:2489–2500;
 (t) Zweig JE, Kim DE, Newhouse TR. *Chem Rev*. 2017;117:11680–11752.
- (a) Nicolaou KC, Bulger PG, Sarlah D. *Angew Chem Int Ed*. 2005;44:4442–4489;
 (b) Heravi MM, Hashemi E, Ghobadi N. *Curr Org Chem*. 2013;17:2192–2224;
 (c) Bai Y, Davis DC, Dai M. *J Org Chem*. 2017;82:2319–2328.
- Liang Q, Wu Q, Jiang J, et al. *J Biol Chem*. 2011;286:26470–26479.
- (a) Liang Q, Yu F, Cui X, et al. *Arch Pharm Res (Seoul)*. 2013;36:890–896;
 (b) Bateman HR, Liang Q, Fan D, Rodriguez V, Lessner SM. *PLoS One*. 2013;8, e70500;
 (c) Deng X, Zhang Y, Jiang F, et al. *Virol J*. 2015;12:108;
 (d) Wang M, Xiu L, Diao J, Wei L, Sun J. *Eur J Pharmacol*. 2015;769:79–85;
 (e) Liang Q, Dong S, Lei L, et al. *Cytokine*. 2015;75:302–329;
 (f) Dattaroy D, Seth RK, Das S, et al. *Am J Physiol Gastrointest Liver Physiol*. 2016;310:G510–G525;
 (g) Kumar A, Fan D, Dipette DJ, Singh US. *PLoS One*. 2014;9, e96343.
- Wang Y, Wang C, Wang Y, Dong L, Sun J. *RSC Adv*. 2015;5:12354–12357.
- Yu H, Tang X, Tong L, Yao M, Liang Q, Wang X. *Synlett*. 2017;28:1187–1190.
- Dess DB, Martin JC. *J Org Chem*. 1983;48:4155–4156.
- Terao Y, Fukuoka Y, Satoh T, Miura M, Nomura M. *Tetrahedron Lett*. 2002;43: 101–104.
- Lessi M, Masini T, Nucara L, Bellina F, Rossi R. *Adv Synth Catal*. 2011;353: 501–507.
- Snieckus V. *Chem Rev*. 1990;90:879–933.
- For a similar autoredox process between catechol and iminoquinone, see: Hosokawa S, Matsushita K, Tokimatsu S, Toriumi T, Suzuki Y, Tatsuta K *Tetrahedron Lett*. 2010;51:5532–5536.