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Repurposing human PDE4 inhibitors for neglected tropical diseases: Design, synthesis and evaluation of cilomilast analogues as *Trypanosoma brucei* PDEB1 inhibitors



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ABSTRACT

A medicinal chemistry exploration of the human phosphodiesterase 4 (hPDE4) inhibitor cilomilast (**1**) was undertaken in order to identify inhibitors of phosphodiesterase B1 of *Trypanosoma brucei* (TbrPDEB1). *T. brucei* is the parasite which causes African sleeping sickness, a neglected tropical disease that affects thousands each year, and TbrPDEB1 has been shown to be an essential target of therapeutic relevance. Noting that **1** is a weak inhibitor of TbrPDEB1, we report the design and synthesis of analogs of this compound, culminating in **12b**, a sub-micromolar inhibitor of TbrPDEB1 that shows modest inhibition of *T. brucei* proliferation.

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Neglected tropical diseases (NTDs) are a group of pathologies that mainly affect the poor in the developing world. The World Health Organization has defined 17 different NTDs, and this list includes human African trypanosomiasis (HAT), or sleeping sickness, which is a parasitic disease caused by the protozoan parasite *Trypanosoma brucei*. Currently, 50 million people are at risk within rural sub-Saharan Africa and the estimated incidence is around 10,000 cases a year.¹ Melarsoprol and eflornithine, the current front-line therapies for advanced HAT, have been the drugs of choice for the treatment of second stage HAT. Though effective, these drugs are limited by serious and sometimes lethal side effects, inconvenient route of administration and increasing incidence of drug resistance. Though new efforts have produced new combination therapies (such as nifurtimox/eflornithine combination therapy (NECT)),² further work is needed to develop the next generation of therapies that are safe, effective, and inexpensive. With this in mind, our research efforts are focused on repurposing established classes of agents that inhibit druggable human targets.³

Phosphodiesterases (PDEs) are one such target; a family of enzymes that inactivate the cellular signaling agents cAMP or cGMP. In recent years it was shown that cAMP plays a key role in *T. brucei* biology. *T. brucei* expresses five cAMP-specific PDEs,

two of which (TbrPDEB1 and TbrPDEB2) have been shown to be together essential for parasite proliferation by both RNAi and small molecules.^{4,5} Initial screening and early optimization experiments have uncovered a susceptibility of TbrPDEB1 and B2 to human PDE4 (hPDE4) inhibitor chemotypes, and that the inhibition of these enzymes leads to parasite death in vitro. For example, we showed that piclamilast (**2**, Fig. 1), a hPDE4 inhibitor, inhibited TbrPDEB1 and B2 (IC₅₀ 4.7 and 11.4 μM, respectively), which translated to inhibition of *T. brucei* cell growth (EC₅₀ = 9.6 μM).⁵ At the same time, others identified **3** as a potent TbrPDE inhibitor via a high-throughput screening campaign.⁶ This racemic compound remains the most potent TbrPDE inhibitor described to date, despite several further reports.^{7–9} A key limitation of all the TbrPDEB inhibitors identified to date is the lack of selectivity over hPDE4, which is likely to lead to various characteristic PDE4 side effects, such as nausea and emesis.

We were surprised at the divergence in TbrPDEB1 activity between closely related hPDE4 inhibitors: roflumilast (**4**) a close analog of **2**, was completely inactive.⁵ Thus, for the purpose of studying a wider variety of hPDE4 inhibitors as starting points for TbrPDEB inhibitors, we investigated cilomilast, **1**, a related hPDE4 inhibitor. Compound **1** (Ariflo, SB-207,499) is an orally active and selective hPDE4 inhibitor developed by GlaxoSmithKline for the treatment of respiratory disorders such as chronic obstructive pulmonary disease (COPD).^{10,11} This compound has a

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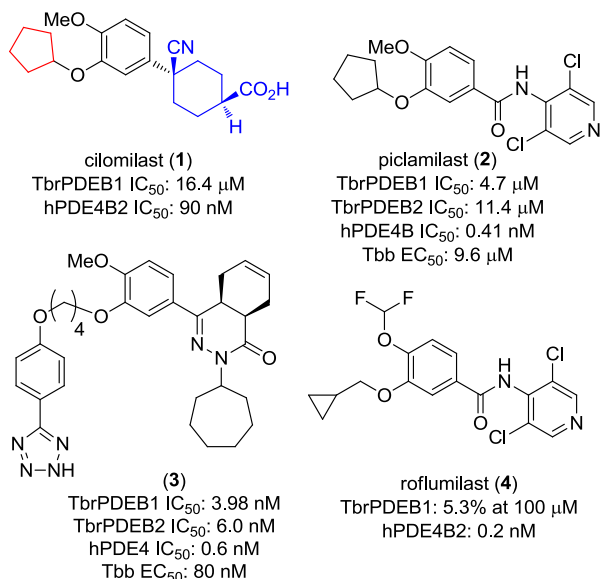


Figure 1. The structure of **1**, highlighting the tail (red) and head region (blue) explored in this work. Also shown are other hPDE4 inhibitors previously studied as inhibitors of TbrPDEB1.^{5,6} Human PDE4 data shown is from previous Letters.^{13,14}

reported IC₅₀ of 84 nM against hPDE4B,¹² and we observed an IC₅₀ against TbrPDEB1 of 16.4 μM. Given the prior art of repurposing hPDE4 inhibitors for TbrPDEB1, we felt that this result warranted additional medicinal chemistry explorations for trypanosomal PDE inhibitors.

Our investigation into the SAR of **1** as a TbrPDEB1 inhibitor involved first assessing the relative stereochemistry of the head-group (Fig. 1, blue), as well as the importance of the carboxylate functionality. We also wished to determine whether a stereochemically simplified headgroup replacement could be achieved. Secondly, a key structural feature of the TbrPDEB1 binding site, predicted by homology modeling and confirmed by crystallography,^{5,15} is a pocket adjacent to the binding site (termed the 'parasite-' or 'P-pocket') that is deeper in comparison to the same region in hPDE4. Thus, guided by the SAR studies of the catechol diethers **2** and **3** reported previously that were intended to explore the 'parasite pocket' of the enzyme, we now report exploration of the cyclopentyl ether (Fig. 1, red) to longer, chain-extended versions, and we then studied the protein–ligand interactions in silico by carrying out molecular docking using the recently published crystal structure of TbrPDEB1.¹⁵

Initial analogues **5**–**8** (where R¹ = cyclopentyl) were prepared by the procedure shown in Scheme 1, based on the previously published preparation for **1**.¹¹ Analogs where R¹ = benzyl were synthesized using an analogous route (see Supporting information). In the interest of exploring simplified headgroup replacements, piperidine analogues were also synthesized (Scheme S2, Supporting information).

We opted to first test compounds at 10 μM concentrations; those that were above 50% inhibition at this concentration were subjected to dose–response experiments. We have previously noted similarity between compound activity against TbrPDEB1 and B2. Thus, for efficiency, we focused our first round of compound assays on TbrPDEB1, and assumed similar (within 2–3 fold activity) against TbrPDEB2.

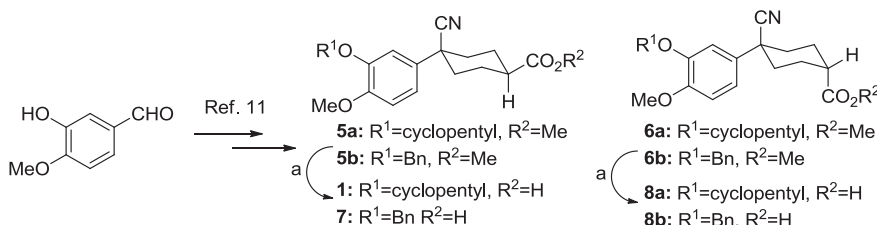
While compound **1** is a 16.4 μM inhibitor of TbrPDEB1, the esters **5a** and **5b** were below the minimum percent inhibition cut-off to obtain an IC₅₀ (Table 1). This is consistent with the SAR for hPDE4 previously reported.¹¹ The benzyl analog of cilomilast (compound **7**) inhibits TbrPDEB1 with activity similar to **1**, though it retains some potency against hPDE4 (IC₅₀ = 0.54 μM). Notably, the compounds with a *cis*-orientation of the cyano and carboxylate groups are consistently more potent (compare **1** and **7** with **8a** and **8b**). Replacement of the cyclohexane headgroup with a piperidine ring resulted in significant loss of activity (Table S3, Supporting information). Others have shown that preparation of analogs with catechol substituents that project into the P-pocket (cf. **3**) can lead to improved potency and selectivity.^{6,7} We assumed that our catechol-derived analogs bind in a similar orientation as **2** and **3**, and used this hypothesis to design analogs of **7**. Thus, we replaced the phenyl group with three pyridine isomers (**10a**–**c**). As shown in Scheme 2, ester **5b** was debenzylated using catalytic hydrogenation to provide phenol **9**. This could be alkylated with the appropriate bromomethylpyridine, providing **10a**–**c** following basic hydrolysis. These compounds were much less active (Table 2), and one (**10b**) was not soluble enough in the assay conditions to obtain inhibition data.

We then prepared chain extended analogs of **7** by alkylation of **9** using various dibromoalkanes to provide the bromides **11**, which could be utilized to alkylate a range of phenols. This provided analogs **12** after hydrolysis. (The synthesis and characterization of **13**, the ester precursor to **12b** is included in the Supporting information.)

The direct analog of **3** (**12a**) showed an approximate two-fold improvement over **1** (IC₅₀ = 7.9 μM Table 2). The unadorned phenyl analog (**12b**), with the same spacer length as **12a**, was more than 17-fold more potent than **1**. Alteration of the chain length (**12c** and **12d**) led to loss in potency, leading to compounds approximately equipotent to **1**. Compound **12b** (IC₅₀ of 0.95 μM) is the most active compound in the series to date, being 17 times more active than **1** against TbrPDEB1, though it is 25-fold more potent against hPDE4 (IC₅₀ = 0.038 μM). This represents a selectivity improvement over **1** (which is 182-fold more potent against hPDE4 than TbrPDEB1).

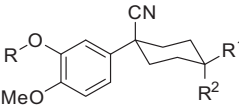
Exploration of *para*-substitution of the phenyl group with a range of electron donating, withdrawing, and neutral substituents resulted in loss of potency (**12e**–**h**).

To determine whether compounds **1** and **12b** had effects in *T. brucei* growth, we tested these for dose–response using an Alamar blue cell viability assay,¹⁶ and found that, while neither compound inhibited growth of mammalian cells (NIH 3T3, TC₅₀ >50 μM), we observed weak activity for **12b** (EC₅₀ = 26 μM) in *T. brucei* cellular



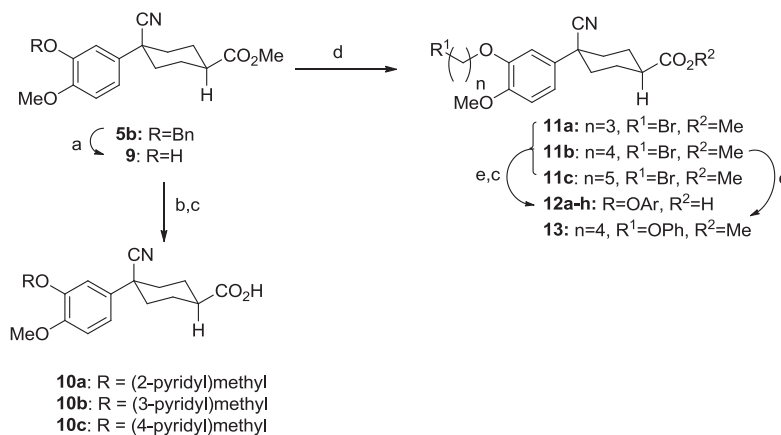
Scheme 1. Reagents and conditions: (a) LiOH, H₂O, MeOH, THF, rt, 2 h.

Table 1
Cilomilast analogs tested against TbrPDEB1

					
Compd	R	R ¹	R ²	TbrPDEB1 ^a (%)	TbrPDEB1 IC ₅₀ ^b (μM)
1		COOH	H		16 ± 6.65
5a		COOCH ₃	H	31 ± 8	
5b		COOCH ₃	H	40 ± 14	
8a		H	COOH	9.3 ± 9.3	
8b		H	COOH	20 ± 4.3	
6a		H	COOCH ₃	26 ± 3.6	
6b		H	COOCH ₃	28 ± 8.9	
7		COOH	H	77 ± 6.4	12 ± 1.18

^a Inhibitor concentration 10.0 μM, tested in at least two independent replicate experiments all error was within 15%.

^b Tested in at least two independent replicate experiments.



Scheme 2. Reagents and conditions: (a) H₂, Pd/C, MeOH, rt, 3 h; (b) appropriate (bromomethyl)pyridinium bromide, K₂CO₃, DMF, 60 °C, 4 h; (c) LiOH, H₂O, MeOH, THF, rt, 2 h; (d) appropriate dibromoalkane, DMF, 60 °C, 4 h; (e) appropriate phenol, K₂CO₃, DMF, 60 °C, 4 h.

assays. Compound **1** showed no effect upon *T. brucei* growth (EC₅₀ > 50 μM). This is surprising, given the close concordance we and others observed between TbrPDEB1 enzyme inhibition and cellular growth inhibition with compounds **2** and **3**.^{5,6}

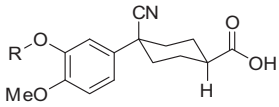
We considered two possible explanations for the lack of cellular activity. First, the essentiality of both TbrPDEB1 and B2 were demonstrated, as simultaneous knockdown by RNAi was required in order to impact on cellular proliferation.⁴ We have previously shown that inhibitors have a similar inhibitory profile when tested against both TbrPDEB1 and B2,⁵ which is perhaps not unexpected given the high sequence similarity in the binding site, and, based on this, we have been screening only against TbrPDEB1. In order to rule out the possible situation where **12b** inhibits only TbrPDEB1,

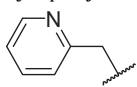
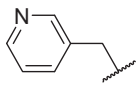
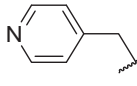
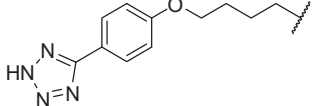
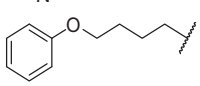
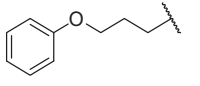
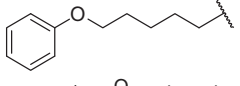
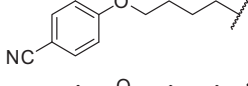
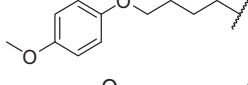
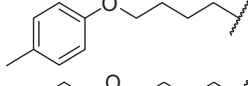
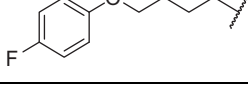
allowing a compensatory effect from TbrPDEB2, we tested **12b** against TbrPDEB2, and observed an IC₅₀ of 2.1 μM, which is approximately equivalent to the activity against TbrPDEB1.

A second possible explanation for the lack of cellular activity of **1** and **12b**, despite the structural similarity between **1–3** and **12b**, could be limited trypanosome cell permeability caused by the negatively charged carboxylate anion.¹⁷ We anticipated that compound **13**, the ester precursor of **12b** would likely display improved cellular penetration and act as a prodrug in trypanosome cells. Though **11d** is inactive (>100 μM) against TbrPDEB1 and B2, we observed activity in *T. brucei* cell cultures (EC₅₀ = 9.8 μM), which is suggestive that the uncharged ester can permeate the parasite cells and be hydrolyzed to the active species (**12b**).

Table 2

Cilomilast analogs tested against TbrPDEB1 varying catechol substituents



Compd	R	TbrPDEB1 ^a (%)	TbrPDEB1 IC ₅₀ ^c (μM)
1	Cyclopentyl		16.4 ± 6.7
10a		33 ± 12	
10b		nd ^b	
10c		11 ± 4.4	
12a		69 ± 1.4	7.9 ± 4.63
12b		96 ± 0.66 ^c	0.95 ± 0.14
12c		66 ± 2.2	11.0 ± 0.13
12d		54 ± 8	9.8 ± 1.27
12e		87 ± 12	3.6 ± 0.3
12f		79 ± 3.4	3.5 ± 0.41
12g		74 ± 1.7	4.9 ± 2.81
12h		79 ± 0.42 ^c	9.2 ± 5.1

^a Inhibitor concentration 10.0 μM, tested in at least two independent replicate experiments, all error was within 15%.^b nd = no data due to insufficient solubility in the assay conditions.^c Tested in at least two independent replicate experiments.

We looked towards molecular modeling to help explain our observations. Two crystal structures have been published for **1**: (i) a complex with hPDE4B (1XLX) and (ii) a complex with hPDE4D (1XOM).¹⁸ The latter crystal structure, with a higher resolution, displays two possible binding modes for the bound ligand where the cyclohexyl showed two alternative conformations (Supporting information, Fig. S1). The structural differences between the human and the parasite phosphodiesterase are important for repositioning the human enzyme inhibitor to a trypanosomal enzyme inhibitor. For this reason we compare the enzyme active sites in the TbrPDEB1 and hPDE4D crystal structures in the top panel of Figure 2.¹⁵

The predicted conformation of **3** bound to TbrPDEB1, overlaid with **1** as observed in the crystal structure of a hPDE4D (1XOM) is shown in Supporting information, Figure S2. This overlay reveals similar interactions for the methoxy group of these two ligands with the glutamine residue in the back pocket (Gln874 in TbrPDEB1) and rationalized our efforts to explore the P-pocket of

the enzyme by replacing the cyclopentyl of **1** with different groups able to fill the P-pocket.

Docking (POSIT v.1.0.2., OpenEye Scientific Software, Santa Fe, NM) was used to rationalize the results explained above. Firstly, we observe that the compounds with a *cis*-orientation of the cyano and carboxylate functionality are more potent than the *trans*-isomers. This can be explained by the observation that, in the *cis*-isomer, the cyano group is predicted to occupy an area at the entrance of the active site and the carboxylate group is in close contact with the magnesium ion (Fig. 2A). The carboxylate group is in fact predicted to interact with this metal, explaining the loss of potency resulting from the removal or substitution of this group with an ester, leading to loss of a predicted electrostatic interaction. Following docking, the distal phenyl group of compound **12b** is optimally placed in the P-pocket of the enzyme, as shown in Figure 2B. The phenyl ring packs against two methionine residues and a tyrosine residue, and, in addition, there is possibly a weak hydrogen bond between the C-alpha hydrogen of Gly873 and the oxygen atom of

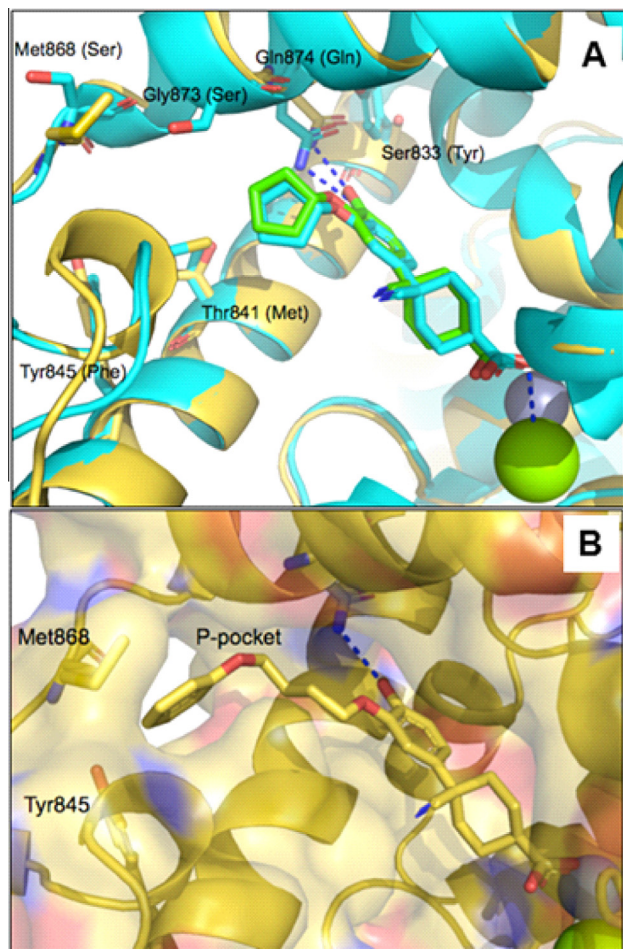


Figure 2. (A) The docked conformation of **1** (green carbon atoms) in TbrPDEB1 (4I15, gold carbon atoms) overlaid with one of the two bound conformations of the same ligand reported in the crystal structure of the human PDE4D-1 complex (1XOM, cyan). Amino acid side chain differences are shown as sticks. TbrPDEB1 residue numbering is used, with the corresponding hPDE4D residues in brackets. The interactions of the ligand with Gln 874 and the Mg^{2+} ion are displayed in blue dashed lines. (B) Docked conformation of **12b** in TbrPDEB1. The surface of the active site and the P-pocket are shown. Mg^{2+} and Zn^{2+} are shown as green and grey spheres respectively. Two residues defining the P-pocket (Met868 and Tyr845) are also shown in sticks. In both figures, Mg^{2+} and Zn^{2+} ions in the active site are shown as green and grey spheres, respectively. Images were generated using PyMol (version 1.5.0.4, Schrödinger).

the phenoxy group of this ligand (Fig. S3, Supporting information).¹⁹ Visual inspection of the poses obtained by molecular docking confirms that the ideal linker connecting the two phenyl groups of the ligand contains four methylene groups, explaining the ~10-fold loss in potency observed for **12c** and **12d**.

Substitutions in the para position of the distal phenyl group of **12b** do not increase potency (Table 2). Modeling suggests that a small polar substituent in para position could have been tolerated by interacting with the hydroxyl group of Tyr845, and there was a strong rationale for introducing the tetrazole-containing substituent (**12a**) given the good potency of compound **3**. However the, inconsistencies in the SAR between these two series may be rationalized by slight differences in binding conformation, leading to differences in positioning of the P-pocket substituents (see e.g., Fig. S2).

Finally, the 1000-fold higher activity of **3** over the matched analog **12a** is clearly due to the bicyclic headgroup; the cycloheptyl N-substituent in this compound is predicted to fill the pocket in the entrance of the active site much better than the cyclohexyl

system of cilomilast derivatives (Fig. S2, Supporting information). Further, the substituted bicyclic ring system is predicted to give increased lipophilic interactions with apolar side chains (such as a methionine and two phenylalanine residues). In our docked pose, the carbonyl group of the ligand is predicted to interact with the magnesium ion.

In summary, we have identified a hPDE4 inhibitor chemotype that can be repurposed to discover new inhibitors of TbrPDEB1, one of which shows sub-micromolar activity toward TbrPDEB1 and a modest cellular growth inhibition of *T. brucei*. As with other repurposed hPDE4 chemotypes, however, we are faced with a formidable barrier of selectivity over hPDE4 which we have yet to surmount. On the other hand, nonetheless, we have shown an ability to improve compound potency within this chemotype, and this knowledge will be useful in informing other PDE inhibitor repurposing programs that are continuing.

Acknowledgments

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Supplementary data

Compound syntheses and a tabulation of the compounds with their Northeastern registry numbers and screening data is available in Supporting information. The screening data has been made freely available as a shared data set at www.collaborativedrugdiscovery.com. Also included are biological assay and molecular modeling details. This Supporting information is available free of charge via the Internet at <http://pubs.acs.org>.

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.bmcl.2014.07.063>.

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