

Revealing chlorinated ethene transformation hotspots in a nitrate-impacted hyporheic zone



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ABSTRACT

Hyporheic zones are increasingly thought of as natural bioreactors, capable of transforming and attenuating groundwater pollutants present in diffuse baseflow. An underappreciated scenario in the understanding of contaminant fate in hyporheic zones is the interaction between point-source trichloroethene (TCE) plumes and ubiquitous, non-point source pollutants such as nitrate. This study aims to conceptualise critical biogeochemical gradients in the hyporheic zone which govern the export potential of these redox-sensitive pollutants from carbon-poor, oxic aquifers. Within the TCE plume discharge zone, discrete vertical profiling of the upper 100 cm of sediment pore water chemistry revealed an 80% increase in dissolved organic carbon (DOC) concentrations and 20–60 cm thick hypoxic zones ($<2 \text{ mg O}_2 \text{ L}^{-1}$) within which most reactive transport was observed. A 33% reduction of nitrate concentrations coincided with elevated pore water nitrous oxide concentrations as well as the appearance of manganese and the TCE metabolite *cis*-1,2-dichloroethene (cDCE). Elevated groundwater nitrate concentrations ($>50 \text{ mg L}^{-1}$) create a large stoichiometric demand for bioavailable DOC in discharging groundwater. With the benefit of a high-resolution grid of pore water samplers investigating the shallowest 30 cm of hypoxic groundwater flow paths, we identified DOC-rich hotspots associated with submerged vegetation (*Ranunculus* spp.), where low-energy metabolic processes such as mineral dissolution/reduction, methanogenesis and ammonification dominate. Using a chlorine index metric, we show that enhanced TCE to cDCE transformation takes place within these biogeochemical hotspots, highlighting their relevance for natural plume attenuation.

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1. Introduction

Chlorinated ethenes (CEs) such as trichloroethene (TCE) and its metabolites *cis*-1,2-dichloroethene (cDCE) and vinyl chloride (VC) are among the most common volatile organic contaminants (VOCs) detected in groundwater (Shapiro et al., 2004; Rivett et al., 2012). These organohalides are prominent chemical stressors in surface water ecosystems impacted by legacy industrial sources (Roy et al., 2018; Sonne et al., 2018). Dissolved-phase CE plumes often

continue to migrate in aquifers with limited sorption capacity and frequently discharge with groundwater baseflow to streams and rivers (Weatherill et al., 2018). Back-diffusion from low permeability media to groundwater means that many plumes will continue to persist well into or even beyond this current century (Seyedabbasi et al., 2012). Hence, given their environmental health significance and widespread abundance, it is little surprise that CEs are among the most prevalent VOC class reported in surface water to date (e.g. Yamamoto, 2014; Wittlingerová et al., 2016).

In complex landscapes with long histories of diverse land use practices, the concurrence of multiple groundwater pollution incidences is increasingly common. For instance, the natural

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attenuation and fate of point-source CE plumes which co-mingle with nitrate (NO_3^-) from non-point agricultural sources has not been widely considered at field scale to date (e.g. Bennett et al., 2007; Lu et al., 2017). In recent decades, global groundwater NO_3^- concentrations have been steadily rising as a result of increasing agricultural intensification (Gu et al., 2013; Wang et al., 2016). NO_3^- is a mobile anion which only weakly adsorbs to clay minerals (Meghdadi, 2018), often resulting in high loadings to vulnerable aquifers in agricultural regions (Ascott et al., 2017). This groundwater excess of NO_3^- is a well-known global environmental concern linked to eutrophication of receiving waters (Boyer et al., 2006) and blue-bay syndrome (methaemoglobinaemia) in drinking water (Addiscott and Benjamin, 2004).

Before reaching surface water, these groundwater pollutants must transit the aquifer-river interface or hyporheic zone (HZ) where groundwater and surface water interact (Boano et al., 2014; Cardenas, 2015). The HZ offers considerable promise as a passive biobarrier for in-situ ‘treatment’ of a broad range of organic groundwater pollutants (Schaper et al., 2018; Gilevska et al., 2019) including CEs (Weatherill et al., 2018). HZ sediments are often naturally rich in organic matter, resulting in hypoxic or anoxic pore water conditions (Atashgahi et al., 2015) where CEs (such as TCE) and NO_3^- are important terminal electron acceptors (TEAs) in dissimilatory microbial metabolism. Here, fermentation of dissolved organic carbon (DOC) supplies electrons via dissolved hydrogen (H_2) which drives a thermodynamic series of terminal electron-accepting processes (TEAPs) (Heimann et al., 2009). Under DOC-limiting conditions, this TEAP series tends to follow an ecological succession which can be predicted by the Gibb's free energy yield (ΔG°) [kJ mol^{-1}] of key redox couples: $\text{O}_2 \rightarrow \text{H}_2\text{O}$ [−238]; $\text{NO}_3^- \rightarrow \text{N}_2$ [−240]; $\text{Mn}^{\text{IV}} \rightarrow \text{Mn}^{\text{II}}$; [−185]; $\text{Fe}^{\text{III}} \rightarrow \text{Fe}^{\text{II}}$ [−126]; $\text{TCE} \rightarrow \text{cDCE}$ [−121]; $\text{cDCE} \rightarrow \text{VC}$ [−101]; $\text{SO}_4^{2-} \rightarrow \text{HS}^-$ [−48]; $\text{HCO}_3^- \rightarrow \text{CH}_4$ [−32.8] (Wiedemeier et al., 1998; under conditions reported therein).

From a thermodynamic perspective, NO_3^- reduction (denitrification) offers considerably higher free energy yields than reductive dechlorination of TCE and cDCE in heterotrophic metabolism (Weatherill et al., 2018). Hence, we hypothesise that elevated

groundwater NO_3^- will detrimentally impact the biogeochemical transformation capacity of HZs where plumes discharge through agriculturally intensive landscapes. To test this hypothesis, the present study aims to conceptualise critical biogeochemical gradients which govern the HZ transformation potential of a TCE plume co-mingled with non-point source NO_3^- . We apply a ‘bottom-up’ (Lansdown et al., 2015) multi-scale hydrochemical monitoring approach to: (a) resolve in-situ biogeochemical gradients controlling chemical fluxes from groundwater (b) identify potential hot-spots of multi-pollutant transformation and (c) evaluate the impact of regionally elevated NO_3^- concentrations on chlorinated ethene transformation in the HZ.

2. Materials and methods

2.1. Study area characteristics

The study area is located in the lowland River Tern catchment ($2^\circ 53' \text{W}$, $52^\circ 86' \text{N}$) in Shropshire (UK) where regional land use is dominated by intensive agriculture (Krause et al., 2013). The catchment is underlain by highly permeable Permo-Triassic sandstones with groundwater flow through a high porosity matrix (Shepley and Streetly, 2007) (Fig. 1). In this tributary of the River Severn, 76% of long-term river flow is derived from groundwater storage (Marsh and Hannaford, 2008). The present study focuses on a 40 m river reach that has been previously identified as the discharge zone of a deep-seated groundwater TCE plume (Fig. 1) (Weatherill et al., 2014). The unconfined regional sandstone aquifer is unlikely to significantly retard migration of the plume (Smith and Lerner, 2008).

2.2. Multi-scale hydrochemical monitoring approach

Depth-discrete groundwater samples were collected from four sandstone bedrock boreholes (Fig. 1) in March 2012 with passive grab sampling (HydraSleeve™) under ambient borehole flow conditions (Weatherill et al., 2014). A 40 m reach-scale network of multi-level mini-piezometers (MP1-MP25) (see Supporting

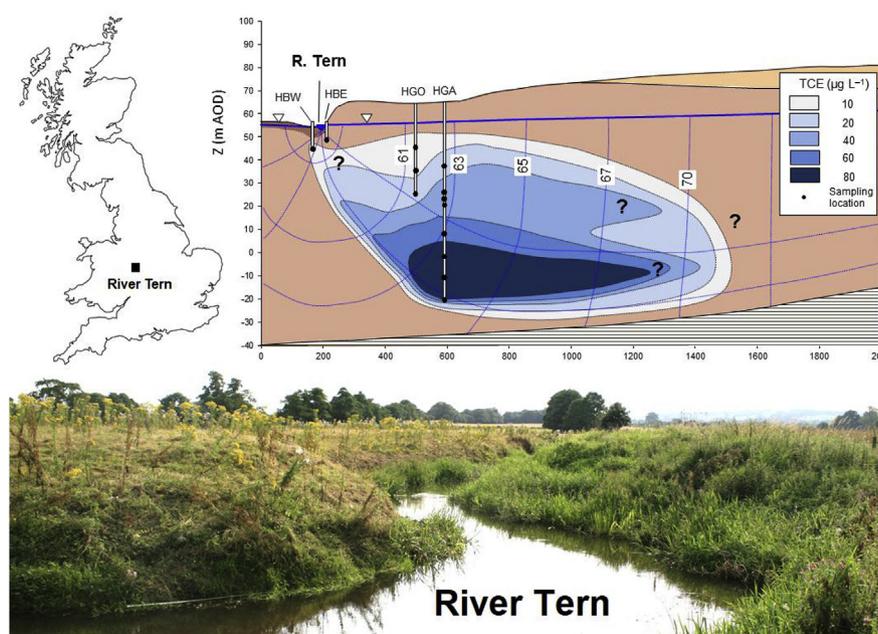


Fig. 1. Study area location and conceptual site model of regional aquifer and TCE plume and its inferred discharge zone at the River Tern (adapted from Weatherill et al., 2014). Black dots denote sampling depths in open boreholes using passive methods.

Information Appendix A; Fig. S1) deployed in the riverbed were sampled in early August 2012 during a summer baseflow recession period (S2). Reach-scale pore water chemistry samples were extracted using dedicated air-tight syringes from PTFE sample tubes with point screens in contact with sediment at 20, 40, 60, 80 and 100 cm below bed level (S1) (Weatherill et al., 2014). Surface water was collected as grab samples from 20 cm above the bed level from the upstream, mid-point and downstream sections of MP network (S1). To provide a high-resolution horizontal window on shallow pore water chemistry, a rectangular grid (0.6 × 2.7 m) of pore water samplers (PW1–PW13) with point-screens at 5, 10, 20 and 30 cm depth was also included (S1, S2). In-situ luminescent dissolved oxygen (DO) was measured in syringe and grab samples using an optical probe (Hach-Lange, UK) with a detection limit of 0.01 mg O₂ L⁻¹.

2.3. Laboratory analysis

VOC samples were analysed using a headspace GC-MS-SIM method described in Weatherill et al. (2014). Dissolved metabolic gas samples were collected in gas-tight Exetainer (Labco, UK) vials fixed by a 50 mol ZnCl solution to inhibit microbial activity. Methane (CH₄) and nitrous oxide (N₂O) were determined using N₂ headspace GC-FID/μECD (Agilent GC7890A series) equipped with a 1 mL sample loop. Equivalent dissolved phase concentrations were estimated from headspace gas volumetric concentrations using Henry's law (Comer-Warner et al., 2018). Equivalent mass concentration detection limits were <0.01 μg L⁻¹ for both gases. This GC method was not suitable to detect ethene, the non-chlorinated end-product of TCE reductive dechlorination.

Nitrate (NO₃⁻), nitrite (NO₂⁻), sulfate (SO₄²⁻) and chloride (Cl⁻), were determined by anion chromatography on a Dionex ICS1000 (Dionex Corporation, UK). Ammonium (as N-NH₄⁺) was analysed using an automated indophenol blue LCK304 method with a detection limit of 0.015 mg N L⁻¹ (Hach-Lange, Germany). Total dissolved iron (Fe), manganese (Mn) and silicon (Si) were determined from inline filtered (0.45 μm) and acidified samples using inductively coupled argon plasma optical emission spectroscopy (ICP–OES) (Varian Vista Pro MPX) with detection limits of <0.1 mg L⁻¹ for all elements. DOC samples were filtered inline by 0.7 μm pre-combusted (550 °C) glass-fibre filters to pre-combusted glass sample vials acidified with HCl and analysed on a Shimadzu TOC-Vcpn analyser (Shimadzu Corporation, Japan) with a detection limit of 0.2 mg C L⁻¹.

2.4. Data analysis

Statistical analyses were performed with SPSS v19 (IBM, USA). Variable distributions were tested using the Shapiro-Wilk test of normality to determine whether parametric or non-parametric statistics were appropriate. Differences in means between groups were analysed using paired T-tests or one-way analysis of variance

(ANOVA) for normally or log-normally distributed variables. All means reported were back-transformed from their logarithms. Non-parametric Mann-Whitney U-tests were used to test differences in medians for datasets which were neither normally nor log-normally distributed. Principal component analysis (PCA) was performed on a subset of pore water samples (n = 27) from the high-resolution pore water grid (S1, S2). PCA is advocated as a key line of evidence for biogeochemical interpretation of aquifer CE natural attenuation potential in recent guidance literature (Tarnawski et al., 2015).

3. Results and discussion

3.1. Aquifer hydrochemistry

Groundwater chemistry results are presented in Table 1 cDCE, VC, Fe, Mn NH₄⁺ and NO₂⁻ were all below their detection limits and are not shown. TCE was the only CE detected in the sandstone aquifer with the highest concentrations observed at the 80 m deep HGA abstraction borehole, approximately 500 m distance from the river (Fig. 1). The persistently elevated TCE concentrations (>150 μg L⁻¹) at depth in this open borehole are believed to be representative of a deep-seated bedrock plume (Weatherill et al., 2014). The plume is inferred to have a migration pathway along a curved axis between the HGA/HGO boreholes in the up-gradient aquifer and HBE/HBW boreholes on the riverbank (Fig. 1) (Weatherill et al., 2014). The groundwater environment of the up-gradient plume was rich in high-energy TEAs including DO (median: 5.6 mg L⁻¹) and NO₃⁻ (median: 55.3 mg L⁻¹) with moderate concentrations of SO₄²⁻ (median: 34.2 mg L⁻¹). Groundwater NO₃⁻ was notably elevated in the riverbank boreholes (73–81 mg L⁻¹). Dissolved N₂O was oversaturated by up to 35 times for all groundwater samples with a median concentration of 18 μg L⁻¹. The highest N₂O levels (up to 47 μg L⁻¹) were associated with the shallower depths of the up-gradient sandstone aquifer. DOC concentrations were very low in the aquifer (<2 mg L⁻¹), with highly undersaturated dissolved CH₄ concentrations with respect to air equilibrium (<0.1 μg L⁻¹).

3.2. Vertical chemical gradients in the hyporheic zone

An overview of maximum pore water TCE and cDCE concentrations in the uppermost metre of riverbed sediment are presented in Fig. 2 with all CE results reported in Supporting Information Appendix B (Table B1). cDCE was detected in 10 samples with a maximum of 6.9 μg L⁻¹ at MP7 at 40 cm. VC was only confirmed twice at 20 cm in MP18 (0.5 μg L⁻¹) and at MP23 (0.7 μg L⁻¹). This plume spatial variability in the riverbed sediments is consistent with previous monitoring results in the discharge zone (Weatherill et al., 2014).

Vertical depth-chemistry profiles are presented in Fig. 3 with comparison of pore water median concentrations at the base

Table 1
Permo-Triassic sandstone groundwater hydrochemistry. Refer to Fig. 1 for location of monitoring boreholes.

Borehole ID	pH	EC	DO	TCE	NO ₃	SO ₄	Cl	DOC	CH ₄	N ₂ O
	pH units	μS cm ⁻¹	mg O ₂ L ⁻¹	μg L ⁻¹	mg L ⁻¹	mg L ⁻¹	mg L ⁻¹	mg C L ⁻¹	μg L ⁻¹	μg L ⁻¹
HGA 35 m	7.6	623	5.1	89	66.2	24.8	32.6	2.19	0.08	44.3
HGA 50 m	7.6	556	6.6	137	53.1	23.1	33.0	1.64	0.09	16.1
HGA 80 m	7.5	545	6.2	162	66.7	20	34.2	1.35	0.08	18.1
HGO 20m	7.8	498	5.6	48.8	55.3	21.2	37.9	2.01	0.09	28.3
HGO 40m	7.7	502	5	74.4	52.8	20.5	36.0	1.58	0.08	20
HBE	7.1	753	6.4	26.9	81.1	28.4	29.6	2.14	0.07	5.9
HBW	7.3	661	6.8	9.38	73	22	37.5	1.94	0.09	17.2

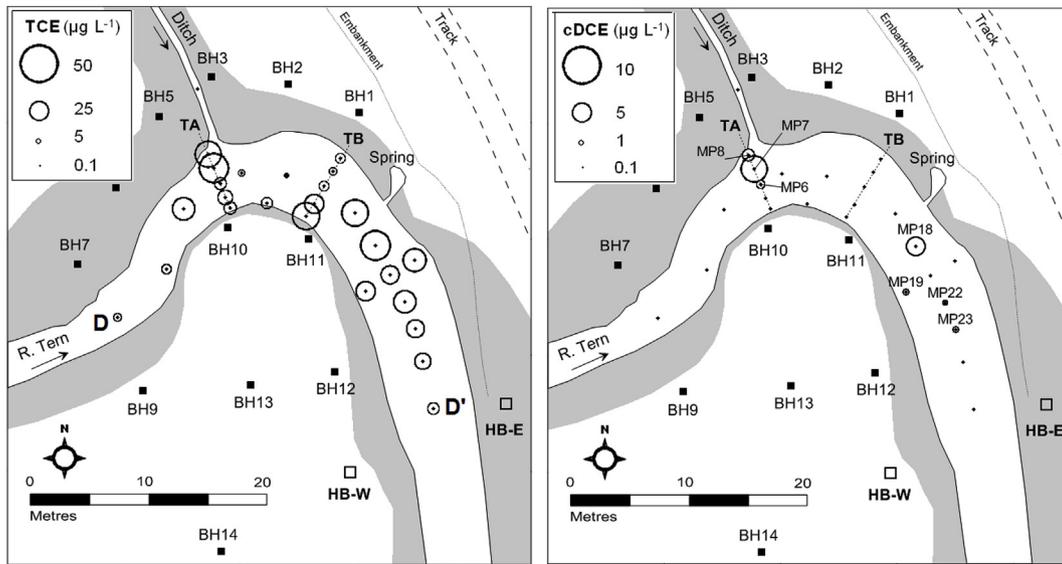


Fig. 2. Spatial variability of maximum TCE and cDCE at mini-piezometer locations in the plume discharge zone in August 2012. Cross-channel plume concentration gradients are observed at the TA and TB mini-piezometer transects with the highest TCE concentrations occurring downstream of the TB transect. In-situ reductive dechlorination is indicated by cDCE detections with the highest concentrations at MP7 and MP18.

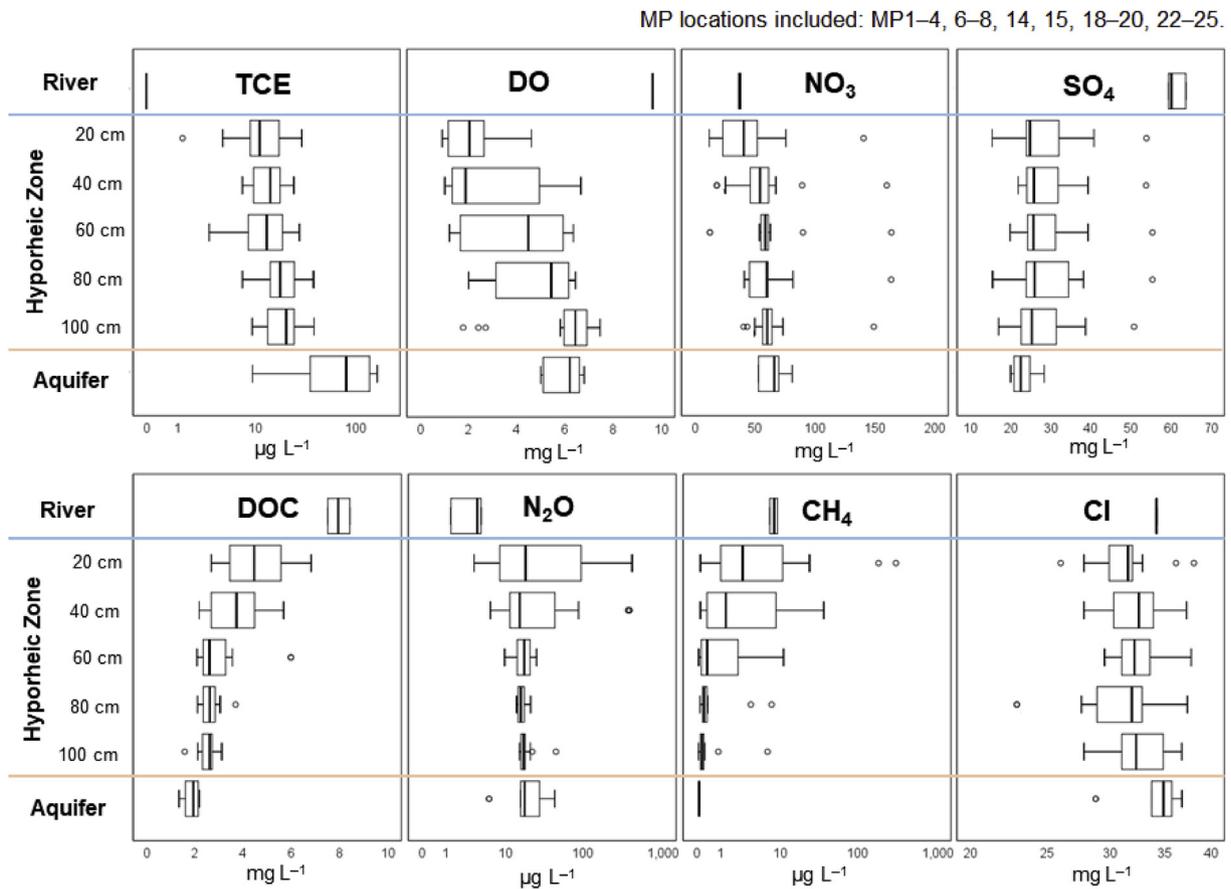


Fig. 3. Vertical depth-concentration profiles for mini-piezometer locations used to compare medians (n = 16) in comparison with the up-gradient aquifer (GW) (n = 7) and surface water (SW) (n = 3).

Table 2
Comparison of median solute concentrations using Mann-Whitney U-tests from selected mini-piezometer samples at the bottom (100 cm) and top (20 cm) of the hyporheic zone (n = 16) including the aquifer (n = 7) and river (n = 3) for comparison. Significant changes are expressed as percentages.

Parameter	units	Aquifer	MP _{100cm}	MP _{20cm}	River	Sig. change	p
TCE	µg L ⁻¹	74.4	21.1	12.0	<0.01	-43%	0.01
DO	mg O ₂ L ⁻¹	6.3	6.4	1.9	9.6	-71%	<0.01
NO ₃	mg L ⁻¹	60.8	60.3	40.1	37.4	-33%	<0.01
SO ₄	mg L ⁻¹	22.6	25.4	28.1	61.0	–	>0.05
Cl	mg L ⁻¹	34.2	32.3	31.6	34.3	–	>0.05
DOC	mg C L ⁻¹	1.98	2.6	4.7	7.96	+80%	<0.01
N ₂ O	µg L ⁻¹	17.7	18.3	17.6	3.2	–	>0.05
CH ₄	µg L ⁻¹	0.08	0.17	2.86	8.22	+1582%	<0.01

(100 cm) and top (20 cm) of the riverbed sediment sequence reported in Table 2. We assume that groundwater discharge through the HZ is predominantly vertical and that observed chemical gradients represent bulk attenuation along groundwater flow paths through riverbed pore water pathways (Krause et al., 2013). TCE concentrations exhibited an overall decline during transport through the HZ. High-energy TEAs (DO and NO₃⁻) and SO₄²⁻ concentrations in deep (100 cm) riverbed pore water samples were similar to the underlying aquifer (Table 2). A step change in DO concentrations from 60 to 40 cm to shallower sample depths with hypoxic conditions (e.g. <2 mg O₂ L⁻¹) was observed for many MP locations including the cross-channel TA transect and MP18-23 (Fig. 2).

NO₃⁻ concentrations declined by one third from the aquifer median during discharge through the HZ. Samples from a group of mini-piezometers (MP12, MP15 and MP16) exhibited exceptionally high NO₃⁻ concentrations (>130 mg L⁻¹) which were not associated with the up-gradient plume (Table 1). Overall, groundwater flow paths through the HZ were not a net source of dissolved N₂O in the plume discharge zone. However, six hypoxic samples from 20 to 40 cm depth had elevated N₂O concentrations (>200 µg L⁻¹) including MP6, MP7, MP18 and MP23. SO₄²⁻ concentrations did not vary significantly during transport through the HZ.

Pathways through the HZ were observed to be a significant source of DOC and CH₄ with marked increases from 40 to 20 cm for DOC and 60–40 cm for CH₄. These samples coincide with the hypoxic locations at Transect TA and MP18-MP23 and detections of dissolved Mn (up to 9.6 mg L⁻¹) and occasionally Fe. Under the near-neutral groundwater pH conditions (Table 1), soluble Fe and Mn are assumed to be mostly a product of microbially-mediated dissolution/reduction of Fe^{III}/Mn^{IV} mineral phases present in the sediment solids (Tarnawski et al., 2015).

The multi-level chemical profiles obtained at MP7, MP18 and MP223 (Fig. 2) were co-located with 100 cm sediment core samples from Weatherill et al. (2014). These cores provide a physical context for observed biogeochemical gradients below the sediment-water interface (Fig. 4). At each location, discrete changes in biogeochemical conditions takes place in the upper 20–60 cm layer of sediment, which is composed of medium sands (MP18 and MP23) and silty peat (MP7). Here, large declines in DO and NO₃⁻ are associated with elevated DOC, CH₄ and N₂O and the appearance of Mn and cDCE (Figs. 2 and 3). Our observations suggest that these hypoxic zones in the HZ can deplete high-energy TEAs from groundwater sufficiently to enable lower energy TEAPs to proceed allowing limited reductive dechlorination of TCE to cDCE.

3.3. High-resolution pore water sampler grid

A consistently reactive hypoxic zone was selected for a follow up high-resolution investigation using a rectangular grid pore water samplers (S2). The spatial variability of selected parameters are

presented in S3 with all results reported in Appendix B (Table B2). The grid was designed to target horizontal chemical gradients in the area of MP7 (Fig. 4) where the highest cDCE concentrations have been observed in previous work (Weatherill et al., 2014). The riverbed within the grid area was composed of medium quartz sands influenced by submerged macrophyte cover (*Ranunculus* spp.). Comparison of solute concentration means between 5, 10 and 20 cm sample depths using single factor ANOVA did not reveal any statistically significant 'depth effect' (p = >0.05) between depth-grouped samples. These results confirm that the sampler network had intercepted pore water above the critical vertical redox gradient in the HZ (Figs. 3 and 4).

TCE was present in all but four samples within the network with a maximum of 29 µg L⁻¹ at PW7 10 cm cDCE was detectable in 75% of samples with six locations exceeding the MP network maximum of 7 µg L⁻¹ (Fig. 4) reaching up 16 µg L⁻¹ at PW9 5 cm. VC was above detection limit in 27% of samples with a maximum of 2.3 µg L⁻¹ also at PW9 5 cm. All PW samples had DO concentrations ≤2 mg O₂ L⁻¹ with little variation across the network. NO₃⁻ was more spatially variable than DO reaching a maximum of 33 mg L⁻¹ at PW5 10 cm. This high energy TEA was below detection limit in 13% samples with a further 34% showing NO₃⁻ less than 5 mg L⁻¹. N₂O was spatially variable with concentrations exceeding 100 µg L⁻¹ in six samples mostly located in the upstream samplers. Very low concentrations of N₂O (<1 µg L⁻¹) were associated with locations which were also characterised by low NO₃⁻ concentrations in the downstream group of samples. These samplers also exhibited elevated NH₄⁺ (>1 mg N L⁻¹). SO₄²⁻ exhibited a greater range of concentrations than NO₃⁻ with a maximum of 66.4 mg L⁻¹ at PW1 10 cm. This SO₄²⁻ concentration is comparable to the overlying river water (Table 1) and may indicate surface water infiltration to the HZ at the upstream samplers.

Many samples from the downstream group of samplers exhibited high DOC concentrations (>10 mg L⁻¹) considerably exceeding that of river water (Table 2) and those from the MP network (Table 2, Figs. 3 and 4). As a product of low energy mineral-reducing TEAPs, dissolved Mn was widespread with a large number of downstream samples exhibiting elevated Mn concentrations (>10 mg L⁻¹). Fe was also present with a range of 1–4 mg L⁻¹ and a maximum of 12.7 mg L⁻¹ at PW11 30 cm. Many samples from the downstream part of the network exhibited CH₄ concentrations in excess of 1,000 µg L⁻¹ or more than twice that of the MP network maximum (Fig. 3).

Paired two sample T-tests were used to compare spatial means from paired upstream (PW1–PW5) and downstream (PW9–PW13) sample locations (Table 3). Significant declines (p = <0.05) are observed from upstream to downstream samples for TCE, NO₃⁻ and N₂O coupled with significant increases in DOC, Mn and CH₄. These results suggest that high and low energy TEAPs may be spatially organised at sub-metre horizontal scales within shallow HZ sediments.

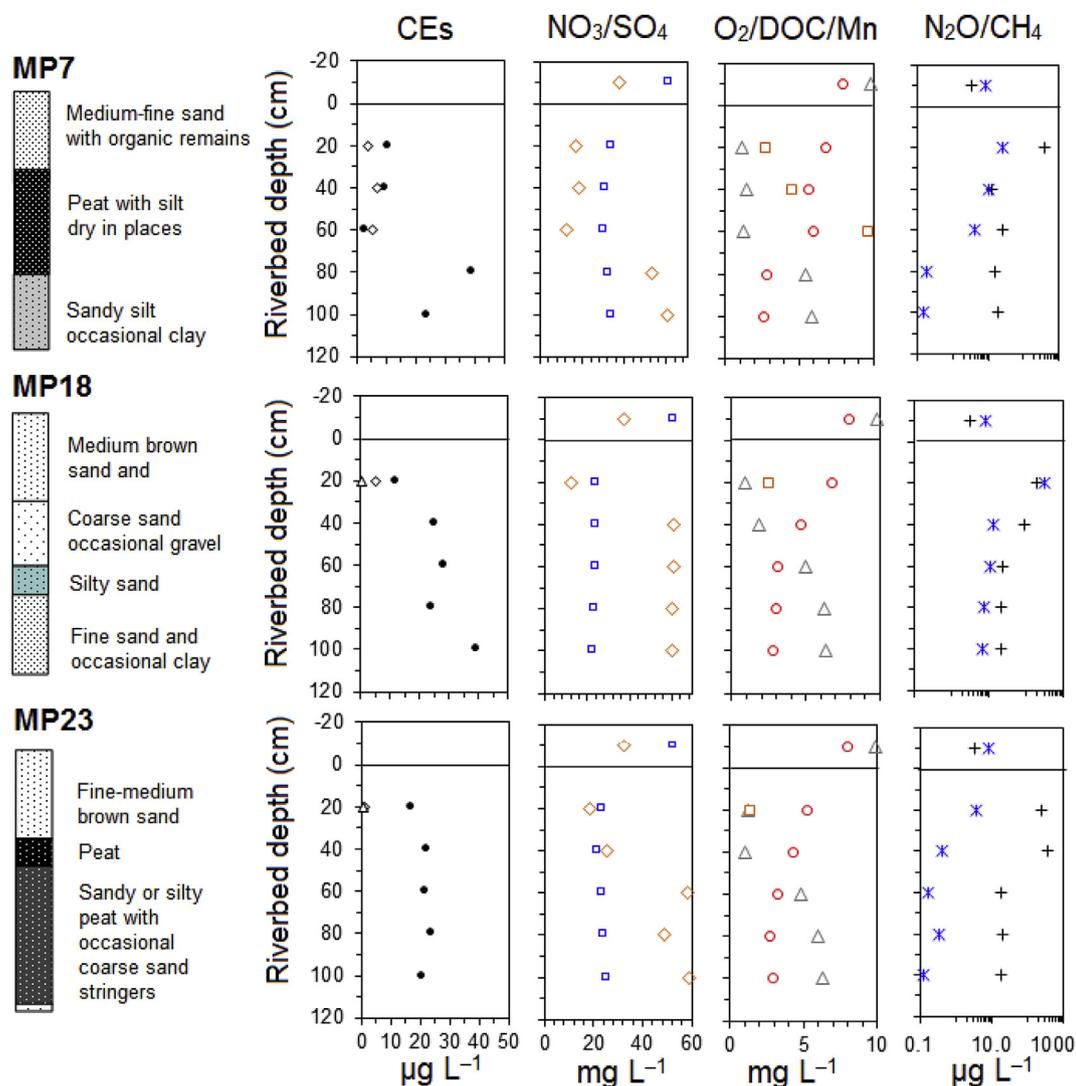


Fig. 4. Representative pore water biogeochemical profiles and riverbed sediment properties. TCE (black diamonds); cDCE (open black diamonds); VC (open black triangles); nitrate (open green diamonds); sulfate (open blue squares); DO (open triangles); DOC (open circles), Mn (open orange squares); blue stars (methane) and black crosses (nitrous oxide). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

Table 3

Comparison of selected mean solute concentrations at paired upstream/downstream samples of the pore water grid using paired T-tests (n = 14) (standard deviations in parentheses). Significant changes are expressed as percentages.

Parameter	units	PW _{upstream}	PW _{downstream}	Sig. change	p
NO ₃	mg L ⁻¹	14.1 (2.03)	5.1 (1.47)	-64%	<0.05
SO ₄	mg L ⁻¹	38.8 (1.38)	26.5 (1.53)	-32%	0.032
Cl	mg L ⁻¹	29.9 (1.11)	32.6 (1.06)	-	>0.05
Mn	mg L ⁻¹	2.07 (1.89)	6.11 (3.07)	+195%	<0.05
DOC	mg C L ⁻¹	4.53 (1.38)	6.89 (1.58)	+32%	<0.05
N ₂ O	µg L ⁻¹	31.5 (5.43)	6.93 (4.45)	-78%	<0.05
CH ₄	µg L ⁻¹	11.6 (2.90)	266.1 (2.15)	+2196%	<0.05

3.4. Evaluating in-situ plume transformation extent

The chlorine index (CI) is a useful metric integrating the stoichiometric mass balance of parent compound to daughter products into a single number (Harkness et al., 2012; Freitas et al., 2015). The CI negates the confounding effects of dilution and dispersion to evaluate parent compound transformation extent:

$$CI_{TCE} = \frac{\sum W_i C_i}{\sum C_i}$$

where W_i is the number of chlorine atoms in the CE molecule and C_i is the molar concentration of the CE species present. The CI approach is a cost-effective alternative to compound-specific stable isotope analysis (CSIA) (e.g. Gilevska et al., 2019) for single-parent compound plumes (Freitas et al., 2015) such as the one in this study. The relationship between total plume mass (in nmol) and CI_{TCE} are presented in Fig. 5 (samples where $CI_{TCE} = 3$ are omitted for clarity). Because it was not possible to include ethene, the minimum CI_{TCE} possible is 1 in this study. From this figure, it can be seen that all mini-piezometer (blue squares) and many samples from the upstream plot-scale grid (open circles) show similar weak dechlorination extents with CI_{TCE} values > 2.5 (partial TCE to cDCE transformation). A greater degree of TCE dechlorination is indicated for most samples from the downstream (black circles) and some mid-point samples (green circles) with a CI_{TCE} 1.6–2.5 (Fig. 5) where much of the plume mass is transformed to cDCE with limited cDCE reduction to VC.

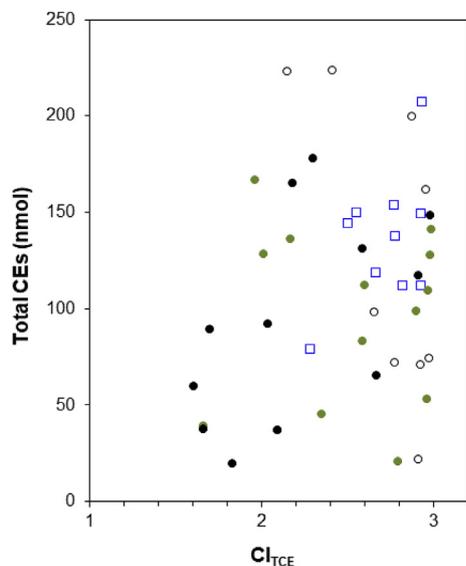


Fig. 5. Relationship between total pore water CEs (in nmol) and the chlorine index of TCE (Cl_{TCE}) in all riverbed samples where $Cl_{TCE} < 3$. Blue squares (reach-scale mini-piezometer samples); open circles (upstream PW samples); green circles (mid-point PW samples); black circles (downstream PW samples). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

3.5. Principal component analysis of pore water chemistry

Here, we use PCA to interpret the hypoxic biogeochemical environment in which reductive dechlorination occurs with concomitant TEAPs that cycle nitrogen, carbon, sulfur and minerals in the HZ (Tarnawski et al., 2015). PC1 and PC2 (Fig. 6) represent linear combinations of the original variables from 27 network samples which explain 73% of the observed variance. PC1 explains 44.3% and is strongly associated with Mn ($r = 0.92$), Si ($r = 0.91$), CH_4 ($r = 0.84$), NH_4^+ ($r = 0.8$) and to a lesser extent DOC ($r = 0.63$). PC1 is interpreted as a biogeochemical *metabolite factor* associated which is co-linear with the end-products of low energy TEAPs. This factor is spatially associated with most downstream and some mid-point sample locations, where pore water chemistry is dominated by mineral dissolution, Mn^{IV} reduction and methanogenesis. Although NH_4^+ is strongly correlated, PC1 is unrelated to NO_3^- which suggests that ammonification of organic matter is the dominant nitrogen cycling process rather than dissimilatory nitrate reduction to ammonium (DNRA) (Rivett et al., 2008). PC1 negatively correlates with Cl_{TCE} ($r = -0.73$) and to a lesser degree SO_4^{2-} ($r = -0.6$). Hence, in this low-energy biogeochemical environment, TCE is a favourable TEA where sediment/pore water hydrochemical interactions are enhanced.

PC2 accounts for 28.8% of the variance observed and is strongly associated with oxidised nitrogen species including NO_2^- and N_2O ($r = 0.88$) and NO_3^- ($r = 0.76$). Given the depleted NO_3^- concentrations and hypoxic pore water conditions present, NO_2^- and N_2O are likely to be intermediates of in-situ denitrification (Rivett et al., 2008; McAleer et al., 2017). NH_4^+ only very weakly correlates with this factor ($r = 0.34$) which suggests that DNRA may play a limited role in this environment. Hence, PC2 is interpreted as a *denitrification factor* which is spatially associated with most of the upstream and some mid-point samples. This factor is positively correlated with Cl_{TCE} ($r = 0.47$) and negatively correlated with DOC ($r = -0.5$). PC2 represents active NO_3^- reduction as a high energy TEAP which is spatially independent of low energy carbon, nitrogen and mineral cycling in the HZ.

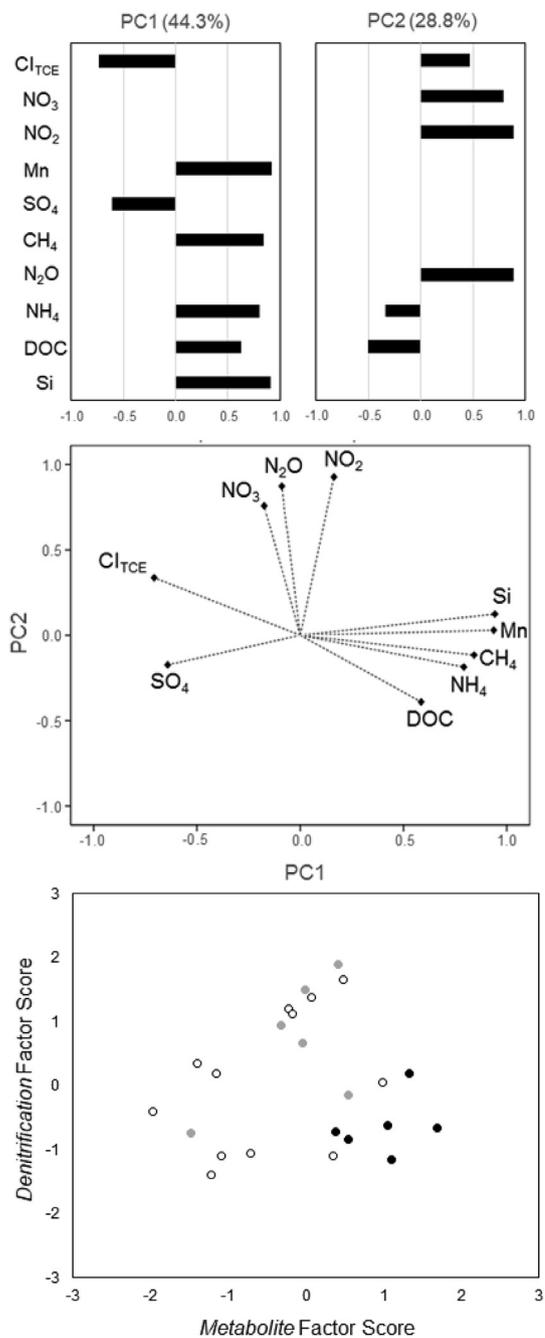


Fig. 6. Principle component analysis of high resolution pore water chemistry including factor loadings on original variables, ordination biplot and factor scores. The values in parenthesis is the percentage variance explained by PC1 and PC2. Factor scores are grouped as upstream (open circles), mid-point (green circles) and downstream (black circles) sample locations.

4. Discussion

4.1. Large scale hydrochemical gradients in the hyporheic zone

Reach-scale observations from multi-level profiles within the top metre of bed sediments suggest that the HZ is an important sink for high-energy groundwater TEAs (Table 1) with bulk attenuation of 71% for DO and 33% for NO_3^- (Table 2). Pore water TCE concentrations were observed to decline by 44% in the plume discharge zone (Table 2). The deeper riverbed is characterised by

discontinuous lenses of peat, clay and silt overlying the bedrock aquifer which may be up to several metres thick (Weatherill et al., 2014). From Fig. 5, clear differences can be seen in the vertical profiles of TCE where peat is absent (MP18) and present (MP7). Given that metabolites (cDCE and VC) were detected in just 9.5% of MP network samples, much of the observed pore water attenuation may be attributed to non-reactive sorptive-diffusive transport processes (e.g. Rivett et al., 2019).

Before reaching surface water, chemical fluxes from groundwater must transit variable thickness bedforms composed of well-sorted quartz sands with porosities of 20% (Weatherill et al., 2014; Weatherill, 2015). Heat flow experiments have shown that this pore water domain is in close hydraulic continuity with the overlying surface water column (Angermann et al., 2012) due to the presence of the flow-confining structures beneath (Gomez-Velez et al., 2014). Aquifer and river water Cl^- concentrations did not contrast significantly (Fig. 3) and hence Cl^- could not be used as a conservative tracer to delineate hyporheic exchange zone extent (e.g. Freitas et al., 2015). On the other hand, SO_4^{2-} concentrations differed considerably (Fig. 3). Nonetheless, no evidence of surface water mixing to a depth of 20 cm can be seen in the SO_4^{2-} profiles (Fig. 3) under baseflow conditions. This precludes dilution from surface water infiltration as a factor in the observed TCE attenuation under baseflow conditions in the MP network (e.g. Hamonts et al., 2012).

Our findings suggest that in-situ reactive transport takes place within discrete hypoxic envelopes associated with the TA transect MP locations (40–60 cm thick) and MP18–MP23 where it is thinner (20 cm). This biogeochemically active zone is mostly associated with the quartz sands layer and some peat-rich lenses in the underlying cohesive deposits (Fig. 4). At MP6, MP7, MP18 and MP23, step-change declines in NO_3^- are associated with greater than 10-fold increases in pore water N_2O concentrations ($>200 \mu\text{g L}^{-1}$) (Figs. 3 and 4). N_2O production in the HZ is thought to be indicative of partial NO_3^- reduction where pore water residence times are insufficient to allow complete transformation to N_2 (Quick et al., 2016). These hypoxic zones are also associated with an 80% increase in pore water DOC concentrations and a greater than 20-fold increase in CH_4 from the aquifer background (Table 2). The presence of CH_4 and Mn indicate that low-energy TEAPs take place where high energy TEAs are locally depleted (e.g. $\text{DO} < 2 \text{ mg O}_2 \text{ L}^{-1}$; $\text{NO}_3^- < 4 \text{ mg L}^{-1}$) (Fig. 4). These conditions appear to be capable of supporting partial reductive dechlorination of TCE to cDCE (e.g. $\text{Cl}_{\text{TCE}} > 2.5$) (Fig. 5).

4.2. Evidence of multi-pollutant transformation hotspots

Evidence of discrete horizontal organisation of dissimilatory TEAPs at sub-metre scales was observed within the rectangular grid (Table 3). At the time of sampling, much of the riverbed at the TA transect of the MP network as well as parts of the wider river corridor was occluded by submerged *Ranunculus* spp. stands. In most upstream PW network samples, active denitrification is suggested by depleted NO_3^- concentrations ($<4\text{--}30 \text{ mg L}^{-1}$) with elevated N_2O ($>100 \mu\text{g L}^{-1}$) and NO_2^- ($0.4\text{--}2 \text{ mg L}^{-1}$). In some downstream and some mid-point samples, oxidised nitrogen species are mostly absent and the pore water chemistry is dominated by the metabolites of low-energy TEAPs including mineral dissolution/reduction, methanogenesis and ammonification (Fig. 6). From a thermodynamic perspective, this low-energy metabolic regime presents a favourable environment for plume transformation which is demonstrated by the low chlorine indices observed in the sub-group of PW samples where $\text{Cl}_{\text{TCE}} = 1.6\text{--}2.5$ (Fig. 5). Overall, the reductive dechlorination potential of the hyporheic zone appears to be limited to cDCE production with only

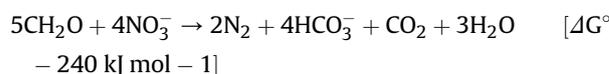
minor concentrations of VC detected which is consistent with previous results (Weatherill et al., 2014). This incomplete natural attenuation may be ascribed to the presence of residual DO concentrations, a lack of suitable cDCE dechlorinator populations or competition for reducing equivalents by metal-reducing bacteria (Chambon et al., 2013; Paul et al., 2016). The presence of both NH_4^+ and CH_4 in hypoxic samples may favour alternative cometabolic pathways for cDCE and VC mineralisation at the interface between hypoxic and oxic flow paths (Mattes et al., 2010). CSIA and molecular microbial techniques may offer additional lines of evidence to elucidate metabolic pathways and the activities of specific degraders (Badin et al., 2016; Ottosen et al., 2019) and are advocated for future work.

DOC concentrations exceeding that of surface water and the underlying pore water (e.g. $>10 \text{ mg L}^{-1}$) were observed in many samples from the PW network. This is consistent with other reports of vegetated riverbeds as important sources of pore water DOC with intensified hotspots of carbon cycling in the HZ as a result (e.g. Trimmer et al., 2009; Ullah et al., 2014). Submerged macrophytes are thought to modify their local aquatic environment by reducing river flow velocities resulting in increased fine sediment deposition in spring and summer months (Sand-Jensen, 1998; Heppell et al., 2009) with a consequent reduction in bed permeability and an increase in contaminant exposure times in reactive zones (Oldham et al., 2013). Our results suggest that the hypoxic sediments at the mid and down-stream parts of the network were influenced by organic-rich fine sediment derived from allochthonous surface water catchment sources (Ballantine et al., 2008), resulting in elevated pore water DOC, Mn, Si and NH_4^+ concentrations. In addition, delivery of root exudate DOC in the rooting zone which cuts across the HZ may play a role in enhanced biogeochemical cycling in vegetated sediments (Ullah et al., 2014). Our observations show that both N_2O and CH_4 are useful indicators for hotspots of high and low-energy TEAP activity respectively in the HZ.

4.3. Implications of elevated nitrate on TCE transformation potential

The large metabolic energy gain from reduction of NO_3^- implies that when DOC is in excess, denitrification will take place in preference to lower energy reductive dechlorination reactions in the HZ. Liu et al. (2015) demonstrated that NO_3^- additions had an inhibitory effect on low energy TEAPs (SO_4^{2-} reduction and methanogenesis) in sewer sediments. Denitrifying bacteria can maintain pore water H_2 thresholds ($<0.1 \text{ nmol}$) far lower than necessary for reductive dechlorination (Weatherill et al., 2018). However, considerable overlap in H_2 thresholds are observed experimental studies under metal-reducing conditions with mineral bioavailability playing an important role (Paul et al., 2016). Our findings suggest that reduction of NO_3^- and TCE to cDCE dechlorination occur independently of one another which is supported by the PCA results in Fig. 6.

Egli (2010) suggest a minimum DOC concentration of 2 mg L^{-1} is required to initiate catabolic genes involved in microbial contaminant transformation. Throughout the up-gradient aquifer, DOC concentrations were at or below this threshold (Table 1). The DOC resources required to reduce high-energy groundwater TEAs (DO and NO_3^-) are estimated using the following stoichiometry (Tesoriero and Puckett, 2011):



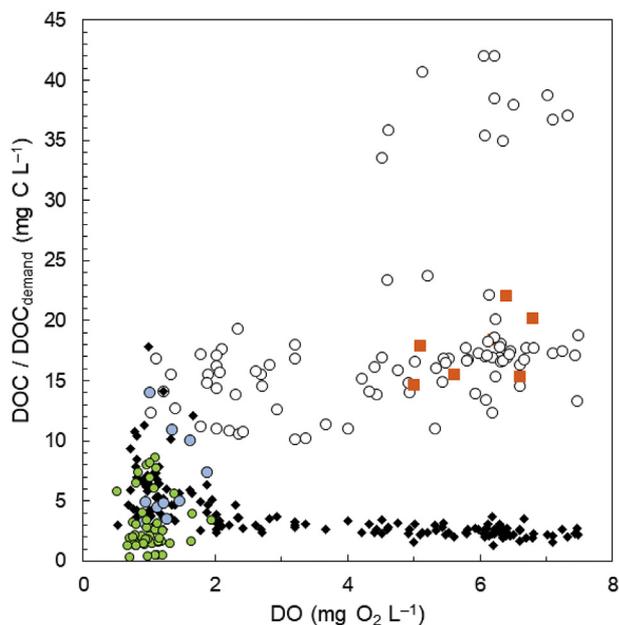


Fig. 7. Calculated DOC demand for reduction of dissolved oxygen and nitrate in samples for the sandstone aquifer (orange squares); MP samples (open circles); MP samples showing TCE degradation (blue circles) and PW samples (green circles). Black diamonds denote observed DOC concentrations for each sample. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

With this stoichiometry, it is possible to estimate the regional DOC demand (mg C L^{-1}) imposed by high-energy TEAPs at overlapping spatial scales within the monitoring networks. Fig. 7 plots observed DO concentration against high-energy DOC demand from DO/NO_3^- concentrations recorded in samples. Given the limited solubility of O_2 in groundwater, the high-energy DOC demand is dominated by the elevated NO_3^- background (Table 1, Fig. 3). By plotting observed DOC concentration for each sample (denoted by the black diamonds) with the calculated DOC demand, it is possible to identify locations in the HZ where DOC is in excess (Fig. 7). NO_3^- -DOC stoichiometric relationships exert important controls on groundwater redox conditions that govern nitrogen export patterns at landscape scale (Taylor and Townsend, 2012; Helton et al., 2015). It is clear that the DOC required to reduce elevated groundwater NO_3^- produces an electron donor-limited environment except for a subset of hypoxic PW samples. In these locations, lower-energy metabolism is favoured including reductive dechlorination of TCE to cDCE. Therefore, elevated NO_3^- levels are likely to impose a critical redox ‘buffer’ (e.g. Diem et al., 2013) which must first be overcome when CE plumes traverse intensive agricultural regions underlain by vulnerable oxic groundwater systems.

5. Conclusions

With the application of a multi-scale ‘bottom-up’ monitoring approach, we have shown that biogeochemical processes in the HZ can locally modify chemical fluxes and redox conditions in discharging groundwater. Biogeochemically active hypoxic pathways through the uppermost 20–60 cm of sediment pore water are capable of reducing bulk groundwater discharge of NO_3^- by one third. In these zones, active denitrification is indicated by depleted NO_3^- and elevated dissolved N_2O concentrations. Using the chlorine index of TCE as a metric to evaluate in-situ plume transformation, we have shown that enhanced dechlorination of TCE to cDCE occurs locally within DOC-rich hotspots where pore water chemistry is

dominated by the end-products of low-energy microbial metabolism (Mn , CH_4 and NH_4).

Our results highlight the importance of DOC-rich pore water associated with vegetated riverbeds for seasonal ecosystem service provision in the passive ‘treatment’ of groundwater pollutants discharging from carbon-poor oxic aquifers. This local electron donor excess is capable of overcoming the large stoichiometric demands for carbon resources posed by high background concentrations of NO_3^- . Here, an ecological succession of TEAPs is enabled where reductive dechlorination of CE species becomes thermodynamically favoured. Below a critical DOC threshold, reduction of elevated groundwater NO_3^- inhibits the transformation of CEs in dissimilatory microbial metabolism. The inclusion pore water N_2O and CH_4 in combination with DO and traditional redox indicators to identify respective high and low energy microbial metabolic regimes is advocated further.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.watres.2019.05.083>.

Declaration of interests

☒ The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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