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Synthesis of [1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium and [1,2,4]triazolo[5,1-*b*][1,3]thiazin-4-ium salts via regioselective electrophilic cyclization of 3-[(2-alken-1-yl)sulfanyl]-4*H*-1,2,4-triazoles

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Abstract: A convenient procedure for the regioselective preparation of [1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium **10** and [1,2,4]triazolo[5,1-*b*][1,3]thiazin-4-ium salts **9** via regioselective electrophilic cyclization of 3-[(2-alken-1-yl)sulfanyl]-4*H*-1,2,4-triazoles **3** is reported. Direction of electrophilic heterocyclization strongly depends on nature of the alkenyl substituent.

Keywords: electrophilic cyclization; halogen; regioselectivity; selenium tetrahalogenides; tellurium tetrahalogenides; [1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium; [1,2,4]triazolo[5,1-*b*][1,3]thiazin-4-ium.

Introduction

Among the large number of condensed derivatives of the symmetric triazoles many compounds show a wide range of biological activities: antimicrobial and antifungal [1–5], anti-inflammatory [6–8], and anti-convulsive properties [5, 9], among others [10]. Considering these observations, it was envisaged to search effective ways for synthesis of new condensed derivatives of symmetrical triazoles, which can possess valuable properties. In recent years, hetero-annulation processes based on electrophilic halocyclization have proved to be useful in producing various heterocycles including furan [11–14], pyrrole [11, 15], quinoline [16, 17], and a large number of complex heterocyclic compounds [18–30]. The regioselectivity of electrophilic

halocyclization of 1,2,4-triazole derivatives was investigated as a function of the reaction conditions [25–27], nature of electrophilic agents [25, 27, 29, 30] and nature of the heterocyclic nucleophilic center in the molecule [26, 29]. On this basis, it was reasoned that introduction of various groups into an allyl substituent of 1,2,4-triazole system can also influence regioselectivity of the annelation reactions.

Results and discussion

Synthesis of [1,3]thiazolo[2,3-*c*][1,2,4]triazoles upon halogenation of 3-allylsulfanyl-4,5-diphenyl-3*H*-1,2,4-triazole was discussed in our previous work [25]. In this work the regioselectivity of electrophilic cyclization of the alkenyl derivatives of 3-sulfanyl-1,2,4-triazole **3** was investigated. Compounds **3** were obtained by the reaction of an excess of the corresponding alkenyl chloride **2** with alkaline alcoholic solutions of triazoles **1**. Methylallyl chloride **2a** and cinnamyl chloride **2b** were used in the reaction conducted in boiling ethanol. Target sulfanyl ethers **3** precipitated after cooling the reaction mixture.

The structure of compounds **3a–d** was confirmed by using spectral methods.

Bromination of (methylallyl)sulfanyl compounds **3a,b** furnished condensed monobromides **4a,b**. The reactions were carried out at 15°C for 3 h with a ratio of Br₂/**3** of 2:1. It should be noted that the nature of solvent (chloroform, glacial acetic acid) did not influence the yields of salts **4a,b**. A similar pattern was observed in the reaction of iodine with (methylallyl)sulfanyl derivative **3a**. This treatment gave iodide **5a** as the cyclization product. On the other hand, condensed salts **7a–d** were obtained upon electrophilic cyclization of compound **3a** using selenium and tellurium tetrahalogenides **6a–d**. Solutions of the reagents **6a–d** in acetic acid were prepared from a chalcogen oxide in the presence of an excess amount of hydrochloric or hydrobromic acid.

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In comparison to the outcome of the reaction of methylallyl derivatives, the regioselectivity process of halocyclization of the cinnamyl analogs **3b,d** is different as the cyclization is accompanied by an annelation to a six-membered thiazine system to form [1,2,4]triazolo[5,1-*b*] [1,3]thiazin-8-ium bromides **8a,b** and [1,2,4]triazolo[5,1-*b*] [1,3]thiazin-8-ium iodides **9a,b**. The difference in halocyclization selectivity can be explained by a powerful steric effect of the aromatic substituent in the cinnamyl fragment.

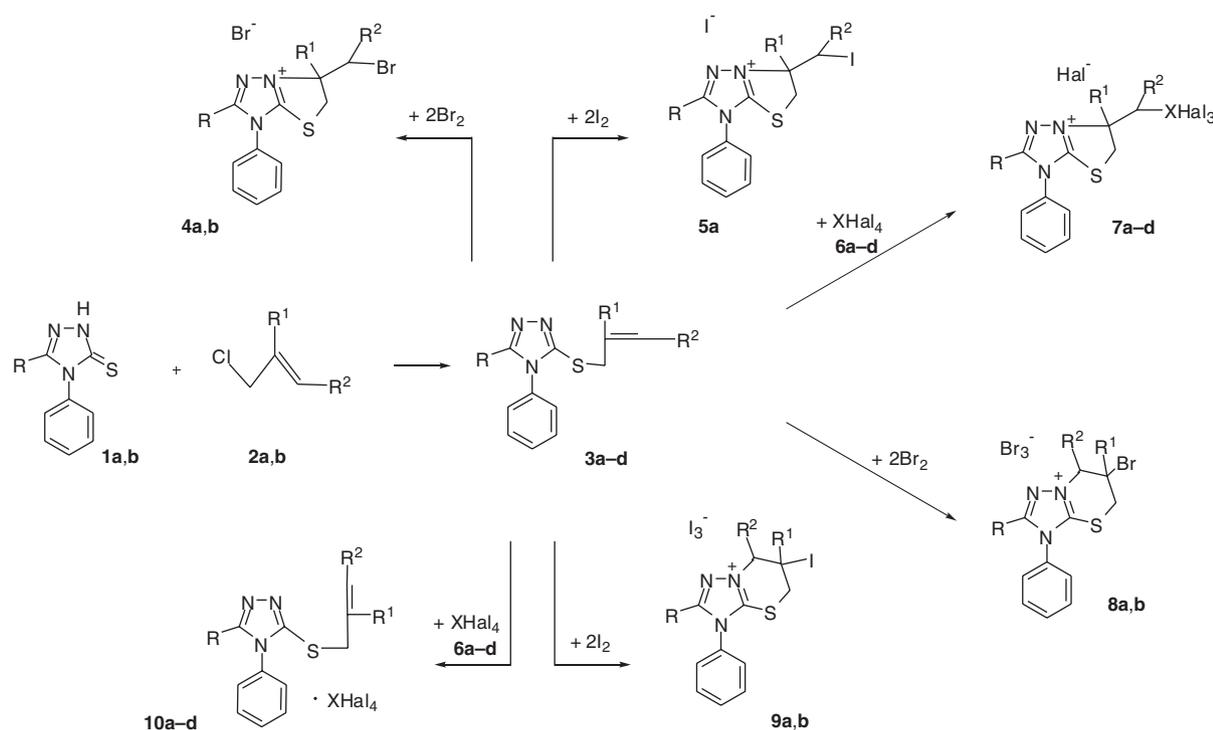
In contrast to the treatment with halogens, the reaction of tetrahalides of selenium and tellurium **6a-d** with cinnamyl derivatives does not cause annelation to form an additional heterocyclic system. This reaction yields stoichiometric yellowish adducts **10a-d**. The spectral characteristics of these adducts are almost identical to those of the substrates **3b,d**. This outcome can be explained in terms of the deactivating influence of steric factors of the

cinnamyl substituent and selenium/tellurium tetrahalides. It should be noted that heating of compounds **10a-d** for 7 h did not lead to their cyclization, and the starting complexes **10a-d** were isolated in high yields.

All compounds **4-10** were characterized by means of NMR spectroscopy including ^1H , ^{13}C , ^{79}Se , NOE and COSY experiments, mass spectrometry and elemental analysis. The proposed structures are fully consistent with the experimental data.

Conclusions

Methylallylsulfanyl- and cinnamylsulfanyl-substituted 1,2,4-triazoles **3a-d** were synthesized and their diverse electrophilic transformations were studied. The nature of the alkenyl fragment exerts a strong influence on the product structure.



- 1a:** R = Ph
1b: R = 4-O₂N-C₆H₄
2a: R¹ = Me; R² = H
2b: R¹ = H; R² = Ph
3a: R = Ph; R¹ = Me; R² = H
3b: R = Ph; R¹ = H; R² = Ph
3c: R = 4-O₂N-C₆H₄; R¹ = Me; R² = H
3d: R = 4-O₂N-C₆H₄; R¹ = H; R² = Ph
4a: R = Ph; R¹ = Me; R² = H
4b: R = 4-O₂N-C₆H₄; R¹ = Me; R² = H
5a: R = Ph; R¹ = Me; R² = H
6a: X = Se; Hal = Cl
6b: X = Te; Hal = Cl
6c: X = Se; Hal = Br
6d: X = Te; Hal = Br
7a: R = Ph; R¹ = Me; R² = H; X = Se; Hal = Cl
7b: R = Ph; R¹ = Me; R² = H; X = Te; Hal = Cl
7c: R = Ph; R¹ = Me; R² = H; X = Se; Hal = Br
7d: R = Ph; R¹ = Me; R² = H; X = Te; Hal = Br
8a: R = Ph; R¹ = H; R² = Ph
8b: R = 4-O₂N-C₆H₄; R¹ = H; R² = Ph
9a: R = Ph; R¹ = H; R² = Ph
9b: R = 4-O₂N-C₆H₄; R¹ = H; R² = Ph
10a: R = Ph; R¹ = H; R² = Ph; X = Se; Hal = Cl
10b: R = Ph; R¹ = H; R² = Ph; X = Te; Hal = Cl
10c: R = Ph; R¹ = H; R² = Ph; X = Se; Hal = Br
10d: R = Ph; R¹ = H; R² = Ph; X = Te; Hal = Br

Scheme 1

Experimental

¹H NMR (300 MHz or 400 MHz) and ¹³C NMR (75 MHz or 100 MHz) spectra were recorded in DMSO-*d*₆ on a Varian VXR-300 or Bruker DPX-400 instruments; the ⁷⁵Se NMR spectrum of **7c** was recorded on a Bruker DPX-400 instrument. 2D-NOESY and COSY experiments were carried out for compounds **7d**, **8b** in CDCl₃ on a Varian Mercury-400 instrument. Microanalyses were performed by the microanalytical unit of the Institute of Organic Chemistry of the National Academy of Sciences (Kyiv, Ukraine). The EI mass spectrum of compound **7d** was obtained using an Agilent 1100 LCMSD SL instrument. Melting points were determined on a Kofler block instrument and were not corrected.

General procedure for the preparation of substituted 4*H*-1,2,4-triazoles **3a–d**

A substituted 2,4-dihydro-3*H*-1,2,4-triazole-3-thione **1a** or **1b** (10.0 mmol) was dissolved in ethanol (20 mL) with heating and the solution was cooled and treated with an alkenyl halogenide **2a** or **2b** (12.0 mmol) in ethanol (5 mL). The mixture was heated under reflux for 1 h. The precipitated product was filtered, washed with ether and dried.

3-[(2-Methylprop-2-en-1-yl)sulfanyl]-4,5-diphenyl-4*H*-1,2,4-triazole (3a**)** This compound was obtained from **1a** and **2a**; yield 80% of colorless crystals; mp 140°C; ¹H NMR (300 MHz): δ 1.70 (s, 3H), 3.76 (s, 2H), 4.83 (s, 1H), 4.91 (s, 1H), 7.35–7.55 (m, 10H). Anal. Calcd for C₁₈H₁₇N₃S: C, 70.33; H, 5.57; N, 13.67; S, 10.43. Found: C, 70.70; H, 5.61; N, 13.52; S, 10.29.

3,4-Diphenyl-5-[(3-phenylprop-2-en-1-yl)sulfanyl]-4*H*-1,2,4-triazole (3b**)** This compound was obtained from **1a** and **2b**; yield 52% of colorless crystals; mp 152–154°C; ¹H NMR (300 MHz): δ 3.97 (d, *J* = 4.2 Hz, 2H), 6.36 (m, 1H), 6.58 (d, *J* = 9.3 Hz, 1H), 7.23–7.52 (m, 15H). Anal. Calcd for C₂₃H₁₉N₃S: C, 74.77; H, 5.18; N, 11.37; S, 8.68. Found: C, 75.01; H, 5.11; N, 11.21; S, 8.50.

3-[(2-Methylprop-2-en-1-yl)sulfanyl]-5-(4-nitrophenyl)-4-phenyl-4*H*-1,2,4-triazole (3c**)** This compound was obtained from **1b** and **2a**; yield 57% of yellowish crystals; mp 168–170°C; ¹H NMR (300 MHz): δ 1.72 (s, 3H), 3.82 (s, 2H), 4.85 (s, 1H), 4.94 (s, 1H), 7.42–7.68 (m, 7H), 8.21 (d, *J* = 8.1 Hz, 2H). Anal. Calcd for C₁₈H₁₆N₄O₂S: C, 61.35; H, 4.58; N, 15.90; S, 9.10. Found: C, 61.48; H, 4.49; N, 15.72; S, 9.14.

3-(4-Nitrophenyl)-4-phenyl-5-[(3-phenylprop-2-en-1-yl)sulfanyl]-4*H*-1,2,4-triazole (3d**)** This compound was obtained from **1b** and **2b**; yield 64% of yellowish crystals; mp 170–172°C; ¹H NMR (300 MHz): δ 3.98 (d, *J* = 4.5 Hz, 2H), 6.32 (m, 1H), 6.58 (d, *J* = 9.9 Hz, 1H), 7.20–7.57 (m, 12H), 8.15 (d, *J* = 5.4 Hz, 2H); ¹³C NMR (75 MHz): δ 153.2, 153.0, 148.1, 136.5, 133.9, 133.6, 133.0, 130.7, 130.5, 129.3, 129.0, 128.2, 128.0, 126.7, 124.8, 124.1, 35.2. Anal. Calcd for C₂₃H₁₈N₄O₂S: C, 66.65; H, 4.38; N, 13.52; S, 7.74. Found: C, 66.79; H, 4.16; N, 13.32; S, 7.70.

General procedure for the preparation of fused salts **4**, **5**, **7–9** and complexes **10**

A solution of bromine or iodine (20.0 mmol) or a solution of a tetrahalogenide of selenium or tellurium **6a–d** (10.0 mmol) in acetic acid

was added dropwise to the solution of a triazole **3a–d** (10.0 mmol) in acetic acid with constant stirring at room temperature. The resultant solid product was filtered, washed with acetone (**4**, **5**, **8**, **9**) or ether (**7**, **10**) and dried.

6-(Bromomethyl)-6-methyl-2,3-diphenyl-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium bromide (4a**)** This compound was obtained by bromination of **3a**; yield 78% of a white powder; mp 285–287°C; ¹H NMR (300 MHz): δ 2.01 (s, 3H), 4.28–4.34 (m, 2H), 4.48 (dd, *J* = 31.5, 12.0 Hz, 2H), 7.46–7.75 (m, 10H); ¹³C NMR (75 MHz): δ 160.1, 157.5, 132.6, 131.4, 129.6, 126.9, 123.9, 69.1, 48.2, 37.9, 23.3. Anal. Calcd for C₁₈H₁₇Br₂N₃S: C, 46.27; H, 3.67; Br, 34.20; N, 8.99; S, 6.86. Found: C, 45.98; H, 3.52; Br, 34.52; N, 8.84; S, 6.79.

6-(Bromomethyl)-6-methyl-2-(4-nitrophenyl)-3-phenyl-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium bromide (4b**)** This compound was obtained by bromination of **3c**; yield 56% of yellowish powder; mp 259–261°C; ¹H NMR (400 MHz): δ 1.96 (s, 3H), 4.21–4.25 (m, 2H), 4.45 (dd, *J* = 30.0, 11.6 Hz, 2H), 7.65–7.73 (m, 7H), 8.35 (d, *J* = 9.2 Hz, 2H). Anal. Calcd for C₁₈H₁₆Br₂N₄O₂S: C, 42.21; H, 3.15; Br, 31.20; N, 10.94; S, 6.26. Found: C, 42.02; H, 3.11; Br, 31.55; N, 10.78; S, 6.17.

6-(Iodomethyl)-6-methyl-2,3-diphenyl-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium iodide (5a**)** This compound was obtained by iodination of **3a**; yield 61% of a brown powder; mp 248–250°C; ¹H NMR (300 MHz): δ 1.98 (s, 3H), 3.95–4.04 (m, 2H), 4.41 (dd, *J* = 27.3, 12.0 Hz, 2H), 7.39–7.72 (m, 10H). Anal. Calcd for C₁₈H₁₇I₂N₃S: C, 38.52; H, 3.05; I, 45.22; N, 7.49; S, 5.71. Found: C, 38.79; H, 2.99; I, 45.56; N, 7.33; S, 5.68.

6-Methyl-2,3-diphenyl-6-[(trichloro-λ⁴-selenanyl)methyl]-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium chloride (7a**)** This compound was obtained from **3a** and **6a**; yield 39% of a white powder; mp 220°C (dec); ¹H NMR (400 MHz): δ 2.01 (s, 3H), 3.96 (d, *J* = 9.0 Hz, 1H), 4.07 (d, *J* = 8.7 Hz, 1H), 4.41 (d, *J* = 11.1 Hz, 1H), 4.97 (d, *J* = 11.1 Hz, 1H), 7.22–7.41 (m, 10H). Anal. Calcd for C₁₈H₁₇Cl₄N₃SSe: C, 40.93; H, 3.24; Cl, 26.85; N, 7.96. Found: C, 41.20; H, 3.29; Cl, 26.68; N, 8.02.

6-Methyl-2,3-diphenyl-6-[(trichloro-λ⁴-telluranyl)methyl]-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium chloride (7b**)** This compound was obtained from **3a** and **6b**; yield 45% of a white powder; mp 255°C (dec); ¹H NMR (400 MHz): δ 2.21 (s, 3H), 4.06 (d, *J* = 9.0 Hz, 1H), 4.35 (d, *J* = 12.0 Hz, 1H), 4.49 (d, *J* = 9.0 Hz, 1H), 5.17 (d, *J* = 12.0 Hz, 1H), 7.34–7.91 (m, 10H). Anal. Calcd for C₁₈H₁₇N₃S: C, 37.48; H, 2.97; Cl, 24.58; N, 7.28. Found: C, 37.72; H, 3.03; Cl, 24.28; N, 7.39.

6-Methyl-2,3-diphenyl-6-[(tribromo-λ⁴-selenanyl)methyl]-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium bromide (7c**)** This compound was obtained from **3a** and **6c**; yield 44% of a yellowish powder; mp 210°C (dec); ¹H NMR (400 MHz): δ 2.02 (s, 3H), 4.35 (d, *J* = 11.1 Hz, 1H), 4.48 (d, *J* = 9.0 Hz, 1H), 4.59 (d, *J* = 9.0 Hz, 1H), 4.90 (d, *J* = 11.1 Hz, 1H), 7.24–7.50 (m, 10H); ¹³C NMR (100 MHz): δ 159.5, 156.7, 132.4, 129.7, 126.5, 123.9, 70.9, 49.3, 45.4, 25.2; ⁷⁵Se NMR (DMSO-*d*₆): δ 660. Anal. Calcd for C₁₈H₁₇Br₃N₃SSe: C, 30.62; H, 2.43; Br, 45.27; N, 5.95. Found: C, 30.68; H, 2.42; Br, 45.36; N, 5.84.

6-Methyl-2,3-diphenyl-6-[(tribromo-λ⁴-telluranyl)methyl]-5,6-dihydro-3*H*-[1,3]thiazolo[3,2-*b*][1,2,4]triazol-7-ium bromide (7d**)** This compound was obtained from **3a** and **6d**; yield 59% of an orange powder; mp 267°C (dec); ¹H NMR (400 MHz): δ 2.21 (s, 3H),

3.94 (d, $J = 11.1$ Hz, 1H), 4.14 (d, $J = 9.0$ Hz, 1H), 4.24 (d, $J = 11.1$ Hz, 1H), 4.91 (d, $J = 9.0$ Hz, 1H), 7.46–7.67 (m, 10H); ^{13}C NMR (100 MHz): δ 158.2, 157.6, 132.5, 123.3, 131.2, 129.8, 129.7, 126.7, 123.8, 67.8, 50.1, 40.0, 25.0; MS: m/z 354.0 (60), 326.0 (70), 309.8 (50), 308.0 (100), 253.8 (25), 238.2 (25). Anal. Calcd for $\text{C}_{18}\text{H}_{17}\text{Br}_4\text{N}_3\text{S}\text{Te}$: C, 28.65; H, 2.27; Br, 42.35; N, 5.57. Found: C, 28.69; H, 2.25; Br, 42.22; N, 5.61.

6-Bromo-2,3,7-triphenyl-3,5,6,7-tetrahydro[1,2,4]triazolo[5,1-*b*][1,3]thiazin-8-ium tribromide (8a) This compound was obtained by bromination of **3b**; yield 44% of a white powder; mp 181–183°C; ^1H NMR (300 MHz): δ 3.72 (d, $J = 12.0$ Hz, 1H), 3.87 (dd, $J = 13.6, 7.2$ Hz, 1H), 5.45 (m, 1H), 6.33 (d, $J = 4.0$ Hz, 1H), 7.37–7.81 (m, 15H); ^{13}C NMR (75 MHz): δ 153.7, 153.6, 136.1, 133.0, 131.6, 130.3, 129.8, 129.6, 128.4, 128.2, 123.1, 69.0, 45.2, 32.1. Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{Br}_4\text{N}_3\text{S}$: C, 40.09; H, 2.78; Br, 46.38; N, 6.10; S, 4.65. Found: C, 40.21; H, 2.84; Br, 46.02; N, 6.18; S, 4.70.

6-Bromo-2-(4-nitrophenyl)-3,7-diphenyl-3,5,6,7-tetrahydro[1,2,4]triazolo[5,1-*b*][1,3]thiazin-8-ium tribromide (8b) This compound was obtained by bromination of **3d**; yield 90% of a yellowish powder; mp 115–120°C; ^1H NMR (400 MHz): δ 3.80 (d, $J = 11.8$ Hz, 1H), 3.97 (dd, $J = 13.6, 7.1$ Hz, 1H), 5.50 (m, 1H), 6.39 (d, $J = 5.2$ Hz, 1H), 7.42–7.98 (m, 12H), 8.29 (d, $J = 8.8$ Hz, 2H); ^{13}C NMR (100 MHz, DMSO- d_6): δ 153.8, 151.6, 149.7, 135.4, 132.7, 131.2, 131.1, 130.9, 129.9, 129.3, 128.7, 128.3, 127.8, 124.4, 69.0, 45.0, 32.4. Anal. Calcd for $\text{C}_{23}\text{H}_{18}\text{Br}_4\text{N}_4\text{O}_2\text{S}$: C, 37.63; H, 2.47; Br, 43.54; N, 7.63; S, 4.37. Found: C, 37.72; H, 2.51; Br, 43.22; N, 7.69; S, 4.40.

6-Iodo-2,3,7-triphenyl-3,5,6,7-tetrahydro[1,2,4]triazolo[5,1-*b*][1,3]thiazin-8-ium triiodide (9a) This compound was obtained by iodination of **3b**; yield 57% of a brown powder; mp 190–192°C; ^1H NMR (300 MHz): δ 3.76 (d, $J = 14.9$ Hz, 1H), 3.92 (dd, $J = 13.5, 6.9$ Hz, 1H), 5.48 (m, 1H), 6.35 (d, $J = 5.2$ Hz, 1H), 7.33–7.91 (m, 15H). Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{I}_4\text{N}_3\text{S}$: C, 31.50; H, 2.18; I, 57.87; N, 4.79; S, 3.66. Found: C, 31.67; H, 2.21; I, 57.72; N, 4.84; S, 3.68.

6-Iodo-2-(4-nitrophenyl)-3,7-diphenyl-3,5,6,7-tetrahydro[1,2,4]triazolo[5,1-*b*][1,3]thiazin-8-ium triiodide (9b) This compound was obtained by iodination of **3d**; yield 67% of a brown powder; mp 132–134°C; ^1H NMR (300 MHz): δ 3.79–3.90 (m, 1H), 3.97 (dd, $J = 15.2, 7.0$ Hz, 1H), 5.25 (m, 1H), 6.23 (d, $J = 7.0$ Hz, 1H), 7.46–7.67 (m, 7H), 7.77 (s, 5H), 8.28 (d, $J = 8.9$ Hz, 2H). Anal. Calcd for $\text{C}_{23}\text{H}_{18}\text{I}_4\text{N}_4\text{O}_2\text{S}$: C, 29.96; H, 1.97; I, 55.05; N, 6.08; S, 3.48. Found: C, 30.16; H, 2.01; I, 54.90; N, 6.12; S, 3.44.

Molecular complex 10a This compound was obtained from **3b** and **6a**; yield 32% of colorless crystals; mp 220°C (dec); ^1H NMR (300 MHz): δ 3.98 (d, $J = 4.2$ Hz, 2H), 6.38 (m, 1H), 6.61 (d, $J = 9.3$ Hz, 1H), 7.18–7.53 (m, 15H). Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{Cl}_4\text{N}_3\text{S}\text{Se}$: C, 46.80; H, 3.24; Cl, 24.03; N, 7.12. Found: C, 47.06; H, 3.20; Cl, 24.34; N, 7.26.

Molecular complex 10b This compound was obtained from **3b** and **6b**; yield 45% (colorless crystals); mp 255°C (dec); ^1H NMR (300 MHz): δ 3.98 (d, $J = 4.2$ Hz, 2H), 6.38 (m, 1H), 6.60 (d, $J = 9.3$ Hz, 1H), 7.22–7.56 (m, 15H). Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{Cl}_4\text{N}_3\text{S}\text{Te}$: C, 43.24; H, 3.00; Cl, 22.20; N, 6.58. Found: C, 43.52; H, 3.08; Cl, 22.43; N, 6.50.

Molecular complex 10c This compound was obtained from **3b** and **6c**; yield 27% of a colorless crystals; mp 222°C (dec); ^1H NMR (300 MHz): δ 3.98 (d, $J = 4.2$ Hz, 2H), 6.36 (m, 1H), 6.61 (d, $J = 9.3$ Hz,

1H), 7.19–7.53 (m, 15H). Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{Br}_4\text{N}_3\text{S}\text{Se}$: C, 35.97; H, 2.49; Br, 41.61; N, 5.47. Found: C, 36.18; H, 2.45; Br, 41.94; N, 5.33.

Molecular complex 10d This compound was obtained from **3b** and **6d**; yield 39% of a colorless crystals; mp 267°C (dec); ^1H NMR (300 MHz): δ 3.98 (d, $J = 4.2$ Hz, 2H), 6.38 (m, 1H), 6.58 (d, $J = 9.3$ Hz, 1H), 7.26–7.58 (m, 15H). Anal. Calcd for $\text{C}_{23}\text{H}_{19}\text{Br}_4\text{N}_3\text{S}\text{Te}$: C, 33.82; H, 2.34; Br, 39.14; N, 5.15. Found: C, 33.98; H, 2.29; Br, 39.46; N, 5.11.

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