

Mohammad Piltan*

Preparation of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones using ZrO₂ nanoparticles as a catalyst under solvent-free conditions

<https://doi.org/10.1515/hc-2017-0142>

Received July 4, 2017; accepted August 10, 2017; previously published online September 23, 2017

Abstract: Zirconium oxide nanoparticles are an efficient catalyst for the preparation of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones via a three-component reaction of phthalhydrazide, aromatic aldehydes and malononitrile under solvent-free conditions.

Keywords: aromatic aldehydes; multicomponent reaction; nanoparticles; phthalhydrazide; solvent-free; zirconium oxide.

Introduction

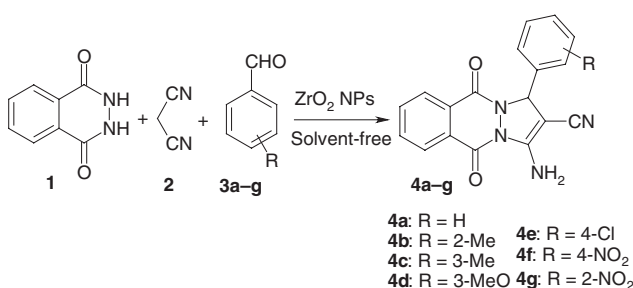
Multi-component reactions (MCRs) have become a powerful, fast and convenient approach for the synthesis of complex heterocyclic compounds from simple and readily available starting materials [1]. MCRs have advantages over classical reaction strategies including greater efficiency and shorter reaction time [2, 3]. Nanometer-sized particles have recently been considered as an alternative matrix for supporting catalytic reactions. Reusable nanoscale heterogeneous catalysts show high catalytic activity due to a high specific surface area and low-coordination sites [4].

1*H*-Pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives containing fused hydrazine are an important class of pharmaceutical compounds that exhibit a wide spectrum of biological activities [5–7]. Such compounds are relatively difficult to synthesize [8–17].

Therefore, the development of simple methods for the synthesis of pyrazolo[1,2-*b*]phthalazine-5,10-diones is important. In continuation of our efforts to develop a new methodology for the synthesis of heterocyclic compounds [18–21], we now report the use of ZrO₂ nanoparticles as an efficient catalyst for the preparation of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones **4a–g** from phthalhydrazide, aromatic aldehydes and malononitrile (Scheme 1).

Results and discussion

Initially, different reaction parameters for the synthesis of compound **4a** by the three-component condensation reaction using benzaldehyde as a model reaction were optimized. The scrutinized parameters were temperature, time of reaction solvents, including ethanol, tetrahydrofuran, acetonitrile, dichloromethane and water, and solvent-free conditions. The best results were obtained under solvent-free conditions. When 10 mol%, 15 mol% and 20 mol% of ZrO₂ nanoparticles were used, the yields of **4a** were 82%, 91% and 91%, respectively, and the yields were much lower in any solvent. Only 20% of **4a** was obtained in the absence of a catalyst. Under optimized solvent-free conditions, the reaction was conducted with 15 mol% of the catalyst at 100°C for 45 min and furnished **4a** with a 91% yield. The reactions with other aromatic aldehydes conducted using this optimized procedure were also similarly effective. All reactions were completed within 35–45 min and afforded good yields of products **4b–g**. The aromatic aldehydes substituted with either electron-donating or electron-withdrawing groups underwent the reaction smoothly and afforded the products in good yields. As seen from Table 1, the method compares well with other known preparations of the model product **4a**. The suggested mechanism for the formation of products **4** involves generation of the intermediate compounds **5–7**, as shown in Scheme 2.

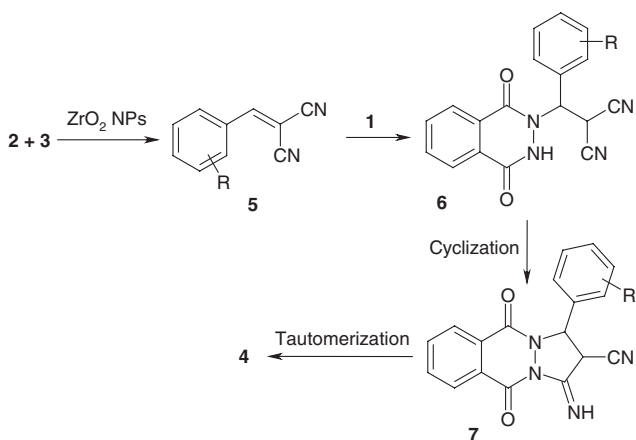


Scheme 1 Synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones **4a–g**.

*Corresponding author: Mohammad Piltan, Department of Chemistry, Faculty of Science, Islamic Azad University, Sanandaj Branch, Sanandaj 618, Iran, e-mail: mohammadpiltan@yahoo.com

Table 1 Comparison of the results of different preparations of **4a**.

Entry	Catalyst	Conditions	Time (min)	Yield (%) [Ref]
1	p-TSA (30 mol%)	[Bmim]Br/100°C	180	94 [8]
2	TBBDA/PBBS	EtOH, 100°C	15/40	89/65 [9]
3	Et ₃ N (20 mol%)	EtOH, 50°C, ultrasound	60	90 [10]
4	[Bmim]OH (20 mol%)	MW, 100 W, 45°C	4	96 [11]
5	CuI NPs	Solvent-free, 70°C	27	91 [12]
7	APTES-MNPs (10 mol%)	Solvent-free, rt	19	92 [14]
8	NiCl ₂ ·6H ₂ O	EtOH, reflux	180	87 [16]
9	AL-KIT-6	EtOH, 60°C	240	93 [17]
10	ZrO ₂ NPs	Solvent-free, 100°C	40	91 (present work)

**Scheme 2** A plausible mechanism for the formation of compounds **4**.

Conclusions

The reaction between phthalhydrazide, malononitrile and aromatic aldehydes in the presence of nano-ZrO₂ offers a simple one-pot access to pyrazolo[1,2-*a*][1,2,4]triazole-1,3-dione derivatives of potential synthetic and pharmacological interest. The simplicity, easy workup, as well as the safety and reusability of the catalyst are advantages of this method.

Experimental

Melting points were determined on an Electrothermal 9100 apparatus and are not corrected. Infrared (IR) spectra were recorded in KBr pellets on a Shimadzu 460 spectrometer. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded on a Bruker AVANCE 400 spectrometer using DMSO-*d*₆ as a solvent. Electron impact mass spectra were obtained at 70 eV on a Finnigan MAT 8430 mass spectrometer. Elemental analyses for C, H and N were obtained on a Heraeus CHNO Rapid analyzer. All commercially available chemicals and reagents were used without further purification. In all experiments, ZrO₂ (5–25 nm, Plasma Chem GmbH) was used.

General procedure for the preparation of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones **4a–g**

A mixture of phthalhydrazide (1 mmol), aromatic aldehyde (1 mmol), malononitrile (1 mmol) and ZrO₂ nanoparticles (15 mol%) was stirred at 100°C under solvent-free conditions for 35–45 min. After completion of the reaction, the resultant precipitate was crystallized from MeOH to afford pure product **4**.

3-Amino-1-phenyl-5,10-dihydro-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4a**)** Yellow powder; yield 0.29 g (91%); mp 276–278°C; IR: 3361, 3292, 3248, 2198, 1659, 1567 cm⁻¹; ¹H NMR: δ 6.12 (s, 1H, CH), 7.31–7.46 (m, 5H, ArH), 7.92–8.26 (m, 6H, ArH and NH₂); ¹³C NMR: δ 61.8, 62.8, 116.3, 126.6, 126.7, 127.1, 128.2, 128.5, 128.7, 129.1, 133.7, 134.6, 138.4, 149.8, 153.6, 156.8; EI-MS: *m/z* 316. Anal. Calcd for C₁₈H₁₂N₄O₂ (316.32): C 68.35, H 3.82, N 17.71. Found: C 68.28, H 3.75, N 17.78.

3-Amino-1-(2-methylphenyl)-5,10-dihydro-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4b**)** Yellow powder; yield 0.29 g (88%); mp 247–249°C; IR: 3383, 3300, 2207, 1659, 1603 cm⁻¹; ¹H NMR: δ 2.43 (3H, s, CH₃), 6.30 (1H, s, CH), 7.17–7.28 (4H, m, Ar), 7.96–8.25 (6H, m, Ar and NH₂); ¹³C NMR: δ 20.1, 61.2, 62.9, 122.0, 126.7, 127.2, 127.7, 128.1, 128.3, 128.5, 129.0, 130.3, 133.8, 134.8, 135.2, 136.5, 153.7, 154.2, 156.9; EI-MS: *m/z* 330. Anal. Calcd for C₁₉H₁₄N₄O₂ (330.35): C 69.09, H 4.27, N 16.95. Found: C 68.95, H 4.25, N 16.96.

3-Amino-1-(3-methylphenyl)-5,10-dihydro-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4c**)** Yellow powder; yield 0.29 g (89%); mp 250–252°C; IR: 3361, 3259, 2194, 1658, 1570 cm⁻¹; ¹H NMR: δ 2.25 (3H, s, CH₃), 6.08 (1H, s, CH), 7.11–7.22 (4H, m, Ar), 7.97–8.26 (6H, m, Ar and NH₂); ¹³C NMR: δ 21.0, 61.2, 62.8, 122.0, 122.3, 127.0, 127.4, 127.9, 128.4, 128.6, 128.9, 131.4, 134.2, 134.7, 135.8, 136.4, 154.1, 154.4, 157.3; EI-MS: *m/z* 330.

Anal. Calcd for C₁₉H₁₄N₄O₂ (330.35): C 69.09, H 4.27, N 16.95. Found: C 69.12, H 4.18, N 16.85.

3-Amino-1-(3-methoxyphenyl)-5,10-dihydro-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4d**)** Yellow powder; yield 0.30 g (87%); mp 247–249°C; IR: 3363, 3260, 2189, 1655, 1567 cm⁻¹; ¹H NMR: δ 3.76 (s, 3H, OCH₃); 6.15 (s, 1H), 6.90 (d, ³*J* = 8.0 Hz, 1H), 6.96–6.99 (m, 2H), 7.26 (t, ³*J* = 7.5 Hz, 1H), 7.65–7.99 (m, 2H), 8.09–8.12 (m, 3H, NH₂, CH), 8.25–8.28 (m, 1H); ¹³C NMR:

δ 55.7, 61.2, 62.6, 112.8, 114.5, 119.2, 124.3, 126.8, 127.2, 128.9, 129.1, 129.2, 129.6, 134.3, 137.6, 150.6, 153.2, 157.1, 160.0; EI-MS: m/z 346. Anal. Calcd for $C_{18}H_{11}N_5O_4$ (346.35): C 65.89, H 4.07, N 16.18. Found: C 65.78, H 4.15, N 16.24.

3-Amino-1-(4-chlorophenyl)-5,10-dihydro-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4e) Yellow powder; yield 0.33 g (93%); mp 273–275°C; IR: 3362, 3256, 2188, 1661, 1652 cm^{-1} ; 1H NMR: δ 6.17 (1H, s, CH), 7.43–7.54 (4H, m, Ar), 7.94–8.18 (6H, m, Ar and NH_2); ^{13}C NMR: δ 61.7, 62.6, 116.7, 126.8, 127.7, 128.2, 128.7, 129.3, 129.4, 133.3, 134.0, 134.7, 137.4, 150.2, 154.1, 157.3; EI-MS: m/z 350. Anal. Calcd for $C_{18}H_{11}N_5O_4Cl$ (350.77): C 61.64, H 3.16, N 15.97. Found: C 61.53, H 3.13, N 16.03.

3-Amino-1-(4-nitrophenyl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4f) Yellow powder; yield 0.35 g (96%); mp 226–228°C; IR: 3371, 3301, 3079, 2185, 1662, 1554 cm^{-1} ; 1H NMR: δ 6.30 (s, 1H); 7.81 (d, $^3J = 8.6$ Hz, 2H), 7.97–8.09 (m, 3H), 8.22 (brs, 2H, NH_2), 8.24 (d, $^3J = 8.6$ Hz, 2H), 8.26–8.29 (m, 1H); ^{13}C NMR δ 61.5, 62.9, 116.5, 126.1, 127.9, 128.1, 128.8, 129.1, 129.4, 134.5, 134.9, 146.4, 147.6, 150.8, 154.0, 157.2; EI-MS: m/z 361. Anal. Calcd for $C_{18}H_{11}N_5O_4$ (361.32): C 59.84, H 3.07, N 19.38. Found: C 60.13, H 3.13, N 19.23.

3-Amino-1-(2-nitrophenyl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitrile (4g) Yellow powder; yield 0.34 g (95%); mp 261–263°C; IR: 3380, 3165, 2192, 1694, 1655 cm^{-1} ; 1H NMR: δ 6.64 (1H, s, CH), 7.59–8.28 (10H, m, Ar and NH_2); ^{13}C NMR δ 61.1, 62.8, 106.5, 124.1, 126.8, 127.3, 128.1, 128.9, 129.2, 129.7, 133.4, 134.0, 134.6, 135.0, 148.4, 151.8, 154.2, 157.0; EI-MS: m/z 361. Anal. Calcd for $C_{18}H_{11}N_5O_4$ (361.32): C 59.84, H 3.07, N 19.38. Found: C 59.98, H 3.15, N 19.16.

Acknowledgments: We gratefully acknowledge financial support from the Research Council of Islamic Azad University, Sanandaj Branch.

References

- [1] Zhu, J.; Bienayme, H. *Multicomponent Reactions*; Wiley, Weinheim, Germany, 2005.
- [2] Weber, L. Multi-component reactions and evolutionary chemistry. *Drug. Discov. Today*. **2002**, *7*, 143–147.
- [3] Tejedor, D. F.; Tellado, G. Synthesis using aromatic homolytic substitution recent advances. *Chem. Soc. Rev.* **2007**, *36*, 484–491.
- [4] Pacchioni, G. Quantum chemistry of oxide surfaces: from chemisorptions to the identification of the structure and nature of point defects on MgO. *Surf. Rev. Lett.* **2000**, *7*, 277–306.
- [5] Vaughan, W. R. The chemistry of the phthalazines. *Chem. Rev.* **1948**, *43*, 447–508.
- [6] Prakash, O.; Kumar, R.; Parkash, V. Synthesis and antifungal activity of some new 3-hydroxy-2-(1-phenyl-3-aryl-4-pyrazolyl) chromones. *Eur. J. Med. Chem.* **2008**, *43*, 435–440.
- [7] Al'-Assar, F.; Zelenin, K. N.; Lesiovskaya, E. E.; Bezhan, I. P.; Chakchir, B. A. Synthesis and pharmacological activity of 1-hydroxy, 1-amino-, and 1-hydrazino-substituted 2,3-dihydro-1*H*-pyrazolo[1,2-*a*]pyridazine-5,8-diones and 2,3-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones. *J. Pharm. Chem.* **2002**, *36*, 598–603.
- [8] Ghahremanzadeh, R.; Shakibaei, G. I.; Bazgir, A. An efficient one-pot synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives. *Synlett*. **2008**, *8*, 1129–1132.
- [9] Ghorbani-Vaghei, R.; Noori, S.; Toghræi-Semiromi, Z.; Salimi, Z. One-pot synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives under solvent-free conditions. *RSC Adv.* **2014**, *4*, 47925–47928.
- [10] Nabid, M. R.; Rezaei, S. J. T.; Ghahremanzadeh, R.; Bazgir, A. Ultrasoundassisted one-pot, three-component synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones. *Ultrason. Sonochem.* **2010**, *17*, 159–161.
- [11] Raghuvanshi, D. S.; Singh, K. N. A highly efficient green synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives and their photophysical studies. *Tetrahedron Lett.* **2011**, *52*, 5702–5705.
- [12] Safaei-Ghomi, J.; Shahbazi-Alavi, H.; Ziarati, A.; Teymuri, R.; Saberi, M. R. A highly flexible green synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives with CuI nanoparticles as catalyst under solvent-free conditions. *Chin. Chem. Lett.* **2014**, *25*, 401–405.
- [13] Reddy, M. V.; Jeeong, Y. T. $SnCl_4$ -catalyzed green synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones under solvent-free conditions. *Tetrahedron Lett.* **2013**, *54*, 3546–3549.
- [14] Shaterian, H. R.; Mohammadnia, M. Mild preparation of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives with magnetic Fe_3O_4 nanoparticles coated by (3-aminopropyl)-triethoxysilane as catalyst under ambient and solvent-free conditions. *Res. Chem. Intermed.* **2014**, *40*, 371–383.
- [15] Shaterian, H. R.; Mohammadnia, M. Mild basic ionic liquids catalyzed new fourcomponent synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones. *J. Mol. Liq.* **2012**, *173*, 55–61.
- [16] Song, S. H.; Zhong, J.; He, Y. H.; Guan, Z. One-pot four-component synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives. *Tetrahedron Lett.* **2012**, *53*, 7075–7077.
- [17] Karthikeyan, G.; Pandurangan, A. Post synthesis alumination of KIT-6 materials with Ia3d symmetry and their catalytic efficiency towards multicomponent synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione carbonitriles and carboxylates. *J. Mol. Catal. A: Chem.* **2012**, *361*–362, 58–67.
- [18] Piltan, M.; Kalantari, S. Synthesis of new polysubstituted indolizines from pyridine, bromonitromethane, and dialkyl acetylenedicarboxylates. *J. Heterocycl. Chem.* **2017**, *54*, 301–304.
- [19] Piltan, M.; Kalantari, S. One-pot synthesis of regioselective polysubstituted indolizines via three-component reactions of dialkyl acetylenedicarboxylates, dialkylchloro malonates, and pyridines. *J. Heterocycl. Chem.* **2017**, *54*, 743–748.
- [20] Sharafi, Z.; Piltan, M. Efficient assembly of quinoxaline derivatives from benzene-1,2-diamines, dialkyl acetylenedicarboxylates and ninhydrin. *Heterocycl. Commun.* **2016**, *22*, 53–57.
- [21] Moradi, L.; Piltan, M.; Abasi, G. One-pot synthesis of novel pyrrolo[1,2-*a*]quinoxaline-4(5*H*)-ones using benzene-1,2-diamine, acetylenedicarboxylates, and β -nitrostyrene derivatives. *Helv. Chim. Acta.* **2014**, *97*, 646–651.