

Igor O. Akchurin, Anna I. Yakhutina, Andrei Y. Bochkov, Natalya P. Solovjova and Valerii F. Traven*

Synthesis of novel push-pull fluorescent dyes – 7-(diethylamino)furo[3,2-*c*]coumarin and 7-(diethylamino)thieno[3,2-*c*]coumarin derivatives

<https://doi.org/10.1515/hc-2017-0253>

Received December 7, 2017; accepted December 10, 2017; previously published online March 24, 2018

Abstract: Novel push-pull fluorescent dyes, 7-(diethylamino)furo[3,2-*c*]coumarin and 7-(diethylamino)thieno[3,2-*c*]coumarin derivatives, were designed and synthesized using formyl derivatives of furo- and thieno[3,2-*c*]coumarins as key intermediates. Electron absorption and emission spectra of the dyes were recorded in different solvents. The longest-wave bands in the electron absorption spectra of the dyes are suggested to be of push-pull nature.

Keywords: coumarin; electron absorption spectra; fluorescence; push-pull dyes.

Introduction

Push-pull chromophore systems contain strong electron acceptor(s) (A) and donor(s) (D) attached to a π -conjugated spacer [1]. Such organic compounds are widely used in biotechnology as labels and probes, medical diagnostics [2–4], analytical chemistry [5, 6], material science including optoelectronic [7] and optical data storage devices [8], functional polymers [9] and devices based on application of efficient nonlinear optics (NLO) fluorophores [10, 11]. Study of complex biochemical processes both *in vitro* and *in vivo*, particularly in a real-time manner, often requires the use of spectroscopic probes the fluorescent properties of which significantly change upon chemical reaction, non-covalent interaction or reversible binding with target analytes [12–15]. Various coumarin derivatives show high

fluorescence quantum yields and large Stokes shifts [16]. Coumarin dyes are used as laser dyes [17, 18], organic photosensitizers in dye-sensitized solar cells (DSSC) [19–21] and fluorescent labels [22] in biochemistry [23] or in medicinal chemistry [24].

The examples discussed above mainly deal with non-fused coumarins, while spectral properties of 3,4-annulated coumarins have not been extensively studied. It has been mentioned, however, that some derivatives of thieno[3,2-*c*]coumarin possess high fluorescent quantum yields (close to 100%) and show bathochromic shift of absorption and emission bands as compared with non-condensed analogs [25]. Thus, it can be suggested that annulation of coumarin with five-membered heterocycles such as furan and thiophene may lead to a bathochromic shift of absorption and emission bands and greater Stokes shifts compared to non-fused coumarin derivatives. Herein we report synthesis of new furo[3,2-*c*]coumarin and thieno[3,2-*c*]coumarin derivatives with large delocalized π -electron systems that include π -conjugated donor and acceptor fragments forming together an asymmetrical push-pull fluorophore (Figure 1). More specifically, the electron-releasing diethylamino substituent is located at position 7 of coumarin fragment and different electron-withdrawing groups (EWG) are present at the furan or thiophene ring that is fused to the 3 and 4 coumarin atoms.

Results and discussion

Synthesis

The target push-pull dyes **1a–g** and **2a,b** were prepared as depicted in Schemes 1–3. For the synthesis of dyes **1a–g**, the key starting material, 7-(diethylamino)-4-hydroxycoumarin (**4** in Scheme 1) was prepared by Pechmann reaction [26] of diphenyl malonate **3** with 3-diethylaminophenol. The malonate **3** was prepared in a higher yield of 68% by modification of the earlier reported protocols

*Corresponding author: Valerii F. Traven, D. Mendeleev University of Chemical Technology of Russia, Higher Chemical College of RAS, Miusskaya Square 9, Moscow 125047, Russian Federation, e-mail: valerii.traven@gmail.com

Igor O. Akchurin, Anna I. Yakhutina, Andrei Y. Bochkov and Natalya P. Solovjova: D. Mendeleev University of Chemical Technology of Russia, Higher Chemical College of RAS, Miusskaya Square 9, Moscow 125047, Russian Federation

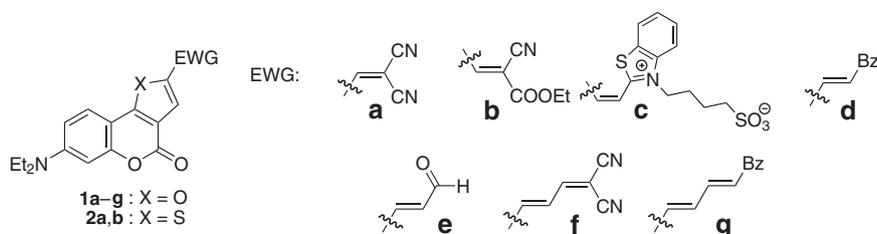
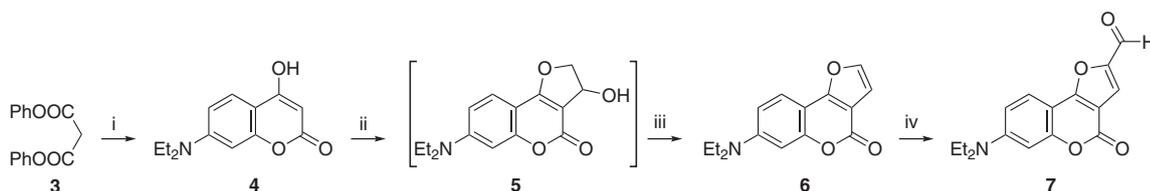


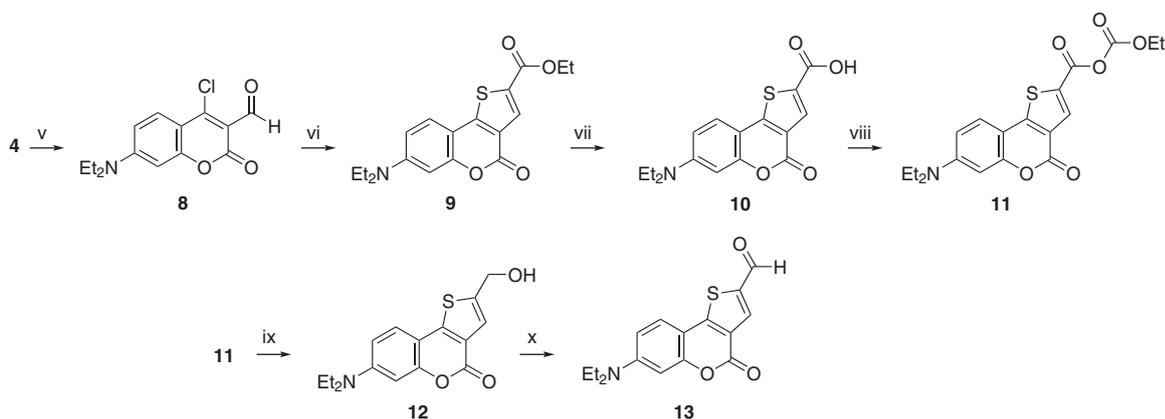
Figure 1 Push-pull fluorescent furocoumarin and thienocoumarin dyes **1a-g** and **2a,b**.

[27–29]. There are numerous methods of furocoumarins synthesis based on cross-coupling reactions of 3-substituted coumarins [30–32]. In this work, 7-(diethylamino)-furo[3,2-c]coumarin (**6**) was prepared following the earlier reported pathway (Scheme 1) [33]. Two-step reaction of the starting coumarin **4** with 2-chloroacetaldehyde in the presence of a potassium carbonate was conducted in one-pot manner without isolation of the intermediate product **5**. However, the diethylamino-substituted compound **6** was obtained with lower yield (40%) than the parent unsubstituted furo[3,2-c]coumarin reported earlier [34]. Vilsmeier-Haack formylation of compound **6** afforded product **7** in 45% yield.

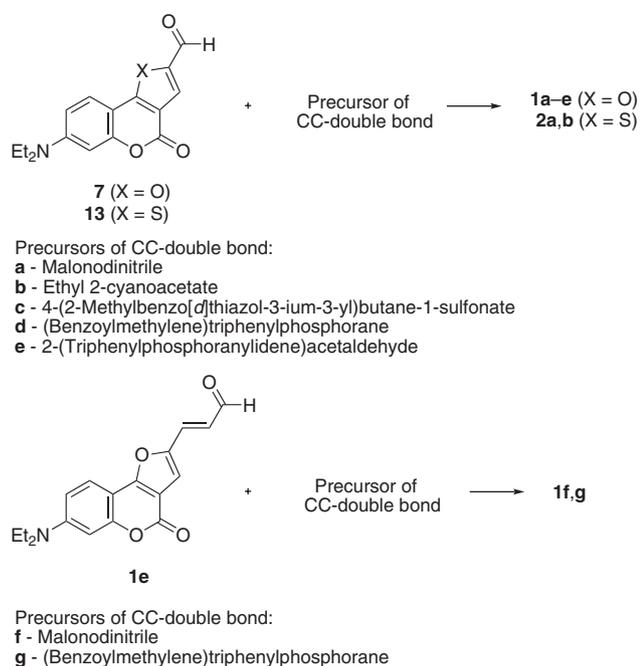
The synthetic route to push-pull dyes **2a,b** also started with substrate **4** but followed a different approach (Scheme 2). First, compound **4** was subjected to Vilsmeier-Haack formylation to give 4-chloro-3-formyl-7-diethylaminocoumarin (**8**) [29]. Then, reaction of compound **8** with ethyl 2-mercaptoacetate in dichloromethane in the presence of diisopropylethylamine (DIPEA) for 7 h afforded the thieno[3,2-c]coumarin **9** [34]. Since attempted reduction of ester **9** directly to the desired alcohol **12** using sodium borohydride failed, another methodology was applied. The successful approach to **12** involved hydrolysis of ester **9** followed by synthesis of a mixed anhydride **11** from the resultant acid **10** and the final reduction of compound **11**



Scheme 1 Reagents and conditions: (i) 3-diethylaminophenol, toluene, reflux, 8 h, yield 56%; (ii) 2-chloroacetaldehyde, aqueous K_2CO_3 , 0°C ; (iii) 40% HCl and (iv) DMF, POCl_3 , 70°C , yield 45%.



Scheme 2 Reagents and conditions: (v) DMF, POCl_3 , room temperature, yield 97%; (vi) ethyl 2-mercaptoacetate, DIPEA, dichloromethane, room temperature, yield 90%; (vii) (1) NaOH, $\text{EtOH}/\text{H}_2\text{O}$, reflux; (2) HCl, 91%; (viii) ethyl chloroformate, NEt_3 , dichloromethane, room temperature, yield 94%; (ix) NaBH_4 , THF, MeOH, yield 60% and (x) DMP, dichloromethane, room temperature, yield 75%.



Scheme 3 Synthesis of push-pull fluorescent furocoumarin and thienocoumarin dyes **1a–g** and **2a,b**.

with sodium borohydride. Oxidation of the alcohol **12** by treatment with Dess-Martin reagent furnished aldehyde **13** in high yield and virtually no side products.

The furo[3,2-*c*]coumarin **7** (Scheme 1) and thieno[3,2-*c*]coumarin **13** (Scheme 2), obtained as discussed above, were the key precursors to the final products **1a–g** and **2a,b** (Scheme 3). Compounds **1a–c,f** and **2a,b** were prepared by the Knoevenagel condensation of **7** or **13** with various methylene compounds as CH-acids. Conversion of aldehyde **7** to conjugated carbonyl derivatives **1d,e** was carried out using Wittig reaction. Compound **1g** was also obtained by treatment of **1e** with a Wittig reagent.

Electron absorption and fluorescence spectra

Electron absorption spectra of the dyes in different solvents show an intense lowest energy charge-transfer (CT) absorption band in the near-UV-visible region. As expected, the location of this band is strongly influenced by the nature of electron-withdrawing substituent EWG attached to the 5-membered heterocycle [35–39]. Since the same electron-releasing group (diethylamino) is present in the coumarin core of all compounds, variation of electron-withdrawing groups at the five-membered ring is of interest for studying the absorption and emission spectra. For example, the absorption maximum of aldehyde **7** ($\lambda_{\text{abs}} = 401$ nm) is shifted bathochromically by

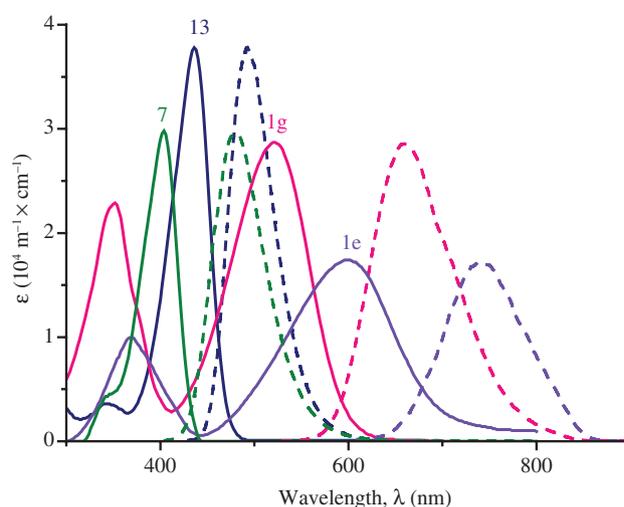


Figure 2 UV-vis absorption (solid lines) and fluorescence (dotted lines) spectra of new push-pull dyes in dichloromethane. Compound numbers are shown.

99 nm for α,β -unsaturated malonitrile **1a** ($\lambda_{\text{abs}} = 500$ nm) and further by 199 nm for α,β -unsaturated 2-benzothiazolium derivative **2c** ($\lambda_{\text{abs}} = 600$ nm). Fluorescence follows a similar pattern. Additional examples are shown in Figure 2. We are currently studying the molecular basis for these phenomena. It can be suggested that they are of push-pull nature.

Conclusions

Novel push-pull fluorescent dyes, 7-(diethylamino)-furo[3,2-*c*]coumarins and 7-(diethylamino)thieno[3,2-*c*]coumarins, were synthesized using the corresponding formyl derivatives as starting materials. Effects of the structures on the spectral characteristics of the dyes are noted. The longest-wave transitions in electron absorption spectra of the new dyes are suggested to be of push-pull nature.

Experimental

Diphenyl malonate (**3**), 7-(diethylamino)-4-hydroxycoumarin (**4**) and 4-chloro-7-(diethylamino)-3-formylcoumarin (**8**) were prepared according to procedures reported by us previously [26]. All commercial reagents were used as received. Anhydrous toluene and dichloromethane were obtained by distillation over P_2O_5 . Dry DMF was prepared by distillation over P_2O_5 under reduced pressure. Solvents used for UV-vis and fluorescence spectroscopy experiments were of spectral grade. Column chromatography was carried out using Macherey-Nagel Kieselgel 60 H silica gel. Analytical thin layer

chromatography was carried out using aluminum-backed plates coated with Macherey-Nagel Alugram SILG/UV₂₅₄; the compounds were visualized under UV light at 254 or 365 nm. Melting points were measured on a Stuart melting point apparatus SMP30 and are uncorrected. The ¹H NMR (500 MHz) and ¹³C NMR (125 MHz) spectra were recorded on a Bruker spectrometer in solvents as indicated. The ESI-HR-MS data were obtained on a Bruker Daltonics MicroTof-Q II instrument operating in positive ionization mode. UV-vis absorption spectra were recorded using an SF-104 spectrophotometer (Interphotophysics LLC, Moscow, Russia).

7-(Diethylamino)-4*H*-furo[3,2-*c*]chromen-4-one (6)

Chloroacetaldehyde (10 mmol, 1.27 mL in 20 mL H₂O) was added dropwise during 30 min to a stirred and cooled (0°C) solution of 7-(diethylamino)-4-hydroxycoumarin (**4**, 2.33 g, 10 mmol) in aqueous K₂CO₃ (2.76 g, 20 mmol, in 40 mL H₂O). The mixture was stirred at room temperature for 30 min, then treated with concentrated hydrochloric acid (4 mL) and stirred at room temperature for an additional 2 h. The mixture was neutralized with K₂CO₃ and extracted with dichloromethane. The extract was concentrated and the residue was chromatographed on silica gel eluting with dichloromethane to give yellow-pink crystals; yield 40%; mp 117–118°C; ¹H NMR (CDCl₃): δ 1.23 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 3.43 (q, 4H, q, *J* = 7.0 Hz, 2 × CH₂), 6.64–6.69 (m, 2H, C(6)H, C(8)H), 6.91 (d, *J* = 2.1 Hz, 1H, CH-furo), 7.48 (d, *J* = 2.1 Hz, 1H, CH-furo), 7.64 (d, *J* = 8.6 Hz, 1H, C(5)H); ¹³C NMR (CDCl₃): δ 11.9, 44.4, 97.8, 101.0, 105.6, 107.7, 108.6, 121.3, 142.3, 149.3, 154.7, 158.7, 158.8. HR-MS (ESI-TOF). Calcd for C₁₅H₁₅NO₃ [M + H]⁺: *m/z* 258.1130. Found: *m/z* 258.1129.

7-(Diethylamino)-4-oxo-4*H*-furo[3,2-*c*]chromene-2-carbaldehyde (7)

POCl₃ (0.75 mL, 8 mmol) was added to a solution of 7-(diethylamino)furo[3,2-*c*]coumarin (**6**, 1.03 g, 4 mmol) in dry DMF (10 mL). The mixture was stirred at 80°C, analyzed by TLC and then poured onto aqueous NH₄OAc after completion of the reaction. The resulting precipitate was purified by silica gel chromatography using petroleum ether/EtOAc (2:1) as eluent to give yellow crystals; yield 45%; mp 143–145°C; ¹H NMR (CDCl₃): δ 1.26 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 3.47 (4H, q, *J* = 7.0 Hz, 2 × CH₂), 6.63 (1H, d, *J* = 2.4 Hz, C(8)H), 6.72 (1H, dd, *J* = 2.4 Hz, *J* = 8.9 Hz, C(6)H), 7.68 (1H, s, CH-furo), 7.80 (1H, d, *J* = 8.9 Hz, C(5)H), 9.70 (1H, s, CHO); ¹³C NMR (CDCl₃): δ 11.8, 44.5, 97.6, 99.2, 106.7, 109.1, 120.1, 122.9, 151.1, 151.3, 156.3, 157.4, 161.2, 176.1. HR-MS (ESI-TOF). Calcd for C₁₆H₁₅NO₄ [M + H]⁺: *m/z* 286.1079. Found: *m/z* 286.1088.

General method for synthesis of compounds 1a–c

A solution of a C-H acid (1 equiv.) and a crystal of EDDA in ethanol was added dropwise to a hot solution of 7-(diethylamino)-4-oxo-4*H*-furo[3,2-*c*]chromene-2-carbaldehyde (**7**, 0.75–1 mmol) in ethanol (5 mL) and the mixture was stirred at room temperature for 2 h. The resultant precipitate was crystallized from ethanol.

2-((7-(Diethylamino)-4-oxo-4*H*-furo[3,2-*c*]chromen-2-yl)methylene)malononitrile (1a) This compound was obtained from **7** (285 mg, 1 mmol) and malonodinitrile (66 mg, 1 mmol) as a red powder; yield 70%; mp 208–210°C; ¹H NMR (CDCl₃): δ 1.28 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 3.49 (4H, q, *J* = 7.0 Hz, 2 × CH₂), 6.61 (1H, d, *J* = 2.4 Hz, C(8)H), 6.73 (1H, dd, *J* = 2.4 Hz, *J* = 9.2 Hz, C(6)H), 7.46 (1H, s, CH-furo), 7.60 (1H, s, CH=), 7.74 (1H, d, *J* = 9.2 Hz, C(5)H); ¹³C NMR (CDCl₃): δ 11.9, 44.7, 92.0, 97.5, 98.6, 108.0, 109.5, 112.4, 113.5, 121.7, 123.5, 140.0, 146.8, 151.7, 156.5, 156.6, 162.2. HR-MS (ESI-TOF). Calcd for C₁₉H₁₅N₃O₃ [M + H]⁺: *m/z* 334.1192. Found: *m/z* 334.1207.

Ethyl 2-cyano-3-(7-(diethylamino)-4-oxo-4*H*-furo[3,2-*c*]chromen-2-yl)acrylate (1b) This compound was obtained from **7** (285 mg, 1 mmol) and ethyl 2-cyanoacetate (113 mg, 1 mmol) as an orange powder; yield 63%; mp 180–181°C; ¹H NMR (CDCl₃): δ 1.27 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 1.43 (3H, t, *J* = 7.0 Hz, CH₃), 3.48 (4H, q, *J* = 7.0 Hz, 2 × CH₂), 4.41 (2H, q, *J* = 7.0 Hz, CH₂), 6.62 (1H, d, *J* = 2.4 Hz, C(8)H), 6.73 (1H, dd, *J* = 2.4 Hz, *J* = 9.2 Hz, C(6)H), 7.61 (1H, s, CH-furo), 7.80 (1H, d, *J* = 9.2 Hz, C(5)H), 7.99 (1H, s, CH=); ¹³C NMR (CDCl₃): δ 11.8, 13.7, 44.6, 62.2, 97.6, 98.8, 99.2, 107.9, 109.3, 114.7, 119.8, 123.3, 136.8, 147.6, 151.2, 156.2, 157.0, 161.6, 162.0. HR-MS (ESI-TOF). Calcd for C₂₁H₂₀N₂O₅ [M + H]⁺: *m/z* 381.1450. Found: *m/z* 381.1470.

4-(2-(7-(Diethylamino)-4-oxo-4*H*-furo[3,2-*c*]chromen-2-yl)-vinyl)benzo[d]thiazol-3-ium-3-yl)butane-1-sulfonate (1c) This compound was obtained from **7** (214 mg, 0.75 mmol) and 4-(2-methylbenzo[d]thiazol-3-ium-3-yl)butane-1-sulfonate (203 mg, 0.75 mmol) in the presence of a crystal of EDDA in hot acetic acid (5 mL). The reaction mixture was heated under reflux for 8 h. The resulting precipitate was filtered off and purified by reverse-phase column chromatography using water/acetonitrile (10:1) as eluent to give dark-purple crystals; yield 69%; mp 237–239°C; ¹H NMR (CDCl₃/DMSO-*d*₆): δ 1.21 (6H, t, *J* = 7.3 Hz, 2 × CH₃), 2.03 (2H, m, CH₂), 2.18 (2H, m, CH₂), 2.83 (2H, t, *J* = 6.7 Hz, CH₂), 3.47 (4H, q, *J* = 7.3 Hz, 2 × CH₂), 4.97 (2H, t, *J* = 4.2 Hz, CH₂), 6.59 (1H, d, *J* = 2.4 Hz, C(8)H), 6.79 (1H, dd, *J* = 2.4 Hz, *J* = 8.2 Hz, C(6)H), 7.71 (1H), 7.78 (1H) (t, *J* = 8.9 Hz, *m,m'*-H(Ph)), 7.73 (1H, s, CH-furo), 8.04 (2H, s, CH=), 8.17 (1H, d, *J* = 8.2 Hz, C(5)H), 8.27 (2H, d, *J* = 8.9 Hz, *o,o'*-H(Ph)). HR-MS (ESI-TOF). Calcd for C₂₈H₂₈N₂O₆S₂ [M + H]⁺: *m/z* 553.1422. Found: *m/z* 553.1424.

General procedure for Wittig reaction of 7

A solution of aldehyde **7** (0.5–2 mmol) and a phosphorane reagent in dichloromethane was stirred at room temperature for 24 h under TLC control of the reaction. After removal of the solvent under reduced pressure, the oily residue was treated with ethanol (40 mL) and the resultant precipitate was crystallized from ethanol.

7-(Diethylamino)-2-(3-oxo-3-phenylprop-1-en-1-yl)-4*H*-furo[3,2-*c*]chromen-4-one (1d) This compound was obtained from **7** (143 mg, 0.5 mmol) and (benzoylmethylene)triphenylphosphorane (209 mg, 0.55 mmol) as yellow crystals; yield 77%; mp 175–177°C; ¹H NMR (CDCl₃): δ 1.26 (t, *J* = 7.0 Hz, 6H, 2 × CH₃), 3.46 (q, *J* = 7.0 Hz, 4H, 2 × CH₂), 6.62 (d, *J* = 2.1 Hz, 1H, C(8)H), 6.70 (dd, *J* = 2.1 Hz, *J* = 8.9 Hz, 1H, C(6)H), 7.15 (s, 1H, CH-furo), 7.53–7.65 (m, 5H, 3H_{ar} + H_α + H_β), 7.76 (d, *J* = 8.9 Hz, 1H, C(5)H), 8.08 (d, *J* = 7.0 Hz, 2H, 2H_{ar}); ¹³C NMR (CDCl₃): δ 11.9, 44.4, 97.6, 99.8, 107.6, 108.7, 113.3, 120.0, 122.0, 128.0 (2C), 128.2 (2C), 129.0, 132.4, 137.5, 150.3, 151.2, 155.6, 157.8, 160.0,

188.9. HR-MS (ESI-TOF). Calcd for $C_{24}H_{21}NO_4$ $[M+H]^+$: m/z 388.1549. Found: m/z 388.1536.

(E)-3-(7-(Diethylamino)-4-oxo-4H-furo[3,2-*c*]chromen-2-yl)-acrylaldehyde (1e) This compound was obtained from **7** (570 mg, 2 mmol) and 2-(triphenylphosphoranylidene)acetaldehyde (608 mg, 2 mmol) as orange crystals; yield 71%; mp 198–199°C; 1H NMR ($CDCl_3$): δ 1.25 (t, $J=7.0$ Hz, 6H, $2 \times CH_3$), 3.46 (q, $J=7.0$ Hz, 4H, $2 \times CH_2$), 6.62 (d, $J=2.1$ Hz, 1H, C(8)H), 6.71–6.76 (m, 2H, d, $J=2.1$ Hz, 1H, C(6)H + C $_{\beta}$ H), 7.19 (s, 1H, CH-furo), 7.27 (d, $J=15.6$ Hz, 1H, C $_{\alpha}$ H), 7.68 (d, $J=8.9$ Hz, 1H, C(5)H), 9.70 (d, $J=7.9$, 1H, CHO); ^{13}C NMR ($CDCl_3$): δ 11.9, 44.5, 97.7, 99.7, 107.7, 109.0, 114.0, 122.1, 126.4, 136.5, 150.0, 150.4, 155.6, 157.6, 160.4, 191.8. HR-MS (ESI-TOF). Calcd for $C_{18}H_{17}NO_4$ $[M+H]^+$: m/z 312.1236. Found: m/z 312.1251.

(E)-2-(3-(7-(Diethylamino)-4-oxo-4H-furo[3,2-*c*]chromen-2-yl)-allylidene)malononitrile (1f) A solution of malonodinitrile and a crystal of EDDA (0.5 mmol, 33 mg) in ethanol was added to a solution of (E)-3-(7-(diethylamino)-4-oxo-4H-furo[3,2-*c*]chromen-2-yl)-acrylaldehyde (**1e**, 156 mg, 0.5 mmol) in hot ethanol/chloroform (2:1, 6 mL). The mixture was stirred at room temperature for 2 h and the precipitated product was crystallized from toluene; purple powder; yield 46%; mp 249.5–250.5°C; 1H NMR ($CDCl_3$): δ 1.27 (6H, t, $J=6.9$ Hz, $2 \times CH_3$), 3.48 (4H, q, $J=6.9$ Hz, $2 \times CH_2$), 6.61 (1H, d, $J=2.1$ Hz, C(8)H), 6.72 (1H, dd, $J=2.1$ Hz, $J=8.9$ Hz, C(6)H), 7.05 (1H, d, $J=14.7$ Hz, CH=), 7.18–7.27 (2H, m, CH-furo, CH=), 7.55 (1H, d, $J=11.9$ Hz, CH=), 7.77 (1H, d, $J=8.9$ Hz, C(5)H); ^{13}C NMR ($CDCl_3$): δ 11.8, 44.9, 81.7, 98.2, 99.8, 108.3, 109.5, 111.5, 113.2, 116.3, 120.5, 122.6, 132.5, 150.1, 150.5, 155.8, 157.2, 157.7, 161.0. HR-MS (ESI-TOF). Calcd for $C_{21}H_{17}N_3O_3$ $[M+H]^+$: m/z 360.1348. Found: m/z 360.1309.

7-(Diethylamino)-2-((1E,3E)-5-oxo-5-phenylpenta-1,3-dien-1-yl)-4H-furo[3,2-*c*]chromen-4-one (1g) (Benzoylmethylene)triphenylphosphorane (209 mg, 0.55 mmol) was added to a solution of aldehyde **1e** (0.5 mmol, 156 mg) in $CHCl_3$ (4 mL). The reaction mixture was stirred at 70°C for 48 h and occasionally analyzed by TLC. The solvent was removed under reduced pressure and an oily residue was treated with ethanol (10 mL) to give a precipitate that was crystallized from ethanol; orange crystals; yield 20%; mp 197.5–199°C; 1H NMR ($CDCl_3$): δ 1.23 (t, $J=7.0$ Hz, 6H, $2 \times CH_3$), 3.43 (q, $J=7.0$ Hz, 4H, $2 \times CH_2$), 6.58 (d, $J=2.4$ Hz, 1H, C(8)H), 6.67 (dd, $J=2.4$ Hz, $J=8.9$ Hz, 1H, C(6)H), 6.80 (d, $J=15.3$ Hz, 1H), 7.15 (d, $J=15.3$ Hz, 1H, (C $_{\alpha}$ H, C $_{\beta}$ H)), 6.89 (s, 1H, CH-furo), 7.04–7.09 (dd, $J=15.3$, $J=12.0$ Hz, 1H, C $_{\beta}$ H), 7.47–7.58 (m, 4H, (3CH $_{ar}$ + C $_{\gamma}$ H)), 7.67 (d, $J=8.9$ Hz, 1H, C(5)H), 7.93–8.01 (m, 2H, CH $_{ar}$); ^{13}C NMR ($CDCl_3$): δ 11.9, 44.3, 97.7, 100.1, 107.5, 108.6, 109.1, 121.7, 125.7, 126.2, 126.4, 127.9 (2C), 128.1 (2C), 132.2, 137.7, 143.0, 150.0, 152.0, 155.3, 158.0, 159.2, 189.6. HR-MS (ESI-TOF). Calcd for $C_{26}H_{23}NO_4$ $[M+H]^+$: m/z 414.1705. Found: m/z 414.1650.

Ethyl 7-(diethylamino)-4-oxo-4H-thieno[3,2-*c*]chromene-2-carboxylate (9)

DIPEA (1.1 mL, 10 mmol) and ethyl 2-mercaptoacetate (2.8 g, 10 mmol) were added to a solution of 4-chloro-7-(diethylamino)-3-formylcoumarin (**8**, 2.8 g, 10 mmol) in dichloromethane (15 mL). The mixture was stirred at room temperature for 7 h and then treated with ethanol (20 mL). The precipitated product was crystallized from ethanol; yellow crystals; yield 90%; mp 160–162°C; 1H NMR ($CDCl_3$): δ 1.25 (t, $J=7.0$ Hz, 6H, $2 \times CH_3$),

1.42 (t, $J=7.2$ Hz, 3H, CH_3), 3.45 (q, $J=7.0$ Hz, 4H, $2 \times CH_2$), 4.40 (q, $J=7.2$ Hz, 2H, CH_2), 6.60 (d, $J=2.1$ Hz, 1H, C(8)H), 6.68 (dd, $J=2.1$ Hz, $J=8.9$ Hz, 1H, C(6)H), 7.50 (d, $J=8.9$ Hz, 1H, C(5)H), 8.19 (s, 1H, CH-furo); ^{13}C NMR ($CDCl_3$): δ 11.8, 13.7, 44.7, 61.1, 98.0, 105.0, 109.2, 120.8, 124.4, 129.9, 132.1, 149.8, 153.5, 153.9, 156.9, 161.1. HR-MS (ESI-TOF). Calcd for $C_{18}H_{19}NO_4S$ $[M+H]^+$: m/z 346.1113. Found: m/z 346.1113.

7-(Diethylamino)-4-oxo-4H-thieno[3,2-*c*]chromene-2-carboxylic acid (10)

Ethyl 7-(diethylamino)-4-oxo-4H-thieno[3,2-*c*]chromene-2-carboxylate (**9**) was added to a solution of sodium hydroxide (5 g, 125 mmol) in ethanol/water mixture (1:1). The resulting suspension was heated under reflux for 2 h and then vigorously stirred at room temperature for 12 h. The mixture was quenched with concentrated hydrochloric acid and the resultant precipitate was crystallized from ethanol; yellow crystals; yield 91%; mp 259–261°C; 1H NMR ($CDCl_3$): δ 1.14 (t, $J=7.0$ Hz, 6H, $2 \times CH_3$), 3.37 (q, $J=7.0$ Hz, 4H, $2 \times CH_2$), 6.53 (br s, 1H, C(8)H), 6.65 (br d, $J=7.0$ Hz, 1H, C(6)H), 7.43 (d, $J=8.9$ Hz, 1H, C(5)H), 7.98 (s, 1H, CH-thieno); ^{13}C NMR ($CDCl_3$): δ 11.7, 13.5, 44.8, 98.0, 105.0, 109.5, 120.7, 124.5, 131.3, 131.5, 149.4, 153.3, 153.6, 156.8, 162.5. HR-MS (ESI-TOF). Calcd for $C_{16}H_{15}NO_4S$ $[M+H]^+$: m/z 318.0800. Found: m/z 318.0846.

(Ethyl carbonic) 7-(diethylamino)-4-oxo-4H-thieno[3,2-*c*]chromene-2-carboxylic anhydride (11)

Triethylamine (1.5 mL, 10.9 mmol) was added to a suspension of 7-(diethylamino)-4-oxo-4H-thieno[3,2-*c*]chromene-2-carboxylic acid (**10**, 2.8 g, 8.8 mmol) in dichloromethane (40 mL). The mixture was then vigorously stirred at room temperature for 30 min, treated with ethyl chloroformate (1 mL, 10.6 mmol) and stirred at room temperature. After the reaction was completed as monitored by TLC analysis, water (200 mL) was added, and the organic layer was separated and concentrated under reduced pressure to give dark-yellow powder, yield 94%.

7-(Diethylamino)-2-(hydroxymethyl)-4H-thieno[3,2-*c*]chromen-4-one (12)

MeOH (5 mL) was added to a suspension of the anhydride **11** (3.2 g, 8.2 mmol) in THF (55 mL) and mixture was vigorously stirred at room temperature for 30 min. Sodium borohydride (0.62 g, 16.4 mmol) was added and the mixture was further stirred at room temperature until gas evolution ceased. The resultant solution was washed with water (2×200 mL). The organic layer was separated, concentrated under reduced pressure, and the residue was subjected to silica gel chromatography using petroleum ether/EtOAc (2:1) as eluent; yellow crystals; yield 60%; mp 121–123°C; 1H NMR ($CDCl_3$): δ 1.25 (t, $J=7.0$ Hz, 6H, $2 \times CH_3$), 3.45 (q, $J=7.0$ Hz, 4H, $2 \times CH_2$), 4.89 (s, 2H, CH_2), 6.65–6.90 (m, 2H, C(8)H + C(6)H), 6.78 (s, 1H), 7.41 (s, 1H, CH-thieno), 7.48 (d, $J=8.5$ Hz, 1H, C(5)H), 14.50 (br s, 1H, OH). HR-MS (ESI-TOF). Calcd for $C_{16}H_{17}NO_3S$ $[M+H]^+$: m/z 304.1007. Found: m/z 304.1049.

7-(Diethylamino)-4-oxo-4*H*-thieno[3,2-*c*]chromene-2-carbaldehyde (**13**)

Dess-Martin periodinane in dichloromethane (15%, 5.4 mL, 2.23 mmol) was added to a solution of 7-(diethylamino)-2-(hydroxymethyl)-4*H*-thieno[3,2-*c*]chromen-4-one (**12**, 676 mg, 2.23 mmol) in dichloromethane (10 mL), and the mixture was stirred at room temperature. After TLC analysis confirmed the absence of substrate **12**, the organic solution was washed with aqueous NaHCO₃ (2 × 200 mL) and concentrated under reduced pressure. The residue was subjected to chromatography on silica gel using petroleum ether/EtOAc (2:1) as eluent; yellow crystals; yield 75%; mp 198–199°C; ¹H NMR (CDCl₃): δ 1.26 (6H, t, *J* = 7.2 Hz, 2 × CH₃), 3.47 (4H, q, *J* = 7.2 Hz, 2 × CH₂), 6.59 (1H, d, *J* = 2.1 Hz, C(8)H), 6.69 (1H, dd, *J* = 2.1 Hz, *J* = 8.9 Hz, C(6)H), 7.55 (1H, d, *J* = 8.9 Hz, C(5)H), 8.18 (1H, s, CH-thieno), 9.94 (1H, s, CHO); ¹³C NMR (CDCl₃): δ 11.9, 44.6, 97.6, 104.6, 109.2, 120.8, 125.0, 136.0, 139.2, 150.6, 154.2, 155.4, 156.9, 181.8. HR-MS (ESI-TOF). Calcd for C₁₆H₁₅N₃O₃S [M + H]⁺: *m/z* 302.0851. Found: *m/z* 302.0901.

General method for synthesis of compounds **2a,b**

Solution of a C-H acid (1 equiv.) and a crystal of EDDA in a solvent indicated below was added dropwise to a solution of 7-(diethylamino)-4-oxo-4*H*-thieno[3,2-*c*]chromene-2-carbaldehyde (**13**, 0.23–0.5 mmol) in hot ethanol (3 mL). The mixture was stirred at room temperature for 2 h and the resultant precipitate was crystallized from ethanol.

2-((7-(Diethylamino)-4-oxo-4*H*-thieno[3,2-*c*]chromen-2-yl)-methylene) malononitrile (2a**)** This compound was obtained from **13** (150 mg, 0.5 mmol) and malonodinitrile (33 mg, 0.5 mmol) in dichloromethane (10 mL) as bright-red crystals; yield 36%; mp 234–236°C; ¹H NMR (CDCl₃): δ 1.28 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 3.49 (4H, q, *J* = 7.0 Hz, 2 × CH₂), 6.49 (1H, d, *J* = 2.3 Hz, C(8)H), 6.71 (1H, dd, *J* = 2.3 Hz, *J* = 9.0 Hz, C(6)H), 7.56 (1H, d, *J* = 9.0 Hz, C(5)H), 7.86 (1H, s, CH=), 8.05 (1H, s, CH-thieno); ¹³C NMR (CDCl₃): δ 11.9, 44.7, 77.0, 97.6, 104.1, 109.6, 112.6, 113.3, 120.5, 125.5, 130.6, 139.4, 149.6, 151.3, 154.7, 156.0, 156.2. HR-MS (ESI-TOF). Calcd for C₁₉H₁₅N₃O₂S [M + H]⁺: *m/z* 350.0963. Found: *m/z* 350.0998.

Ethyl 2-cyano-3-(7-(diethylamino)-4-oxo-4*H*-thieno[3,2-*c*]chromen-2-yl)acrylate (2b**)** This compound was obtained from **13** (210 mg, 0.7 mmol) and ethyl 2-cyanoacetate (79 mg, 0.7 mmol) in dichloromethane (60 mL) as orange crystals; yield 87%; mp 235–237°C; ¹H NMR (CDCl₃): δ 1.27 (6H, t, *J* = 7.0 Hz, 2 × CH₃), 1.42 (3H, t, *J* = 7.0 Hz, CH₃), 3.48 (4H, q, *J* = 7.1 Hz, 2 × CH₂), 4.40 (2H, q, *J* = 7.1 Hz, CH₂), 6.62 (1H, d, *J* = 2.1 Hz, C(8)H), 6.72 (1H, dd, *J* = 2.1 Hz, *J* = 8.9 Hz, C(6)H), 7.58 (1H, d, *J* = 8.9 Hz, C(5)H), 8.06 (1H, s, CH-thieno), 8.35 (1H, s, CH=); ¹³C NMR (CDCl₃): δ 11.8, 13.7, 45.0, 62.2, 98.1, 98.8, 104.9, 109.7, 115.2, 120.7, 125.3, 131.8, 138.2, 145.6, 150.3, 154.3, 155.0, 156.5, 162.0. HR-MS (ESI-TOF). Calcd for C₂₁H₂₀N₂O₄S [M + H]⁺: *m/z* 397.1222. Found: *m/z* 397.1192.

Acknowledgments: This work was funded by the Russian Science Foundation (RSCF), grant no. 17-13-01302.

References

- [1] Kivala, M.; Diederich, F. Acetylene-derived strong organic acceptors for planar and nonplanar push-pull chromophores. *Acc. Chem. Res.* **2009**, *42*, 235–248.
- [2] Kobayashi, H.; Ogawa, M.; Alford, R.; Choyke, P. L.; Urano, Y. New strategies for fluorescent probe design in medical diagnostic imaging. *Chem. Rev.* **2010**, *10*, 2620–2640.
- [3] Monici, M. Cell and tissue autofluorescence research and diagnostic applications. *Biotechnol. Ann. Rev.* **2005**, *11*, 227–256.
- [4] Zeng, Y. S.; Gao, R. C.; Wu, T. W.; Cho, C.; Tan, K. T. Fluorescent probe encapsulated in SNAP-Tag protein cavity to eliminate nonspecific fluorescence and increase detection sensitivity. *Bioconjug. Chem.* **2016**, *27*, 1872–1879.
- [5] Martinez-Manez, R.; Sancenon, F. Fluorogenic and chromogenic chemosensors and reagents for anions. *Chem. Rev.* **2003**, *103*, 4419–4476.
- [6] Liu, Y.; Xiang, K.; Guo, M.; Tian, B.; Zhang, J. A coumarin-based fluorescent probe for the fast detection of Pd⁰ with detection limit. *Tetrahedron Lett.* **2016**, *57*, 1451–1455.
- [7] Dalton, L. K.; Sullivan, P. A.; Bale, D. H. Electric field poled organic electro-optic materials: state of the art and future prospects. *Chem. Rev.* **2010**, *110*, 25–55.
- [8] Jiang, G.; Michinobu, T.; Yuan, W.; Feng, M.; Wen, Y.; Du, S. Crystalline thin films of a donor-substituted cyanoethynylene for nanoscale data recording through intramolecular charge-transfer interaction. *Adv. Mater.* **2005**, *17*, 2170–2173.
- [9] Cho, M. J.; Choi, D. H.; Sullivan, P. A.; Akelaitis, A. J. P.; Dalton, L. R. Recent progress in second-order nonlinear optical polymers and dendrimers. *Prog. Polym. Sci.* **2008**, *33*, 1013–1058.
- [10] Martínez de Baroja, N.; Garín, J.; Orduna, J.; Andreu, R.; Blesa, M. J.; Villacampa, B.; Alicante, R.; Franco, S. Synthesis, characterization, and optical properties of 4*H*-pyran-4-ylidene donor-based chromophores: the relevance of the location of a thiophene ring in the spacer. *J. Org. Chem.* **2012**, *77*, 4634–4644.
- [11] Castro, M. C. A.; Belsley, M.; Raposo, M. M. M. Synthesis and characterization of push-pull bithienylpyrrole NLOphores with enhanced hyperpolarizabilities. *Dyes Pigments* **2016**, *131*, 333–339.
- [12] Sinkeldam, R. W.; Greco, N. J.; Tor, Y. Fluorescent analogs of biomolecular building blocks: design, properties, and applications. *Chem. Rev.* **2010**, *110*, 2579–2619.
- [13] Richard, J.-A.; Massonneau, M.; Renard, P.-Y.; Romieu, A. 7-Hydroxycoumarin-hemicyanine hybrids: a new class of far-red emitting fluorogenic dyes. *Org. Lett.* **2008**, *10*, 4175–4178.
- [14] He, X.; Na, M.-H.; Kim, J.-S.; Lee, G.-Y.; Park, J. Y.; Hoffman, A. S.; Nam, J. O.; Han, S. E.; Sim, G. Y.; Oh, Y. K.; et al. A novel peptide probe for imaging and targeted delivery of liposomal doxorubicin to lung tumor. *Mol. Pharmaceutics* **2011**, *8*, 430–438.
- [15] Morales, A. R.; Frazer, A.; Woodward, A. W.; Ahn-White, H.-Y.; Fonari, A.; Tongwa, P.; Timofeeva, T.; Belfield, K. D. Design, synthesis, and structural and spectroscopic studies of push-pull two-photon absorbing chromophores with acceptor groups of varying strength. *J. Org. Chem.* **2013**, *78*, 1014–1025.
- [16] Wagner, B. D. The use of coumarins as environmentally-sensitive fluorescent probes of heterogeneous inclusion systems. *Molecules* **2009**, *14*, 210–237.

- [17] Brackman, U. Lamdachrome Laser Dyes Data Sheets. 2nd Edition. Lamda Physik GmbH: Gottingen, 1997; pp 153–179.
- [18] Ayyangar, N. R.; Srinivasan, K. V.; Daniel, T. Polycyclic compounds part VII. Synthesis, laser characteristics and dyeing behaviour of 7-diethylamino-2H-1-benzopyran-2-ones. *Dyes Pigments* **1991**, *16*, 197–204.
- [19] Hara, K.; Wang, Z.-S.; Sato, T.; Furube, A.; Katoh, R.; Sugihara, H.; Dan-oh, Y.; Kasada, C.; Shinpo, A.; Suga, S. Oligothiophene-containing coumarin dyes for efficient dye-sensitized solar cells. *J. Phys. Chem. B* **2005**, *109*, 15476–15482.
- [20] Wang, Z.-S.; Cui, Y.; Dan-oh, Y.; Kasada, C.; Shinpo, A.; Hara, K. Molecular design of coumarin dyes for stable and efficient organic dye-sensitized solar cells. *J. Phys. Chem. C* **2008**, *112*, 17011–17017.
- [21] Welsch, K.; Grilj, J.; Vauthey, E.; Aylott, J. W.; Chan, W. C. Protease responsive nanoprobe with tethered fluorogenic peptidyl 3-aryl coumarin substrates. *Chem. Commun.* **2009**, *45*, 671–673.
- [22] Christie, R. M.; Lui, C.-H. Studies of fluorescent dyes: part 2. An investigation of the synthesis and electronic spectral properties of substituted 3-(2'-benzimidazolyl)coumarins. *Dyes Pigments* **2000**, *47*, 79–89.
- [23] Dondon, R.; Khilya, V. P.; Roshal, A. D.; Fery-Forgues, S. UV/vis absorption and fluorescence spectroscopic study of some new 4-hydroxy-7-methoxycoumarin derivatives. Part I: Effect of substitution by a benzo-1,4-dioxanyl or an ethyl furoate group in the 3-position. *New J. Chem.* **1999**, *23*, 923–927.
- [24] Santana, L.; Uriarte, E.; Roleirab, F.; Milhazes, N.; Borges, F. Furocoumarins in medicinal chemistry. Synthesis, natural occurrence and biological activity. *Curr. Med. Chem.* **2004**, *11*, 3239–3261.
- [25] Weibenfels, M.; Hantschmann, A.; Steinfuhrer, T.; Birkner, I. Synthese und eigenschaften von 7-substituierten thieno[3,2-c]-coumarin-2-carbonsaureestern. *Z. Chem.* **1989**, *29*, 166.
- [26] Bochkov, A. Y.; Akchurin, I. O.; Dyachenko, O. A.; Traven, V. F. NIR-fluorescent coumarin-fused BODIPY dyes with large Stokes shifts. *Chem. Commun.* **2013**, *49*, 11653–11655.
- [27] Chen, Y.-S.; Kuo, P.-Y.; Shie, T.-L.; Yang, D.-Y. Structure, reactivity, and application of some triketone derivatives. *Tetrahedron* **2006**, *62*, 9410–9416.
- [28] Knierzinger, A.; Wolfbeis, O. S. Synthesis of fluorescent dyes. IX. New 4-hydroxycoumarins, 4-hydroxy-2-quinolons, 2H,5H-pyrano[3,2-c]benzopyran-2,5-diones and 2H,5H-pyrano[3,2-c]-quinoline-2,5-diones. *J. Heterocycl. Chem.* **1980**, *17*, 225–229.
- [29] Jabin, I.; Revial, G.; Monnier-Benoit, N.; Netchitailo, P. Reaction of cyclohexanone benzylimines with ethylidenemalonate diesters. Diphenyl 2-ethylidenemalonate: a highly electrophilic synthetic equivalent of crotonic esters. *J. Org. Chem.* **2001**, *66*, 256–261.
- [30] Conreaux, D.; Belot, S.; Desbordes, P.; Monteiro, N.; Balme, G. Et₃N-induced demethylation-annulation of 3-alkynyl-4-methoxy-2-pyridones and structurally related compounds in the synthesis of furan-fused heterocycles. *J. Org. Chem.* **2008**, *73*, 8619–8622.
- [31] James, C. A.; Coelho, A. L.; Gevaert, M.; Forgione, P.; Snieckus, V. Combined directed *ortho* and remote metalation-Suzuki cross-coupling strategies. Efficient synthesis of heteroaryl-fused benzopyranones from biaryl *O*-carbamates. *J. Org. Chem.* **2009**, *74*, 4094–4103.
- [32] Raffa, G.; Rusch, M.; Balme, G.; Monteiro, N. A Pd-catalyzed heteroannulation approach to 2,3-disubstituted furo[3,2-c]-coumarins. *Org. Lett.* **2009**, *11*, 2009.
- [33] Majumdar, K. C.; Bhattacharyya, T. Regioselective synthesis of furo[3,2-c][1]benzopyran-4-one and furo[3,2-c]quinolin-4-one. *J. Chem. Res. (S)* **1997**, 244–245.
- [34] Iaroshenko, V. O.; Erben, F.; Mkrtchyan, S.; Hakobyan, A.; Vilches-Herrera, M.; Dudkin, S.; Bunescu, A.; Villinger, A.; Sosnovskikh, V. Y.; Langer, P. 4-Chloro-3-(trifluoroacetyl)- and 4-chloro-3-(methoxalyl)coumarins as novel and efficient building blocks for the regioselective synthesis of 3,4-fused coumarins. *Tetrahedron* **2011**, *67*, 7946–7955.
- [35] Bai, H.; Qian, J.; Tian, H.; Pan, W.; Zhang, L.; Zhang, W. Fluorescent polarity probes for identifying bovine serum albumin: amplification effect of para-substituted benzene. *Dyes Pigments* **2014**, *103*, 1–8.
- [36] Tathe, A. B.; Gupta, V. D.; Sekar, N. Synthesis and combined experimental and computational investigations on spectroscopic and photophysical properties of red emitting 3-styryl coumarins. *Dyes Pigments* **2015**, *119*, 49–55.
- [37] Kim, G.-J.; Yoon, D.-H.; Yun, M.-Y.; Kwon, H.; Ha, H.-J.; Kim, H.-J. Ratiometric fluorescence probes on a Michael acceptor type of coumarin and their application for the multichannel imaging of *in vivo* glutathione. *RSC Adv.* **2014**, *4*, 18731–18736.
- [38] Bangar, R. B.; Varadaragan, T. S. Spectroscopic studies of 7-diethylamino-3-styryl coumarins. *J. Photochem. Photobiol. A Chem.* **1985**, *85*, 263–267.
- [39] Lyong, A. K.; Shin, S. R.; Jun, K.; Park, S. Y. The synthesis and light absorption behavior of novel coumarin chromophores. *J. Korean Chem. Soc.* **2014**, *58*, 297–302.