

# The heat effect of combustion process depending on fuel composition fluctuations

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**Abstract.** This article describes effects of changes in the fuel composition on the combustion process. Assessment of effect made on example of the steam boiler analysis.

## 1. Introduction

The problem of natural resources rational use, energy efficiency and energy saving is relevant at the moment. In this connection, the task of rational and efficient use of natural resources in their combustion is relevant, in particular in thermal power plants. One way of solving this problem is the use of alternative fuels, such as mine gas, landfill gas, biogas, gas of waste water and that is especially true - associated petroleum gas and waste oil refineries.

The problem of using such fuels is in the uncontrollable changes of their composition in time (Table 1) [1].

**Table 1.** The fluctuation ranges of fuel composition.

	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	C <sub>4</sub> H <sub>10</sub>	C <sub>5</sub> H <sub>12</sub> and higher	CO <sub>2</sub>	H <sub>2</sub>	N <sub>2</sub>
<b>Natural gas</b>	80-98	0-1	0-0.5	0-0.5	0-0.5	0-1	2-10	0
<b>Associated petroleum gas</b>	15-90	0-25	0-20	0-10	0-15	0-10	0-5	0-30
<b>Waste of oil refineries</b>	0-35	0-19	0-32	0-40	0-47	0	0-25	0

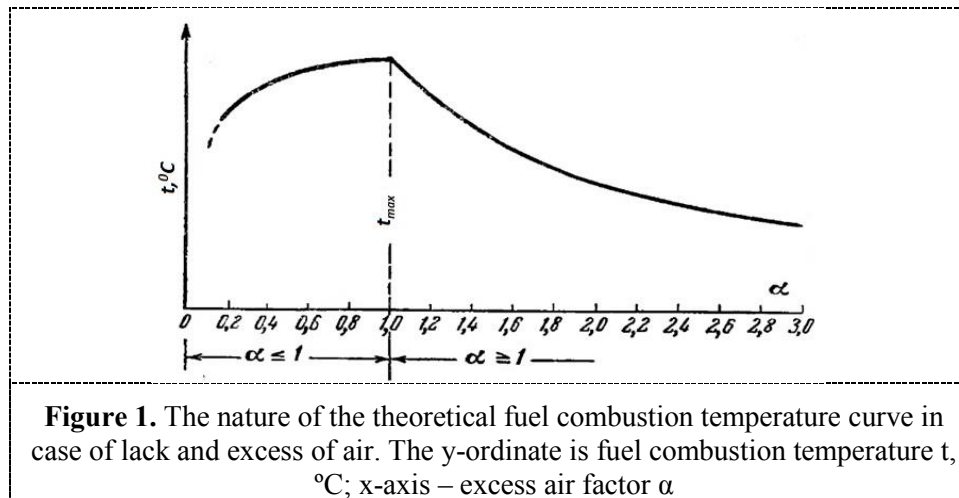
These changes negatively effect on the combustion process, reduce combustion process efficiency, causing damage to the environment [2-4].

The aim of this work was to assessment of the changes effect of fuel composition on the thermal effect of the combustion process.

## 2. Calculation results

The optimality of combustion process (stoichiometric ratio of "fuel-air") can quantify the excess air factor  $\alpha$ . When  $\alpha = 1$  the combustion optimal, combustion temperature maximum (Figure 1) [5].





The fuel composition affects the carbon number of fuel  $n$ , and hence the theoretically required amount of air  $V_0$ , which affects the factor  $\alpha$ .

$$n = \frac{1\text{CH}_4 + 2\text{C}_2\text{H}_6 + 3\text{C}_3\text{H}_8 + \dots}{(100 - B)},$$

where  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$ ,  $\text{C}_3\text{H}_8$ , etc. – the sum of the hydrocarbons components, %;  $B$  - the sum of the ballast components  $\text{CO}_2$  and  $\text{N}_2$ , %.

$$V_0 = 7.13n + 2.28.$$

$$\alpha = V / V_0,$$

where  $V$  is the real amount of air.

Increasing the proportion of fuel heavy fractions with a large carbon number reduces  $\alpha$ , reduction of carbon number - to increase  $\alpha$ . In both cases, the combustion temperature is reduced, optimality of combustion process is impaired.

Assessment of the fuel composition impact on the thermal effect of the combustion process carried out by the example of analysis a steam boiler operating mode. The steam capacity  $D = 6097$  kg/h; enthalpy of water at the inlet  $H_w = 0.508$  MJ/kg; the required enthalpy of steam  $H_s = 2.77$  MJ/kg; boiler efficiency  $\eta = 91.85$  %. The optimal mode: the fuel composition  $\text{CH}_4=60$  %,  $\text{C}_2\text{H}_6=4.5$  %,  $\text{C}_3\text{H}_8=13$  %,  $\text{C}_4\text{H}_{10}=7.5$  %,  $\text{C}_5\text{H}_{12}=4.9$  %,  $\text{CO}_2=0.1$  %,  $\text{N}_2=10$  %; fuel consumption  $G_f = 286.98$  m<sup>3</sup>/h.

The fuel composition affects specific heat of combustion (SHC)  $Q_m$ , that affects the enthalpy of coolant [6].

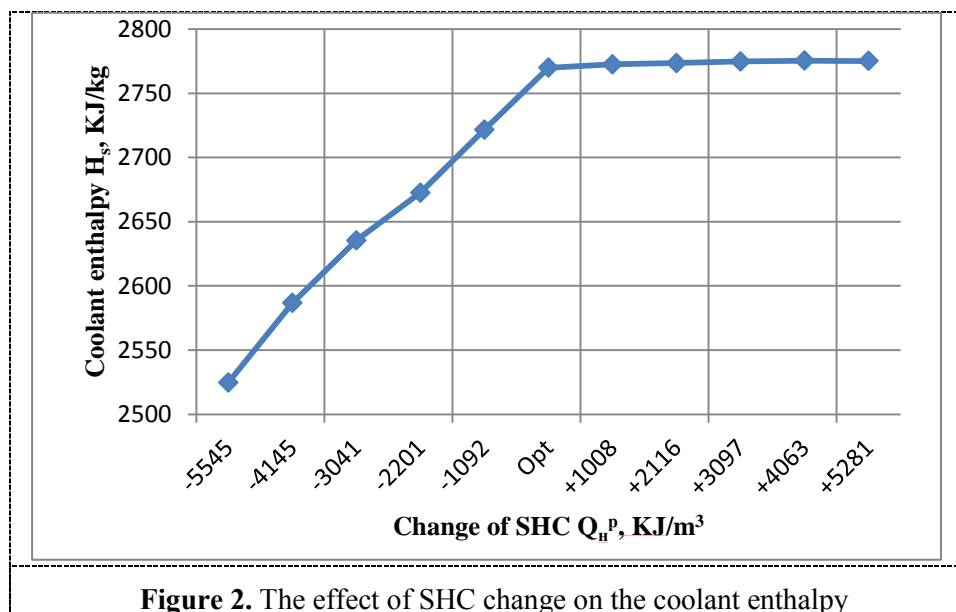
$$Q_m = 0.01(Q_{\text{H}_2\text{S}}\text{H}_2\text{S} + Q_{\text{CO}}\text{CO} + Q_{\text{H}_2}\text{H}_2 + \sum Q_{\text{C}_m\text{H}_n}\text{C}_m\text{H}_n),$$

where  $Q_{\text{H}_2\text{S}}$ ,  $Q_{\text{CO}}$  etc. – combustion heat of gas components.

$$H_s = \frac{G_f Q_m \eta}{100 \cdot D} + H_w.$$

The graph (Figure 2) shows the change in enthalpy of the coolant in case of SHC deviation from the optimum values.

As can be seen from the graphs, in the case of reducing the fuel SHC enthalpy of the coolant begins to decline sharply. This is due to a decreasing in heat release of fuel combustion and the increasing of air excess. In the case of SHC increasing enthalpy of the coolant rises slightly or remains constant. This is because the increase of combustion heat was compensated by combustion incompleteness.



### 3. Conclusion

Uncontrolled changes in fuel composition lead to instability of the thermal effect of the combustion process, which adversely affects the thermal performance of power plants.

Reducing the SHC leads to a sharp decrease in the enthalpy of the coolant at the outlet because of the excess air coefficient increasing. With increasing fuel SHC the coolant enthalpy is not substantially changed due to the growing underburning value.

The work is performed according to the Russian Government Program of Competitive Growth of Kazan Federal University.

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