

Toxicity assessment of polluted sediments using swimming behavior alteration test with *Daphnia magna*

O V Nikitin, E I Nasyrova, V R Nuriakhmetova, N Yu Stepanova, N V Danilova and V Z Latypova

Institute of Environmental Sciences, Kazan Federal University, Kremlevskaya str., 18, 420008, Russia
e-mail: olnova@mail.ru

Abstract. Recently behavioral responses of organisms are increasingly used as a reliable and sensitive tool in aquatic toxicology. Behavior-related endpoints allow efficiently studying the effects of sub-lethal exposure to contaminants. At present behavioural parameters frequently are determined with the use of digital analysis of video recording by computer vision technology. However, most studies evaluate the toxicity of aqueous solutions. Due to methodological difficulties associated with sample preparation not a lot of examples of the studies related to the assessment of toxicity of other environmental objects (wastes, sewage sludges, soils, sediments etc.) by computer vision technology. This paper presents the results of assessment of the swimming behavior alterations of *Daphnia magna* in elutriates from both uncontaminated natural and artificially chromium-contaminated bottom sediments. It was shown, that in elutriate from chromium contaminated bottom sediments (chromium concentration $115 \pm 5.7 \mu\text{g l}^{-1}$) the swimming speed of daphnids was decreases from 0.61 cm s^{-1} (median speed over the period) to 0.50 cm s^{-1} (median speed at the last minute of the experiment). The relocation of *Daphnia* from the culture medium to the extract from the non-polluted sediments does not essential changes the swimming activity.

1. Introduction

Various bioassay methods are actively used for assess the quality of natural waters and other environmental objects as well as the safety of existing or new artificially created chemical substances. In the tests, the evaluation is performed on the basis of observations and experiments with “living sensors” – organisms whose vital activity is disturbed in the toxic conditions [1–3]. Despite the fact that the range of organisms used as tests is very wide, there is no organism that is universally sensitive to all existing toxicants. Therefore, tend to use batteries of tests organisms [4, 5].

Cladocera (an order of minute branchiopod crustaceans) are one of the most frequently used organisms in ecotoxicological studies [6]. A feature of most of them is the filtration method of feeding, in which considerable amounts of water are passed over a specialized filtering structure. Cladocera are able to extract from the water not only the necessary food but also dissolved and suspended solids including the components of contaminants.

In most cases, as test objects used crustaceans *Daphnia magna*, *D. longispina*, *Ceriodaphnia dubia*, *C. affinis* and some others. The most universal test object for sensitivity and adequacy of response to different toxicants can be considered *D. magna*. Biotesting methods using *D. magna* are widely used for the purposes of environmental control, the laboratory culture of *D. magna* is one of the most commonly used in the practice of determining the acute and chronic toxic effects of individual chemicals and water samples with a complex chemical composition. Typically the toxicity is assessed



by the EC50, LC50 end-points and also by a statistically significant change in the parameters of vital activity of daphnids under toxic and control conditions. Mortality of crustaceans is mainly used as a test reaction in acute tests; observations of changes in fertility and quality of offspring are carried out in the study of chronic toxic effects [7].

It is possible to significantly expand the list of recorded parameters by using additional information about the test organism based on its functional characteristics including behavioral responses [8,9]. Moreover, analysis of the literature shows that behavioral reactions yielded significant responses at concentrations of toxicants well below reported LC50 values [10].

Behavioral responses most clearly manifested at the organism level, are between the biochemical and ecosystem levels of biomonitoring. However, at the same time behavioral alterations are directly based on biochemical processes, reflect the adaptation of an individual organism as well as the potential effects on the population level. Behavioral responses appear to compare favourably with biochemical and physiological responses in terms of sensitivity and efficiency. Furthermore, in addition to their integrative nature and ecological importance, the assessment of behavioral responses is a non-destructive method that makes possible continuous long-term environmental monitoring [11, 12].

Behavioral responses of aquatic organisms have been used for decades as methods for environmental monitoring but these types of studies have previously received much less attention than others focused on different subject such as developmental or reproductive toxicology [10]. Probably this was due to the problems in recording and quantification of the behavioral characteristics of test organisms. Many difficulties eventually were removed and at present, in addition to visual registration, a quantitative description of behavioral reactions is carried out by various methods: optical, ultrasonic, electrical, magnetic etc.

At the current stage of the development of technology, it is possible to sensitive determination of the presence of toxic substances in water by the behavioral characteristics of test objects, including such small organisms like *Daphnia*, using digital analysis of video recording [13–16]. To assess the toxicity used such daphnids behavioral characteristics as swimming time and speed, swimming trajectory, hopping frequency, vertical and horizontal distribution, distance travelled, number of turnings and turning angle, resting time, duration of quiescence, sinking rate, gravitaxis, swarming and spinning. Swimming mobility expressed by scalar quantity-speed and its vector quantity-velocity (usually expressed in millimeters per second) is one of the most reliable and widely used parameter of *Daphnia* behavioral activity [16].

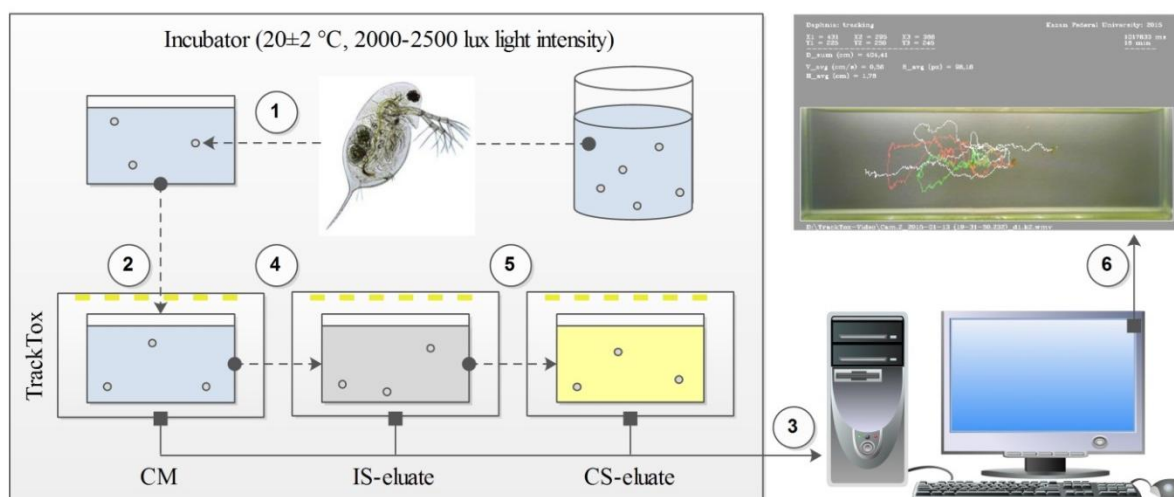
Most studies have evaluated the toxicity of aqueous solutions of pure chemicals by the behavioral alterations of the test organisms. In addition to these studies, it is of interest to estimate the toxicity of other environmental objects (wastes, sewage sludges, soils, sediments etc.) using a digital analysis of the behavioral responses of test organisms exposed to water extracts of solid sample. In this case, there are methodological difficulties associated with sample preparation, such as staining and increasing the turbidity of the water solution. This can restrict the application of computer vision technology since this will make it impossible to recognize objects. The aim of this study was to assess the alterations in the swimming behavior of *D. magna* in elutriates from both uncontaminated natural and artificially chromium-contaminated bottom sediments.

2. Material and methods

Sediment samples for the experiment were collected in 2015 from the Almetyevsk reservoir (54°55'1" N 52°16'24" E). The reservoir is located in the northern part of Almetyevsk (Republic of Tatarstan, Russia). The reservoir exists within the riverbed and the low floodplain of the Steppe Zay River. The reservoir is shallow; the prevailing depth is about 0.9 m. Water area is 1.11 km², the maximum length and width is 1.78 and 1.0 km respectively.

Sediments were dried at the temperature of 105 °C and ground, than they were sieved through a 1 mm mesh. To simulate contaminated bottom sediments in the initial non-polluted sediments ("IS") potassium dichromate (K₂Cr₂O₇) was added in an amount of 1.41 grams per 100 grams of dry weight

The behavioral experiments were performed with *D. magna* Straus, cultured in accordance with the Russian standard procedure [10]. For experiments we used parthenogenetically reproducing laboratory clones of *Daphnia* cultivated at the Department of Applied Ecology at the Kazan Federal University (Kazan, Russia). *D. magna* are cultivated at constant conditions of temperature ($20 \pm 2^\circ\text{C}$), under 2000–2500 lux light intensity and 12/12 h light/dark photoperiod in a laboratory incubator. Daphnids were fed daily with 1–2 mL *Chlorella vulgaris* suspension cultured in Tamiya medium. The daphnids used in this experiment originate from one single female to eliminate variation among clones. *D. magna* individuals were selected according to size (3.23 ± 0.03 mm). Size measurements carried out using an eyepiece micrometer.



0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99

deviation (SD), standard error of the mean (SE), median, and interquartile range were calculated. The swimming speeds were averaged for time intervals (every minute) as well as for the consolidated group of thirty daphnids (number of experiments is N=10). Statistical analysis was performed using STATISTICA 8.0 (StatSoft).

3. Results and discussion

A chemical analysis both of initial non-polluted sediments and water elutriates of them showed a low level of metals contamination (table 1). The chromium content in natural bottom sediments was $23 \pm 4.6 \text{ mg kg}^{-1}$, in artificially contaminated sediments – $9100 \pm 1820 \text{ mg kg}^{-1}$. The concentrations of chromium in “IS-elutriate” and “CS-elutriate” were $90 \pm 4.5 \text{ } \mu\text{g l}^{-1}$ and $2300 \pm 115 \text{ } \mu\text{g l}^{-1}$ respectively. In the latter case, difficulties were encountered with the recognition of daphnids by “TrackTox” (due to staining of the solution), so that the sample was further diluted to a concentration of $115 \pm 5.7 \text{ } \mu\text{g l}^{-1}$ (“CS-elutriate_d”).

Table 1. The concentration of metals in bottom sediments (mg kg^{-1}) and in eluents of them ($\text{ } \mu\text{g l}^{-1}$).

	Cr	Al	Cd	Cu	Fe	Mn	Ni	Pb
IS	23 ± 4.6	2025 ± 526.5	<0.05	15 ± 3	5000 ± 1400	950 ± 285	21 ± 7.35	6.5 ± 1.63
CS	9100 ± 1820	1850 ± 481.0	<0.05	18 ± 3.6	4875 ± 1365	900 ± 270	19 ± 6.65	4.9 ± 1.23
IS-elutriate	24.4 ± 1.22	52 ± 3.12	0.3 ± 0.03	6.5 ± 0.39	81 ± 4.05	20 ± 1.4	3.9 ± 0.23	9 ± 0.72
CS-elutriate	2300 ± 115.0	23 ± 1.38	0.1 ± 0.01	11 ± 0.66	75 ± 3.75	63 ± 4.41	5.5 ± 0.33	11 ± 0.88

The summary results of the *D. magna* behavior alteration experiments in the following sequence: “CM” → “IS-elutriate” → “CS-elutriate_d” are shown in figure 2.

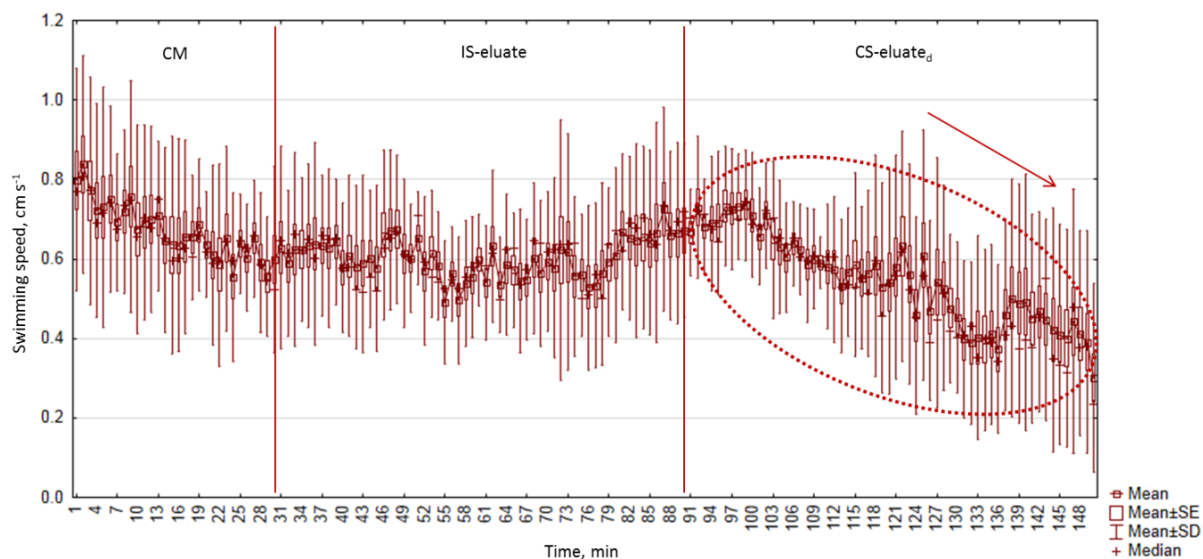


Figure 2. Summary results of the swimming activity of *Daphnia* in a series of experiments.

The average swimming speed of daphnids in culture medium was $0.67 \pm 0.01 \text{ cm s}^{-1}$, after transfer to the “IS-elutriate” it practically did not change and was $0.60 \pm 0.01 \text{ cm s}^{-1}$, after transfer to the “CS-elutriate_d” it further decreased to $0.55 \pm 0.01 \text{ cm s}^{-1}$ (table 2). More detailed information about swimming activity of *Daphnia* can give a per-minute analysis of the dynamics – in this case, the inhibiting effect of chromium can be traced particularly clearly.

Table 2. Parameters of *Daphnia* swimming activity (cm s^{-1}) in different mediums.

Swimming speed	Mean	Min	Max	Std.Dev.	Std.Err.	Q25	Median	Q75
<i>Values in culture medium</i>								
CM _p (whole period)	0.67	0.12	1.65	0.23	0.01	0.52	0.65	0.78
CM _{fm} (first minute)	0.80	0.44	1.43	0.28	0.07	0.55	0.77	1.04
CM _{lm} (last minute)	0.60	0.37	1.29	0.23	0.06	0.43	0.52	0.73
<i>Values in IS-elutriate medium</i>								
IS _p (whole period)	0.60	0.09	1.53	0.20	0.01	0.46	0.61	0.72
IS _{fm} (first minute)	0.63	0.13	1.03	0.25	0.07	0.42	0.65	0.86
IS _{lm} (last minute)	0.67	0.26	1.10	0.22	0.06	0.54	0.72	0.78
<i>Values in CS-elutriate_a medium</i>								
CS _p (whole period)	0.55	0.02	1.38	0.24	0.01	0.37	0.58	0.71
CS _{fm} (first minute)	0.67	0.50	0.86	0.11	0.03	0.57	0.70	0.73
CS _{lm} (last minute)	0.30	0.04	0.81	0.24	0.06	0.10	0.24	0.50

An additional estimate is the toxicity index widely used in practice:

$$T, \% = 100 \times (V_{bc} - V_{ac}) / X_{bc} \quad (1)$$

V_{bc} – is the swimming speed in the baseline, V_{ac} – is the swimming speed in the analyzed conditions. It can be noted that the toxic effect in the series with chromium, despite a pronounced decrease in speed by the end of the experiment, is not revealed by this approach. The toxicity index for the average speed in this case is only 8.3%.

Probably this is a feature of data aggregation, when there is a significant variation in values of speed during the experiment. To eliminate the influence of this factor, it is possible to use for the calculation of the toxicity index not the integral value of the swimming speed for the entire observation period, but only the boundary values that characterize the final state of *Daphnia* (at the last minute), both in the control and in the experimental conditions (table 3).

Table 3. Variants of the toxicity indexes calculated by the different values of swimming speed.

Toxicity index	By mean values	By median values
$(CM_p - IS_p) / CM_p$	10.4%	6.2%
$(IS_p - CS_p) / IS_p$	8.3%	4.9%
$(CM_p - IS_{lm}) / CM_p$	0.0%	–10.8%
$(IS_p - CS_{lm}) / IS_p$	50.0%	60.7%
$(IS_{fm} - CS_{lm}) / IS_{fm}$	52.4%	63.1%
$(CM_{lm} - IS_{lm}) / CM_{lm}$	–11.7%	–38.5%
$(IS_{lm} - CS_{lm}) / IS_{lm}$	55.2%	66.7%

The toxicity index calculated by this way was 55 and 67% for the mean and median speed, respectively. Since the data distribution in the samples differs from the normal one, it is possible to recommend the use of a toxicity index calculated just by the median. The values obtained correspond more accurately to the actual picture of the swimming activity of *Daphnia* observed in the experiment.

Daphnia swimming behavior is complex, multiparametric and it may be characterized by several parameters reflecting changes induced by various compounds on sensitive (i.e. nervous and endocrine) systems [16]. A number of results indicate that various substances may alter swimming speed. In particular, our earlier studies showed that *D. magna* change their behavior under the exposure of potassium dichromate, zinc sulfate, esfenvalerate pesticide, microcystin-LR [17–19]. In the case of the exposure of metal salts, there was a statistically significant decrease, and in the case of a pesticide and cyanotoxin, an increase in the swimming speed of *Daphnia* with a short exposure time of 10–30 minutes.

Some authors also suggested that the swimming speed may be also altered by some metals. In particular, daphnids exposed to sublethal cadmium [20,21] or copper stress [22] tends to manifest inhibited swimming activity when compared to the unexposed, similarly-sized control organisms. The mechanisms by which metals decrease *Daphnia* speed have not been completely clarified, however it could be hypothesized that oxidative stress known to be induced by these substances in other animal species may also be responsible for the toxic effects on the behavior of crustaceans [16].

4. Conclusion

The conducted study showed the possibility of using a biotesting method based on computer vision to assess the quality of contaminated bottom sediments. In the elutriate from chromium contaminated bottom sediments, the *Daphnia* swimming speed decrease from 0.61 cm s^{-1} (median speed over the period) to 0.50 cm s^{-1} (median speed at the last minute of the experiment). The relocation of daphnids from the culture medium to the extract from the non-polluted sediments does not essential changes the swimming activity.

Acknowledgement

The work is performed according to the Russian Government Program of Competitive Growth of Kazan Federal University

References

- [1] Stepanova N 2014 *Int. Multidisciplinary Scientific GeoConference Surveying Geology and Mining Ecology Management SGEM* **2** 215 – 22
- [2] Stepanova N Y, Nikitin O V and Latypova V Z 2016 *International Journal of Pharmacy and Technology* **8** 25096–102
- [3] Szczerbińska N and Gałczyńska M 2015 *Archives of Polish Fisheries* **23** 185–96
- [4] Selivanovskaya S, Galitskaya P, Schnell S and Hung Y-T 2010 *International Journal of Environment and Waste Management* **6** 183–96
- [5] Kokkali V and van Delft W 2014 *TrAC Trends in Analytical Chemistry* **61** 133–55
- [6] 2017 EPA ECOTOX Knowledgebase URL: <https://cfpub.epa.gov/ecotox>
- [7] Blaise C and Féraud J-F 2005 *Small-scale freshwater toxicity investigations vol 1 – Toxicity test methods* (Dordrecht: Springer)
- [8] Borchertding J 2006 *Hydrobiologia* **556** 417–26
- [9] Olkova A S 2016 *Water and Ecology* **4** 75–95
- [10] Melvin S D and Wilson S P 2016 *Chemosphere* **93** 2217–23
- [11] Gerhardt A, Svensson E, Clostermann M and Fridlund B 1994 *Environment International* **20** 209–19
- [12] Gerhardt A 2007 *Human and Ecological Risk Assessment: An International Journal* **13** 481–91
- [13] Jeon J, Kim J H, Fee B C. and Kim S D 2008 *Science of The Total Environment* **389** 545–56
- [14] Noss C, Dabrunz A, Rosenfeldt R R, Lorke A and Schulz R 2013 *PLoS One* **11** e80960
- [15] Chevalier J, Grote M, Keller M, Pandard P and Cachot J J *Environ Anal Toxicol* **4** 246
- [16] Bownik A 2017 *Science of the Total Environment* **601–602** 194–205
- [17] Nikitin O and Latypova V 2014 *Int. Multidisciplinary Scientific GeoConference Surveying Geology and Mining Ecology Management SGEM* vol 2 (Sofia: STEF92 Technology) 85–92
- [18] Nikitin O 2014 *Advances in Environmental Biology* **8** 74–8
- [19] Nikitin O V, Petrova V M and Latypova V Z 2015 *Research Journal of Pharmaceutical, Biological and Chemical Sciences* **6** 1729–36
- [20] Wolf G, Scheundersb P and Selens M 1998 *Comp. Biochem. Physiol.* **120** 99–105
- [21] Baillieul M and Blust R 1999 *Aquat. Toxicol.* **44** 245–54
- [22] Untersteiner H, Kahapka J and Kaiser H 2003 *Aquat. Toxicol.* **65** 435–42