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The Modifications of Electronic and Optical Properties of Bulk Molybdenum Disulfide by Oxygen Substitution

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Abstract. We study the electronic and optical properties of bulk MoS₂ and MoSO systems using the plane-wave method within the generalized gradient approximation. The structural properties show that the O substitution at S sites tunes the *z*-axis Wyckoff positions of all atoms and the bond angles. The electronic band structures show that the substitution tunes the conduction band minimum at 0.8 Σ to M and significantly promotes more localized valence states compared to that of MoS₂ system. The localization, mainly applying to Mo 4*d* and S 3*p* states, is more pronounced above -4 eV. Below -4 eV, O 2*p* states are dominant indicating that they are more stable than S 3*p* states. However, the substitution slightly increases the indirect and direct (K \rightarrow K) bandgap of MoS₂ system. From the optical properties, both systems show the strong optical dichroism. By means of the substitution, the σ_1 width significantly enhances, while the plasmonic-state energy levels decrease. Our result emphasizes that the O substitution significantly tunes the electronic and optical properties of bulk MoS₂ system.

1. Introduction

Molybdenum disulfide (MoS₂) is a semiconducting layered transition-metal dichalcogenide (TMD) which is the most promising two-dimensional (2D) systems for nanoelectronic applications [1-4]. One of the interesting phenomena of the system is that it exhibits the layer-number-dependence of band structure. The bulk and monolayer MoS₂ systems have the indirect bandgap of 1.88 eV [5] and direct bandgap of 1.90 eV [1], respectively. However, the bulk system exhibits the indirect minimum bandgap (E_g) of 1.29 eV [5]. The reduction of the layer thickness from bulk to monolayer systems increases E_g , indicating the tunable band structures [6]. The monolayer system can strongly emit photons. The system exhibits the very high luminescence quantum efficiency of more than 10^4 compared with that of the bulk system. Furthermore, the monolayer system shows the strong photoluminescence (PL) compared to the bilayer and bulk systems [1]. The defect sites in the monolayer MoS₂ can be used to significantly improve its hydrogen evolution activity [7]. Moreover, MoS₂ can also be applied in the supercapacitor by building the MoS₂-H₂O multilayer [8] and for the energy storage by building reduced graphene oxide/MoS₂ hybrids [9].

The doping treatment using the transition metals at Mo site tunes the electronic and magnetic properties of MoS₂ system [10]. The previous experimental report has shown that the *n*-type substitution increases the local chemical affinity, while the *p*-type substitution shows the larger mobility when the electron beam exposes the system. The result emphasizes the application possibility in nanoelectronic devices [11]. The other report has shown the possibility of doping at S sites by using nonmetal and transition metal elements. The doping treatment significantly modifies the magnetic behavior of the



monolayer system. Notably, H, Br, and Cr dopings induce the half-metallic behavior of the system [12]. However, there is no report of optical properties and plasmonic states in MoS₂ system doped at S sites, as O and S are included in one group in the periodic table.

In this paper, we report electronic and optical properties of bulk MoS₂ and MoSO systems. In the MoSO system, one S atom is substituted by one O atom. We employed the plane-wave method within the generalized gradient approximation (GGA) [13]. Firstly, this paper reveals band structures and density of states from the electronic structure calculation. Secondly, the optical properties are calculated in the term of energy-dependent optical conductivity (σ), which is extracted from the imaginary part of dielectric function (ϵ_2). Lastly, the electron-energy loss function (L) also calculated to indicates the plasmonic-state energy levels.

2. Computational details

Quantum ESPRESSO package code [14, 15] was carried out for calculating the electronic and optical properties of unit cells of bulk MoS₂ and MoSO systems. The crystal structure models of both systems visualized by VESTA [16] are presented in figure 1. The structural parameters in this calculation are directly adopted from the previous experimental report. At room temperature, the bulk MoS₂ exhibits the hexagonal crystal structure with the structural parameters of $a = 3.15$ Å, $c = 12.30$ Å, $z_s = 0.621$ within the space group of $P6_3/mmc$ [17]. Each Mo atom is surrounded by O atoms within the triangular prismatic shape. From top view along c -axis, the hexagonal pattern consists of 3 Mo and 3 S atoms [18]. For MoSO system, two O atom substitute one S atom at atomic positions of $(2/3, 1/3, -z)$ and $(1/3, 2/3, -z + 1/2)$ in the unit cell of MoSO system. We employed the Perdew-Burke-Ernzerhof (PBE)-type exchange-correlation functional energy for all atoms within the GGA method [13]. This method was used for various systems in our previous reports [19-21]. All-electron potentials are approached by the norm-conserving pseudopotentials [22-24]. The calculation carried out kinetic cutoff energy of 1.6 keV, a Γ -point-centered k -point mesh of $7 \times 7 \times 2$, and threshold energy of 10 meV. All atomic positions are optimized using the ‘relax’ calculation with threshold force of 50 meV/Å involving the Broyden–Fletcher–Goldfarb–Shanno (BFGS) algorithm [25-28]. On the other hand, the lattice parameters (a and c) are set to be constant.

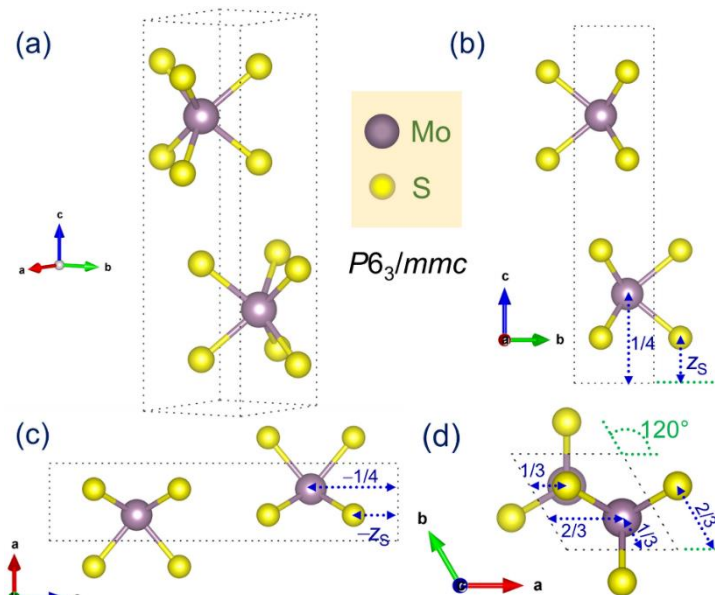


Figure 1. Crystal structure of bulk MoS₂ for views along (a) diagonal, (b) a , (c) b , and (d) c axes. The Wyckoff positions of Mo and S atoms are Mo (2c: $1/3, 2/3, 1/4$) and S (4f: $1/3, 2/3, z_s$).

The Drude-Lorentz model [29] is used in the optical properties calculation. This model covers the interband and intraband parts of complex dielectric functions (ε) [30]. From both parts, ε is divided into real and imaginary parts of dielectric functions (ε_1 , ε_2) which are connected by the Kramers-Kronig relation [31, 32]. Both ε_1 , ε_2 will be reported elsewhere in future. In this paper, we show the imaginary-part optical conductivity (σ_1) expressed as

$$\sigma_1 = \frac{\omega \varepsilon_2}{4\pi} \quad (1)$$

where ω is the photon frequency. We also reveal the loss function (L) spectrum calculated by

$$L = -\text{Im}\left\{\frac{1}{\varepsilon}\right\} = \frac{\varepsilon_2}{\varepsilon_1^2 + \varepsilon_2^2} \quad (2)$$

The energy-dependent L can accurately assign plasmonic-state energy levels [32].

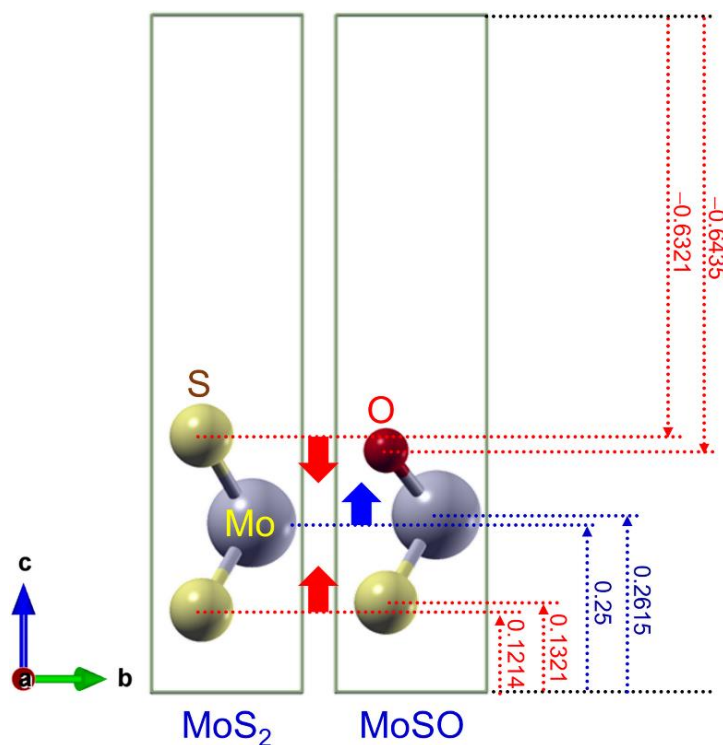


Figure 2. Schematic illustration of shifts of z -axis atomic positions in MoS_2 system induced by the O substitution at S site in MoSO system.

Table 1. Optimized Wyckoff positions of MoS_2 and MoSO systems.

Atom	Atomic positions (x, y, z) ($a = b = 3.15 \text{ \AA}$ and $c = 12.30 \text{ \AA}$)		
	MoS_2		MoSO
	Present work	Previous experiment [17]	
Mo ($2c$)	1/3, 2/3, 1/4	1/3, 2/3, 1/4	1/3, 2/3, 0.2615
S ($4f$)	1/3, 2/3, z_S ($z_S = 0.6214$)	1/3, 2/3, z_S ($z_S = 0.621$)	1/3, 2/3, z_S ($z_S = 0.6321$)
O ($4f$)	—	—	1/3, 2/3, z_O ($z_O = 0.6435$)

3. Results and Discussions

3.1. Structural properties

Firstly, we investigate the structural optimization results of both MoS₂ and MoSO systems. Table 1 presents optimized Wyckoff positions of both systems. As the results, the present calculation well reproduces the experimental results [17]. On the other hand, the O substitution at S sites slightly modifies the atomic positions. The z-axis Wyckoff position of all atoms increases. The result indicates the atomic bonding modifications. Figure 2 presents the schematic illustration of shifts of the z-axis atomic positions, which is visualized by XCrysDEN [33]. We find that the S–Mo–S bond angle of 82.012° decreases to 73.921° as the O–Mo–S bond angle. The previous report shows the S–Mo–S bond angle of 82.207° [17]. Hence, the structural properties of MoS₂ system in the present work is comparable with the previously reported results. We suggest that the structural properties of MoSO system can also be used to predict experimental results.

3.2. Electronic properties

Figure 3 presents the band structures of bulk MoS₂ and MoSO systems. We find the indirect $E_g = 0.94$ eV from Γ to 0.8 Σ points in MoS₂ system, which is lower than that of the previous experimental report, i.e., 1.29 eV from reflectivity [5] and L spectra [34]. The underestimated E_g can be caused by the limitation of GGA in describing the exact Kohn-Sham band structure within GGA [35]. We also find the direct transition at K point of 2.66 eV. The band structures show the delocalized states along Γ –M and K– Γ points, indicated by the delocalized valence-band feature. However, the localized states are found along M–K points, which are indicated by the localized valence-band feature.

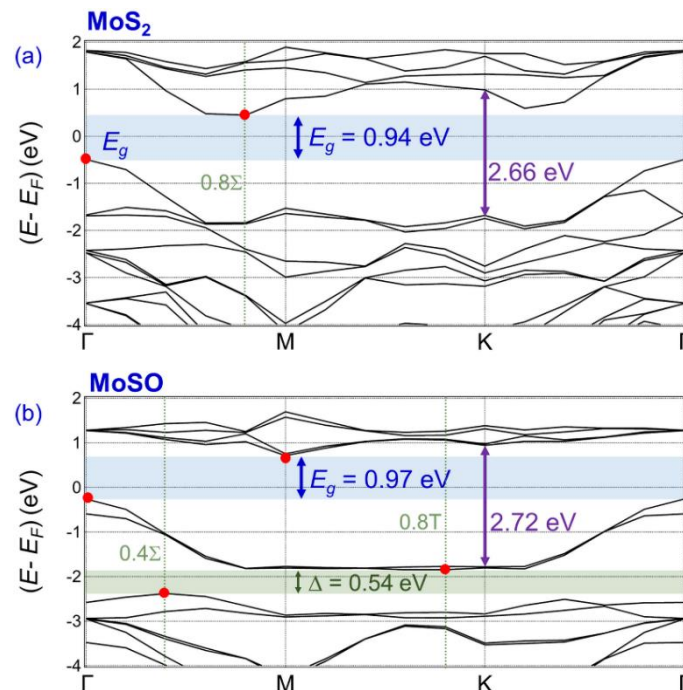


Figure 3. Band structures of bulk (a) MoS₂ and (b) MoSO systems. Indirect and direct bandgaps are assigned. The blue shades denote the minimum bandgap (E_g) of both systems, while the green shade denotes the bandgap (Δ) within the valence band of MoSO system. The red dots denote the points involved in the indirect band transitions. Fermi energy level (E_F) is denoted at 0 eV.

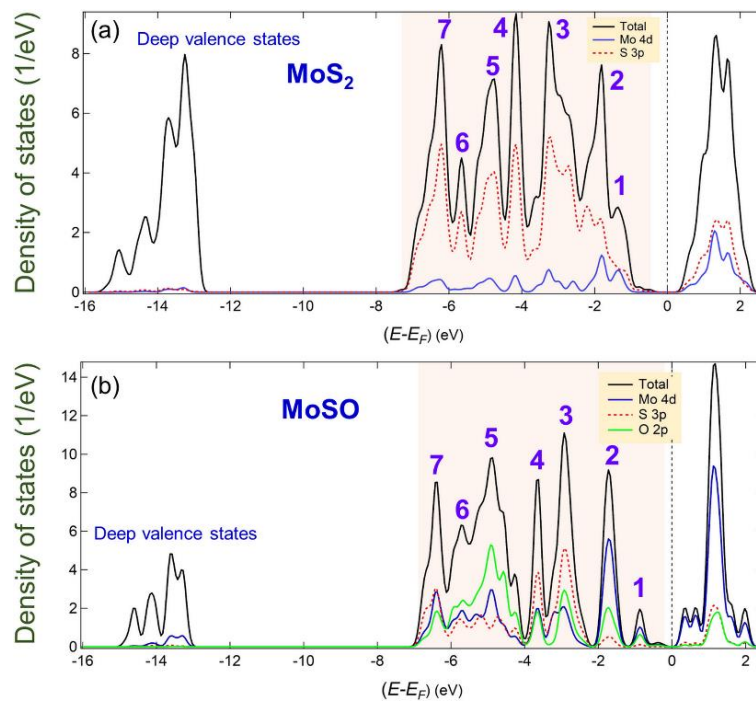


Figure 4. Total and partial density of states of bulk (a) MoS₂ and (b) MoSO systems. Near Fermi energy level (E_F), the width and peaks of valence band are assigned by the soft red shade and numbers (1 to 7), respectively. Fermi energy level (E_F) is denoted at 0 eV.

MoSO system shows the indirect E_g of 0.97 eV from Γ to M points. This result shows that the substitution slightly increases E_g . However, it changes the transition point at the conduction band minimum from 0.8 Σ path to M point. This result is different with that of the monolayer MoS₂, as revealed by the previous first-principle calculation. The O substitution at S site in the monolayer MoS₂ does not only change the transition point but also changes E_g type from direct to indirect types [36]. We also find the in-valence-band interband transition (Δ) of 0.54 eV from 0.4 Σ to 0.8T paths. The presence of Δ is induced by the localized valence band feature at around -2 eV. The band structure also indicates the localized states for all k -points. Hence, the substitution significantly promotes more localized valence states.

Figure 4(a) shows the total and partial density of states (DOS) of bulk MoS₂ system. We find that MoS₂ system shows E_g around Fermi energy level (E_F), which mainly promoted by Mo 4d, followed by S 3p states. Near E_F , the valence band is dominated by S 3p. This result approves the theoretical electron configuration of S²⁻: [37] 3s² 3p⁶. However, the conduction band is dominated by both Mo 4d and S 3p states. The presence of Mo 4d states in the conduction band approves the partially occupied Mo²⁺ 4d² orbital. The presence of S 3p states may be resulted due to the strong hybridization between Mo 4d and S 3p states. We also assign seven peaks (1-7) in the valence band. All seven peaks overlap each other, indicating the dominant delocalized states, as presented in figure 3(a).

Figure 4(b) presents the total and partial density of states (DOS) of bulk MoSO system. We find that E_g is mainly promoted by Mo 4d and O 2p states. The conduction band is dominated by Mo 4d states approving the valence electron configuration of Mo²⁺ 4d². Interestingly, more complex features in the valence band are observable. Seven peaks in the valence band of MoSO system are assigned. We find that the peaks 1-4 are localized above -4 eV, which are in contrast to that of MoS₂ system. This result confirms the localized bands in figure 3(b). However, the peaks 5-7 below -4 eV are delocalized, as also found in MoS₂ system in figure 4(a). Hence, the substitution significantly modifies the valence band structure alignment of MoS₂ system near E_F .

3.3. Optical properties

Figure 5 presents the energy-dependence of σ_1 bulk MoS₂ and MoSO systems. In MoS₂ system, we find different curves of σ_1 along xy plane (σ_{1xy}) and z axis (σ_{1z}), as shown in figure 5(a). This result indicates optical dichroism, which has also been found in α -PbO [38] and TiO₂-based systems [39]. The optical dichroism indicates that the bulk MoS₂ is anisotropic, as previously shown by different dielectric constants along different directions (xy -plane and z -axis) [40]. In region A, two major peaks are observed for both xy -plane and z -axis directions. The peak shape is similar, indicating the weak dichroism in this region. In region B, we find high peak at around 5.7 eV along z axis, while xy plane does not show any significant peak. This result indicates the strong dichroism in the region. Between both regions A and B, a major peak at around 4.5 eV is found for xy plane. Figure 5(b) also shows the optical dichroism in MoSO system. In contrast to MoS₂ system, both regions A and B show strong optical dichroism along both xy -plane and z -axis directions, indicated by the significant different shapes between σ_1 of both systems. Notably, region A of MoSO system shows the multiple peaks at which their intensities are lower than that of the double peaks of MoS₂ system. We suggest that the presence of multi-peaks are induced by the transitions between the localized valence states near E_F to conduction states. Furthermore, σ_1 width (~ 7.16 eV) of MoSO system is wider than that of MoS₂ system (~ 6.39 eV), indicating the wider absorption range in MoSO system.

Figure 6 presents L of bulk MoS₂ and MoSO systems. In the L spectrum, the peaks represent plasma energy as the oscillatory response of delocalized valence electrons due to the plasma excitation. This excitation is induced by the inelastic scattering between the atomic electrons in a material and an electron beam. The excitation results when the speed of electron beam is higher than the Fermi velocity [41]. Figure 6(a) shows two L peaks at 11.3 and 12.1 eV along xy -plane (L_{xy}) in MoS₂ system. At the lower energy level of 10.9 eV along z axis, we find a higher peak compared to both other peaks. All these peaks imply the plasmonic states. This result indicates the dichroism of plasmonic state. By considering the difference between energy levels for the peaks at 11.3 and 10.9 eV, the dichroism is 0.4 eV. Figure 6(b) shows two L peaks at 10.9 and 11.3 eV along xy -plane (L_{xy}) in MoSO system. A peak at 9.5 eV is also found along z axis (L_z). By considering the difference between energy levels for the peaks at 10.9 and 9.5 eV, the dichroism of MoSO system is 1.4 eV, which indicates that MoSO system exhibit the higher plasmonic-state dichroism than that of MoS₂ system. Additionally, plasmonic-state width is also increased by the substitution (green shade).

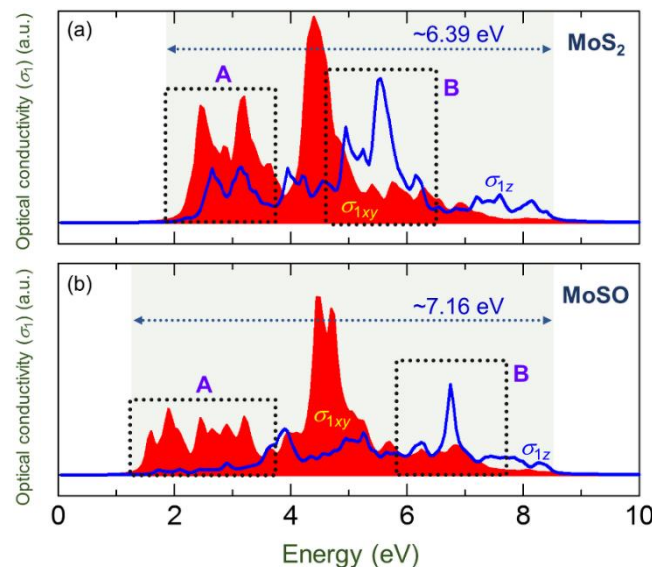


Figure 5. Optical conductivity (σ_1) of bulk (a) MoS₂ and (b) MoSO systems. Red face and blue line denote σ_1 along xy plane (σ_{1xy}) and z axis (σ_{1z}), respectively. The σ_1 width is assigned by the green shade. The dashed squares (A and B) emphasize the significant σ_1 modification.

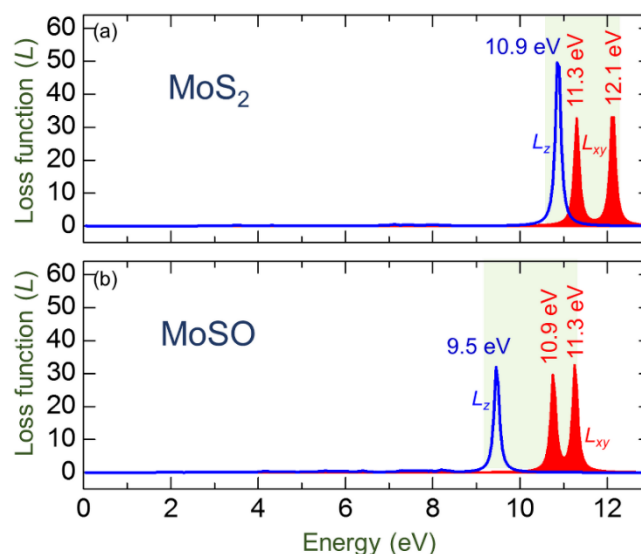


Figure 6. Electron-energy loss function (L) of bulk (a) MoS_2 and (b) MoSO systems. Red face and blue line denote L along xy plane (L_{xy}) and z axis (L_z), respectively. The green shade denotes the plasmonic-state width.

4. Conclusions

We have studied the electronic and optical properties of bulk MoS_2 and MoSO systems using the GGA. As the results, the structural properties show that the O substitution modifies the z -axis Wyckoff positions of atoms as well as the bond angles between cations (Mo) and anions (O or S). The substitution also tunes the conduction band minimum and provokes more localized valence states compared to that of MoS_2 system. We find that the localization is more pronounced above -4 eV, which mainly comes from Mo $4d$ and S $3p$ states. On the other hand, O $2p$ states are dominant below -4 eV. However, there is only a small shift of E_g due to the substitution. Furthermore, the optical properties show that both systems show the strong optical dichroism of σ_1 and L . The substitution significantly increases the σ_1 width and decreases the plasmonic-state energy levels. This study shows the significant tuning of the electronic and optical properties of bulk MoS_2 system by the O substitution at S sites.

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