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The Effect of Annealing Temperature of ZnO Compact Layer and TiO₂ Mesoporous on Photo-Supercapacitor Performance

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Abstract. Photo-supercapacitor is a combination of solar cells and supercapacitor which intensively being developed. Photo-supercapacitor performance is influenced by the efficiency of solar cells and storing and releasing capacity of supercapacitors. Any types of either solar section or supercapacitor sections could be used. In the DSSC solar cell, one of the influential variables is the performance of the photoanode. The photoanode with semiconducting metal oxides play a role in charges mobility and light absorption process, which are influenced by crystal morphology and structures. The common metal oxides used are ZnO and TiO₂ which show high electron mobility, wide band gap, and good optical properties. This work is designed to investigate the effect of annealing temperature of the composite layer of ZnO and mesoporous TiO₂ on structure, morphology, optical absorption, and photo-supercapacitor performance. The ZnO compact layer was deposited onto the FTO substrate by a spin coating method with various annealing temperature. The mesoporous TiO₂ layer was deposited on top of the ZnO compact layer by means of screen printing method. The construction of photo-supercapacitor model comprises DSSC and BaTiO₃-PVDF symmetric supercapacitor which integrated by using aluminum substrate. Characterization was done using XRD, SEM, UV-Vis, and I-V solar simulators for the performance of photo-supercapacitor.

Keywords: Photo-supercapacitor, annealing temperature, ZnO compact layer, mesoporous TiO₂.

1. Introduction

Generation and storage electrical energy device such as photo-supercapacitor have been the subject of many research interest in recent years [1–4]. Photo-supercapacitor is interesting to be developed because of it's perform the combination of solar cell and supercapacitor simultaneously, so other conventional devices and size can be reduced [2–6]. One of the photo-supercapacitor is a combination of dye-sensitized solar cell (DSSC) and supercapacitor based on electric double layer capacitors (EDLC) [1,4,6–10]. The combination of those two devices may affect not only of each section but also the interconnection between them.

Several sections which influenced the DSSC efficiency are metal oxide at photoanode, photosensitizer (dye), electrolyte, and counter electrodes [11–14]. The photoanode electrode with metal



oxide effect on collecting and transferring electrons from the dye to transparent conductive oxide (TCO). There are many studies have been conducted to optimize metal oxide materials with various n-type semiconductors such as TiO₂, ZnO, and SnO₂ [12,15–19]. TiO₂ is the main candidate for DSSC photoanode because of its wide band gap, thermal stability, non-toxic, and stable under sunlight [12,20,21]. In addition, TiO₂ in nanometer size and porous morphology not only good for dye absorption but also has electron interfacial recombination faster than other metal oxides [22–24]. However, TiO₂ shows quite low electron mobility, which has an impact on conversion efficiency [25]. One way to increase electron mobility in TiO₂ needs to be modified its morphology or composites with other metal oxides.

The morphological structure of mesopore TiO₂ apply as photoanodes DSSC can improve conversion efficiency, but the interface between mesopore TiO₂ with TCO such as indium tin oxide (ITO) or fluorine tin oxide (FTO) is low [26,27]. It is necessary to modify the photoanode structure by introducing a compact layer such as ZnO between mesopore TiO₂ and TCO. In addition, ZnO can increase electron mobility in DSSC photoanodes because ZnO has high electron mobility (115-155 cm²V⁻¹s⁻¹) [28]. Many works have been performed in combining TiO₂ with ZnO with core-shell structure [29], composite TiO₂ nanoparticles with ZnO nanorod [18], ZnO nanowire / TiO₂ core-shell nanosheet [30], compact layers TiO₂ [31], and ZnO nanoflower composites with TiO₂ nanorod [28]. Those efforts aimed to enhance the DSSC efficiency.

The efficiency energy conversion is expected to increase by combining TiO₂ mesopore with compact layer ZnO at DSSC's photoanode. In the long run, it affects the performance of photo-supercapacitor. One type of EDLC supercapacitor by using BaTiO₃ is exciting to be studied. BaTiO₃ is a ferroelectric ceramic which exhibits an excellent of pyro, piezo, and capacitance properties [32,33]. However, since ceramic is a brittle and rigid material which need to be combined with polymers such as polyvinylidene fluoride (PVDF) which has a flexible, easily composited, and has good capacitance [32]. One type of photo-supercapacitor with the best design is three photo-supercapacitor electrodes exhibit maximum voltage, high power, and energy density, have high efficiency, fast charge and discharge, and durable [1,4,7,34]. The interface between mesoporous TiO₂ with TCO and electron mobility in DSSC photoanodes for mesopore TiO₂ is too low to be implemented. We report three gates photo-supercapacitor performance by implementing a compact layer ZnO with mesoporous TiO₂ and ZnO rods combined with symmetric supercapacitor PVDF-BaTiO₃.

2. Methods

An FTO-glass substrate was firstly cleaned following a standard protocol. Compact layer ZnO was deposited on FTO-glass substrate using a spin coating method. ZnO solution was prepared using Zinc acetate dehydrate (Zn(CH₃COO)₂·2H₂O) Sigma-Aldrich 99% in ethanol and added 0.5 mL Monoethanolamine (MEA) Sigma-Aldrich 99% as a stabilizer. Mesoporous TiO₂ was layered on compact layer ZnO by screen printing method. Mesoporous TiO₂ paste was prepared by TiO₂ (P25) Sigma-Aldrich 99%, Sodium dodecyl sulfate (SDS), Polyethylene glycol 6000 (PEG) in distilled water. The ZnO solution was deposited on the mesoporous TiO₂ layer by a spin coating method. FTO/ZnO/TiO₂-ZnO was dried at 100 °C followed by annealed for a various temperature of 300, 350, and 400 °C for 2 hours. Photoanode electrode FTO/ZnO/ TiO₂-ZnO with the codes for each annealing temperature were ZnTi 300, ZnTi 350, and ZnTi 400 respectively. The counter electrode was prepared by carbon tape at aluminum foil substrate, then heated at 250 °C for 30 min.

Supercapacitor was designed by symmetry PVDF-BaTiO₃ on aluminum foil substrate with hydrogel electrolyte and Polytetrafluoroethylene (PTFE) as a separator. A PVDF-BaTiO₃ paste was deposited on an aluminum substrate by slip casting. Hydrogel electrolyte was prepared by polyvinyl alcohol (PVA), potassium hydroxide (KOH), and potassium iodide (KI) in distilled water. Photo-supercapacitor was designed by modifying the design as described by Lau et al. [1] is represented by Figure 1. DSSC was constructed by sandwiching photoanode and counter electrode with quercetin and beta-carotene dye and I⁻/I³⁻ electrolyte. DSSC was combined with symmetric supercapacitor become photo-supercapacitor.

The structure and morphology of the nanostructured photoanodes were characterized by the use of a scanning electron microscopy (SEM Merk FEI, Type: Inspect-S50). The X-ray diffraction (XRD) pattern was recorded using an XRD E'Xpert Pro PANalytical with Cu- α ($\lambda = 1.54060 \text{ \AA}$, 35 mA, dan 40 kV) at 2θ in the range of $10\text{-}80^\circ$. Optical properties of the photoanode film were scanned at a wavelength of 200-800 nm with UV-Vis UV1700. Photovoltaic and photo-charging were measured for current vs. time (I-t) and voltage vs. time (V-t) by Keithley 6515B Electrometer under 20 mW/cm^2 illumination.

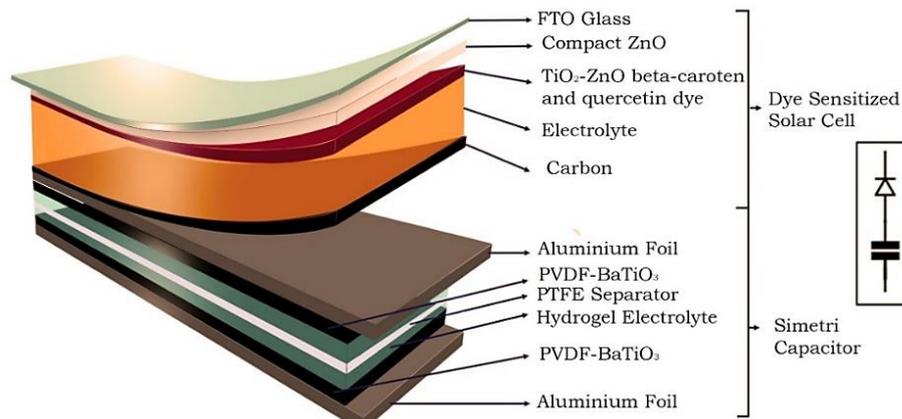


Figure 1. Schematic of photo-supercapacitor device modified from ref. [1].

3. Results and Discussion

XRD pattern of the photoanode electrode with various annealing temperature shows in Figure 2. All the Bragg peaks in the XRD pattern can be indexed well to the anatase TiO_2 phase COD 2310710 [35], ZnO phase COD 9004178 [36], and FTO JCPDS card no. 01-0625 [37]. Anatase TiO_2 has a tetragonal crystal system with $I4_1/amd$ space group and ZnO crystallized under $P6_3mc$ space group. The variation in the annealing temperature does not look significant for the ZnO phase due to compact layer ZnO at the bottom layer [37,38].

Rietveld refinement was calculated for the composite of ZnO and TiO_2 (later indicated as ZnTi) 400 XRD pattern using GSAS software shows in Figure 3. It is shown that TiO_2 lattice parameters fall to $a = b = 3.750 \text{ \AA}$ and $c = 9.5053 \text{ \AA}$. The obtained lattice parameters of ZnO phase is $a = b = 3.2494 \text{ \AA}$ dan $c = 5.2038 \text{ \AA}$. Reliability Rietveld refinement between calculation and experiment data ZnTi 400 are $R_p = 0.1999$, $R_{wp} = 0.3005$, dan $\chi^2 = 1.413$. Figure 3 (b) and (c) shows a crystal structure of TiO_2 and ZnO respectively using VESTA software.

XRD pattern of supercapacitor electrode with PVDF- BaTiO_3 is shown in Figure 4. All the diffraction peak in the XRD pattern is appropriate with PVDF phase CSD LILXIN05, BaTiO_3 phase COD 4124842, and Aluminum foil. BaTiO_3 phase fit to cubic crystal structure with $Pm\text{-}3m$ space group and PVDF phase has orthorhombic crystal system with $Cm2m$ space group. Rietveld refinement of XRD pattern has been done using GSAS software. Kisi parameter of BaTiO_3 phase is $a = b = c = 4.025419 \text{ \AA}$ and PVDF phase are $a = 8.5822$, $b = 4.9102$, $c = 2.559 \text{ \AA}$. Good reliability Rietveld refinement are $R_p = 0.2741$, $R_{wp} = 0.1906$, dan $\chi^2 = 1.483$. Crystal structure of BaTiO_3 and PVDF shows in Figure 4 (b) and (c) respectively. Equation 1 represents the Scherrer formula used to calculate the crystal size of BaTiO_3 is 41.3 nm .

$$d = \frac{k\lambda}{\beta \cos \theta} \quad (1)$$

where d is crystal size, k is shape factor, λ is wavelength Cu- α , β is full width at half maximum (FWHM), and θ is the Bragg position.

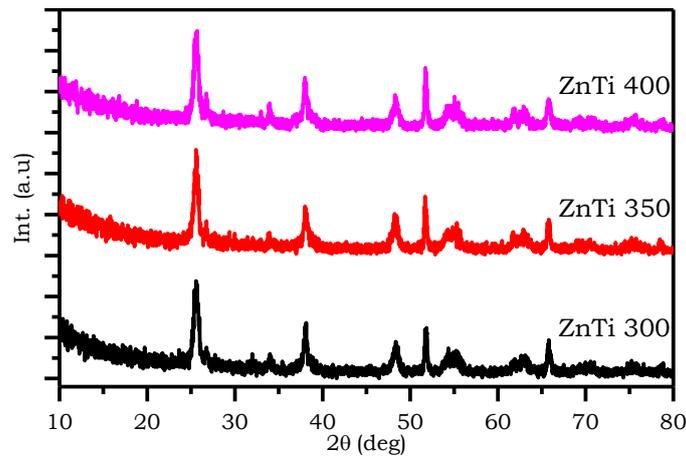


Figure 2. XRD pattern of photoanode electrode with various annealing temperature

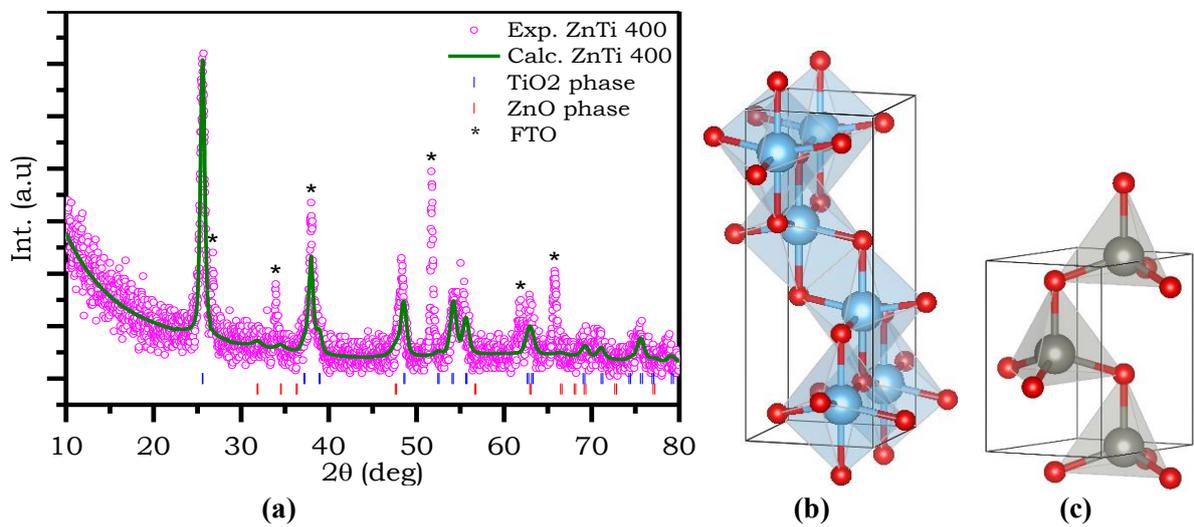


Figure 3. Rietveld refinement of ZnTi 400 using GSAS (a), crystal structure of TiO₂ (b) and ZnO (c)

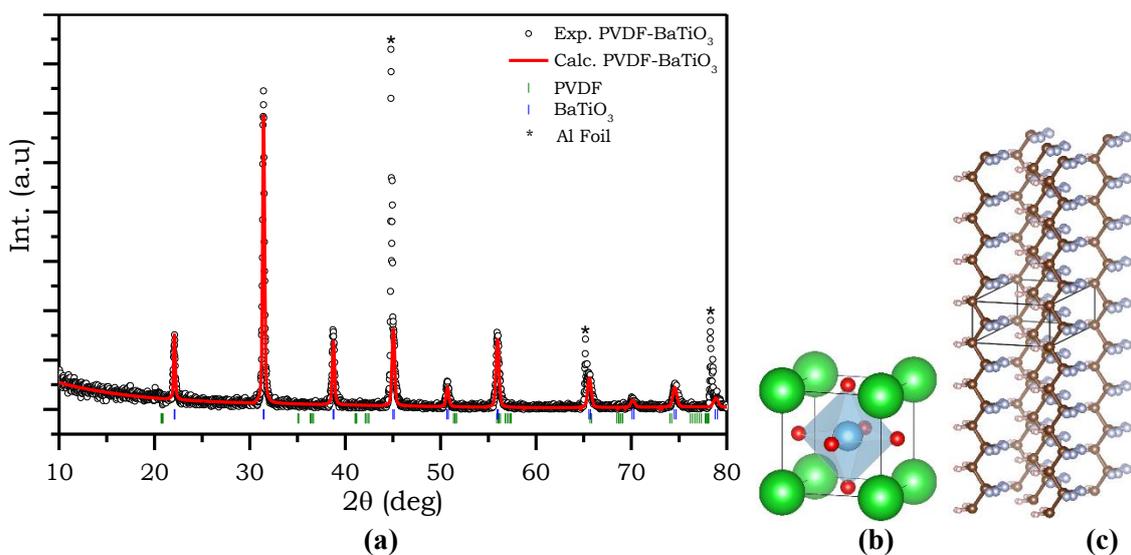


Figure 4. XRD pattern with Rietveld refinement using GSAS of PVDF-BaTiO₃ (a), crystal structure of BaTiO₃ (b), and PVDF (c)

Figure 5 shows the top-view SEM of the photoanode electrode at the various annealing temperature. The morphology of TiO₂ is mesoporous and ZnO growth to nanorods. The annealing temperature is linearly influencing crystal growth of ZnO nanorods diameter is shown in insert picture Figure 5. Crystal growth of ZnO at compact layer increase interfacing between mesoporous TiO₂ and FTO substrate to provide electron mobility increase [31,39,40].

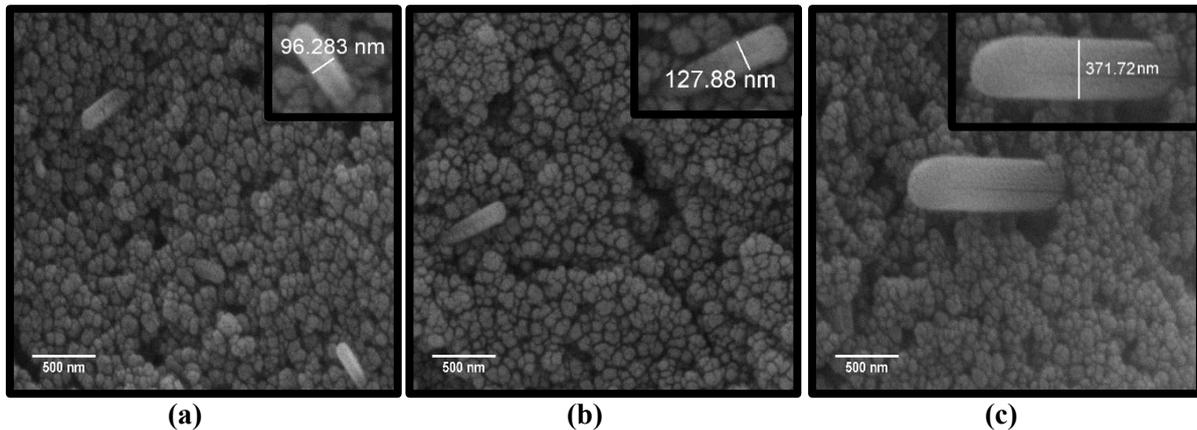


Figure 5. SEM image of photoanode electrode with various annealing temperature 300 (a), 350 (b), and 400 °C (c)

Figure 6 (a) shows SEM image top-view of supercapacitor electrode (PVDF-BaTiO₃) on aluminum foil substrate. PVDF-BaTiO₃ has a uniform particle size. The porosity analysis has been done using previous work [41] by approximation area under the curve that density porosity is 0.70. The porous morphological of supercapacitor electrode can increase the capacitance of the supercapacitor [42].

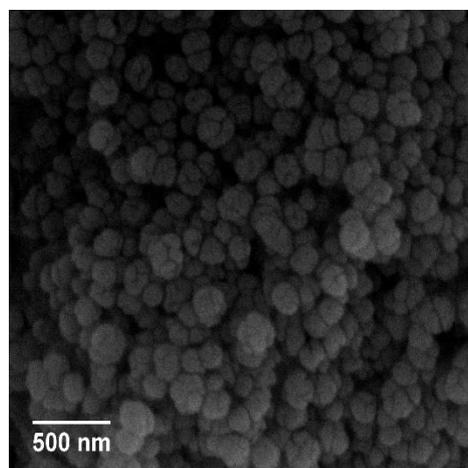


Figure 6. SEM image of PVDF-BaTiO₃

The optical properties of the photoanode electrode with various layer and the annealing temperature were characterized using UV-Vis at a wavelength between 200-800 nm shows in Figure 7a and Figure 8a respectively. The absorbance at semiconductor material is related to the energy used by an electron to excite from valence band to conduction band [43]. The direct transition of the electron is expressed by Equation 2 [11].

$$(\alpha h\nu) = A(h\nu - E_g)^{1/2} \quad (2)$$

where α , $h\nu$, E_g , and A are absorbance coefficient, photon energy, band gap, and constant respectively. Tauc plot method was used to analyze band gap by intercept at x -axis from graph $h\nu$ vs. $(\alpha h\nu)^2$ [44].

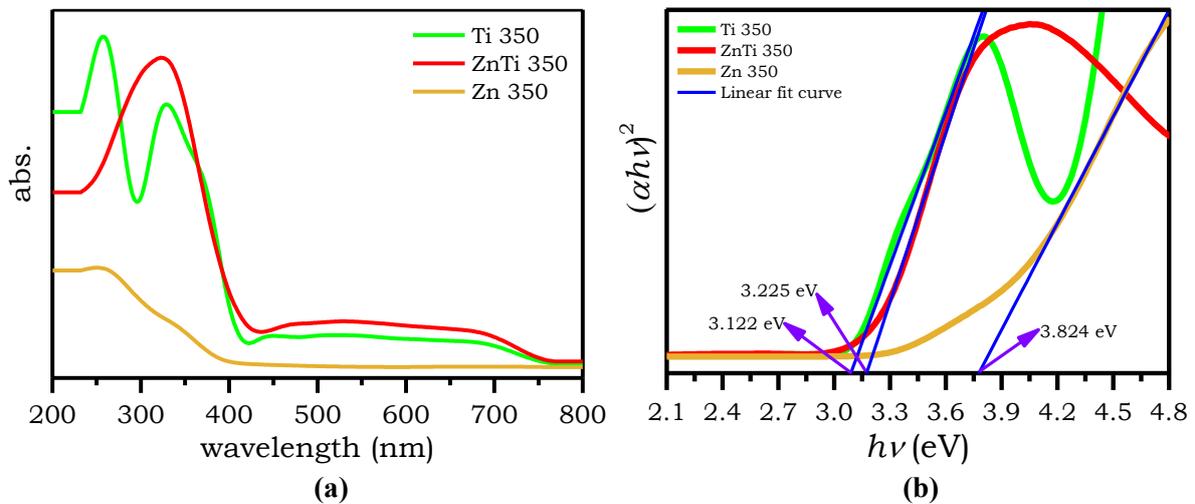


Figure 7. Absorbance spectra of TiO₂ without compact layer ZnO (Ti 350), Compact layer ZnO (Zn 350), and TiO₂-ZnO with compact layer ZnO (ZnTi 350) on FTO substrate (a), and Tauc plot (b)

Figure 7a represents spectral absorbance of a various layer that indicates decreasing absorbance coefficient at wavelength 400-800 nm. This phenomenon indicates that photon energy was used by the electron to excite from valence band to conduction band. The magnitude energy corresponding to the band gap of each semiconductor. The compact layer ZnO has $E_g = 3.824$ eV [45] and TiO₂ layer without compact layer ZnO has $E_g = 3.122$ eV [46,47]. TiO₂-ZnO layer with compact layer ZnO has $E_g = 3.225$ eV.

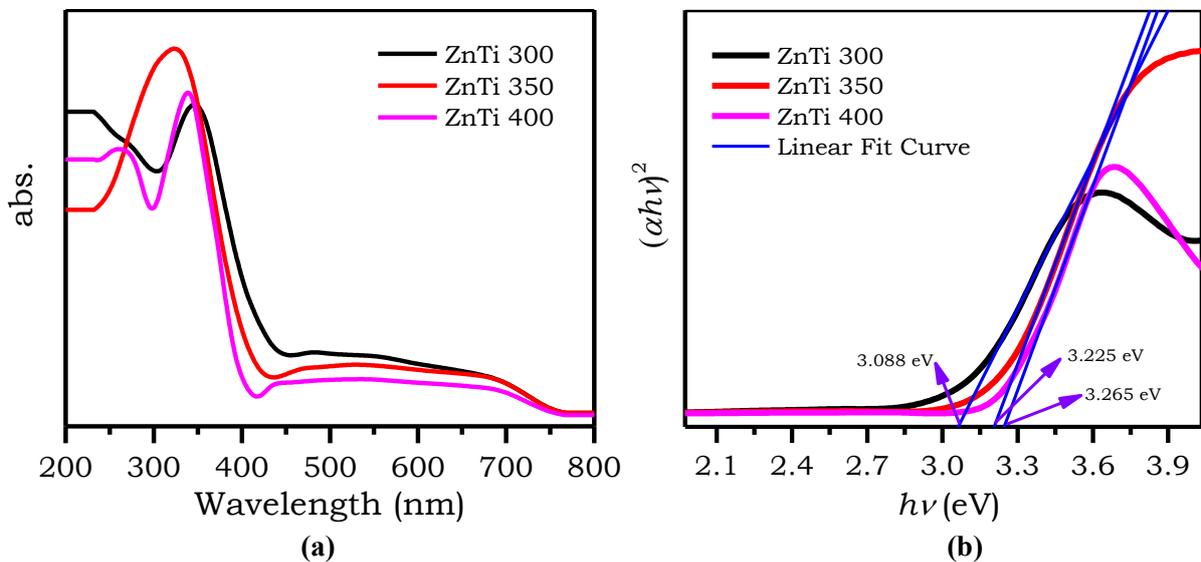


Figure 8. Absorbance spectra of photoanode electrode (ZnTi) with various annealing temperature (a) and Tauc plot for band gap calculation (b)

Figure 8a shows the absorbance spectral photoanode electrode of ZnTi with various annealing temperature. There is a shift of decreasing absorbance coefficient from wavelength 450 nm becomes 400 nm. The annealing temperature is increased band gap of ZnTi is showed by Figure 8b. Increasing band gap indicate crystal growth of compact layer ZnO.

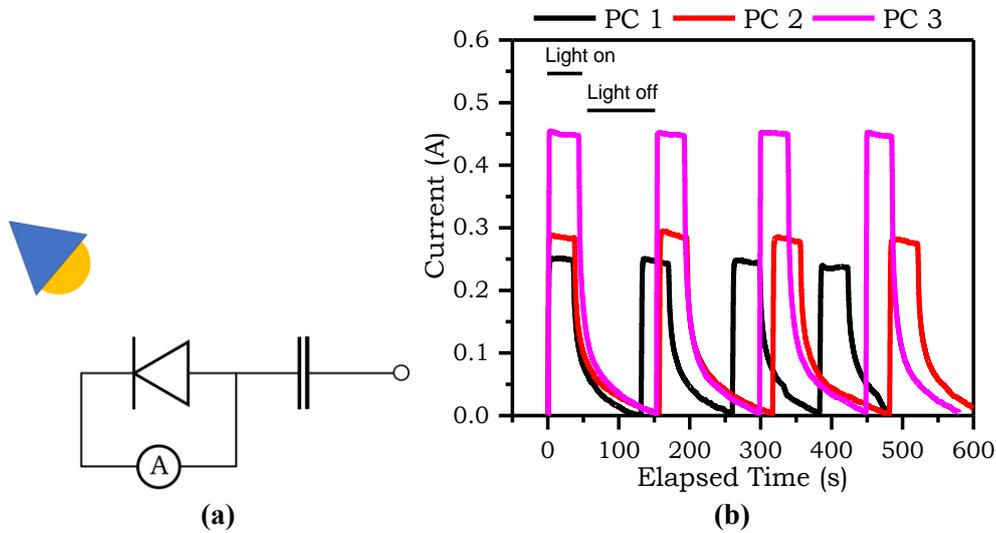


Figure 9. Schematic of photocurrent measurement (a) and I-t graph of photo-supercapacitor (b)

Current vs. elapsed time (I-t) of photo-supercapacitor was measured with the scheme is showed by Figure 9a under 20 mW/cm^2 illumination. Figure 9b shows the photocurrent characteristic of photo-supercapacitor (PC) with various annealing temperature $300, 350,$ and $400 \text{ }^\circ\text{C}$ is coded by PC 1, PC 2, and PC 3 respectively. The annealing temperature influence I_{sc} (short-circuit current) of photo-supercapacitor. This corresponds to the crystal growth, energy gap, and the interface between the substrate and metal oxides ZnO and TiO_2 . The compact layer ZnO increase interface between mesoporous TiO_2 and FTO substrate that makes electron mobility and photocurrent increase [31,39,40].

Photo-charge of photo-supercapacitor was measured with schematic is showed in Figure 10a. Current vs. elapsed time (I-t) at PC 3 with capacitance supercapacitor 5nF with active area $2 \times 1.5 \text{ cm}^2$ is represented by Figure 10b. The constant current charge supercapacitor is 0.423 A . Figure 10c shows photo charge and discharge of supercapacitor is showed by voltage vs. elapsed time (V-t) graph. The charge and discharge of supercapacitor correspond to characteristic decrease exponentially.

Charge-discharge performance of PC 3 corresponds to the transient voltage at series RC circuit [48,49]. Equation 3 express charging while Equation 4 express discharge of the capacitor. Figure 11 shows the voltage vs. elapsed time (V-t) graph and the fitting curve. Voltage maximum V_{max} of charging supercapacitor is 40.26 mV with the time constant $\tau = 15.04 \text{ s}$ and potential breakdown is 10 mV that is larger than those in previous work [2]. This is related to the time of charge and discharge in photo-supercapacitor.

$$V_c(t) = V_{\max} \left(1 - \exp\left(-\frac{(t-t_0)}{\tau}\right) \right) \quad (3)$$

$$V_c(t) = V_{\max} \exp\left(-\frac{(t-t_0)}{\tau}\right) \quad (4)$$

where V_c is time-dependent capacitor voltage, V_{max} is saturated voltage, t is time, and τ time constant.

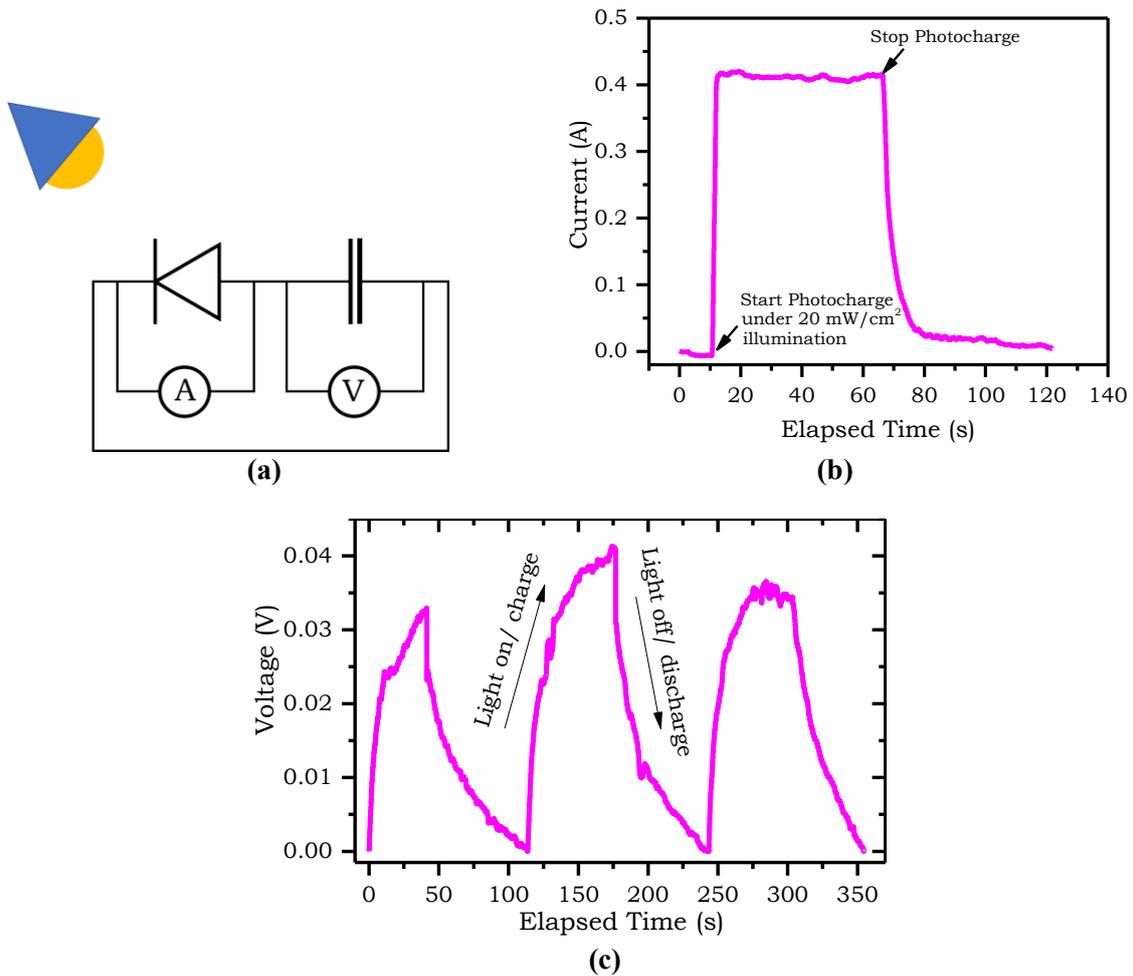


Figure 10. Schematic photocurrent and charge-discharge measurement (a), photocurrent performance (b), and charge-discharge (c) of photo-supercapacitor (PC 3)

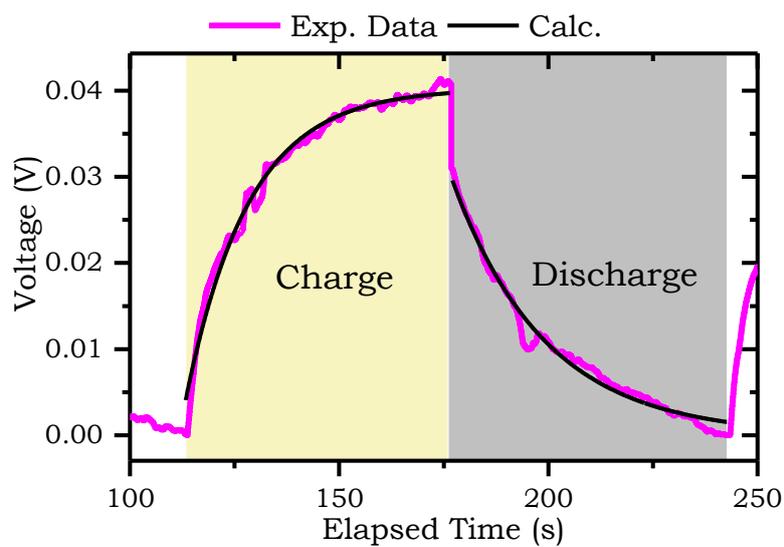


Figure 11. Charge-discharge and fitting curve of capacitor performance for charge and discharge

4. Conclusion

The temperature annealing affects various aspects of photo-supercapacitor performance. This annealing temperature increases the activation energy for crystal growth which improves the interface between the FTO and TiO₂ mesopore. This condition increases the E_g and I_{sc} as well as the stability of the current generated in the photocurrent characteristic.

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