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## Addition of feldspart minerals as a substitution material on ceramics clay manufacture

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## Addition of feldspart minerals as a substitution material on ceramics clay manufacture

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**Abstract.** The use of clay minerals as the main raw material for pottery and sand additives has many weaknesses, among others, easy to crack, have high water absorption and low compressive strength. Therefore it is necessary to develop a better method of making traditional ceramics by providing additional materials such as mineral feldspar. Mineral feldspart can be used as a ceramic raw material because it is functional as filler and glass forming phases. Clay and mineral feldspart raw materials were analyzed using XRF method, followed by mixing with the composition of clay soil mixture as much as 65% compared to feldspart by 35%, printed with the shape of cube size 50x50x50mm. Drying in oven with temperature 110°C for 8 hours. Combustion in muffle furnace at temperature variation is 800°C, 900°C, 950°C and 1000°C and holding time is 2 hours, 3 hours, 4 hours and 5 hours. Test of compressive strength and porosity is done to see the physical properties of pottery ceramics and characterization by using XRF and XRD. The result of compressive strength test shows the temperature of 1000°C with 3 hours reaches the optimum value of compressive strength which is 9.52 Mpa, while the porosity value reaches the optimum point at heating temperature 1000°C with time of 5 hours that is equal to 16,19%. Characterization of XRF shows that ceramic pottery contains SiO<sub>2</sub> of 64.074%, and Al<sub>2</sub>O<sub>3</sub> of 25.287% according to XRD characterization indicates the dominant phase formed by mineral quartz or silica oxide (SiO<sub>2</sub>).

### 1. Introduction

Ceramic clay (earthenware) is a part of ceramics, this term in the modern context encompassing a very large inorganic material, ceramic clay containing nonmetallic and metallic elements made with various manufacturing techniques (the traditional ceramic is clay) that is burned at a certain temperature up to the chemically has been transformed into a permanent form. The mechanical properties of ceramics clay are usually a strong material, hard but fragile and easy to broken. Furthermore ceramic clay corrosion resistance, have low density and high melting point. The quality of ceramic clay affected by raw materials used and temperature heating [1]. Main raw material of ceramic clay are clay or clay minerals. This is according to Jone et.al (2015), the quality of pottery depends on the physical properties of clay minerals as the main raw material and sand as an additional raw material [2]. In the wet conditions clay mineral have plastic properties, but in a dry conditions become hard and when it burned become solid and strong [3]. The clay composed of oxide compounds such as SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> which are materials for the skeleton and form the ceramic body. Application of clay minerals as a main material for ceramic clay and sand as additives has many weaknesses, such as easy to crack, have high water absorption and have low compressive strength [4]. Therefore it is necessary to develop a better method of making traditional ceramics by providing additional materials such as mineral feldspart. Mineral feldspart can be used as ceramic material because it is functional as filler and glass forming [5]. Feldspart stone is the raw material needed for the production of ceramic tiles, household appliances, glazes, etc., because when it burned at temperature 1120-1300°C give effect of porosity change, bulk density and water absorption on pottery [6]. The industrial of ceramics and glass are the



main consumers of feldspar. In the manufacture of glass, feldspar used as a flux agent and as a source of alumina ( $\text{Al}_2\text{O}_3$ ), alkali ( $\text{Na}_2\text{O}$  and  $\text{K}_2\text{O}$ ), and silica [7]. When ceramic clay mixed with feldspar minerals are molded and burned the microstructure will be change from the pottery product to mullite ( $3\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2$ ) and insoluble quartz crystal ( $\text{SiO}_2$ ) that functioning into glass phase formation [8]. Based on the research of Indiani et.al (2009) the use of feldspar minerals in the manufacture of ceramics are useful as a melting agent to form glass phase at temperatures below  $1100^\circ\text{C}$  [9]. Research conducted by Rivai (2016), reported that after passing temperature  $600^\circ\text{C}$  physical and chemical clay will changes to be hard, solid and can't be destroyed by water [1]. This change is called ceramic change. This conclusion is reinforced by research conducted Ratri et.al (2008), at temperatures between  $400$ - $600^\circ\text{C}$  chemically bonded water and other substances in clay minerals will evaporate [4]. After the temperature more than  $600^\circ\text{C}$  clay will change into a solid, hard and permanent mineral. In the range of temperature  $900^\circ\text{C}$  the mullite phase will be formed. Mullite is a very stable compound that makes ceramic clay are hard, compact, and solid. So the ceramic clay has a high compressive strength and low absorption

## 2. Methode

In this research, raw material of clay used from Rejosari, Natar, Lampung Selatan, and the feldspar mineral from Nyukang Harjo, Lampung Tengah. Those raw materials were analyzed using XRF method as seen at the Table 1.

**Table 1.** XRF analysis results of the raw materials ceramic clay

Elements	Chemical Composition (%)	
	Clay	Feldspar
$\text{SiO}_2$	50.55	67.30
$\text{Al}_2\text{O}_3$	32.61	14.02
$\text{Fe}_2\text{O}_3$	2.43	2.24
$\text{K}_2\text{O}$	0.53	4.59
$\text{TiO}_2$	0.19	0.13
$\text{CaO}$	0.77	2.25
$\text{P}_2\text{O}_5$	0.14	3.20
$\text{NiO}$	0.12	0.020
$\text{SO}_3$	0.018	0.069
$\text{MnO}$	0.006	0.044
$\text{MgO}$	0.15	0.059
$\text{Cr}_2\text{O}_3$	0.003	0.015
$\text{SrO}$	0.002	0.020
$\text{PbO}$	0.004	0.006
$\text{CuO}$	0.001	0.007
$\text{Na}_2\text{O}$	1.05	0.005

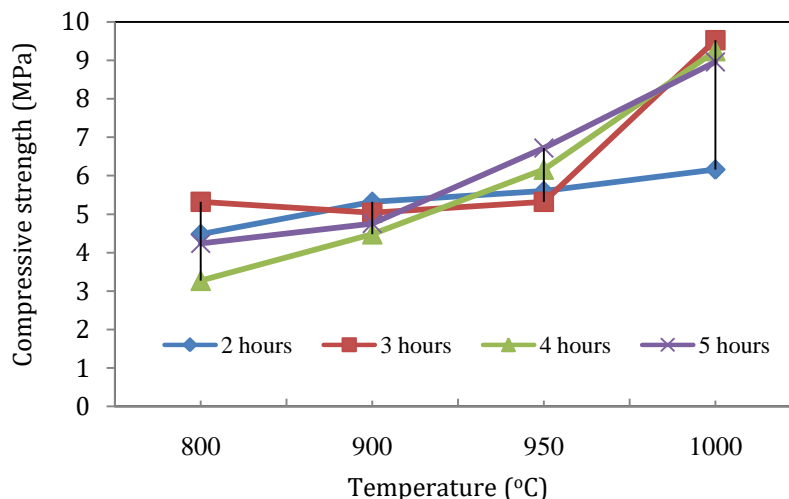
Next, ceramic clay material is cleaned from impurities, dried in the oven with temperature  $110^\circ\text{C}$  or with sun for 3 hours and sieved with 20 pass mesh sifting.

Be pondered according to the mixing composition between clay and feldspar with ratio clay: feldspar = 65%:35%, stirring using mixer machine for 5 minutes with 10% water added from it weight. And printed with cube shape size  $50 \times 50 \times 50 \text{ mm}$ . Drying in oven with  $110^\circ\text{C}$  temperature for 8 hours. Heating in the muffle furnace at temperature variation of  $800^\circ\text{C}$ ,  $900^\circ\text{C}$ ,  $950^\circ\text{C}$  and  $1000^\circ\text{C}$  and

the holding time 2 hours, 3 hours, 4 hours and 5 hours, It cooled and compressive strength tested, at the Civil Engineering laboratory in University of Lampung, The porosity test and characterization with XRF and XRD methods were performed in the laboratory of BPTM – LIPI Lampung.

### 3. Result and Discussion

The results of ceramic clay compressive strength testing can be seen at the graphic bellow in Figure 1.



**Figure 1.** Graph of the comparison between compressive strength value to the temperature and heating time of ceramic clay

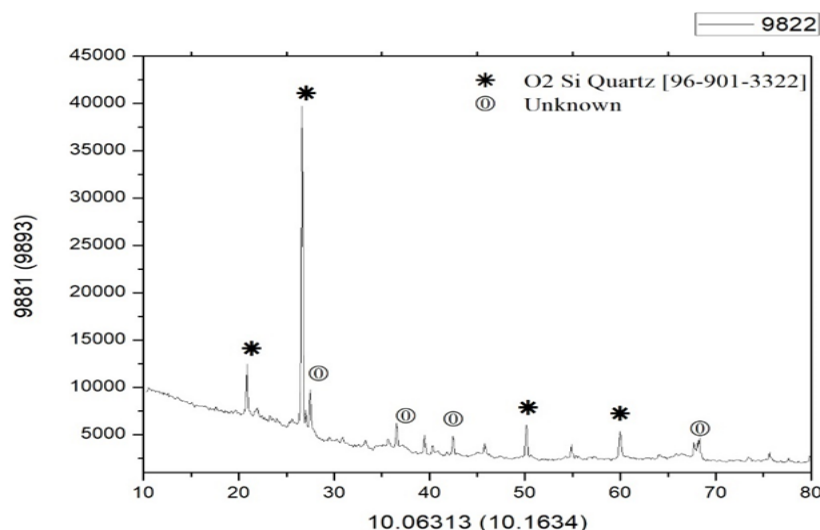
Testing porosity of physical properties in ceramic clay is to know the empty space in the sample. Yafie et.al (2014), reveals that shaft is an empty space or cavity formed in a compacting process that causes air to be trapped between powders during sintering [11]. The existence of a shaft inside a material will affect its mechanical properties. At 5hour variation time have porosity value respectively, at heating temperature 800°C with 5 hours is 20.85%, its value down at heating temperature 900°C with 5 hours is 20.43%, and back down at heating temperature 950°C with 5 hours is 18.89% and reached the optimum value at heating temperature of 1000°C 5 hours is 16.19%. The porosity value of an object greatly affects the value of its compressive strength. The more pores on the ceramic clay the strength of a material will decrease. The results of a study published by Gonggo et.al (2013) explain that the magnitude of the porosity value is due to the feldspart minerals that act as a fuser not yet completely fused to fill the cavities between the clay mineral granules [12]. Ceramic that have been heating at 900°C still contain minerals from feldspart. To reduce the porosity of ceramic required combustion at temperatures above 900°C, because the felders of feldspart begin to melt at a temperature of 900°C-1100°C. This is in accordance with this study, where the increasing temperature and heating time decreases the porosity percentage. Because there has been a good melting process to fill other pore spaces on ceramic, although the melting process has not been perfect.

Ceramic clay XRF characterization as performed to determine the amount of oxide contained in the sample in percent are presented in Table 2.

**Table 2.** X-ray fluorescence characterization results of ceramic clay with some variations of temperature and heating time.

Elements (%)	Heating Time (°C)				
	800 (3 hours)	900 (3 hours)	950 (3 hours)	1000 (3 hours)	1000 (5 hours)
SiO <sub>2</sub>	61.169	58.218	60.066	60.162	64.074
Al <sub>2</sub> O <sub>3</sub>	29.138	30.661	28.918	27.901	25.287
Fe <sub>2</sub> O <sub>3</sub>	4.863	5.208	5.252	5.732	5.358
TiO <sub>2</sub>	0.732	0.742	0.779	0.732	0.719
CaO	0.641	0.961	1.182	0.681	0.635
K <sub>2</sub> O	2.695	2.971	2.958	2.532	3.114
P <sub>2</sub> O <sub>5</sub>	0.466	0.477	0.488	0.481	0.469

Based on the analysis in Table 2, it is also known that the ratio of the highest compound content of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> this proves that there is no significant phase change. For at each combustion temperature with 3 hours, the percentage of both compounds is in cosmetic numbers. At 800°C temperature the amount of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> is compounds 90.307%, at 900°C the amount of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> is compounds 88.899%, at 950°C the amount of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> is compounds 88.984% and at 1000°C the amount of SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> is compounds 89.361%. The high compounds of Fe<sub>2</sub>O<sub>3</sub>, in the range 4.863%, 5.208%, 5.252% and 5.358% are the oxide that gives red color when heating as evidenced in the reddish sample [12]. This is also supported by the XRD characterization results in the sample.

**Figure 2.** XRD Graph of ceramic clay with heating temperature 1000°C 3 hours

The results of XRD characterization in temperature variation 1000°C 3 hours indicating that the dominant phase formed just silica oxide (SiO<sub>2</sub>) with the highest peak at  $2\theta = 26.64^\circ$ . The research of Ratri et.al (2008) also expressed opinion that regarding the formation of a mullite phase formed above a temperature of 1000°C [4]. But in this study, ceramic clay heating at 1000°C 3 hours only formed silica oxide. Based on the XRD graph, it is known that the mullite phase (2Al<sub>2</sub>O<sub>3</sub>·2SiO<sub>2</sub>) ceramic clay heating at 1000°C 3 hours has not been formed. The glass phase is formed more by the feldspar grains ceramic clay will decrease on 1100°C temperatures. The pores remained small the at 1150°C, because

the silica granules occurs by causing more glass phase. At 1200°C temperature, the ceramic body matures with a porosity almost toward zero. In the pseudomorphic form of the feldspar grains can be seen a large number of mullite needles formed resulting from alumina diffusing into the clay through the glass phase.

#### 4. Conclusion

The value of the optimum compressive strength occurs at a temperature of 1000°C with 3 hours time is 9.52 MPa. This is supported by the porosity value of 17.032%. Because the higher of compressive strength value is that the lower of the porosity percentage. Based on of XRD analysis result that the heating optimum at 1000°C with 3 hours and at 1000°C with 5 hours mullite phase ( $2\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2$ ) not formed it just inversion change from low quartz to high quartz.

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