

Bis(imino)pyridine (BIMP) Fe(II) catalyses one-pot green condensation of resorcinol, malononitrile, aromatic aldehydes and cyclohexanone

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Abstract. A novel, efficient and green approach for the synthesis of pyranopyridine derivatives through one-pot, four-component reaction of resorcinol, malononitrile, aromatic aldehydes and cyclohexanone using bis(imino)pyridine (BIMP) Fe(II) catalyst under solvent-free and ultrasonic irradiation is described.

Keywords. Pyranopyridine; four-component; one-pot; BIMP catalyst.

1. Introduction

Benzopyrano[2,3-*b*]pyridines exhibit cancer chemopreventive,¹ antibacterial (including antitubercular),² antimyopic,³ hypotensive,⁴ antirheumatic,⁵ and anti-asthmatic activities.⁶ During the course of studies on the development of new procedures to synthesize substituted pyranopyridine, a few procedures have been reported, for instance, two-step synthesis of pyrano[2,3-*b*]pyridine derivatives has been reported by Dushyant *et al.*⁶ 4-[(*N*-Imidazol-2-ylmethyl)anilino]pyranopyridine derivatives were synthesized by Sunkyung *et al.*⁷ Synthesis of 2-Aryl-4H-pyrano[2,3-*b*]pyridin-4-ones has been reported by Khlebnikov *et al.*⁸ However, due to the economical and atom efficiency issues the development of a one-pot, efficient, rapid and convenient protocol for the multicomponent synthesis of pyranopyridines is of remarkable interest.

Ultrasound has been increasingly used in organic synthesis in the last three decades. It has been demonstrated as an alternative energy source for organic reactions ordinarily accomplished by heating. The use of ultrasound irradiation technique for activating various reactions is well-documented in the literature such as Reformatsky reaction,⁹ Pinacol coupling,¹⁰ Ullmann condensation¹¹ and Suzuki cross-coupling.¹²

Schiff base catalysts are versatile catalysts which have been used in many various fields of organic chemistry. Recently, we have studied the catalytic activity and application of 2,6-bis(imino)pyridine catalysts based on iron(II) possessing different substituents.¹³

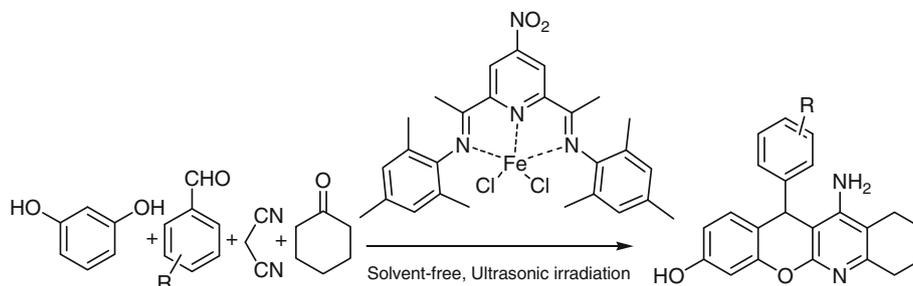
In continuation of our study on the development of new routes in heterocyclic synthesis via novel one-pot MCRs,^{14,15} here a novel four-component reaction for the synthesis of pyrano[2,3-*b*]pyridine derivatives is reported. Four-component, one-pot reaction of resorcinol, malononitrile, aromatic aldehydes and cyclohexanone was performed in the presence of catalytic amounts of (4-nitro-2,6-diacetylpyridinebis(2,4,6-trimethylanil)) FeCl₂ under solvent-free conditions using ultrasonic irradiation (scheme 1).

2. Experimental

2.1 Materials and methods

Chemicals were either prepared in our laboratories or purchased from Merck, Fluka and Aldrich Chemical Companies. All yields refer to isolated products. IR spectra were recorded on a Shimadzu-IR 470 spectrophotometer. ¹H NMR spectra was recorded on a Bruker 100 MHz spectrometer in chloroform as the solvent and TMS as internal standard. Flash column chromatography was performed with 300 and 400 meshes silica gel and analytical thin layer chromatography was performed on pre-coated silica gel plates (60F-254). Sonication was performed in a Shanghai Branson-CQX ultrasonic cleaner with a frequency of 40 kHz and a nominal power of 100 W. The reaction vessel placed inside the ultrasonic bath. Elemental analyses were performed on Thermo Finnigan EA1112 elemental analyser. The procedure for synthesis of 4-nitro-2,6-diacetylpyridinebis(2,4,6-trimethylanil)) FeCl₂ is reported elsewhere.¹³

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Scheme 1. Multicomponent synthesis of pyrano[2,3-*b*]pyridine derivatives.

2.2 General procedure for synthesis of pyrano[2,3-*b*]pyridine derivatives

In a round bottom flask a mixture of resorcinol (1 mmol), malononitrile (1.1 mmol), aromatic aldehyde (1 mmol) and cyclohexanone (1 mmol) was mixed with 4-nitro-2,6-diacetylpyridinebis(2,4,6-trimethylanil) FeCl₂ (0.1 mmol) and the mixture was irradiated under ultrasonic waves at room temperature for an appropriate time as indicated in table 1. The progress of the reaction was monitored by TLC. After completion the reaction, the catalyst was filtered, the reaction mixture was then extracted with dichloromethane (3 × 10 mL). The organic layer was then washed with water and dried

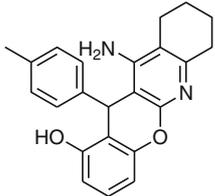
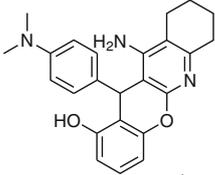
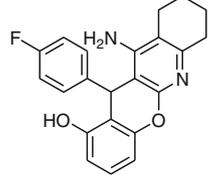
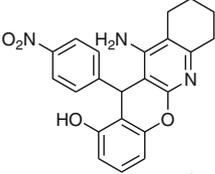
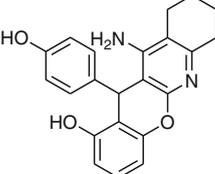
over anhydrous Na₂SO₄. Organic solvent was evaporated under reduced pressure and the obtained product was purified by column chromatography to afford the pure products in good to excellent yields. Spectral data for the selected products are as follows:

2.2a *11-Amino-8,9,10,12-tetrahydro-12-phenyl-7H-chromeno[2,3-*b*]quinolin-1-ol* (entry 1): 310–311°C. IR (KBr): 3440, 3365, 3018, 1614, 1527, 1238, 1111 cm⁻¹. ¹H NMR (100 MHz, DMSO-*d*₆): δ = 1.60 (m, 4H, CH₂), 2.30 (m, 2H, CH₂), 2.62 (m, 2H, CH₂), 5.33 (s, 1H, CH), 6.75–7.45 (m, 10H, ArH, NH₂), 9.60 (s, 1H, OH). Anal. Calcd for C₂₂H₂₀N₂O₂: C, 76.72; H, 5.85; N, 8.13. Found: C, 75.66; H, 5.99; N, 7.95.

Table 1. Results of pyranopyridine derivatives synthesis.^a

Entry	Product	Time (min)	Yield (%) ^b
1		35	90
2		32	90
3		36	88
4		42	82

Table 1. (continued).

Entry	Product	Time (min)	Yield (%) ^b
5		40	90
6		42	88
7		33	90
8		30	92
9		45	84

^aThe reactions were carried out under solvent-free, ultrasonication in the presence of 10 mol% catalyst. ^bIsolated yields.

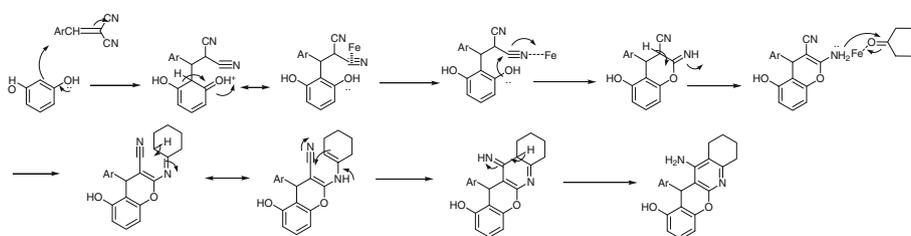
2.2b 11-Amino-12-(4-bromophenyl)-8,9,10,12-tetrahydro-7H-chromeno[2,3-b]quinolin-1-ol (entry 2): mp: 278–279°C. IR (KBr): 3434, 3376, 2977, 1607, 1488, 1206, 1167 cm⁻¹. ¹H NMR (100 MHz, DMSO-*d*₆): δ = 1.64 (m, 4H, CH₂), 2.28 (m, 2H, CH₂), 2.55 (m, 2H, CH₂), 5.40 (s, 1H, CH), 6.35 (br, 2H, NH₂), 6.65–7.40 (m, 7H, ArH), 9.80 (s, 1H, OH). Anal. Calcd for C₂₂H₁₉BrN₂O₂: C, 62.42; H, 4.52; N, 6.62. Found: C, 62.17; H, 4.44; N, 6.73.

2.2c 11-Amino-8,9,10,12-tetrahydro-12-(4-methoxyphenyl)-7H-chromeno[2,3-b]quinolin-1-ol (entry 4): mp: 293–295°C. IR (KBr): 3475, 3388, 2907, 1619, 1455, 1223, 1166 cm⁻¹. ¹H NMR (100 MHz, DMSO-*d*₆): δ = 1.55 (m, 4H, CH₂), 2.25 (m, 2H, CH₂), 2.40 (m, 2H, CH₂), 3.75 (s, 3H, OCH₃), 5.55 (s, 1H, CH), 6.25 (s, 2H, NH₂), 6.6–7.4 (m, 7H, ArH), 9.75 (s, 1H, OH). Anal. Calcd for C₂₃H₂₂N₂O₃: C, 73.78; H, 5.92; N, 7.48. Found: C, 73.52; H, 5.80; N, 7.34.

2.2d 11-Amino-8,9,10,12-tetrahydro-12-*p*-tolyl-7H-chromeno[2,3-b]quinolin-1-ol (entry 5): mp: 285–286. IR (KBr): 3455, 3365, 2910, 1613, 1433, 1218, 1165 cm⁻¹. ¹H NMR (100 MHz, DMSO-*d*₆): δ = 1.60 (m, 4H, CH₂), 2.20 (s, 3H, CH₃), 2.35 (m, 2H, CH₂), 2.55 (m, 2H, CH₂), 5.40 (s, 1H, CH), 6.10 (s, 2H, NH₂), 6.45–7.20 (m, 7H, ArH), 9.63 (s, 1H, OH). Anal. Calcd for C₂₃H₂₂N₂O₂: C, 77.07; H, 6.19; N, 7.82. Found: C, 76.73; H, 6.30; N, 7.71.

3. Results and discussion

Here, a novel, one-pot, four-component coupling of resorcinol, malononitrile, aromatic aldehydes and cyclohexanone to prepare pyrano[2,3-*b*]pyridine derivatives using 4-Nitro-2,6-diacetylpyridinebis(2,4,6-trimethylanil) FeCl₂ under solvent-free using ultrasonic irradiation is reported. First of all, the one-pot reaction of resorcinol, malononitrile, benzaldehydes and cyclohexanone was carried out under solvent-free



Scheme 2. Reasonable mechanism for the one-pot synthesis of pyrano[2,3-*b*]pyridines.

and ultrasonic irradiation. The reaction was considered as a standard model reaction. In order to evaluate the effect of ultrasonic irradiation (table 1, compound 1), firstly, the model reaction was examined without ultrasound at room temperature. Almost moderate yield (44%) with prolonged reaction time (5 h) was found, and when the reaction was carried out under ultrasonication at room temperature, excellent yield of the corresponding pyranopyridine was obtained (90%) within short reaction time (35 min). Furthermore, no by-product was detected when the reaction was carried out under ultrasonic irradiation.

The scope and generality of this process is illustrated with respect to various aromatic aldehydes. Aromatic aldehydes substituted with either electron-donating or electron-withdrawing groups have undergone smooth reaction and gave the corresponding pyrano[2,3-*b*]pyridine in moderate to excellent yields.

The proposed mechanism is depicted in scheme 2. The aldehyde first is condensed with malononitrile to afford α -cyanocinnamionitrile by Knoevenagel addition. Resorcinol C-alkylation gives an intermediate which cyclizes via nucleophilic attack of O atom on the cyano moiety following to protonation and rearrangement to produce the intermediate of 2-amino-3-cyano-7-hydroxy-4-substituted-4H-chromene which could be isolated to investigate the mechanism. Subsequently, cyclohexanone reacted with the chromene and as shown the pyrano[2,3-*b*]pyridine is synthesized through heterocyclization.

4. Conclusion

Four-component green coupling reactions of resorcinol, malononitrile, aromatic aldehydes and cyclohexanone

efficiently proceeded in the presence of (4-nitro-2,6-diacetylpyridinebis(2,4,6-trimethylanil)) FeCl_2 under ultrasonic irradiation and solvent-free conditions. The catalyst exhibited high activity and the reactions were clean and highly selective affording exclusively pyrano[2,3-*b*]pyridine in moderate to excellent yields.

References

1. Azuine M A, Tokuda H, Takayasu J, Enjyo F, Mukainaka T, Konoshima T, Nishino H and Kapadia G 2004 *J. Pharmacol. Res.* **49** 161
2. Srivastava S K, Tripathi R P and Ramachandran R 2005 *J. Biol. Chem.* **280** 30273
3. Toshiro S and Noriko W 1995 *Eur. Pat. Appl.* EP 647445 A1 19950412
4. Sangshetti J, Kokaren N, Kotharkara S A and Shinde D B 2008 *J. Chem. Sci.* **120** 463
5. Ukawa K, Ishiguro T, Kurik H and Nohara A 1985 *Chem. Pharm. Bull.* **33** 4432
6. Dushyant S R and Krishna N S 2010 *ARKIVOC* 305
7. Sunkyung L, Sun M C, Kyu Y Y, Nakjeong K and Chang H 2005 *Bull. Korean Chem. Soc.* **26** 4
8. Khlebinkov V, Patel K, Zhou X, Reddy M M, Su Z, Chiacchia F S and Hansen H C 2005 *Tetrahedron* **65** 6932
9. Singh V, Sapehiyia V and Kad G L 2003 *Synthesis* **2** 198
10. Li J T, Bian Y J, Zang H J and Li T S 2002 *Synth. Commun.* **32** 547
11. Robin M, Pique V, Faure R and Glay J P 2002 *J. Heterocycl. Chem.* **39** 1083
12. Rajagopal R, Jarikote D V and Srinivasan K V 2002 *Chem. Commun.* 616
13. Zohuri G H, Seyedi S M, Sandaroos R, Damavandi S and Mohammadi A 2010 *Catal. Lett.* **140** 160
14. Damavandi S 2011 *Heterocycl. Commun.* **17** 79
15. Eshghi H, Zohuri G H and Damavandi S 2011 *European J. Chem.* **2** 100