

Differentiation potentiates oxidant injury to mitochondria by hydrogen peroxide in Friend's erythroleukemia cells

Marina Comelli, Giovanna Lippe, Irene Mavelli*

Department of Biomedical Sciences and Technologies, University of Udine, Via Gervasutta 48, 33100 Udine, Italy

Received 10 June 1994; revised version received 25 July 1994

Abstract Oxidative damage to mitochondrial functions was investigated upon non-lethal treatment with H₂O₂ of Friend's erythroleukemia cells induced to differentiate, in comparison with the parental cell line. Both respiration and maximal ATP synthase capacity were more severely diminished by H₂O₂ in induced cells. The effects were mediated by intracellular redox-active iron and OH[•] radicals. Specifically, the mechanisms of the selective oxidant injury to F₀F₁ ATP synthase observed in differentiating cells likely involved impairment of F₀-F₁ coupling sensitive to oligomycin. We suggest a Fenton-like reaction of H₂O₂ with iron ions, more available in the differentiating cells, as occurring at the surface and/or in the lipid bulk phase of the inner mitochondrial membrane, thus injuring subunits responsible for the coupling of F₀F₁ ATP synthase through generation in situ of the actual damaging species. Besides, we propose heme iron as the most likely candidate for such reaction in induced cells actively synthesizing heme. In accordance, pretreatment of uninduced cells with hemin made H₂O₂-damage qualitatively identical.

Key words: Hydrogen peroxide; Oxidative phosphorylation; Heme; Iron; F₀F₁ ATPsynthase; Friend's erythroleukemia cells

1. Introduction

The exposure of Friend's virus-infected murine erythroleukemia cells (FELC) to HMBA initiates a coordinated erythroid differentiation program during which FELC synthesize characteristic proteins of mature erythrocytes [1], notably hemoglobin [2,3]. HMBA-induced FELC undergo substantial modifications in iron metabolism concomitant with the increase of heme biosynthesis [4–7]. We recently demonstrated that heme-synthesizing FELC were more susceptible to the oxidative insult inflicted by the anthracycline antibiotic daunomycin via H₂O₂ generation in a way closely related to their increased cellular iron levels [8]. Accordingly, iron is known to participate in a Fenton-type reaction with H₂O₂ producing highly reactive hydroxyl radical (OH[•]). As mitochondria were shown to require iron [9], as well as to be major intracellular targets in the mechanisms of oxidant-mediated injury to some tumor cell lines [10], we expected that alterations in the redox-active iron pool of mitochondria could be associated to FELC differentiation and could be critical to enhance the injury by H₂O₂. Interestingly, we recently documented H₂O₂ to determine inactivation with a strict requirement for iron of purified mitochondrial F₁ ATPase [11] and F₀F₁ ATP synthase [12]. The aim of the present investigation was to evaluate whether early mitochondrial dysfunctions were caused by H₂O₂ to FELC and whether differentiation could influence H₂O₂ effects on mitochondrial components, particularly on ATP synthase. Then, mitochondrial functions of both parental and induced FELC were assayed after incubation with H₂O₂ determining no more than

10% of cell mortality. The results showed that the maximal ATP synthase capacity was more severely impaired by H₂O₂ in HMBA-induced FELC. Differential effects were also observed on basal and FCCP-uncoupled respiration rates, as well as, and more notably, on the oligomycin-sensitive respiration rate, investigated as a measure of mitochondrial ATP synthesis flux. Evidence was reported that such effects were due to greater availability of redox-active iron, likely heme iron, at the level of mitochondrial membranes of heme-synthesizing FELC.

2. Materials and methods

FELC, clone 3CL8 *in vitro*, kindly provided by Dr. Belardelli, Istituto Superiore di Sanità, Rome, Italy, were grown in RPMI 1640 medium (ICN Biomedicals, Inc. Costa Mesa, CA) supplemented with 10% heat-inactivated fetal calf serum (Seromed, Biochrom KG, Berlin, Germany) and maintained in logarithmic phase with appropriate dilutions every 3–4 days. To induce differentiation, cells from confluent cultures were seeded in complete medium added with 5 mM HMBA (Sigma, St. Louis, MO) and grown for 120 h, with a refeeding (including fresh inducer) after 96 h. Parallel cultures grown under identical conditions, except for the absence of HMBA, served as controls in all experiments. The erythroid differentiation was assessed by evaluating the increase (ca. 20-fold) of hemoglobin content of cells on the basis of the intensity of the Soret band in 17,000 × g supernatants from sonicated homogenates [8]. Cell viability, evaluated by the Trypan blue dye exclusion test, never was lower than 95%. The oxidant treatment was performed by exposing cells at confluence, suspended at 2 × 10⁶ cells/ml in PBS (20 mM sodium phosphate pH 7.4, 140 mM NaCl, 5.4 mM KCl, 0.8 mM MgSO₄) containing 5 mM glucose, to 0.25 mM H₂O₂ for 30 min; cell mortality never exceeded 10%. In some experiments 1 h preincubation was carried out in the presence of 20 mM DFO (Ciba Geigy, Basle, Switzerland), or 20 mM FO prepared by addition of ferric chloride to yield 95% saturation of DFO, or 100 μM DMSO (Merck, Darmstadt, Germany), or 100 μM BHT (Sigma). The effects of exogenous hemin on H₂O₂-susceptibility of uninduced FELC were investigated by incubating cell suspensions (2 × 10⁶ cells/ml) with 25 μM hemin (Sigma) for 30 min; the drained pellets were then resuspended at the initial density and exposed to H₂O₂. For the measurements of mitochondrial oxygen uptake by intact cells [10], FELC were suspended at 5–10 × 10⁶ cells/ml in respiration buffer (0.25 M sucrose, 10 mM HEPES, 5 mM phosphate, 1 mM MgCl₂, 2 mM EGTA, 5 mM glucose, pH 7.4) at 37°C and transferred into the polarographic cell (1 ml) of a Y.S.I. oxygraph

*Corresponding author. Fax: (39) (432) 600 828.

Abbreviations: FELC, Friend's erythroleukemia cells; HMBA, *N,N'*-hexamethylene bisacetamide; PBS, phosphate-buffered saline; HEPES, *N*-(2-hydroxyethyl) piperazine-*N'*-(2-ethanesulfonic acid); EGTA, ethylene glycol-bis (β -aminoethyl ether)*N,N,N,N'*-tetraacetic acid; RCR, respiratory control ratio; FCCP, carbonyl cyanide *p*-trifluoromethoxyphenyl hydrazone; BHT, butylated hydroxytoluene; DMSO, dimethyl sulfoxide; DFO, desferrioxamine methanesulphonate; FO, ferrioxamine.

equipped with a Clark electrode (model 53, Yellow Springs Instrument Co., Yellow Springs, OH). Oxygen consumption was completely suppressed by 1 mM KCN and was taken to be a direct measure of mitochondrial respiration. After 6–8 min recording of basal respiration rate, 10 μ M oligomycin (Sigma), or 50 μ M FCCP (Sigma), or 75 μ M atractyloside (Sigma) were added. RCR was calculated as ratio of the rate of FCCP-stimulated respiration to that of basal respiration. The maximal efficiency of ATP hydrolysis by mitochondria was assayed as Das and Harris [13], by measuring NADH oxidation with a continuous spectrophotometric method at 340 nm, in the presence of a coupled lactate dehydrogenase-pyruvate kinase ATP regenerating system at 37° C and pH 7.4, on homogenates prepared as follows. Cell suspensions (5×10^7 cells/ml) in 20 mM HEPES, 1 mM MgCl₂, 2 mM EGTA, pH 7.0, were subjected to brief controlled sonication with a 301 Sonic Dismembrator (Artek System Co.): disruption of mitochondrial membranes and formation of submitochondrial vesicles were monitored during sonication by evaluating the increase in ATPase activity and the decline in the effect of 5 μ M FCCP or 50 μ M atractyloside on such activity. Possible interferences of Ca²⁺- and Na⁺/K⁺-ATPases were minimized by using an assay buffer containing 2 mM EGTA and less than 5 mM Na⁺ [13]. Very low sensitivity to 10 μ M sodium orthovanadate or to 2 mM ouabain (Sigma) were found (ca. 9% or 6%) in both parental and induced FELC, indicating that over 85% of the measured ATPase activity was mitochondrial in origin. Oligomycin-sensitive ATPase activity was assayed in the presence of 4 μ M oligomycin and considered as maximal ATP synthesis capacity.

3. Results and discussion

3.1. Impairment of mitochondrial respiration of FELC by H₂O₂

The rate of mitochondrial oxygen consumption by intact FELC with glucose as the sole exogenous substrate was measured in the absence or in the presence of the uncoupler FCCP, as well as of specific inhibitors of adenine nucleotide translocase and F₀F₁ ATP synthase, i.e. atractyloside and oligomycin, to investigate the effects of non-lethal treatment with H₂O₂ on the basal respiration, on the respiration uncoupled from the proton electrochemical gradient, on the resting respiration detectable under non-phosphorylating conditions and on the respiration coupled to ATP synthesis. The results are shown in Table 1A).

Basal respiration was largely inhibited, as for other glucose-fed intact cells [14,15], by oligomycin (ca. 70%) or by atractyloside (ca. 60%) in both parental and HMBA-induced FELC. This indicated that electron transfer was similarly coupled to ATP synthesis and transport in both cell types, despite the lower values of the respiration rates observed in induced cells when expressed per cell ratio, reflecting the reduction of cell volume and protein content of such cells [1,8]. Upon H₂O₂ exposure of the parental FELC a 40% decrease was observed in the basal respiration rate, which still showed full sensitivity to oligomycin or to atractyloside. Moreover, a 39% decline in FCCP-uncoupled respiration rate occurred, indicating an impairment inflicted by the oxidant to the respiratory chain. On the other hand, the FCCP stimulating effect on the basal respiration (see RCR values) showed that the proton-motive force of the mitochondrial membrane was not diminished. The resting respiration measured in the presence of oligomycin (state 4), or of atractyloside (not shown), containing components of oxygen consumption coupled to ion transport [14], was inhibited by 34%, suggesting H₂O₂ may have caused a decrease in membrane conductance [16]. Oligomycin- and atractyloside-sensitive respiration rates, reflecting ADP phosphorylation flux, were also declined by 43% and 41%, respectively. In the case of differentiating FELC, H₂O₂ caused significantly larger ($P < 0.001$) decrease in the basal respiration rate (54% decrease). Even the decrease in FCCP-uncoupled respiration rate (56%) was significantly greater ($P < 0.001$) than that observed in the parental cells, indicating a significant additional damage to the electron transport chain. Nevertheless, similarly to control cells, the resting respiration declined only by 40%, and the effect of the uncoupler on H₂O₂-inhibited basal oxygen consumption (see RCR) suggested that the proton electrochemical gradient still affected the respiration. Furthermore, the most notable differential effect observed in heme-synthesizing FELC upon H₂O₂-exposure was on ADP phosphorylation, as oligomycin- and atractyloside-sensitive respiration rates declined by 65%. Oli-

Table 1
Effects of HMBA-induced differentiation of FELC on dysfunctions caused by H₂O₂ to mitochondrial respiration and maximal ATP synthesis capacity

	Parental cells		HMBA-induced cells	
	No treatment	H ₂ O ₂	No treatment	H ₂ O ₂
(A)				
Basal respiration rate	1.92 ± 0.09	1.16 ± 0.14*	1.22 ± 0.09	0.56 ± 0.08*
FCCP-uncoupled respiration rate	2.93 ± 0.09	1.78 ± 0.03*	1.88 ± 0.07	0.82 ± 0.07*
State 4 respiration rate	0.56 ± 0.06	0.37 ± 0.09*	0.38 ± 0.06	0.27 ± 0.05*
Oligomycin-sensitive respiration rate	1.36 ± 0.06 (71%)	0.79 ± 0.09* (68%)	0.84 ± 0.07 (69%)	0.29 ± 0.04* (52%)*
Atractyloside-sensitive respiration rate	1.19 ± 0.05 (62%)	0.70 ± 0.08* (60%)	0.75 ± 0.06 (61%)	0.26 ± 0.04* (47%)*
RCR	1.53 ± 0.09	1.53 ± 0.11	1.54 ± 0.12	1.46 ± 0.08
(B)				
ATPase activity	18.9 ± 1.9	12.1 ± 1.2*	9.2 ± 1.6	5.9 ± 1.0*
Oligomycin-sensitive ATPase activity	13.8 ± 1.5 (73%)	8.2 ± 1.0* (67%)	6.4 ± 0.9 (70%)	2.9 ± 0.9* (50%)*

FELC were cultured in the absence or in the presence of 5 mM HMBA. Then, cells at confluence were incubated in PBS for 30 min with or without 0.25 mM H₂O₂. The rate of mitochondrial oxygen consumption by intact cells (A) and maximal ATPase activity of briefly sonicated cell suspensions (B) were assayed. Data of respiration rates are expressed as nmol O₂/min/10⁶ cells. Oligomycin- and atractyloside-sensitive respiration rates were calculated as difference between the basal respiration rate and that detected in the presence of 10 μ M oligomycin (state 4) and 75 μ M atractyloside, respectively. The values in parentheses are % of oligomycin-sensitive or atractyloside-sensitive oxygen consumption rate taking the basal respiration as 100%. RCR was calculated as the ratio of FCCP-uncoupled respiration rate, divided by the rate of the basal respiration. The values of both total and oligomycin-sensitive maximal ATPase activity are expressed as nmol ATP/min/10⁶ cells. The values in parentheses are % of oligomycin-sensitive activity with respect to the total ATPase activity taken as 100%. All data represent means ± S.D. of 10 different experiments in which the assays were at least in duplicate. *Significantly different from the appropriate controls in the absence of H₂O₂, $P < 0.001$, Student's *t*-test.

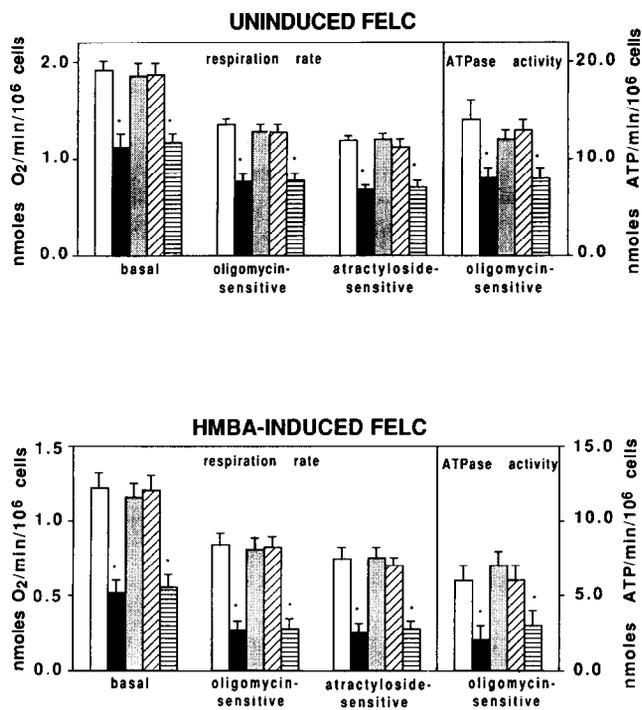


Fig. 1. Preventive effect of DFO or DMSO and ineffectiveness of BHT on H_2O_2 -dependent damage to respiration rate and oligomycin-sensitive ATPase activity in HMBA-induced and uninduced FELC. Cells from confluent cultures were treated with 0.25 mM H_2O_2 for 30 min after 1 h preincubation in PBS + 5 mM glucose in the absence (full columns) or in the presence of 20 mM DFO (dotted columns), 100 μ M DMSO (diagonal columns), 100 μ M BHT (dashed columns). Control experiments without H_2O_2 either in the absence or in the presence of DFO, or DMSO, or BHT were also performed and reported as empty columns. The measurements of rates of basal, oligomycin- and atractyloside-sensitive respiration and of oligomycin-sensitive maximal ATPase activity were carried out as specified in section 2. Data are mean values \pm S.D. of at least 3 different experiments in which the assays were performed in duplicate. * $P < 0.001$, Student's t -test, with respect to the appropriate controls in the absence of H_2O_2 (empty columns).

gomylin- and atractyloside-sensitivity of H_2O_2 -inhibited basal respiration was then diminished to about 50%, suggesting that the coupling of the ATP synthesis to the respiration was impaired by H_2O_2 in this case, but not in undifferentiated cells, where, as a matter of fact, the full sensitivity was maintained. Assuming a high degree of control of 'phosphorylation flux' over basal respiration and a non-ohmic proton conductance of the mitochondrial inner membrane, that were suggested as general phenomena in intact cells [14,16], taking also into account that FCCP- uncoupled respiration rates were greater than the basal rates of respiration, the results indicated that the impairment of ADP phosphorylation caused by H_2O_2 , mainly in differentiating FELC, appeared to be related to damage to the ATP synthesis/transport system rather than to electron transport.

3.2. Damage by H_2O_2 to maximal mitochondrial ATP synthase capacity of FELC

We next investigated the susceptibility to H_2O_2 of the maximal ATP synthase capacity of the mitochondria, which we assayed by measuring the ATP hydrolysis rate at saturating ATP concentration sustained by homogenates prepared by brief controlled sonication of FELC, either induced to differen-

tiolate or not. In both cell types, treated or not with H_2O_2 , the addition of FCCP or atractyloside had no effect on the ATPase activity measured, indicating that the sonication of cell suspensions provided a correct and complete exposure of the enzyme complex through the formation of submitochondrial vesicles. The results are reported in Table IB), where the oligomycin-sensitive ATPase activity represents the maximal ATP synthase capacity. 70% sensitivity to oligomycin was observed (see values in parentheses) in both parental and HMBA-induced cells, pointing to a correct coupling of F_0 and F_1 moieties of F_0F_1 ATP synthase complex in both cases, despite the lower value of ATPase activity observed in induced cells. The exposure to H_2O_2 apparently caused a comparable significant decrease in ATPase activity of both cell types. Conversely, % oligomycin-sensitivity of ATPase activity was diminished only in the differentiating FELC from 70% to 50%, so that the maximal ATP synthesizing capacity more markedly declined in this case (55% decrease with respect to 41% in the parental cells). These results indicated that H_2O_2 was able to affect F_0F_1 ATP synthase complex in intact FELC, in a way apparently stronger in heme-synthesizing cells and presumably different from a qualitatively point of view. In fact, in this case subunits responsible for the proper assembling of F_0 and F_1 may have been damaged, whereas in control cells the full oligomycin-sensitivity of ATPase activity excluded this possibility.

3.3. Protection experiments with cell-permeable 'anti-oxy-radical' agents

Fig. 1 shows that 1 h preincubation of intact cells with the iron chelator DFO, as well as with the OH^\cdot radical scavenger DMSO, completely prevented the damage caused by H_2O_2 to both control and differentiating FELC, indicating that intracellular redox-active chelatable iron and the OH^\cdot radical were involved in both cases. In accordance with this hypothesis was the observed ineffectiveness of FO, 95% iron-saturated DFO derivative, in preventing H_2O_2 -damage (not shown), which ruled out direct reactions of DFO with H_2O_2 , OH^\cdot , or other oxy-radicals, as responsible for the protective effect under our conditions. On the other hand, the lipid antioxidant BHT never showed any protection, suggesting that lipid peroxidation did not occur to such extent as to mediate the impairment of mitochondrial functions. Then, it is tempting to argue that, upon the non-lethal treatment of FELC with H_2O_2 which we carried out, the mitochondrial dysfunctions observed in both parental and differentiating cells were due to iron-catalyzed peroxidative attack to proteins, rather than to lipids, of the inner membrane. The results of Davies and coworkers [17], showing oxidative inactivation of electron transport chain components and ATPase of submitochondrial particles as independent of lipid peroxidation, support our hypothesis.

3.4. The selective injury caused by H_2O_2 to mitochondrial functions of heme-synthesizing FELC was mimicked by pre-exposure of uninduced FELC to hemin

We have recently shown that the iron content of HMBA-induced FELC was apparently increased with respect to the uninduced cells by about 3-fold, with non-hemoglobin-bound iron being only slightly augmented, while hemoglobin-bound iron was more than 20-fold greater [8]. In accordance, it was reasonable that non-heme chelatable iron affected in both cell types H_2O_2 -dependent damage to mitochondrial functions, as

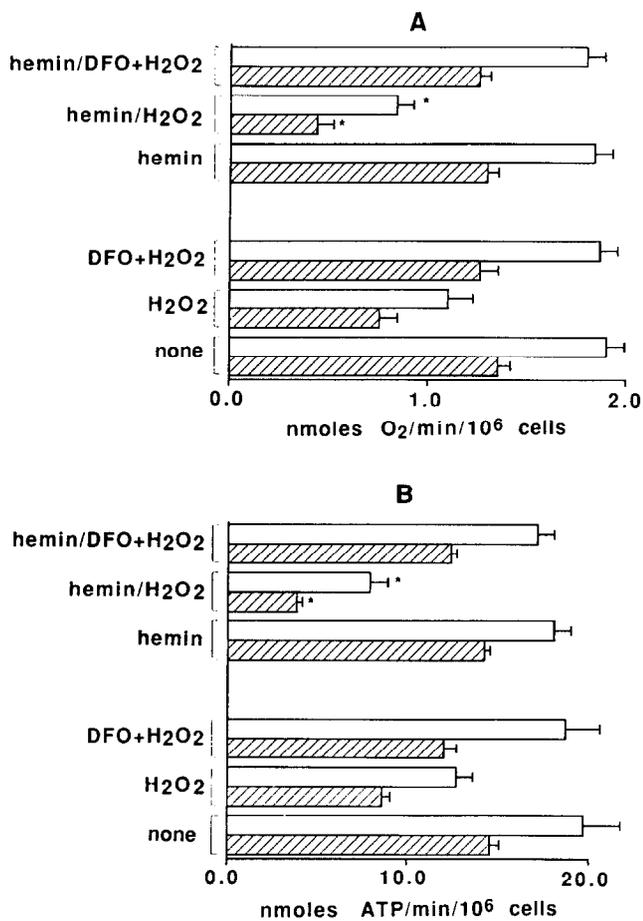


Fig. 2. Pre-exposure to exogenous hemin potentiates the mitochondrial dysfunctions caused by H₂O₂ to uninduced FELC. Uninduced FELC (2×10^6 cells/ml) were preincubated in PBS + 5 mM glucose for 30 min in the presence of 25 μ M hemin (hemin). Cell mortality never exceeded 5%. Then, the drained pellet was resuspended at the initial density and exposed to 0.25 mM H₂O₂ for 30 min after 1 hour incubation with 20 mM DFO (hemin/DFO + H₂O₂) or without DFO (hemin/H₂O₂). Control experiments were performed in the absence of hemin and indicated as: none, H₂O₂, DFO + H₂O₂. Panel A shows the basal respiration rate (empty columns) and the oligomycin-sensitive respiration rate (diagonal columns) of intact FELC. Panel B reports the maximal ATPase activity of sonicated cell suspensions: total activity (empty columns) and oligomycin-sensitive activity (diagonal columns). Data are mean values \pm S.D. of 3 different experiments in which the assays were performed at least in duplicate. *Significantly different from the appropriate controls in the absence of hemin, $P < 0.001$, Student's *t*-test.

suggested by the results of the protection experiments above reported, indicating that the oxidative insult occurred in the parental cells as mediated, as a matter of fact, by chelatable iron. Conversely, we suggest, considering the augmented heme synthesis, that an increase in heme levels of the inner mitochondrial membrane could be a significant factor in determining the selective injury caused by H₂O₂ to differentiating FELC, whereby the metalloporphyrin promoted OH[•] radical formation at the membrane surface and/or in the lipid bulk phase. This hypothesis was supported by the results shown in Fig. 2, demonstrating that by treating uninduced FELC with hemin before the exposure to H₂O₂ (hemin/H₂O₂) the effects of the oxidant, both on respiration rate (panel A) and on maximal ATPase activity (panel B), became qualitatively identical to those before described for the heme-synthesizing cells. Specifi-

cally, as for the damage to F₀F₁ ATP synthase, it appeared to result in a decline in the % oligomycin-sensitivity of ATPase activity (from 68% to 49%) and then to alter some oligomycin-sensitivity responsible subunits of the complex. In this respect, it is noteworthy to emphasize that we recently demonstrated that H₂O₂ was not able to cause any damage affecting the correct oligomycin-sensitive coupling of the components F₀ and F₁ assembled in the purified native enzyme complex, when it was exposed to the oxidant in the presence of redox-active adventitious iron ions [12]. Taking into account such results, hemin, as a lipid-soluble iron chelate, could be likely supposed to provide redox-active iron ions in a more appropriate form accessible to ATP synthase subunits responsible for the coupling. Such subunits should thus be damaged directly, and/or through oxidative injury to adenine nucleotide translocase, whose conformational rearrangements may be transmitted to F₀ moiety of the complex [18]. Finally, it should be considered that heme has been long documented to efficiently react with H₂O₂ leading to the formation of hyper-valent iron species and/or OH[•] radicals [19], as well as of protein-derived radicals of hemoproteins [20], or of proteins like albumin present together hemin in the bulk solution [21]. The reaction was reported to eventually release iron from heme (22). In this regard, because DFO does not chelate iron from hemin *in vitro* [23], the preventive effect of DFO shown in Fig. 2 (hemin/DFO+H₂O₂) indicates that liberation of iron from hemin at the level of the inner mitochondrial membranes was required for H₂O₂-damage to occur, rather than direct interaction of hemin iron with H₂O₂ and target protein simultaneously. This may be considered a possible mechanism for the protective effect exhibited by DFO on the selective damage to HMBA-induced FELC, supporting our hypothesis that it was mediated by heme.

3.5. Conclusions

The results reported show that, due to the modifications in their iron metabolism associated to the augmentation of heme synthesis, FELC committed to the erythroid differentiation provided us with a good model to investigate the role of intracellular iron in the mechanisms of oxidative damage to cells and specifically to mitochondria. The potentiating effects by heme iron evidenced in intact FELC toward the injury caused by H₂O₂ to proteins of inner mitochondrial membranes, and particularly to the F₀F₁ ATP synthase complex, suggest that the selective damage inflicted by H₂O₂ to HMBA-induced FELC may be explained by the presence of a suitable amount of heme at the level of the mitochondrial membranes of these cells. The occurrence of such condition can be reasonably expected on the basis of the modifications of heme metabolism described as concomitant with the differentiation [24,25].

Acknowledgements: Supported by 'Consiglio Nazionale delle Ricerche' C.N.R., 'Ministero Università e Ricerca Scientifica e Tecnologica' M.U.R.S.T. (40%, 60%), and C.N.R. Targeted Project 'Applicazioni Cliniche Ricerca Oncologica: A.C.R.O.'. The expert photographic work of Dr. Giancarlo Cro 'Studio Fotografico Controluce' is gratefully acknowledged.

References

- [1] Marks, P.A. and Rifkind, R.A. (1978) *Annu. Rev. Biochem.* 47, 419-448.

- [2] Hu, Y.H., Gardner, J., Aisen, P. and Skoultch, A.I. (1977) *Science* 197, 559–561.
- [3] Marks, P.A., Sheffery, M. and Rifkind, R.A. (1987) *Cancer Res.* 47, 659–666.
- [4] Battistini, A., Marziali, G., Albertini, R., Habeyswallner, D., Bulgarini, D., Coccia, E.M., Fiorucci, G., Romeo, G., Orsatti, R., Testa, U., Affabris, E., Peschele, C. and Rossi, G.B. (1991) *J. Biol. Chem.* 266, 528–535.
- [5] Beaumont, C., Jain, S.K., Bogard, M., Nordmann, Y. and Drysdale, J. (1987) *J. Biol. Chem.* 262, 10619–10623.
- [6] Coccia, E.M., Profita, V., Fiorucci, G., Romeo, G., Affabris, E., Testa, U., Henze, M. and Battistoni, A. (1992) *Mol. Cell. Biol.* 12, 3015–3022.
- [7] Conder, L.H., Woodard, S.I. and Dailey, H.A. (1991) *Biochem. J.* 275, 321–326.
- [8] Pietrangeli, P., Steinkuhler, C., Marcocci, L., Pedersen, J.Z., Mondovi, B. and Mavelli, I. (1994) *Biochim. Biophys. Acta*, in press.
- [9] Weaver, J. and Pollack, S. (1990) *Biochem. J.* 271, 463–466.
- [10] Hyslop, P.A., Hinshaw, D.B., Wayne, A.H., Schraufstatter, I.U., Sauerheber, R.D., Spragg, R.G., Jackson, J.H. and Cochrane, C.G. (1988) *J. Biol. Chem.* 263, 1665–1675.
- [11] Lippe, G., Comelli, M., Mazzilis, D., Dabbeni Sala, F. and Mavelli, I. (1991) *Biochem. Biophys. Res. Commun.* 181, 764–770.
- [12] Lippe, G., Londero, D., Dabbeni Sala and Mavelli, I. (1993) *Biochem. Mol. Biol. Int.* 30, 1061–1070.
- [13] Das, A.M. and Harris, D.A. (1990) *Biochem. J.* 266, 355–361.
- [14] Brown, G.C., Lakin-Thomas, P.L. and Brand, M.D. (1990) *Eur. J. Biochem.* 192, 355–362.
- [15] van der Valk, P., Gille, J.J.P., van der Plas, L.H.W., Jongkind, J.F., Verkerk, A., Konings, A.W.T. and Joenie, H. (1988) *Free Rad. Biol. Med.* 4, 354–356.
- [16] Nobes, C.D., Brown, G.C., Olive, P.N. and Brand, M.D. (1990) *J. Biol. Chem.* 265, 12903–12909.
- [17] Zhang, Y., Marcillat, O., Giulivi, C., Ernster, L. and Davies, K.J.A. (1990) *J. Biol. Chem.* 265, 16330–16336.
- [18] Ziegler, M. and Penefsky, H.S. (1993) *J. Biol. Chem.* 268, 25320–25328.
- [19] Harel, S. and Kanner, J. (1988) *Free Rad. Res. Commun.* 5, 21–33.
- [20] Davies, M.J. (1991) *Biochim. Biophys. Acta* 1077, 86–90.
- [21] Nohl, H. and Stolze, K. (1993) *Free Rad. Biol. Med.* 15, 257–263.
- [22] Gutteridge, J.M.C. (1986) *FEBS Lett.* 201, 291–295.
- [23] Rouault, T., Rao, K., Harford, J., Mattia, E., Klausner, R.D. (1985) *J. Biol. Chem.* 260, 14862–14866.
- [24] Ching Lo, S., Aft, R. and Mueller, G.C. (1981) *Cancer Res.* 41, 864–870.
- [25] Fujita, H., Yamamoto, M., Yamagami, T., Hayashi, N., Bishop, T.R., De Verneuil, H., Yoshinaga, T., Shibahara, S., Morimoto, R. and Sassa, S. (1991) *Biochim. Biophys. Acta* 1090, 311–316.