

Purification and spectral characterization of a *b*-type cytochrome from the plasma membrane of the archaebacterium *Sulfolobus acidocaldarius*

Marita Becker and Gunter Schäfer

Institute of Biochemistry, Medical University of Lubeck, Lubeck, Germany

Received 24 July 1991, revised version received 20 August 1991

For the first time the purification of a heme-*b* containing cytochrome from the plasma membrane of an extremely thermoacidophilic archaebacterium is described. The detergent solubilized 30 kDa polypeptide contains two heme-*b* centers and one copper ion. According to its low temperature spectra and CO-binding properties, it is likely to function as a cytochrome-*o* like terminal oxidase in the membrane. The purified cytochrome does not retain catalytic activity, however.

Cytochrome-*b*, Archaebacteria, *Sulfolobus*, Electron transport, Respiration

1 INTRODUCTION

The extremely thermoacidophilic archaebacterium *Sulfolobus acidocaldarius* grows aerobically and has been shown to carry out respiration-linked proton extrusion [1]. The electrochemical gradient of H⁺ across its membrane is likely to be used for chemiosmotically driven ATP synthesis [1,2]. It is unknown, however, which components of the respiratory system are acting as effective proton pumps allowing for a pH gradient >3. It has also been shown that differential cyanide sensitivity indicates the action of more than one terminal oxidase. The membrane-residing respiratory system besides NADH-dehydrogenase [3] and succinate dehydrogenase [4] contains only *a*- and *b*-type cytochromes while *c*-type cytochromes are absent [5–7]. From these we have purified and characterized a novel cytochrome-*aa*₃ [8,9], acting as a 'quinol oxidase' and using caldariella quinone as substrate.

The isolation of *b*-type cytochromes revealed to be unusually discouraging due to the instability under the harsh conditions necessary for dissociation of the firmly associated respiratory complexes from the membrane. A partial purification has been preliminarily reported [10]. Here we describe the isolation and spectral characterization of a purified *b*-type cytochrome from *Sulfolobus acidocaldarius*, likely to function as a cytochrome-*o* in the intact membrane.

2 MATERIALS AND METHODS

2.1 Growth of bacteria

Sulfolobus acidocaldarius DSM 639 cells were grown aerobically in Brock's medium [11] supplemented with 10 mM K₂SO₄, 0.2% sucrose, 0.1% yeast extract (Gibco), pH 2.5, at 75°C. Cells were harvested in the early stationary phase, washed once in a buffer containing 50 mM imidazole, 1 mM malonate, pH 7.0, and sedimented for 15 min at 6500 × *g* at 4°C. The pellet was stored at –80°C in the same buffer additionally containing 50% glycerol.

2.2 Membrane preparation

Membranes were prepared as described [9], with the exception that a buffer containing 50 mM imidazole, 1 mM malonate, pH 7.0 was used and the cells were disrupted in a Manton Gaulin press for 5 min at 5 × 10⁷ Pa.

2.3 Purification procedure

Membrane extraction and protein purification were performed at 4°C. First the membranes were incubated in a buffer containing 50 mM imidazole, 20 mM K₂P₂O₇, pH 7.5, for 1 h at a final protein concentration of about 10 mg/ml. After sedimentation at 200 000 × *g* for 50 min, the resulting pellet was suspended in 50 mM imidazole (final protein concentration of 20 mg/ml) and extracted for 75 min in the presence of 1% dodecylmaltoide (DM). After 1 h sedimentation at 150 000 × *g*, the supernatant was concentrated (Amicon, PM10 membrane), gently shaken with Bio-Beads SM2 for at least 2 h, and subsequently passed through a hydroxyapatite column (2.6 × 26 cm), equilibrated with 0.75 M KH₂PO₄, 0.05% DM, 0.02% *N*-dodecyl-*N*,*N*-dimethylammonio-3-propane-sulfonate (SB12), pH 7.5, (flow rate 40 ml/h). Protein and heme in eluting fractions were monitored at 280 and 420 nm, respectively. Heme-containing peak fractions were analyzed by their dithionite-reduced minus oxidized difference spectra. Cytochrome containing fractions were concentrated, the buffer exchanged to 50 mM KH₂PO₄, pH 7.5, and loaded on a second hydroxyapatite column (3 × 14 cm) equilibrated with 50 mM KH₂PO₄, 0.05% DM, 0.02% SB12, pH 7.5 (flow rate 30 ml/h). When 100 ml of the equilibration buffer had passed through, a linear phosphate gradient (50–500 mM KH₂PO₄) in 0.05% DM, 0.02% SB12, pH 7.5, was applied. Cytochrome-*b*-enriched fractions were pooled and concentrated. In case of a residual cytochrome-*a* contamination the material was rechromatographed on hydroxyapatite under analogous conditions. Subsequently the material was loaded on a H/Load/600-

Correspondence address: G. Schäfer, Institut für Biochemie, Medizinische Universität zu Lübeck, Ratzeburger Allee 160, W-2400 Lübeck, Germany. Fax: (49) (451) 500 4034.

Superdex 200 column using a buffer containing 50 mM imidazole, 150 mM NaCl, 0.2% SB12, pH 7.5 (flow rate 1 ml/min). The final step of purification was performed on a MonoQ-5/5 column with a linear gradient from 0 to 0.5 M NaCl in 50 mM imidazole, 0.2% SB12, pH 7.5.

2.4 Spectroscopy

A Hewlett Packard HP8450A diode array spectrophotometer was used for spectra at room temperature, low temperature spectra were measured with a Sigma ZWSII dual wavelength spectrophotometer equipped with a self-designed low-temperature device. Pyridine heme-chrome spectra were recorded according to [12]. Copper content was determined using a Hitachi 180-80 Zeeman atomic absorption spectrophotometer set at 324.8 nm.

2.5 Other methods

Hydroxyapatite was prepared according to [13]. Membrane protein was determined by the biuret method [14], membrane extracted proteins according to Lowry [15] in the presence of SDS. PAGE were performed by the Laemmli procedure [16] on 15% gels, proteins were visualised by silver staining [17].

2.6 Materials

H/Load/600-Superdex 200 and Mono-Q-5/5 columns were obtained from Pharmacia. All chemicals were purchased from Merck, Sigma, Serva, Biomol or Fluka.

3 RESULTS

Previous spectroscopic studies have shown that more than one heme-*b* center is present in the membranes of *Sulfolobus*, as indicated by an absorption maximum at 565.5 nm and the pronounced shoulder at 558 nm in reduced-oxidized difference spectra taken at room temperature [9]. Partial reduction by ascorbate and spectral subtraction from the fully dithionite-reduced state suggested a third compound to be present absorbing at 562 nm. In Fig. 1, a low-temperature spectrum of freshly prepared *Sulfolobus* membranes is shown where indeed three individual maxima in the α -region of *b*-type cytochromes can be clearly distinguished. At low temperature the absorptions are shifted slightly to lower wave-

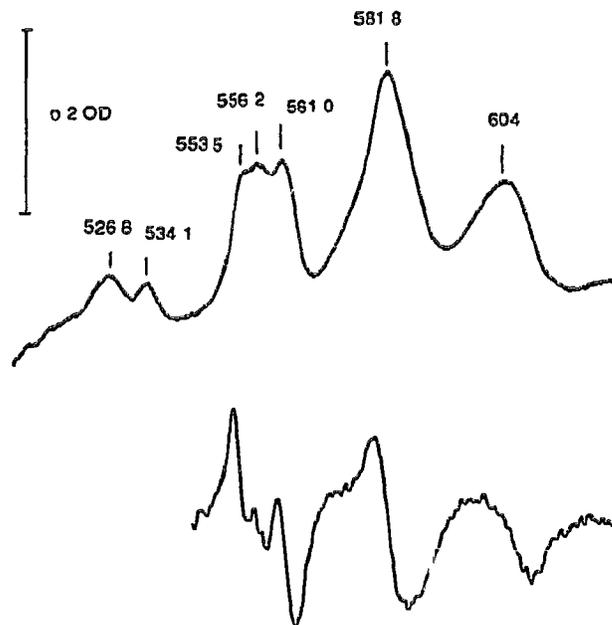


Fig. 1 Reduced minus oxidized difference spectrum of a membrane suspension from *Sulfolobus acidocaldarius* taken at liquid nitrogen temperature, 1.5 mm light path, the medium contained 50 mM imidazole, pH 7.5, 30% glycerol and 2.9 mg protein/ml final concentration. The reduced sample was prepared by addition of sodium dithionite. The bottom trace shows the 1st derivative, clearly indicating three inflection points in the α -band region of heme-*b*.

lengths located at 553.5, 556.2 and 561 nm, respectively. Also the β -bands are nicely developed. In addition, the spectrum shows the typical band of the aa_3 -oxidase and the prominent peak at 582 nm which could not be functionally attributed so far though being the most significant spectral band. While we have tentatively assigned it as a cytochrome- a_1 [6,7], it was suggested to be a constituent part of a terminal aa_3 -type oxidase described for another species of *Sulfolobus* [18]. During preparation of *b*-type hemoproteins from solubilized membranes the separation from this latter

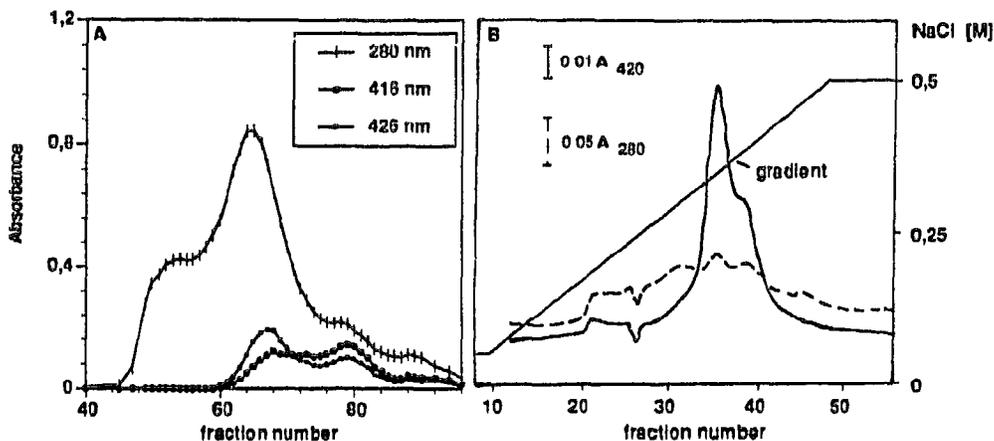


Fig. 2 Elution profiles of chromatographic separation steps. (A) Superdex-200 column, the fractions were tested individually at 280 nm for protein, at 416 nm for oxidized, and at 426 nm for reduced heme absorption, respectively. It was observed that during the course of preparation the hemes can be partially reduced. (B) Mono-Q 5/5 column, in this case protein and heme were monitored automatically at 280 nm for protein and 420 nm as an intermediate wavelength for the hemes. Other conditions as described in the text.

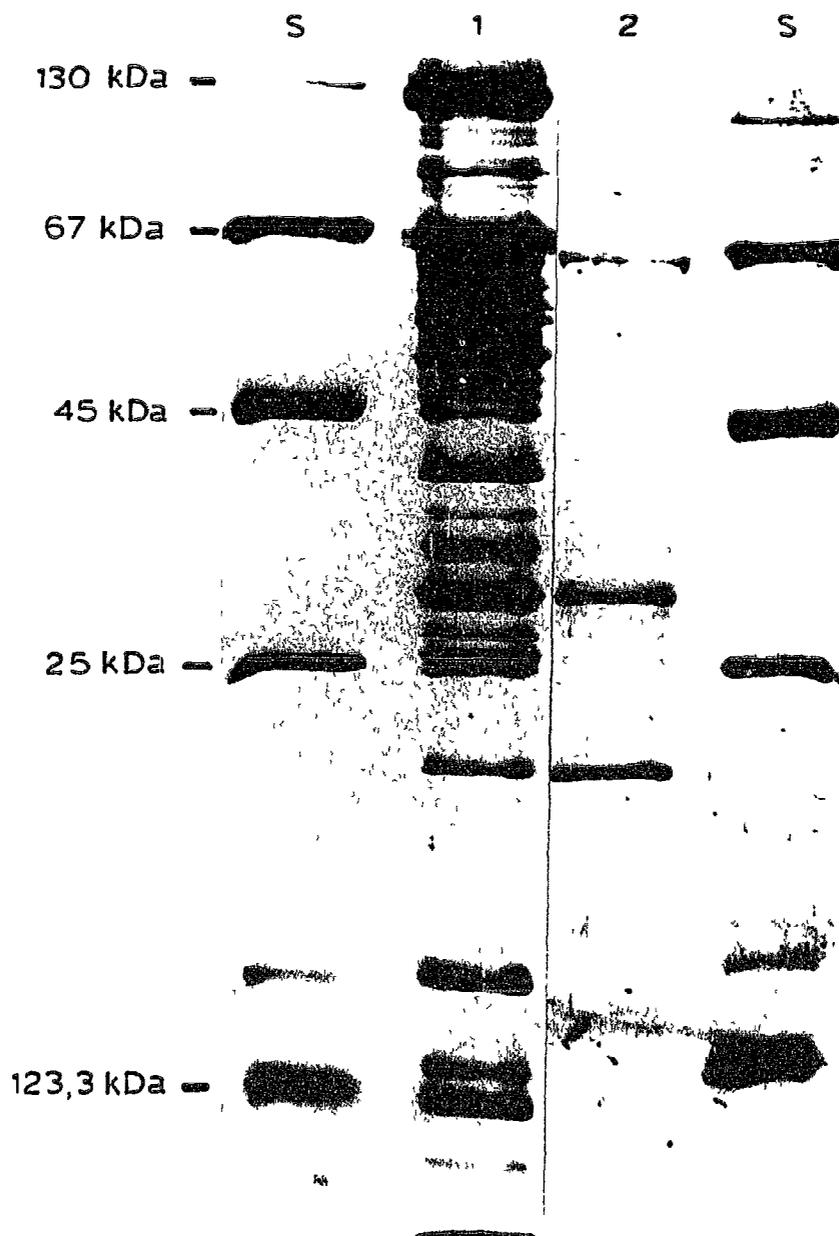


Fig. 3 SDS-gel electrophoresis of membrane extract (lane 1, 10 $\mu\text{g/ml}$) and purified cytochrome-*b* (lane 2, 3 $\mu\text{g/ml}$) from *Sulfolobus acidocaldarius*. Protein bands were visualized by silver staining. S = molecular weight markers (β -galactosidase 130 kDa, bovine serum albumin 67 kDa, ovalbumin 45 kDa, chymotrypsin 25 kDa, and cytochrome-*c* 12.3 kDa)

compound was the most critical challenge. No detergent was found which preferentially solubilized *b*-type cytochromes. In contrast, most detergents failed to preserve the native state of hemoproteins as concluded from the disappearance of spectral resolution resulting in only one coalescent heme-*b* absorption maximum at 560 nm (not shown). This was not the case when membranes were solubilized by 1% dodecylmaltoside after preextraction with 20 mM pyrophosphate which removed 7–15% of loosely membrane associated proteins (for details see Methods). On a first hydroxyapatite column, a change of detergent was achieved together with partial retention of *a*-type cytochromes. The

second hydroxyapatite column removed a large portion of contaminating protein from a slowly migrating broad heme-containing band. Actually, one of the major obstacles is the obviously strong association of the various respiratory carriers resulting in a smear of components over a wide range of fractions, thus, an inverse gradual distribution of *a*- and *b*-containing hemoproteins over a broad peak could be achieved, rather than a clear separation. A subsequent gel chromatographic step on a Superdex-200 column had the purpose to change the detergent again to a higher concentration of SB-12 only, yielding two peaks, one of which indicated a high specific enrichment of heme-*b*. Anion ex-

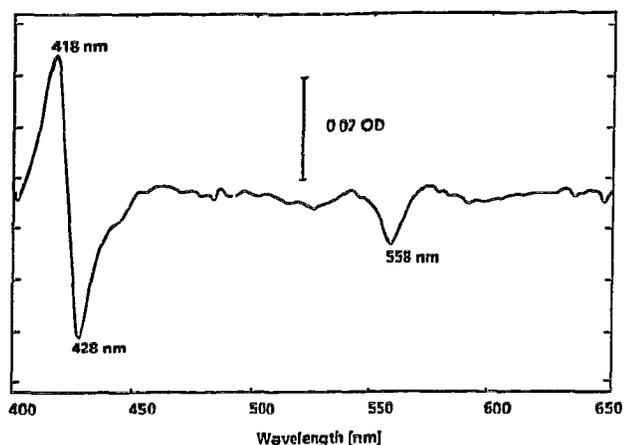


Fig 4 CO/reduced minus reduced spectrum of purified cytochrome-*b* from *Sulfolobus acidocaldarius*. Protein concentration was 0.045 mg/ml in a buffer containing 50 mM imidazole, 0.2% SB12, pH 7.5, $d = 1$ cm

change chromatography on a Mono-Q column finally gave a fraction of *b*-type cytochrome, free of any contaminating heme-*a* as verified by difference spectroscopy and pyridine hemochrome analysis. Elution profiles of the last two separation steps are given in Fig. 2. From its elution volume on a calibrated S-200 gel chromatography column a molecular mass of 48 kDa was calculated for the detergent-embedded cytochrome. Fig. 3 shows a silver stained SDS-gel of the preparation compared to the membrane extract. The band of 30–31 kDa was identified to carry the heme-*b* by heme-staining of gels run in the absence of mercaptoethanol, allowing to prevent a complete loss of the prosthetic iron porphyrin during SDS gel electrophoresis. A smaller peptide of about 18–20 kDa was frequently observed as well as a tendency to form aggregates of about 67 kDa apparent molecular mass. It is unclear whether or not this polypeptide is an additional constituent part of the cytochrome-*b* complex.

Table I gives a typical purification protocol. Though a high degree of enrichment was achieved, it also illustrates the poor yield of purified cytochrome. Obviously also some loss of heme-*b* occurred, especially during the 2nd and 3rd column steps. The final product shows a

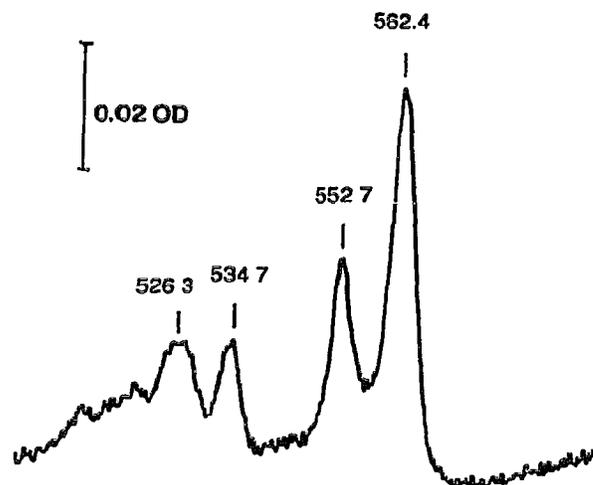


Fig 5 Low-temperature reduced minus oxidized difference spectrum of purified cytochrome-*b*. The sample was oxidized by ferricyanide (0.6 mM), after the spectrum was run and stored as a reference, the sample was reduced by a small grain of dithionite and the spectrum scanned again at liquid nitrogen temperature, light path 1 mm, 0.8 mg protein/ml in a buffer containing 0.2 M potassium phosphate pH 6.5, 30% glycerol

typical cytochrome-*b* reduced minus oxidized difference spectrum at room temperature, with maxima at 427 nm, 558 nm and a pronounced shoulder at 566 nm. It binds carbon monoxide, yielding a CO/reduced minus reduced difference spectrum typical for terminal oxidases as given in Fig. 4. The characteristic maximum at 418 nm and troughs at 428 and 558 nm are clearly developed, resembling the typical spectrum of CO-binding heme-*b* containing cytochromes. In Fig. 5 a low-temperature difference-spectrum (dithionite-reduced minus oxidized) is shown, indicating the presence of two heme-*b* centers. It parallels nicely spectra of well-characterized *o*-type cytochromes [19], displaying at 554 and 561 nm the same maxima in the α -band region as found in intact *Sulfolobus* membranes (Fig. 1). The purified cytochrome also contained tightly bound copper withstanding extensive dialysis against 10 mM EDTA. From several preparations an average copper content of 0.4–0.6 mol/mol heme-*b* was determined, suggesting that the comparatively small 30 kDa-polypeptide hosts two

Table I
Typical purification protocol for cytochrome-*b* from *Sulfolobus acidocaldarius*

Purification state	Total protein (mg)	Specific heme- <i>b</i> content (nmol/mg)	Purification factor <i>n</i>
Membranes	715	0.29	1
Preextract	55	–	–
DM extract	163	0.63	2.2
(1) Hydroxyapatite column	66	1.39	4.8
(2) Hydroxyapatite column	11	1.74	6
Gel filtration	1.22	4.65	18
Anion exchange	0.033	39.2	134

es and one copper associated with one of the hemes expected for an *o*-type terminal oxidase. Despite similarities no significant catalytic activity as a oxidase or a TMPD-oxidase could be verified. Although an almost negligible activity with caldariella quinone could be detected, the preparation is likely to be inactive due to the loss of either additional subunits, or specific lipids essential for catalytic activity. Addition of soybean asolectins or a crude *Sulfolobus* lipid extract did not improve catalytic activity. It should be mentioned, however, that in membrane extracts by redox titrations the presence of high- and low-potential heme cytochromes could be established which would give values for a functional *o*-type cytochrome [6,20]

DISCUSSION

The presence of *b*-type cytochromes in aerobically grown archaeobacteria has been shown for *Halobacterium cutirubrum* [21] and *H. halobium* [22,23]. None of these has been purified so far and, moreover, their redox potentials differed significantly when determined either in solution or in membranes, respectively. Nevertheless, at least one of those might represent a terminal oxidase as concluded from its CO-binding properties. Here for the first time isolation and spectral characterization of a highly purified *b*-type cytochrome from a thermoacidophilic archaeobacterium, *Sulfolobus acidocaldarius*, could be demonstrated. Though its heme-*b* content, copper content, and CO-binding properties resemble those of Q-oxidizing *o*-type cytochromes, a corresponding catalytic activity could not be preserved. An important difference to known *o*-type cytochromes [24] is the apparent molecular mass of the isolated polypeptide carrying the heme centers. In this regard it falls rather to *b*-type cytochromes acting as intermediary iron carriers. However, since more than only one terminal oxidase is present in this archaeobacterial genus, together with the fact that a low molecular weight quinol-oxidase was identified [9], the assumption of a rather simple *o*-type terminal oxidase may be justified. Actually, the partially cyanide-insensitive reduction at low inhibitor concentration, sufficient to block cytochrome-*aa*₃ completely, would be in line with this concept. Besides that, it was shown in previous studies [6] that a high-potential *b*-type cytochrome (E° 300 mV) could be detected by redox-titration in membrane extracts from *Sulfolobus*, supporting the existence of an alternative terminal oxidase apart from cytochrome-*aa*₃. In addition, qualitative observations indicating that the ratio of *a*-type versus *b*-type cytochromes in *Sulfolobus* varies with oxygen availability (unpublished results). Caldariella quinone serves as an intermediary pool for reducing equivalents in the

membrane of *Sulfolobus* and may provide the specific reductant for both terminal oxidases presumably differing in oxygen affinity. Regrettably, it was impossible to follow the catalytic activity during purification due to competition with the highly active *aa*₃-oxidase. The failure to isolate the inferred *o*-type cytochrome in active form may result from the loss of specific lipids or of other constituent polypeptides for an active complex. The latter is not unlikely, taking into account an obvious tendency of *Sulfolobus* membrane proteins to desintegrate under conditions necessary to resolve the extraordinary rigid membrane structure. Sequencing of the *b*-polypeptide is in progress and is expected to allow a definite functional assignment by comparison to known heme-*b* containing cytochromes.

Acknowledgements Our thanks are due to Mrs Doris Mutschall for skilful technical assistance

REFERENCES

- [1] Moll, R. and Schafer, G. (1988) FEBS Lett 232,359-363
- [2] Lubben, M. and Schafer, G. (1989) J Bact 171, 6106-6116
- [3] Wakao, H., Wakagi, T. and Oshima, T. (1987) J Biochem (Tokyo) 102, 255-262
- [4] Moll, R. and Schafer, G. (1989) Biol Chem Hoppe-Seyler 370, 936
- [5] Anemüller, S., Lubben, M. and Schafer, G. (1985) FEBS Lett 193, 83-87
- [6] Schäfer, G., Anemüller, S., Moll, R., Meyer, W. and Lübben, M. (1990) FEMS Microbiol Rev 75, 335-348
- [7] Schäfer, G., Lübben, M. and Anemüller, S. (1990) Biochim Biophys Acta 1018, 271-274
- [8] Anemüller, S. and Schäfer, G. (1989) FEBS Lett 244, 451-455
- [9] Anemüller, S. and Schäfer, G. (1990) Eur J Biochem 191, 653-657
- [10] Becker, M. and Schafer, G. (1990) 6th EBEC Short Reports, 102 Elsevier Amsterdam
- [11] Brock, T.D., Brock, K.M., Belly, R.T. and Weiss, R.L. (1972) Arch Microbiol 84, 54-68
- [12] Williams, J.N. (1964) Arch Biochem Biophys 107, 537-543
- [13] Thiselius, A., Hierten, S. and Levin Ö. (1956) Arch Biochem Biophys 65, 132-155
- [14] Watters, C. (1978) Anal Biochem 88, 695-698
- [15] Peterson, G.L. (1977) Anal Biochem 83, 346-356
- [16] Laemmli, U.K. (1970) Nature 227, 680-685
- [17] Heukeshoven, J. and Dernick, R. (1985) Electrophoresis 6, 103-112
- [18] Wakagi, T., Yamauchi, T., Oshima, T., Müller, M., Azzi, A. and Sonc, N. (1990) Biochem Biophys Res Commun 165, 1110-1114
- [19] Kita, K., Konishi, K. and Anraku, Y. (1984) J Biol Chem 259, 3368-3374
- [20] Jones, C.W. and Poole, R.K. (1985) Methods Microbiol 18, 285-328
- [21] Lanyi, J.K. (1968) Arch Biochem Biophys 128, 716-724
- [22] Hallberg, C. and Baltischeffsky, H. (1981) FEBS Lett 125, 201-204
- [23] Hallberg-Gladin, C. and Calmsjo, A. (1989) Arch Biochem Biophys 272, 130-136
- [24] Anraku, Y. and Gennis, R.B. (1987) Trends Biochem Sci 12, 262-266