

# Regulation of photosynthetic light harvesting by nitrogen assimilation in the green alga *Selenastrum minutum*

David H. Turpin\* and Doug Bruce

\*Department of Biology, Queen's University, Kingston, Ont. K7L 3N6 and Department of Biological Sciences, Brock University, St. Catharines, Ont. L2S 3A1, Canada

Received 6 February 1990

The interaction of whole cell metabolism with the distribution of excitation energy between photosystem 2 (PS2) and photosystem 1 (PS1), the light state transition, was investigated in vivo in the green alga *Selenastrum minutum*. Nitrogen limited cells of *S. minutum* were presented with a pulse of either  $\text{NH}_4^+$  or  $\text{NO}_3^-$  in the light. As shown previously,  $\text{CO}_2$  fixation is inhibited and high rates of N assimilation ensue [(1986) Plant Physiol. 81, 273–279].  $\text{NH}_4^+$  assimilation has a much higher requirement ratio for ATP/NADPH than either  $\text{CO}_2$  or  $\text{NO}_3^-$  assimilation and thus drastically increases the demand for ATP relative to reducing power. Room temperature chlorophyll *a* fluorescence kinetic measurements showed that a reversible non-photochemical quenching of PS2 fluorescence accompanied the assimilation of  $\text{NH}_4^+$  but not the assimilation of  $\text{NO}_3^-$  or  $\text{CO}_2$ . 77K fluorescence emission spectra taken from samples removed at regular intervals during  $\text{NH}_4^+$  assimilation showed that the non-photochemical quenching of PS2 was accompanied by a complementary increase in the fluorescence yield of PS1, characteristic of a transition to state 2. Our data suggests that *S. minutum* responds to the increased demand for ATP/NADPH during  $\text{NH}_4^+$  assimilation by inducing the light state transition to direct more excitation energy to PS1 at the expense of PS2 to increase the production of ATP by cyclic electron transport.

Photosynthesis; State transition; Chlorophyll fluorescence; Nitrogen assimilation; *Selenastrum minutum*

## 1. INTRODUCTION

In oxygenic photosynthesis the distribution of excitation energy between photosystem 2 (PS2) and photosystem 1 (PS1) is regulated by the light state transition [1,2]. The mechanism of the state transition in green algae and higher plants involves a complementary change in the association of the chl *a/b* light harvesting complex (LHCII) with PS2 and PS1. Reduction of plastoquinone (PQ) and/or the cytochrome (Cyt)  $b_6/f$  complex by excess PS2 activity or chlororespiration [3] in green algae activates a thylakoid membrane-bound kinase which phosphorylates LHCII. The phosphorylation of LHCII is believed to decrease its association with PS2 and increase its association with PS1 via a lateral migration of the phospho-LHCII in the thylakoid membrane. The resulting state is characterized by decreased PS2 activity and increased PS1 activity and is called state 2. Conversely, oxidation of PQ and/or the Cyt  $b_6/f$  complex by excess PS1 activity leads to inactivation of the kinase and background phosphatase activity cleaves the phospho-LHCII which then returns to PS2. The re-association of LHCII with PS2 increases PS2 activity at the expense of PS1 activity and defines state 1. The state transition is most often thought of as a regulatory mechanism which serves to

optimize the relative activities of PS2 and PS1 under environmental conditions which result in preferential excitation of either PS2 or PS1 [1,2].

It has been shown in intact chloroplasts [4,5] that the state transition is affected by metabolism. In intact maize chloroplasts [4] the degree of phosphorylation observed on transition to state 2 was decreased by the addition of oxaloacetate (which requires reducing equivalents but no ATP) and was increased by pyruvate (which requires ATP but no reducing equivalents). It was suggested by Fernyhough et al. [4] that the demand for NADPH would directly affect the redox state of PQ and thus control the activity of the LHCII kinase as described above. In this way, as suggested by Horton [6], an increased demand for ATP relative to NADPH would trigger LHCII phosphorylation, increased PS1 activity and increased production of ATP via cyclic electron transport.

Horton and Lee [5] have also shown the distribution of excitation energy between PS2 and PS1 (as determined by the degree of Emerson enhancement) to be dependent on the electron acceptor used in intact spinach chloroplasts.

When the green alga *Selenastrum minutum* is cultured under N limitation its capacity for nitrogen assimilation increases [7]. N resupply results in rates of N assimilation into protein which can exceed control rates of photosynthetic carbon fixation. The onset of these rapid rates of N assimilation causes a shut down

Correspondence address: D. Bruce, Department of Biological Sciences, Brock University, St. Catharines, Ont., Canada L2S 3A1

of carbon fixation and the mobilization of starch to provide carbon skeletons for amino acid synthesis [8,9]. The reductant and ATP requirements associated with  $\text{CO}_2$ ,  $\text{NH}_4^+$  and  $\text{NO}_3^-$  assimilation vary and are shown in Table I. The fixation of  $\text{CO}_2$  into carbohydrate by the Calvin cycle has an ATP/NADPH consumption ratio of 1.5. The reduction of  $\text{NO}_3^-$  and its assimilation into protein has an ATP/NADPH requirement ratio of 1.2 while the ATP/NADPH requirement ratio climbs to 5 for the assimilation of  $\text{NH}_4^+$  into protein.

A summary of the photosynthetic characteristics of N limited *S. minutum* and the effects of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  assimilation is shown in Table II. The addition of  $\text{NH}_4^+$  or  $\text{NO}_3^-$  to these cells results in high rates of N assimilation [10] which correspond to a decline in the concentration of ribulose biphosphate (RuBP) below the active site density of ribulosebiphosphate carboxylase (Rubisco) [11]. This RuBP limitation of Rubisco causes a rapid drop in the rate of photosynthetic carbon fixation [11,12]. For  $\text{NH}_4^+$  this decline in carbon fixation decreases the demands for photo-generated reductant drastically as illustrated by the decline in gross  $\text{O}_2$  evolution. However, there is no change in gross  $\text{O}_2$  evolution after addition of  $\text{NO}_3^-$ , because  $\text{NO}_3^-/\text{NO}_2^-$  replaces  $\text{CO}_2$  as the terminal electron acceptor in non-cyclic electron flow [12].

The supply of  $\text{NH}_4^+$  or  $\text{NO}_3^-$  to N limited *S. minutum* causes the cells to change from a physiological state characterized by  $\text{CO}_2$  fixation to one in which N assimilation is the major metabolic pathway. However, only  $\text{NH}_4^+$  assimilation is characterized by a dramatic increase in the requirement for ATP relative to NADPH.

In the present study we exploit the changes in metabolism induced by resupply of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  to investigate metabolic control of the light state transition by the green alga *S. minutum* in vivo. We show that the metabolic demands induced by  $\text{NH}_4^+$  assimilation regulate the distribution of excitation energy in photosynthesis to control the production of ATP and NADPH.

## 2. MATERIALS AND METHODS

*Selenastrum minutum* (UTEX 2459) was cultured in  $\text{NO}_3^-$  limited chemostats as previously described [8]. Cells were harvested by centrifugation and used at a final concentration between 4–5  $\mu\text{g}$  chl  $\text{ml}^{-1}$ . Forty ml of cells were placed in a water-jacketed cuvette for room temperature fluorescence assay.

Room temperature fluorescence emission was monitored continuously during  $\text{NH}_4^+$  and  $\text{NO}_3^-$  assimilation with a PAM fluorimeter (H. Walz Inst., Effeltrich, FRG). The PAM fluorimeter separates the effects of the redox state of PS2 (photochemical quenching) on fluorescence emission from other effects (including state transitions) which contribute to non-photochemical quenching [13]. Frequency sensitive detection ensures that only fluorescence excited by the fluorimeters modulated light source is monitored. The intensity of this source is too low to stimulate PS2 photochemistry and in the

absence of additional illumination the minimal fluorescence yield characteristic of open reaction centres ( $F_0$ ) is determined. Periodically the sample is presented with a brief flash of light intense enough to saturate the photochemistry of PS2. Whilst this light is on, the modulated beam measures the fluorescence level characteristic of closed reaction centres ( $(F_v)s$ ). The addition of a subsaturating light source will induce variable fluorescence ( $F_v$ ) which lies between  $F_0$  and  $(F_v)s$ .

For each experiment cells were initially allowed to dark-adapt for 30 min and  $F_0$  and  $(F_v)s$  were determined for the dark-adapted cells. Cells were then illuminated with white light from a 300 W projector lamp attenuated to an incident intensity of  $78 \mu\text{E m}^{-2} \text{s}^{-1}$  with neutral density filters.  $F_v$  induced by this light was monitored continuously and  $(F_v)s$  was determined periodically with saturating flashes until stable values were attained. Either  $\text{NH}_4\text{Cl}$  or  $\text{NaNO}_3$  was subsequently added to a final concentration of 1 mM and  $F_v$  and  $(F_v)s$  were monitored until the added N was depleted and fluorescence levels were recovered to pre N-addition levels. The cuvette was finally darkened and  $F_v$  monitored until a stable level was obtained.

77K fluorescence emission spectra were determined with a home built fluorimeter based on an optical multichannel analyser (EG&G PARC, New Jersey, USA) described previously [14]. Samples (100  $\mu\text{l}$ ) were removed from the water-jacketed cuvette at regular intervals during the time course of each experiment, quickly placed into capillary tube sample holders and frozen in liquid nitrogen for later 77K fluorescence emission assay. The averages of 20 scans were taken over the surface of each capillary tube and the relative yields of fluorescence emission between identical samples were accurate to an SE of 4%. Fluorescence yield changes between samples could thus be detected with confidence and normalization of the fluorescence emission spectra to intrinsic or extrinsic fluorescence markers was not required.

## 3. RESULTS AND DISCUSSION

The distribution of excitation energy between PS2 and PS1 can be estimated by measuring the fluorescence emission from PS2 and PS1. Room temperature fluorescence arises primarily from PS2, however, both PS2 and PS1 fluorescence can be measured at 77K. Room temperature fluorescence emission is sensitive to the amount of excitation energy reaching PS2, the redox state of PS2 and the electrochemical gradient across the thylakoid membrane [15,16]. As described in section 2, we measured room temperature fluorescence emission with a PAM fluorimeter to separate the effects of the redox state of PS2 (photochemical quenching) on fluorescence emission from non-photochemical quenching (which includes state transitions).

Fig. 1 shows the room temperature fluorescence emission trace recorded during the assimilation of  $\text{NH}_4^+$ . Cells were dark-adapted for 30 min before the  $F_0$  determination shown at the start of the trace. At the arrow marked 'light on' the cells were illuminated with  $78 \mu\text{E m}^{-2} \text{s}^{-1}$  of white light causing the induction of  $F_v$  indicative of the photoreduction of  $Q_a$  (the primary quinone electron acceptor of PS2). Determinations of  $(F_v)s$ , the light saturated fluorescence yield characteristic of fully reduced  $Q_a$ , appear as large spikes superimposed on the  $F_v$  trace.  $(F_v)s^{\text{dark}}$  refers to the light-saturated fluorescence yield of dark-adapted cells determined just prior to illumination.

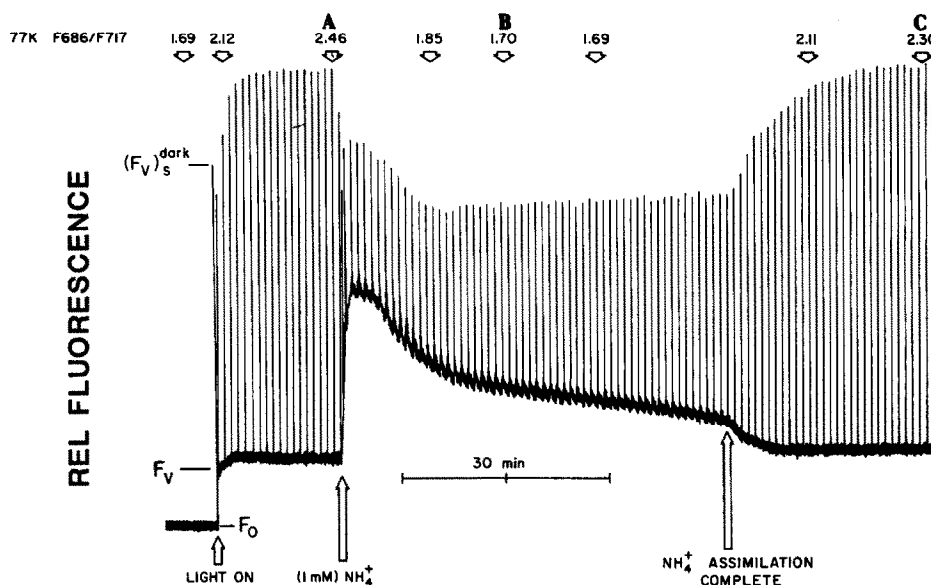


Fig. 1. Room temperature fluorescence induction measured during the time course of  $\text{NH}_4^+$  assimilation by N limited *S. minutum* with a PAM fluorimeter as described in the text. Variable fluorescence ( $F_v$ ) is shown by the continuous trace and the light saturated fluorescence yield ( $F_v$ )s is given by the amplitude of the spikes shown on the continuous trace. During the course of this experiment samples were removed periodically for concurrent 77K fluorescence emission determination. The ratio of PS2 to PS1 fluorescence emission at 77K (F686/F717) for each of these samples is noted at the time of sample removal above the PAM trace. The 77K emission spectra of samples A, B and C are shown in Fig. 2b.

As ( $F_v$ )s is determined when  $Q_a$  is fully reduced, its amplitude is independent of photochemical quenching and is proportional to the amount of excitation energy reaching PS2 minus all non-photochemical quenching processes. ( $F_v$ )s was observed to increase in the light over a time period of about 12 min to a level 25% higher than the dark level. This is indicative of either an increase in the amount of excitation energy reaching PS2 or the relief of a dark non-photochemical quenching state. The relative distribution of excitation energy between PS2 and PS1 can be estimated from 77K fluorescence emission spectra [17,18]. Over this same time period the ratio of PS2 (686 nm) to PS1 (717 nm) fluorescence emission at 77K (F686/F717) also increased from a dark value of 1.69 to 2.46 in the light. The 77K data indicated that a change in the amount of excitation energy reaching PS2 was responsible for the increase in ( $F_v$ )s. In Fig. 2a the 77K fluorescence emission spectra clearly show that the increase in ( $F_v$ )s at room temperature accompanying the dark to light transition was correlated to an increase in PS2 (686 nm, 696 nm) emission and a decrease in PS1 emission (717 nm) at 77K. Such a complementary change in PS2 and PS1 fluorescence emission is characteristic of a state 2 to state 1 transition and confirms earlier reports that green algae [19,20], like cyanobacteria [14], are in state 2 in the dark and convert to state 1 upon illumination.

As reported previously [12], the initiation of  $\text{NH}_4^+$  assimilation in Fig. 1 resulted in a rapid increase in  $F_v$  accompanied by a rapid oscillation of ( $F_v$ )s. The in-

crease in  $F_v$  reflects reduction of  $Q_a$ , most likely due to a build up of NADPH resulting from the decreased requirement for reductant during  $\text{NH}_4^+$  assimilation. The initial oscillation of ( $F_v$ )s reflects a transient increase in non-photochemical quenching [12]. Over the next 20 min both  $F_v$  and ( $F_v$ )s declined at a similar rate. Following this initial decline,  $F_v$  continued to slowly decrease while ( $F_v$ )s maintained a low steady state until all of the added  $\text{NH}_4^+$  had been assimilated. The large

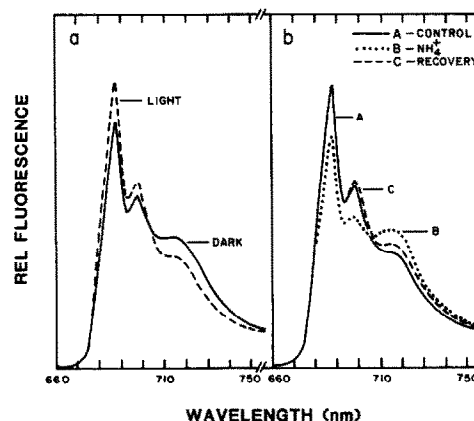


Fig. 2. 77K fluorescence emission spectra of *S. minutum*. (a) Cells dark-adapted for 20 min are compared to cells light-adapted ( $78 \mu\text{E m}^{-2} \text{s}^{-1}$ ) for 20 min at room temperature prior to freezing in liquid nitrogen. (b) Fluorescence emission spectra of light-adapted cells before (A), during (B) and after (C)  $\text{NH}_4^+$  assimilation. These spectra were obtained from samples taken at the times marked A, B and C in Fig. 1. The excitation wavelength was 435 nm. The spectra have not been normalized.

increase in non-photochemical quenching demonstrated by the decline of ( $F_v$ )s was mirrored by a decrease in F686/F717 from 2.46 just prior to addition of  $\text{NH}_4^+$  to 1.70 after about 20 min of assimilation. As described above the 77K data indicate that the non-photochemical quenching of PS2 was indicative of a decrease in the amount of excitation energy reaching PS2. After  $\text{NH}_4^+$  assimilation was complete  $F_v$  and ( $F_v$ )s recovered to their pre  $\text{NH}_4^+$  levels as did F686/F717. The addition of  $\text{NH}_4^+$  does not change fluorescence or metabolism in the presence of the inhibitors of  $\text{NH}_4^+$  assimilation, MSX and azaserine [7,10].

The 77K fluorescence spectra of Fig. 2b clearly show that the reversible decrease in ( $F_v$ )s induced by  $\text{NH}_4^+$  assimilation was accompanied by a reversible decrease in PS2 fluorescence emission (686 nm, 696 nm) and increase in PS1 emission (717 nm). These spectra confirm a complementary change in the distribution of energy between PS2 and PS1 and demonstrate a reversible transition from state 1 during light driven  $\text{CO}_2$  assimilation to state 2 during light driven  $\text{NH}_4^+$  assimilation.

In the control experiment (not shown),  $\text{NO}_3^-$  assimilation resulted in relatively small changes in room temperature and 77K fluorescence emission. There was no evidence for a change in the distribution of excitation energy between PS2 and PS1 during  $\text{NO}_3^-$  assimilation. As discussed earlier both the inhibition of  $\text{CO}_2$  fixation and rates of N assimilation have been shown to be very similar for  $\text{NO}_3^-$  and  $\text{NH}_4^+$  but only  $\text{NH}_4^+$  assimilation is accompanied by a decline in linear electron flow (oxygen evolution) (Table II).

Our data show that assimilation of  $\text{NH}_4^+$  in the light induces a reversible transition to state 2 (an increase in the amount of excitation energy reaching PS1 and decrease of that reaching PS2). This change should stimulate cyclic electron transport around PS1 at the expense of linear electron transport and thus increase the chloroplasts production of ATP relative to NADPH, matching the requirements of  $\text{NH}_4^+$  assimilation. The  $\text{NH}_4^+$  induced transition to state 2 is likely triggered by the reduction of PQ and/or the Cyt  $b_6/f$  complex which results from the decreased requirement for photogenerated reductant during  $\text{NH}_4^+$  assimilation. Reduction of the electron transport carriers was demonstrated by the initial increase in  $F_v$  induced by  $\text{NH}_4^+$ . The assimilation of  $\text{NO}_3^-$  occurs at similar rates to and inhibits  $\text{CO}_2$  fixation as much as  $\text{NH}_4^+$  assimilation but does not induce state transitions. However, the assimilation of  $\text{NO}_3^-$  requires photogenerated reductant and therefore does not limit linear electron transport or cause reduction of PQ and/or the Cyt  $b_6/f$  complex.

Light-induced state transitions in green algae are probably rare under environmental light conditions as the similarity in the absorption characteristics of the antenna pigments of PS2 and PS1 would make a strong imbalance in energy distribution unlikely.

Table I

The reductant and ATP requirements associated with  $\text{CO}_2$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  assimilation. Reductant requirements (NAD(P)H or ferridoxin) are represented as electron pairs,  $2e^-$

Substrate	$2e^-$	ATP	ATP/ $2e^-$
$\text{CO}_2^a$	2	3	1.5
$\text{NO}_3^-^b$	5	6	1.2
$\text{NH}_4^+^c$	1	5	5

<sup>a</sup> The fixation of one molecule of  $\text{CO}_2$  via ribulose biphosphate carboxylase produces 2 molecules of 3-phosphoglycerate. The subsequent conversion to triose phosphate requires 2ATP and 2NADPH. Regeneration of ribulose biphosphate requires an additional ATP

<sup>b</sup> We assume the transport of  $\text{NO}_3^-$  into the cell requires 1 ATP. Reduction of  $\text{NO}_3^-$  to  $\text{NH}_4^+$  requires 4 NAD(P)H. The subsequent assimilation via glutamine synthetase/glutamine 2 oxoglutarate aminotransferase requires ATP and 2 ferridoxin or one electron pair. Four additional ATP equivalents are required for the addition of an amino acid to a polypeptide chain

<sup>c</sup> We assume  $\text{NH}_3$  enters the cell via diffusion. Assimilation of  $\text{NH}_4^+$  via glutamine synthetase/glutamine 2 oxoglutarate aminotransferase requires 1 ATP and 2 ferridoxin (one electron pair). Subsequent incorporation of an amino acid into a poly peptide chain requires 4 ATP equivalents. If we assumed  $\text{NH}_4^+$  transport into the cell required ATP then the ATP/ $2e^-$  ratio would increase from 5 to 6

Our data support the idea of a metabolic control of state transitions and demonstrate this control in vivo under physiological conditions. We show that the state transition is most likely triggered by the redox state of intersystem electron carriers controlled directly by the whole cells downstream requirements for photosynthetic reducing power. Planktonic algae are frequently confronted by nutrient deficiencies in natural systems [21] where nutrient supply can be of a transient nature [22,23]. A transitory supply of nutrients can induce large changes in metabolic demands and, as demonstrated in this work, metabolic regulation of the light state transition can take maximal advantage of a transient nutrient source.

Table II

The effects of N assimilation on photosynthetic metabolism by the N limited green alga *Selenastrum minutum*

Treatment	Rate of N assimilation <sup>a</sup> [9]	Gross $\text{CO}_2$ fixation <sup>a</sup> [11]	Gross $\text{O}_2$ evolution <sup>a</sup> [11]	[RuBP] <sup>b</sup> [6,10]	Rubisco active site density <sup>b</sup> [6]
Control	0	125	115	109	60
+ $\text{NH}_4^+$	170	40	60	40	60
+ $\text{NO}_3^-$	155	35	115	20	60

<sup>a</sup> Values in  $\mu\text{mol} \cdot \text{mg}^{-1} \text{ chl} \cdot \text{h}^{-1}$

<sup>b</sup> Values in  $\text{nmol} \cdot \text{mg}^{-1} \text{ chl}$

*Acknowledgements:* We would like to thank B.G. Winsborrow for her help. This research was supported by NSERC operating grants to D.H.T. and D.B.

## REFERENCES

- [1] Williams, W.P. and Allen, J.F. (1987) *Photosyn. Res.* 13, 19–45.
- [2] Fork, D.C. and Satoh, K. (1986) *Annu. Rev. Plant Physiol.* 37, 335–361.
- [3] Wollman, F.-A. and Delepelaire, J. (1984) *J. Cell Biol.* 98, 1–7.
- [4] Fernyhough, P., Foyer, C. and Horton, P. (1983) *Biochim. Biophys. Acta* 725, 155–161.
- [5] Horton, P. and Lee, P. (1986) *Photosyn. Res.* 10, 297–302.
- [6] Horton, P. (1987) in: *Progress in Photosynthesis Research* (Biggins, J. ed.) pp.681–688, Nijhoff, Dordrecht.
- [7] Elrif, I.R. and Turpin, D.H. (1986) *Plant Physiol.* 81, 273–279.
- [8] Elrif, I.R. and Turpin, D.H. (1987) *Plant Physiol.* 83, 97–104.
- [9] Weger, H.G., Birch, D.G., Elrif, I.R. and Turpin, D.H. (1988) *Plant Physiol.* 86, 688–692.
- [10] Weger, H.G. and Turpin, D.H. (1989) *Plant Physiol.* 89, 409–415.
- [11] Elrif, I.R., Holmes, J.J., Weger, H.G., Mayo, W.P. and Turpin, D.H. (1988) *Plant Physiol.* 87, 395–401.
- [12] Holmes, J.J., Weger, H.G. and Turpin, D.H. (1989) *Plant Physiol.* 91, 331–337.
- [13] Schreiber, U., Schliwa, U. and Bilger, W. (1986) *Photosyn. Res.* 10, 51–62.
- [14] Bruce, D., Brimble, S. and Bryant, D.A. (1989) *Biochim. Biophys. Acta* 974, 66–73.
- [15] Krause, G.H. and Weiss, E. (1984) *Photosynth. Res.* 5, 139–157.
- [16] Quick, W.P. and Horton, P. (1984) *Proc. R. Soc. Lond. B* 220, 371–382.
- [17] Butler, W.L. (1978) *Annu. Rev. Plant Physiol.* 29, 345–378.
- [18] Krause, G.H. and Behrend, U. (1983) *Biochim. Biophys. Acta* 723, 176–181.
- [19] Sane, P.V., Furtado, D., Desai, T.S. and Tatake, V.G. (1982) *Z. Naturforsch.* 37c, 458–463.
- [20] Satoh, K. and Fork, D. (1983) *Photochem. Photobiol.* 37, 429–434.
- [21] Dougdale, R.C. (1967) *Limnol. Oceanogr.* 12, 685–695.
- [22] McCarthy, J.J. and Goldman, J.C. (1979) *Science* 203, 670–672.
- [23] Goldman, J.C., McCarthy, J.J. and Peavey, D.G. (1979) *Nature* 279, 210–215.