

# An electrogenic uniport mediates light-dependent $\text{Ca}^{2+}$ influx into intact spinach chloroplasts

Georg Kreimer, Michael Melkonian and Erwin Latzko\*

*Botanisches Institut der Universität Münster, Schloßgarten 3, D-4400 Münster, FRG*

Received 19 November 1984

Light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts, measured with the metallochromic indicator arsenazo III, is stimulated by uncouplers (FCCP, CCCP, nigericin) and inhibited by ruthenium red. The data presented demonstrate that light-dependent  $\text{Ca}^{2+}$  influx into chloroplasts is electrogenic and mediated by a uniport-type carrier. The characteristics of the carrier system are similar to those of the  $\text{Ca}^{2+}$  uniport of mitochondria.

*Spinach    Chloroplast     $\text{Ca}^{2+}$     Electrogenic uptake    Uniport*

## 1. INTRODUCTION

Calcium exhibits specific functions in the regulation of growth, development, and metabolism in plants [1,2]. It may affect cellular processes either directly or through the  $\text{Ca}^{2+}$ -calmodulin regulatory system [1-4]. Light mediates the regulation of several chloroplast enzymes, some of which are  $\text{Ca}^{2+}$ -dependent enzymes [3,4]. In addition, the existence of specific  $\text{Ca}^{2+}$ -binding sites at photosystem II [5,6] and the involvement of  $\text{Ca}^{2+}$  in the spillover-phenomenon [7] suggest an important role for calcium within the chloroplast. Recently it has been shown that  $\text{Ca}^{2+}$  influx into wheat and spinach chloroplasts is light-dependent [8,9]. In this study we characterize the light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts as an electrogenic process mediated by a uniport-type carrier system.

**Abbreviations:** CCCP, carbonyl cyanide *m*-chlorophenyl hydrazone; FCCP, carbonyl cyanide *p*-trifluoromethoxyphenylhydrazone; DCMU, 3-(3,4-dichlorophenyl)-1, 1-dimethylurea;  $\Delta \bar{\mu}_{\text{H}^+}$  proton motive force;  $\Delta E$ , membrane potential;  $\Delta \text{pH}$ , transmembrane pH gradient; PGA, 3-phosphoglyceric acid

\* To whom correspondence should be addressed

## 2. MATERIALS AND METHODS

### 2.1. Chemicals

Percoll was obtained from Pharmacia; arsenazo III, valinomycin, nigericin, CCCP, and ATP from Sigma; FCCP from Serva; vanadate from EGA-Chemie, and Chelex-100-200 mesh was from Bio-Rad. Spinach calmodulin was kindly provided by M. Burchert (this laboratory). All other chemicals were from Merck and of analytical grade.

### 2.2. Preparation and purification of spinach chloroplasts

Spinach was grown as in [10]. Chloroplasts were isolated as in [11] except that the resuspension medium contained 330 mM sorbitol, 1 mM  $\text{MgCl}_2$  and 50 mM Hepes-KOH (pH 7.0). For further purification discontinuous percoll gradients were used. The gradient medium contained 330 mM sorbitol, 20 mM KCl, 20 mM Hepes-Tris (pH 7.0). In 12 ml tubes 4.5 ml of chloroplast suspension was layered on top of 1.5 ml 65% (v/v) and 4 ml 40% percoll and centrifuged for 3 min at  $1500 \times g$ . Intact chloroplasts were collected with a Büchler Auto-Densi-Flow II gradient fractionator, diluted with an equal volume of resuspension medium and pelleted by a 50 s centrifugation at  $750 \times g$  to remove the percoll. The pellet was resuspended in

a medium (treated with Chelex-100) which contained 330 mM sorbitol, 20 mM Hepes-Tris (pH 7.0), 20 mM KCl and 1 mM  $\text{MgCl}_2$  (the latter added after Chelex-100 treatment). Purity of the chloroplast suspension was evaluated by (a) electron microscopy, and (b) the marker enzymes catalase [12], cytochrome-c-oxidase [13], nitrate reductase [14], and NADH-cytochrome-c-reductase [15]. No contaminating organelles were detected, and contamination with cytoplasmic enzymes was negligible (e.g., nitrate reductase: 3%). Intactness of chloroplasts was measured as in [16] and was always  $\geq 90\%$ . PGA (0.5 mM)-dependent oxygen evolution was measured according to [17], but NaCl and catalase were omitted. Chlorophyll was determined as in [18].

### 2.3. Measurement of $\text{Ca}^{2+}$ influx

Arsenazo III was purified as in [19]. The standard assay contained 330 mM sorbitol, 20 mM Hepes-Tris (pH 7.0), 20 mM KCl, 2 mM  $\text{MgCl}_2$ , 1 mM ATP, 0.5 mM PGA, 50  $\mu\text{M}$  arsenazo III, 50–60  $\mu\text{M}$   $\text{CaCl}_2$ , and 10–20  $\mu\text{g}$  chlorophyll. Sorbitol, Hepes and KCl solutions were treated with Chelex-100 to reduce the  $\text{Ca}^{2+}$  content. The extinction changes were followed with a double beam spectrophotometer (Sigma ZWS II) at 660–690 nm according to [20]. The standard conditions were 25°C, 3 min preincubation in the dark with the different additions, then illumination with 85  $\text{W} \cdot \text{m}^{-2}$  in a stirred and cooled cuvette. As a control the same assay was done in the presence of 100  $\mu\text{M}$  EGTA without  $\text{CaCl}_2$ . The measured extinction change of the control was subtracted from that measured for the assay (without EGTA). In all experiments, except the time-course experiment,  $\text{Ca}^{2+}$  influx/efflux was measured after a 2 min period of illumination. Internal calibrations were done as in [20] for each different set of conditions. Different free  $\text{Ca}^{2+}$  concentrations were obtained using EGTA buffers.

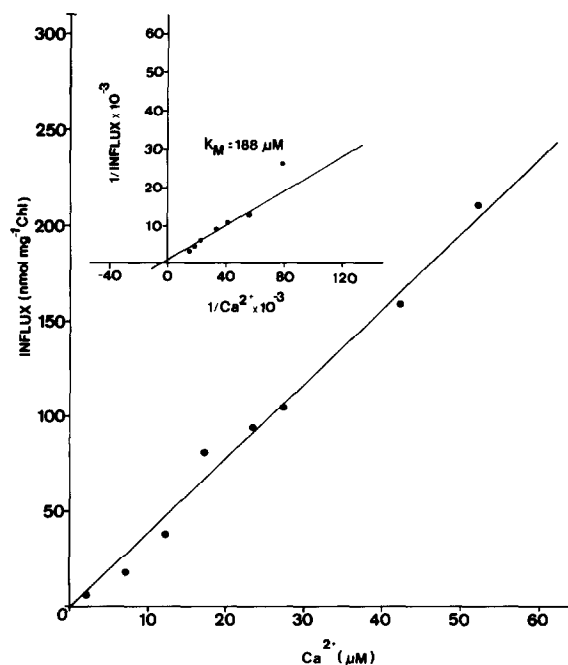
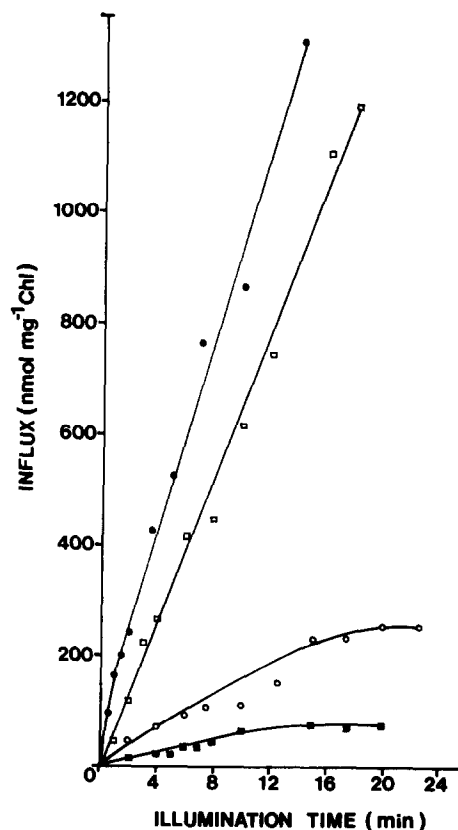


Fig. 1. (a) Time course of light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts at different external  $\text{Ca}^{2+}$  concentrations. Assay conditions as in section 2.3. 50  $\mu\text{M}$  free  $\text{Ca}^{2+}$  (●), 20  $\mu\text{M}$  free  $\text{Ca}^{2+}$  (□), 10  $\mu\text{M}$  free  $\text{Ca}^{2+}$  (○), 5  $\mu\text{M}$  free  $\text{Ca}^{2+}$  (■). (b) Effect of different  $\text{Ca}^{2+}$  concentrations on light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts. Assay conditions as in section 2.3.

### 3. RESULTS

Under our assay conditions a light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts is observed (fig. 1a). The rate and extent of  $\text{Ca}^{2+}$  influx is dependent on the concentration of free  $\text{Ca}^{2+}$  in the assay medium and the duration of illumination. The net influx is saturated after 10 min in the presence of  $5 \mu\text{M}$  free  $\text{Ca}^{2+}$ , and after 15 min in the presence of  $10 \mu\text{M}$  free  $\text{Ca}^{2+}$  in the assay medium. At these two  $\text{Ca}^{2+}$  concentrations the maximum amount of  $\text{Ca}^{2+}$  uptake by the chloroplasts is 80 and  $260 \text{ nmol} \cdot \text{mg}^{-1} \text{ Chl}$ , respectively, which corresponds to 21 and 43% of the total available  $\text{Ca}^{2+}$  in the medium. The observed  $\text{Ca}^{2+}$  influx is not due to chloroplast damage during the assay: in an average experiment chloroplasts were 98% intact before the assay and 96% intact after the assay. Light-dependent  $\text{Ca}^{2+}$  influx is completely abolished by treatment with heat (5 min,  $100^\circ\text{C}$ ) or Triton X-100 (0.25 vol.%). In addition, no  $\text{Ca}^{2+}$  influx is detected after chemical fixation of chloroplasts with 5% glutaraldehyde. Photosynthetic activity of the chloroplasts was measured by PGA-dependent oxygen evolution. The observed rates are identical to that measured using the assay medium of [17].  $\text{Ca}^{2+}$  influx is linear between 0 and  $60 \mu\text{M}$  free  $\text{Ca}^{2+}$  (fig. 1b). The apparent  $k_M$  is  $188 \mu\text{M}$  (fig. 1b). The average rate of  $\text{Ca}^{2+}$  influx under the conditions described in section 2.3. is  $6.6 \mu\text{mol} \cdot \text{mg}^{-1} \cdot \text{h}^{-1}$  ( $n=32$  different chloroplast preparations).

The light-dependent  $\text{Ca}^{2+}$  influx is observed at relatively low light intensities. Half-maximum influx is achieved at  $40\text{--}50 \text{ W} \cdot \text{m}^{-2}$ . DCMU completely inhibits  $\text{Ca}^{2+}$  influx at concentrations  $> 1 \mu\text{M}$ . Spinach calmodulin ( $4.8 \mu\text{g} \cdot \text{ml}^{-1}$ ) and vanadate had no effect on the light-dependent  $\text{Ca}^{2+}$  influx (detailed results not shown). Low concentrations of FCCP and CCCP, which are sufficient to uncouple electron transport, stimulate  $\text{Ca}^{2+}$  influx, whereas higher concentrations, known to inhibit electron transport [21], suppress  $\text{Ca}^{2+}$  influx (fig. 2). The uncoupling  $\text{K}^+/\text{H}^+$  exchanger nigericin also enhances  $\text{Ca}^{2+}$  influx (table 1). Additional KCl (15–30 mM) given 3 min before illumination stimulates  $\text{Ca}^{2+}$  influx (table 1). At standard concentrations of KCl (20 mM) in the assay medium  $\text{Ca}^{2+}$  influx increase following the addition of valinomycin (table 1). When an addi-

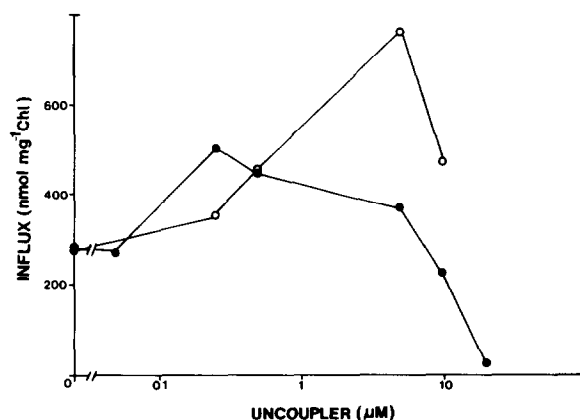


Fig. 2. Effect of uncouplers on light-dependent  $\text{Ca}^{2+}$  influx into spinach chloroplasts. Assay conditions as in section 2.3. FCCP (●), CCCP (○).

tional 30 mM KCl are added, valinomycin suppresses  $\text{Ca}^{2+}$  influx and in several experiments even caused  $\text{Ca}^{2+}$  efflux (table 1). Nigericin abolishes the  $\text{Ca}^{2+}$  efflux induced by valinomycin in the presence of high external KCl concentrations (table 1).

Ruthenium red inhibits light-dependent  $\text{Ca}^{2+}$  influx into spinach chloroplasts (fig. 3). At concentrations of ruthenium red ( $\leq 10 \mu\text{M}$ ) which significantly inhibit  $\text{Ca}^{2+}$  influx, PGA-dependent

Table 1

Effect of valinomycin, nigericin, and combinations of both in relation to external KCl concentrations on light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts

KCl, added to the medium (mM)	Valinomycin ( $\mu\text{M}$ )	Nigericin ( $\mu\text{M}$ )	$\text{Ca}^{2+}$ influx ( $\text{nmol} \cdot \text{mg}^{-1} \text{ Chl}$ )
0	0	0	48.8
15	0	0	213.3
30	0	0	246.8
0	0	0.5	195
0	0.5	0	106.7
0	0.5	0.5	280.4
30	0	0.5	383.9
30	0.5	0	-97.5
30	0.5	0.5	414.4

Assay conditions as in section 2.3.

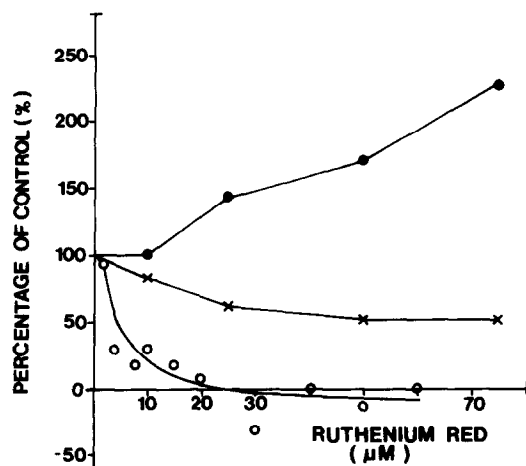


Fig. 3. Inhibition of light-dependent  $\text{Ca}^{2+}$  influx into spinach chloroplasts by ruthenium red. Assay conditions as in section 2.3.  $\text{Ca}^{2+}$  influx/efflux (○), PGA-dependent oxygen evolution (×), ferricyanide-dependent oxygen evolution by broken chloroplasts (●).

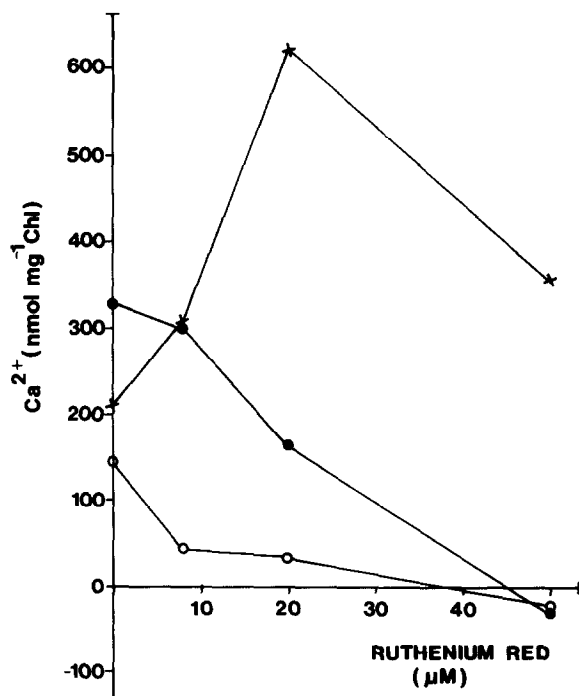


Fig. 4. Effect of ruthenium red on normal, FCCP- and ionophore A 23187-enhanced  $\text{Ca}^{2+}$  influx into spinach chloroplasts. Assay conditions as in section 2.3. Control (○),  $0.5 \mu\text{M}$  FCCP (●),  $10 \mu\text{M}$  A 23187 (×).

oxygen evolution is only slightly affected (fig. 3). To exclude the possibility that the effect of ruthenium red is due to unspecific charge-interactions at the envelope, we tested the ability of ruthenium red to inhibit the enhanced  $\text{Ca}^{2+}$  influx mediated by uncouplers and the  $\text{Ca}^{2+}$ -ionophore A 23187. FCCP-enhanced  $\text{Ca}^{2+}$  influx is inhibited by ruthenium red, while A 23187-mediated  $\text{Ca}^{2+}$  influx is not inhibited by it (fig. 4).

## 4. DISCUSSION

### 4.1. Transport mechanism

The observations made in this study indicate that ATP is not directly involved in light-dependent  $\text{Ca}^{2+}$  influx into intact spinach chloroplasts because: (i)  $\text{Ca}^{2+}$  influx is insensitive to vanadate, a potent inhibitor of ion transport ATPases [22], (ii)  $\text{Ca}^{2+}$  influx is enhanced by uncouplers, and (iii) calmodulin, which stimulates the plasma membrane  $\text{Ca}^{2+}$ -ATPase (23), has no effect on  $\text{Ca}^{2+}$  influx into spinach chloroplasts.

Ruthenium red is known as an inhibitor of electrogenic  $\text{Ca}^{2+}$  influx mediated by uniport-type carriers [24–27]. We can exclude unspecific effects of ruthenium red, since at inhibitory ruthenium red concentrations electron transport is only slightly affected. In addition, unspecific charge interactions of ruthenium red at the chloroplast envelope can be dismissed, since A 23187-mediated  $\text{Ca}^{2+}$  influx could not be inhibited by ruthenium red [26].

Two criteria have been used to distinguish between  $\text{Ca}^{2+}$  influx mediated by a uniport-type carrier and that mediated by a  $\text{Ca}^{2+}/\text{H}^{+}$ -exchange carrier [27]: uniport-type influx is sensitive to ruthenium red and enhanced by uncouplers. These criteria for a uniport-type carrier are met by the light-dependent  $\text{Ca}^{2+}$  influx into spinach chloroplasts.

Protons might act as the counterions to  $\text{Ca}^{2+}$  influx [8] through the uniport, since they are extruded by the chloroplast in the light [17] and their amount would be sufficient to balance  $\text{Ca}^{2+}$  influx.

### 4.2. The driving force

The inhibition of light-dependent  $\text{Ca}^{2+}$  influx by DCMU indicates that  $\Delta\bar{\mu}_{\text{H}^{+}}$  (established by electron transport) most likely triggers  $\text{Ca}^{2+}$  influx. The  $\Delta\text{pH}$  component is not involved, since  $\text{Ca}^{2+}$  influx can be stimulated by different types of un-

couplers. Thus  $\Delta E$  is presumably the driving force, since the  $\Delta E$ -modulating  $K^+$ -ionophore valinomycin stimulates  $Ca^{2+}$  influx at low  $K^+$  levels ( $\Delta E$  is high), and inhibits  $Ca^{2+}$  influx at high  $K^+$  levels ( $\Delta E$  is low). The effect of nigericin is probably due to its uncoupling activity which promotes  $Ca^{2+}$  influx.

#### 4.3. Comparison with mitochondrial $Ca^{2+}$ influx systems

$Ca^{2+}$  influx into animal [25,27] and plant mitochondria [23,24] is mediated by an electrogenic uniport-carrier system similar to that described here for chloroplasts. The influx is sensitive to ruthenium red [24-27] and has been shown to be under the control of  $\Delta E$  [28,29].

#### 4.4. Rates and affinity

The measured  $Ca^{2+}$  influx rate of  $6.6 \mu\text{mol} \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$  for spinach chloroplasts is in agreement with rates measured for chloroplasts in other studies [8,30]. On a protein basis the chloroplastic  $Ca^{2+}$  influx rates are similar to those of plant [31] and rat liver mitochondria [27]. From a comparison of  $k_m$  values it has been concluded that the affinity of the mitochondrial system is at least 10-fold lower than the affinity of the plasma membrane system [23]. However, a comparison of the apparent first-order rate constants ( $V_{\max}/K_m$ ), which reflect the affinity of an enzyme to its substrate at very low substrate concentrations [32], shows that all three  $Ca^{2+}$ -transport systems (plasma membrane, chloroplasts, mitochondria) have similar affinities towards  $Ca^{2+}$  at very low  $Ca^{2+}$ -levels.

#### 4.5. Physiological implications

We can calculate from the maximum amount of  $Ca^{2+}$  influx, assuming a chloroplast volume of  $26 \mu\text{l}$  [33], that the calcium concentration inside the chloroplast increases by 3-9 mM when 5-10  $\mu\text{M}$  free  $Ca^{2+}$  is present in the assay medium. The actual concentration of free  $Ca^{2+}$  inside the chloroplast would, however, be much less. Targets for chloroplastic  $Ca^{2+}$  include light-activated,  $Ca^{2+}$ -regulated enzymes (discussion in [3,4,34,35]). Other possible sites of action include various photochemical reactions [5-7]. We suggest that the  $Ca^{2+}$ -carrier system of the chloroplast-envelope may not only be important for regulation of

$Ca^{2+}$ -dependent processes inside the chloroplast, but it could also affect the  $Ca^{2+}$ -level in the cytosol and therefore exert some influence on other  $Ca^{2+}$ -regulated cellular functions in plants.

#### ACKNOWLEDGEMENTS

The authors are indebted to Dr Joe Holtum for useful discussions. This work was supported by grants from the Deutsche Forschungsgemeinschaft.

#### REFERENCES

- [1] Roux, S.J. and Slocum, R.D. (1982) in: Calcium and Cell Function (Cheung, W.Y. ed.) vol. 3, pp. 409-453, Academic Press, New York.
- [2] Marmé, D. (1983) in: Inorganic Plant Nutrition (Läuchli, A. and Bielecki, R.L. eds) *Encycl. Plant Physiol. New Series*, Vol. 15B, pp. 599-625, Springer Verlag, Berlin.
- [3] Hertig, C. and Wolosiuk, R.A. (1980) *Biochem. Biophys. Res. Commun.* 97, 325-333.
- [4] Jarrett, H.W., Brown, C.J., Black, C.C. and Cormier, M.J. (1982) *J. Biol. Chem.* 257, 13795-13804.
- [5] Barr, R., Troxel, K.S. and Crane, F.L. (1983) *Plant Physiol.* 73, 309-315.
- [6] Ono, T. and Inoue, Y. (1984) *FEBS Lett.* 168, 281-286.
- [7] Gross, E.L. and Hess, S.C. (1974) *Biochim. Biophys. Acta* 339, 334-346.
- [8] Muto, S., Izawa, S. and Miyachi, S. (1982) *FEBS Lett.* 139, 250-254.
- [9] Gimmler, H., Neumann, J. and Steppan, M. (1976) VIIth. Internat. Congr. Photobiol. (abstr.), p. 64, Rome.
- [10] Lilley, R. McC. and Walker, D.A. (1974) *Biochim. Biophys. Acta* 368, 269-278.
- [11] Walker, D.A. (1980) *Methods Enzymol.* 69, 94-104.
- [12] Aebi, H. (1974) in: *Methoden der enzymatischen Analyse* (Bergmeyer, H.U. ed.) vol. 1, 3rd edn, pp. 713-724, Verlag Chemie, Weinheim.
- [13] Schnarrenberger, C., Oeser, A. and Tolbert, N.E. (1971) *Plant Physiol.* 48, 566-574.
- [14] Evans, H.J., and Nason, A. (1953) *Plant Physiol.* 28, 233-254.
- [15] Huang A.H.C. (1979) *Plant Physiol.* 55, 870-874.
- [16] Lilley, R. McC., Fitzgerald, M.P., Rienits, K.G. and Walker, D.A. (1975) *New Phytol.* 75, 1-10.

- [17] Enser, U. and Heber, U. (1980) *Biochim. Biophys. Acta* 592, 577-591.
- [18] Arnon, D.I. (1949) *Plant Physiol.* 24, 1-15.
- [19] Scarpa, A. (1979) *Methods Enzymol.* 56, 301-338.
- [20] Thomas, M.V. (1982) in: *Techniques in Calcium Research* (Treherne, J.E. and Rubery, P.H. eds.) *Biological Techniques Series*, pp. 90-138, Academic Press, London.
- [21] Izawa, S. and Good, N.E. (1972) *Methods Enzymol.* 24, 355-377.
- [22] Bürkler, J. and Solioz, M. (1982) *Ann. NY Acad. Sci.* 402, 422-432.
- [23] Dieter, P. and Marmé, D. (1983) *Planta* 159, 277-281.
- [24] Chen, C.-H. and Lehninger, A.L. (1973) *Arch. Biochem. Biophys.* 157, 183-196.
- [25] Reed, K.C. and Bygrave, F.L. (1974) *Biochem. J.* 140, 143-155.
- [26] Niggli, V., Gazzotti, P. and Carafoli, E. (1978) *Experientia* 34, 1136-1137.
- [27] Bernadi, P. and Azzone, G.F. (1979) *Eur. J. Biochem.* 102, 555-562.
- [28] Bernadi, P. and Azzone, G.F. (1983) *Eur. J. Biochem.* 134, 377-383.
- [29] Bernadi, P. and Azzone, G.F. (1982) *FEBS Lett.* 139, 13-16.
- [30] Demmig, B. and Gimmmler, H. (1979) *Z. Naturforsch.* 34c, 233-241.
- [31] Dieter, P. and Marmé, D. (1980) *Planta* 150, 1-8.
- [32] Cleland, W.W. (1970) in: *The Enzymes* (Boyer, P.D. ed.) vol. II, 3rd ed., pp. 1-65, Academic Press, New York.
- [33] Heldt, H.W., Werdan, K., Milovancev, M. and Geller, G. (1973) *Biochim. Biophys. Acta* 314, 224-241.
- [34] Rosa, L. and Whatley, F.R. (1984) *Plant Physiol.* 75, 131-137.
- [35] Wolosiuk, R.A., Hertig, C.M., Nishizawa, A.N. and Buchanan, B.B. (1982) *FEBS Lett.* 140, 31-35.