

FeCl₃-Catalyzed Synthesis of 2-Methyl-4-Substituted-1,2,3,4-Tetrahydroquinoline Derivatives by the Imino Diels-Alder Reaction

Ye Chen and Yong Rok Lee*

School of Chemical Engineering and Technology, Yeungnam University, Gyeongsan 712-749, Korea. *E-mail: yrlee@yu.ac.kr
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The tetrahydroquinoline nucleus is present in many natural products and these products exhibit a broad range of biological activities.¹ The imino Diels-Alder reaction between *N*-arylimines and dienophiles is a powerful synthetic tool for the construction of nitrogen-containing heterocyclic compounds including tetrahydroquinoline derivatives.² Using an imino Diels-Alder reaction as a key step, a number of methods for the synthesis of tricyclic compounds bearing the tetrahydroquinoline moiety, starting from substituted anilines with 2,3-dihydrofuran or 3,4-dihydro-2*H*-pyran or *N*-protected 2-pyrroline, have been reported (Scheme 1). These reactions have been extensively studied with the use of a variety of Lewis acids such as LiClO₄,³ BF₃·OEt₂,⁴ CuCl₂,⁵ SmI₂,⁶ InCl₃,⁷ SbCl₃,⁸ GbCl₃,⁹ ZrCl₄,¹⁰ and Brønsted acids such as TFA,¹¹ *p*-TsOH,¹² KHSO₄,¹³ and oxalic acid.¹⁴

Recently, several methods for the synthesis of 2-methyl-4-substituted-1,2,3,4-tetrahydroquinoline derivatives through imino Diels-Alder reactions have been developed (Scheme 2). These methods involve coupling reactions of substituted anilines with vinyl ether (Path A)¹⁵ or *N*-vinyl lactams (Path B).^{16,17} These reactions have been accomplished using catalysts such as cerium ammonium nitrate,¹⁵ copper dipyrindine dichloride,¹⁶ and 4-nitro phthalic acid.¹⁷

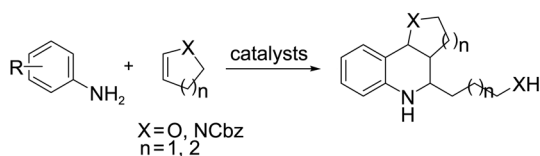
Although several methods for the synthesis of 2-methyl-4-substituted-1,2,3,4-tetrahydroquinoline derivatives have been reported,¹⁵⁻¹⁷ there is still demand for simpler, less toxic, more cost effective, and more environmentally benign catalysts. In a related work, an FeCl₃/NaI mediated reaction of aryl azides with 3,4-dihydro-2*H*-pyran for the synthesis of tricyclic compounds involving the tetrahydroquinoline

moiety has been described as a stoichiometric reaction.¹⁸ Our interest in developing a mild and efficient synthetic method that provides a variety of 2-methyl-4-substituted-1,2,3,4-tetrahydroquinoline derivatives has led us to looking into more convenient and safely usable catalysts. Among these, we think iron(III) chloride is a viable alternative, and may be a promising catalyst for the synthesis of 2-methyl-4-substituted-1,2,3,4-tetrahydroquinoline derivatives due to its easy availability, low price, sustainability, non-toxicity, and environmentally friendly properties.¹⁹ We report herein an FeCl₃-catalyzed one-pot synthesis of biologically interesting 2-methyl-4-substituted-1,2,3,4-tetrahydroquinoline derivatives.

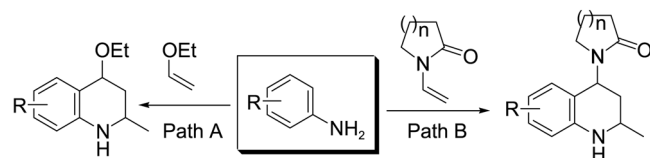
Results and Discussion

The reaction of aniline (**1**, 1.0 mmol) with *N*-vinylpyrrolidinone (**2**, 3.0 mmol) was first examined in the presence of 20 mol % FeCl₃ in several solvents (Table 1). Across the range of solvents tested, the best yield was obtained in acetonitrile, with 74% yield. Interestingly, in this case, only *cis*-adduct was produced. The *cis*-stereochemistry of compound **3** was determined by ¹H NMR analysis and compared directly with the reported data.¹⁵⁻¹⁷

In order to extend the utility of this methodology, further reactions of a series of substituted anilines with *N*-vinylpyrrolidinone and *N*-vinylcaprolactam were examined. These reactions were carried out in the presence of 20 mol % of FeCl₃ in acetonitrile using the conditions described above. The results are summarized in Table 2. The reactions worked



Scheme 1



Scheme 2

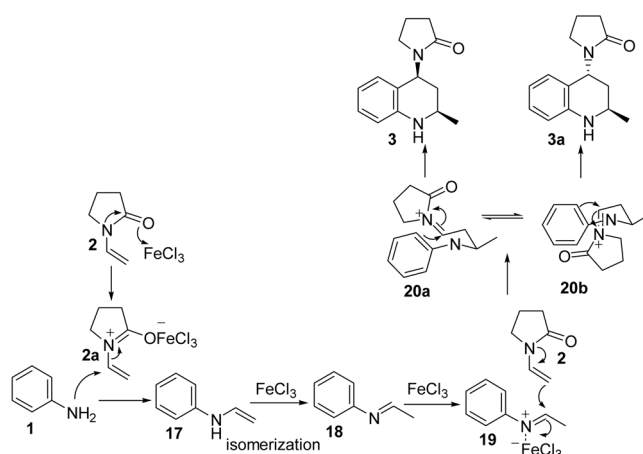
Table 1. Reaction of aniline (**1**) with *N*-vinylpyrrolidinone (**2**) in the presence of FeCl₃ in several solvents

Entry	FeCl ₃ (mol %)	Solvent	Condition	Yield (%)
1	20	toluene	reflux, 20 h	40
2	20	THF	reflux, 8 h	68
3	20	acetonitrile	reflux, 8 h	74
4	20	DMF	100 °C, 10 h	34
5	20	DMSO	100 °C, 10 h	39

Table 2. FeCl₃-catalyzed synthesis of a variety of 2-methyl-4-substituted tetrahydroquinoline derivatives

Entry	Starting material	<i>N</i> -vinyl Compound	Time (h)	Product	Yield (%)
1			10		71
2			10		66
3			10		72
4			8		90
5			10		72
6			8		88
7			8		84
8			10		80
9			8		85
10			8		85
11			8		88
12			10		80

well with substituted anilines bearing either electron-donating or -withdrawing groups on the benzene ring. For example, the treatment of *o*-substituted anilines with *N*-vinylpyrrolidinone provided products **4–7** in 71, 66, 72, and 90% yield, respectively (entries 1–4). Importantly, reactions of *m*-substituted anilines with *N*-vinylpyrrolidinone have not been reported so far. With 3-chloroaniline, inseparable two regioisomers **8** and **9** as a 5:95 ratio were produced in 72% yield (entry 5), whereas reaction with 3-nitroaniline gave **10** in 88% yield, without any detection of other regioisomer (entry 6). Interestingly, **10** was cyclized at the hindered site between the amino and nitro groups. Their structural assignment and the ratio of regioisomers of **8** and **9** were based on ¹H NMR spectroscopic data. The ¹H NMR spectrum of **8** showed 3 peaks on the benzene ring at δ 6.91 (d, *J* = 7.8 Hz), 6.65 (t, *J* = 7.8 Hz), and 6.41 (d, *J* = 7.8 Hz), whereas **9** exhibited 3 peaks at δ 6.69 (d, *J* = 8.1 Hz), 6.57 (dd, *J* = 8.1,

**Scheme 3**

1.8 Hz), and 6.44 (d, 1.8 Hz). *p*-Substituted aniline reacted smoothly with *N*-vinylpyrrolidinone to give the corresponding tetrahydroquinoline **11** in 84% yield, respectively (entry 7). With other *N*-vinylcaprolactam, the desired products **12–16** were produced in 80–88% yield (entries 8–12). These reactions provided rapid access to various 2-methyl-4-substituted tetrahydroquinoline derivatives **4–16** in good yields.

The formation of **3** can be explained by the proposed mechanism as shown in Scheme 3. The aniline (**1**) is first reacted with *N*-vinylpyrrolidinone in the presence of FeCl₃ to give *N*-vinylaniline (**17**), which undergoes isomerization to afford enamine **18**. Next, the FeCl₃-catalyzed imino Diels-Alder reaction of **19** to *N*-vinylpyrrolidinone (**2**) gives product **3**. The imino Diels-Alder reaction is known to proceed *via* a stepwise mechanism,²⁰ so the generation of two intermediates, **20a** and **20b**, with the pseudoequatorial conformation adopted by the methyl group in the chair-like transition state is expected. In the final cyclization step, *cis*-stereochemistry may be observed due to there being a greater preference for an equatorial arrangement of the iminium ion **20a** than for intermediate **20b**.

In summary, efficient one-pot multi-component reactions of substituted anilines with *N*-vinylpyrrolidinone or *N*-vinylcaprolactam were developed in the presence of 20 mol % of FeCl₃ in acetonitrile. These reactions provided rapid access to a variety of 2-methyl-4-substituted tetrahydroquinoline derivatives in good yields.

Experimental

General Procedure for the Synthesis of 2-Methyl-4-amino-1,2,3,4-tetrahydroquinolines. A mixture of the substituted anilines (1.0 mmol), *N*-vinyl-2-pyrrolidinone (3.0 mmol) or *N*-vinylcaprolactam (3.0 mmol) and FeCl₃ (32 mg, 0.2 mmol) in acetonitrile (10 mL) was refluxed, and the reaction progress was monitored by TLC. When the reaction was completed, the reaction was quenched with saturated aqueous NaHCO₃ solution (30 mL) and extracted with ethyl acetate (3 × 30 mL). The combined organic layer was wash-

ed with water (30 mL), dried over anhydrous Na_2SO_4 , and concentrated. The crude material was purified by column chromatography to give products.

cis-1-(2-Methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (3): Yield (74%). mp 74–76 °C; ^1H NMR (600 MHz, CDCl_3) δ 6.96 (1H, t, $J = 7.8$ Hz), 6.77 (1H, d, $J = 7.8$ Hz), 6.60 (1H, d, $J = 7.8$ Hz), 6.47 (1H, d, $J = 7.8$ Hz), 5.51 (1H, dd, $J = 12.0, 6.0$ Hz), 3.70 (1H, br s), 3.54–3.49 (1H, m), 3.21–3.16 (1H, m), 3.12–3.08 (1H, m), 2.50–2.40 (2H, m), 2.00–1.92 (2H, m), 1.91–1.88 (1H, m), 1.70–1.64 (1H, m), 1.17 (3H, d, $J = 6.0$ Hz); ^{13}C NMR (150 MHz, CDCl_3) δ 175.7, 145.9, 128.0, 126.7, 118.8, 117.7, 114.6, 48.0, 46.9, 42.2, 34.1, 31.4, 22.3, 18.2; IR (KBr) 3332, 2953, 1673, 1491, 1285, 1031, 756 cm^{-1} .

cis-1-(2,8-Dimethyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (4): Yield (71%). mp 129–131 °C; ^1H NMR (300 MHz, acetone- d_6) δ 6.71 (1H, d, $J = 7.5$ Hz), 6.51 (1H, d, $J = 7.5$ Hz), 6.36 (1H, t, $J = 7.5$ Hz), 5.31 (1H, dd, $J = 12.0, 6.0$ Hz), 4.13 (1H, br s), 3.48–3.39 (1H, m), 3.14–3.06 (1H, m), 2.95–2.88 (1H, m), 2.27–2.19 (2H, m), 1.94 (3H, s), 1.90–1.82 (2H, m), 1.74–1.66 (1H, m), 1.63–1.51 (1H, m), 1.13 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, acetone- d_6) δ 175.9, 146.0, 130.4, 125.6, 123.1, 120.2, 117.9, 49.6, 48.6, 43.2, 35.6, 32.3, 23.2, 19.6, 18.2; IR (KBr) 3346, 2923, 1676, 1475, 1429, 1300, 1268, 1197, 1164, 1102, 763 cm^{-1} .

cis-1-(8-Methoxy-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl)pyrrolidin-2-one (5): Yield (66%). mp 78–79 °C; ^1H NMR (300 MHz, CDCl_3) δ 6.62–6.57 (2H, m), 6.47 (1H, d, $J = 7.5$ Hz), 5.57 (1H, dd, $J = 12.0, 6.3$ Hz), 4.16 (1H, br s), 3.81 (3H, s), 3.56–3.47 (1H, m), 3.27–3.03 (2H, m), 2.49–2.43 (2H, m), 2.00–1.92 (2H, m), 1.92–1.81 (1H, m), 1.77–1.65 (1H, m), 1.24 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 175.8, 146.4, 136.0, 119.3, 118.9, 116.9, 108.4, 55.5, 48.2, 46.8, 42.5, 34.1, 31.6, 22.4, 18.3; IR (KBr) 3327, 2973, 2919, 1678, 1609, 1584, 1503, 1460, 1430, 1376, 1335, 1291, 1272, 1252, 1235, 1166, 1083, 1063, 968, 844, 767, 787, 767, 743 cm^{-1} .

cis-1-(8-Fluoro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (6): Yield (72%). mp 96–97 °C; ^1H NMR (300 MHz, acetone- d_6) δ 6.76–6.70 (1H, m), 6.50 (1H, d, $J = 7.5$ Hz), 6.45–6.38 (1H, m), 5.34 (1H, dd, $J = 12.0, 6.0$ Hz), 4.75 (1H, br s), 3.55–3.49 (1H, m), 3.23–3.15 (1H, m), 3.00–2.95 (1H, m), 2.29–2.23 (2H, m), 1.97–1.86 (2H, m), 1.80–1.73 (1H, m), 1.71–1.59 (1H, m), 1.18 (3H, d, $J = 6.0$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 174.6, 150.7 (d, $J = 236$ Hz), 134.9 (d, $J = 12.8$ Hz), 121.7 (d, $J = 3.0$ Hz), 115.6 (d, $J = 7.5$ Hz), 113.2, 112.9, 47.6, 46.8, 41.7, 33.8, 30.7, 21.4, 18.1; IR (KBr) 3319, 2969, 1674, 1624, 1503, 1457, 1316, 1274, 1229, 1164, 970, 783, 720 cm^{-1} .

cis-1-(8-Bromo-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (7): Yield (90%). mp 127–128 °C; ^1H NMR (300 MHz, acetone- d_6) δ 7.12 (1H, d, $J = 7.8$ Hz), 6.65 (1H, d, $J = 7.8$ Hz), 6.36 (1H, t, $J = 7.8$ Hz), 5.30 (1H, dd, $J = 12.0, 6.0$ Hz), 4.50 (1H, br s), 3.57–3.51 (1H, m), 3.20–3.11 (1H, m), 2.97–2.90 (1H, m), 2.26–2.16 (2H, m), 1.93–1.83 (2H, m), 1.78–1.71 (1H, m), 1.67–1.55 (1H, m),

1.18 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, acetone- d_6) δ 176.1, 144.7, 132.7, 127.3, 122.8, 118.9, 109.7, 49.5, 48.5, 43.4, 35.1, 32.2, 23.0, 19.6; IR (KBr) 3329, 2963, 1678, 1598, 1499, 1459, 1376, 1339, 1294, 1165, 1135, 972, 795, 762, 733 cm^{-1} .

cis-1-(5-Chloro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (8) and cis-1-(7-Chloro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (9): Yield of 8 and 9 (5:95 ratio, 72%). **Compound 8** (minor): ^1H NMR (300 MHz, CDCl_3) δ 6.91 (1H, d, $J = 7.8$ Hz), 6.65 (1H, d, $J = 7.8$ Hz), 6.41 (1H, t, $J = 7.8$ Hz), 5.26–5.22 (1H, m), 3.78 (1H, br s), 3.35–3.28 (1H, m), 3.12–3.06 (1H, m), 2.96–2.79 (1H, m), 2.41–2.35 (2H, m), 2.25–2.15 (1H, m), 2.04–1.87 (3H, m), 1.18 (3H, d, $J = 6.3$ Hz). **Compound 9** (major): δ 6.69 (1H, d, $J = 8.1$ Hz), 6.57 (1H, dd, $J = 8.1, 1.8$ Hz), 6.44 (1H, d, $J = 1.8$ Hz), 5.46 (1H, dd, $J = 12.0, 6.0$ Hz), 3.78 (1H, br s), 3.60–3.49 (1H, m), 3.25–3.14 (1H, m), 3.11–3.06 (1H, m), 2.50–2.44 (2H, m), 2.05–1.96 (2H, m), 1.92–1.87 (1H, m), 1.72–1.56 (1H, m), 1.19 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 175.9, 162.5, 147.0, 133.6, 128.1, 117.7, 114.2, 47.8, 47.1, 42.3, 33.9, 31.6, 22.4, 18.4; IR (KBr) 3333, 3269, 2960, 1663, 1602, 1492, 1423, 1338, 1311, 1288, 1267, 1198, 1166, 1126, 1088, 919, 908, 849 cm^{-1} .

cis-1-(2-Methyl-5-nitro-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (10): Yield (88%). mp 228–229 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.06 (1H, dd, $J = 8.1, 7.8$ Hz), 6.84 (1H, d, $J = 7.8$ Hz), 6.67 (1H, d, $J = 8.1$ Hz), 5.69 (1H, dd, $J = 11.1, 7.2$ Hz), 4.16 (1H, br s), 3.50–3.43 (1H, m), 3.14–3.06 (1H, m), 2.92–2.85 (1H, m), 2.39–2.14 (2H, m), 2.07–2.00 (1H, m), 1.92–1.81 (2H, m), 1.64–1.52 (1H, m), 1.21 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 174.8, 162.3, 147.8, 128.4, 118.1, 112.7, 111.6, 46.0, 45.7, 42.7, 33.3, 30.7, 21.5, 17.9; IR (KBr) 3267, 2961, 1667, 1609, 1527, 1449, 1330, 1287, 1197, 1076, 978, 779, 726 cm^{-1} .

cis-1-(6-Chloro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) pyrrolidin-2-one (11): Yield (84%). mp 142–143 °C; ^1H NMR (300 MHz, CDCl_3) δ 6.88 (1H, d, $J = 8.5$ Hz), 6.70 (1H, s), 6.38 (1H, d, $J = 8.5$ Hz), 5.43 (1H, dd, $J = 12.0, 6.0$ Hz), 3.88 (1H, br s), 3.50–3.44 (1H, m), 3.22–3.04 (2H, m), 2.54–2.38 (2H, m), 2.03–1.93 (2H, m), 1.88–1.82 (1H, m), 1.69–1.57 (1H, m), 1.15 (3H, d, $J = 6.0$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 175.9, 144.6, 128.1, 126.3, 122.3, 120.5, 115.9, 47.9, 47.0, 42.3, 33.7, 31.4, 22.2, 18.3; IR (KBr) 3334, 1666, 1602, 1488, 1437, 1350, 1312, 1293, 1200, 882, 821 cm^{-1} .

cis-1-(2-Methyl-1,2,3,4-tetrahydroquinoline-4-yl) azepan-2-one (12): Yield (80%). mp 128–130 °C; ^1H NMR (300 MHz, CDCl_3) δ 6.96 (1H, dd, $J = 7.8, 7.5$ Hz), 6.84 (1H, d, $J = 7.8$ Hz), 6.61 (1H, dd, $J = 7.8, 7.5$ Hz), 6.48 (1H, d, $J = 7.5$ Hz), 6.00 (1H, dd, $J = 12.0, 6.3$ Hz), 3.69 (1H, br s), 3.55–3.44 (1H, m), 3.13–2.94 (2H, m), 2.63–2.60 (2H, m), 2.01–1.94 (1H, m), 1.82–1.66 (3H, m), 1.64–1.52 (3H, m), 1.42–1.35 (1H, m), 1.19 (3H, d, $J = 6.3$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 176.9, 146.7, 128.0, 127.5, 120.3, 118.0, 114.8, 51.3, 47.2, 44.9, 37.7, 35.0, 30.2, 29.7, 23.8, 22.5; IR (KBr): 3326, 2929, 1628, 1490, 1443, 1340, 1196, 973, 912,

845, 735 cm⁻¹.

cis-1-(2,8-Dimethyl-1,2,3,4-tetrahydroquinoline-4-yl) azepan-2-one (13): Yield (85%). mp 158-160 °C; ¹H NMR (300 MHz, CDCl₃) δ 6.84 (1H, d, *J* = 7.2 Hz), 6.72 (1H, d, *J* = 7.8 Hz), 6.53 (1H, dd, *J* = 7.8, 7.2 Hz), 6.00 (1H, dd, *J* = 11.7, 6.0 Hz), 3.52 (1H, br s), 3.49-3.45 (1H, m), 3.01-2.90 (2H, m), 2.60-2.57 (2H, m), 2.04 (3H, s), 1.96-1.93 (1H, m), 1.79-1.52 (6H, m), 1.39-1.32 (1H, m), 1.21 (3H, d, *J* = 6.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 176.5, 144.5, 128.8, 125.1, 121.4, 119.4, 117.0, 51.3, 47.0, 44.6, 37.5, 34.6, 30.0, 29.5, 23.5, 22.4, 17.3; IR (KBr) 3355, 2927, 1630, 1474, 1445, 1337, 1293, 1189, 1083, 972, 845, 751 cm⁻¹.

cis-1-(8-Bromo-2-methyl-2,3,4-tetrahydroquinoline-4-yl) azepan-2-one (14): Yield (85%). mp 159-160 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.20 (1H, d, *J* = 7.8 Hz), 6.77 (1H, d, *J* = 7.8 Hz), 6.53 (1H, t, *J* = 7.8 Hz), 6.00 (1H, dd, *J* = 12.0, 6.0 Hz), 4.30 (1H, br s), 3.55-3.50 (1H, m), 3.12-2.90 (2H, m), 2.65-2.52 (2H, m), 1.94-1.91 (1H, m), 1.81-1.53 (6H, m), 1.40-1.32 (1H, m), 1.24 (3H, d, *J* = 6.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 176.7, 143.4, 131.2, 126.4, 121.7, 117.9, 109.0, 51.3, 47.1, 44.8, 37.5, 34.4, 30.1, 29.6, 23.6, 22.2; IR (KBr) 3401, 2929, 1634, 1490, 1454, 1339, 1293, 1195, 1144, 1085, 972, 734 cm⁻¹.

cis-1-(6-Fluoro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) azepan-2-one (15): Yield (88%). mp 96-98 °C; ¹H NMR δ 6.70-6.64 (1H, m), 6.57-6.53 (1H, m), 6.43-6.40 (1H, m), 5.95 (1H, dd, *J* = 11.7, 6.0 Hz), 3.68 (1H, br s), 3.47-3.41 (1H, m), 3.14-3.06 (1H, m), 2.99-2.92 (1H, m), 2.62-2.58 (2H, m), 1.99-1.92 (1H, m), 1.81-1.70 (3H, m), 1.61-1.49 (3H, m), 1.41-1.34 (1H, m), 1.16 (3H, d, *J* = 6.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 176.7, 156.1 (d, *J* = 234 Hz), 142.9, 121.7, 115.7 (d, *J* = 7.5 Hz), 114.8 (d, *J* = 22.5 Hz), 113.5 (d, *J* = 22.5 Hz), 51.2, 47.4, 44.9, 37.6, 34.6, 30.1, 29.7, 23.7, 22.4; IR (KBr) 3331, 2932, 1632, 1498, 1445, 1343, 1259, 1220, 1191, 1147, 1088, 974, 911, 813, 739 cm⁻¹.

cis-1-(6-Chloro-2-methyl-1,2,3,4-tetrahydroquinoline-4-yl) azepan-2-one (16): Yield (80%). mp 141-142 °C; ¹H NMR (300 MHz, CDCl₃) δ 6.91 (1H, dd, *J* = 8.7, 2.4 Hz), 6.79 (1H, d, *J* = 2.4 Hz), 6.41 (1H, d, *J* = 8.7 Hz), 5.96 (1H, dd, *J* = 12.0, 6.0 Hz), 3.77 (1H, br s), 3.55-3.46 (1H, m), 3.19-3.10 (1H, m), 3.00-2.94 (1H, m), 2.71-2.57 (2H, m), 2.02-1.94 (1H, m), 1.87-1.73 (3H, m), 1.63-1.51 (3H, m), 1.43-1.36 (1H, m), 1.18 (3H, d, *J* = 6.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 176.7, 144.9, 127.8, 126.8, 122.4, 121.7, 115.7, 50.9, 47.0, 44.7, 37.5, 34.4, 30.0, 29.6, 23.5, 22.2; IR (KBr) 3332, 2931, 1627, 1487, 1444, 1343, 1303, 1259, 1196, 1091, 973, 814, 735 cm⁻¹.

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References

- (a) Perry, N. B.; Blunt, J. W.; McCombs, J. D.; Munro, M. H. G. *J. Org. Chem.* **1986**, *51*, 5476. (b) Ramesh, M.; Moham, P. S.; Shanmugam, P. *Tetrahedron* **1984**, *40*, 4041. (c) Johnson, J. V.; Rauckman, B. S.; Baccanari, D. P.; Roth, B. *J. Med. Chem.* **1989**, *32*, 1942. (d) Williamson, N. M.; March, D. R.; Ward, A. D. *Tetrahedron Lett.* **1995**, *36*, 7721. (e) Carling, R. W.; Leeson, P. D.; Moseley, A. M.; Smith, J. D.; Saywell, K.; Tricklebank, M. D.; Kemp, J. A.; Marshall, G. R.; Foster, A. C.; Grimwood, S. *Bioorg. Med. Chem. Lett.* **1993**, *3*, 65.
- (a) Crousse, B.; Bégué, J.-P.; Bonnet-Delpon, D. *J. Org. Chem.* **2000**, *65*, 5009. (b) Akiyama, T.; Morita, H.; Fuchibe, K. *J. Am. Chem. Soc.* **2006**, *128*, 13070. (c) Cheng, D.; Zhou, J.; Saiah, E.; Beaton, G. *Org. Lett.* **2002**, *4*, 4411. (d) Shi, M.; Shao, L.-X.; Xu, B. *Org. Lett.* **2003**, *5*, 579. (e) Xue, Z.; Samanta, A.; Whittlesey, B. R.; Mayer, M. F. *Tetrahedron Lett.* **2009**, *50*, 6064. (f) Lavilla, R.; Bernabeu, M. C.; Carranco, I.; Diaz, J. L. *Org. Lett.* **2003**, *5*, 717. (g) Muhuhi, J.; Spaller, M. R. *J. Org. Chem.* **2006**, *71*, 5515. (h) Gaddam, V.; Nagarajan, R. *J. Org. Chem.* **2007**, *72*, 3573. (i) Desrat, S.; van de Weghe, P. *J. Org. Chem.* **2009**, *74*, 6728. (j) Gaddam, V.; Nagarajan, R. *Org. Lett.* **2008**, *10*, 1975.
- Yadav, J. S.; Reddy, B. V. S.; Srinivas, R.; Madhuri, C.; Ramalingam, T. *Synlett* **2001**, 240.
- Kametani, T.; Takeda, H.; Suzuki, Y.; Honda, T. *Synth. Commun.* **1985**, *15*, 499.
- Semwal, A.; Nayak, S. *Synth. Commun.* **2006**, *36*, 227.
- Zhou, F.; Xu, F.; Han, X.; Zhou, J.; Shen, Q. *Eur. J. Org. Chem.* **2007**, 5265.
- (a) Kametani, T.; Takeda, H.; Suzuki, Y.; Honda, T. *Synth. Commun.* **1985**, *15*, 499. (b) Babu, G.; Perumal, P. T. *Tetrahedron Lett.* **1997**, *38*, 5025. (c) Baudelle, R.; Melnyk, P.; Deprez, B.; Tatar, A. *Tetrahedron* **1998**, *54*, 4125. (d) Babu, G.; Nagarajan, R.; Natarajan, R.; Perumal, P. T. *Synthesis* **2000**, 661. (e) Yadav, J. S.; Reddy, B. V. S.; Reddy, J. S. S.; Rao, R. S. *Tetrahedron* **2003**, *59*, 1599.
- (a) Maiti, G.; Kundu, P. *Tetrahedron Lett.* **2006**, *47*, 5733. (b) Maiti, G.; Kundu, P. *Tetrahedron* **2006**, *62*, 5733.
- Ma, Y.; Qian, C.; Xie, M.; Sun, J. *J. Org. Chem.* **1999**, *64*, 6462.
- (a) Mahesh, M.; Venkateshwar Reddy, Ch.; Srinivasa Reddy, K.; Raju, P. V. K.; Narayana Reddy, V. V. *Synth. Commun.* **2004**, *34*, 4089. (b) Das, B.; Reddy, M. R.; Reddy, V. S.; Ramu, R. *Chem. Lett.* **2004**, *33*, 1526.
- Grieco, P. A.; Bahasas, A. *Tetrahedron Lett.* **1988**, *29*, 5855.
- Boger, D. L. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; 5, 451.
- Kumar, R.; Nagarajan, P. T.; Perumal, P. T. *Synthesis* **2004**, 949.
- Nagarajan, R.; Perumal, P. T. *Synth. Commun.* **2001**, *31*, 1733.
- Sridharan, V.; Avendaño, C.; Menéndez, J. C. *Tetrahedron* **2007**, *63*, 673.
- Kavitha, C. S.; Hosamani, K. M.; Harisha, R. S. *Can. J. Chem.* **2010**, *88*, 443.
- Srinivasa, A.; Mahadevan, K. M.; Hulikal, V. *Synth. Commun.* **2009**, *39*, 93.
- Kamal, A.; Prasad, B. R.; Ramana, A. V.; Babu, A. H.; Reddy, K. S. *Tetrahedron Lett.* **2004**, *45*, 3507.
- (a) Correa, A.; Mancheno, O. G.; Bolm, C. *Chem. Soc. Rev.* **2008**, *37*, 1108. (b) Sherry, B. D.; Fürstner, A. *Acc. Chem. Res.* **2008**, *41*, 1500. (c) Liang, Z.; Hou, W.; Du, Y.; Zhang, Y.; Pan, Y.; Mao, D.; Zhao, K. *Org. Lett.* **2009**, *11*, 4978. (d) Fan, J.; Gao, L.; Wang, Z. *Chem. Commun.* **2009**, 5021. (e) Bonnamour, J.; Bolm, C. *Org. Lett.* **2008**, *10*, 2665. (f) Li, H.; Yang, J.; Liu, Y.; Li, Y. *J. Org. Chem.* **2009**, *74*, 6797.
- Jimenez, O.; de la Rosa, G.; Lavilla, R. *Angew. Chem. Int. Ed.* **2005**, *44*, 6521.