

Regioselective Introduction of Homoallylic Amine Moiety to Quinolines: Preparation of Reissert Compound Followed by In-Mediated Allylation of Nitrile

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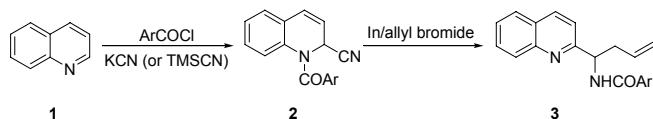
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Homoallylic amines are valuable building blocks in organic synthesis as precursors for β -amino acids and many heterocycles.¹⁻³ In addition, the structural motif is present in numerous biologically important substances.³

During our recent studies on the reaction of allylindium reagents and nitrile moiety-containing substrates⁴ we found that the imine intermediate, generated *in situ* by the reaction of a nitrile and allylindium reagents, could be quenched by a suitable electrophilic quencher in the same molecule to form a cyclic compound.⁴ Inter- and intramolecular quenching of the imine intermediate has been known in limited cases.⁵

Based on the previous results^{4,5} we reasoned out that a regioselective introduction of a homoallylic amine moiety at the 2-position of quinoline could be accomplished by a simple two-step procedure, as shown in Scheme 1. The first step is a synthesis of Reissert compound of quinoline⁶ and the next is an In-mediated allylation of nitrile moiety of the Reissert compound.^{4,7} In the intermediate stage, an intramolecular transfer of benzoyl group would be possible because the dihydroquinoline residue is a good leaving group and the *N*-benzoylimine intermediate could be converted into the homoallylic amine by a consecutive 1,3-H shift (see, path-a in Scheme 2).

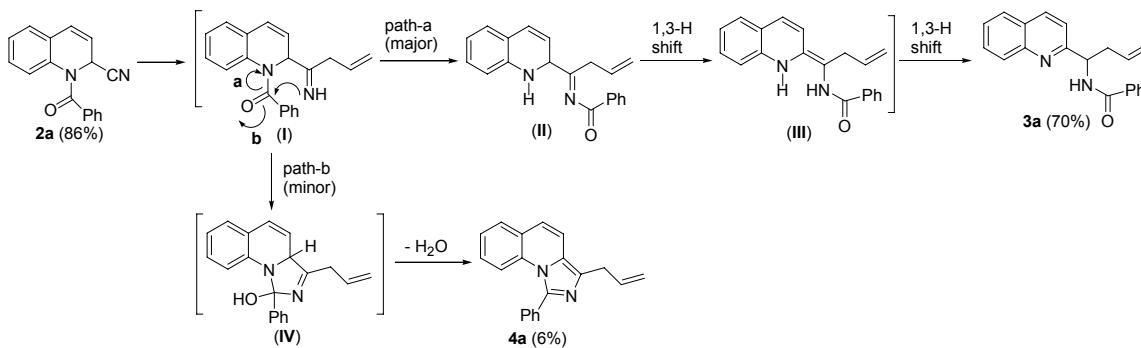


Scheme 1

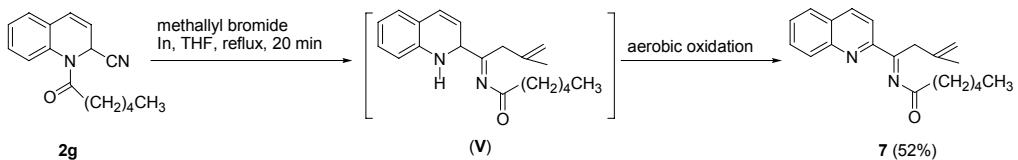
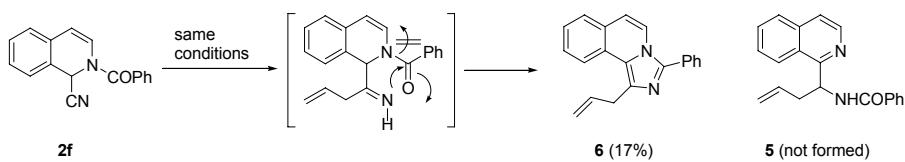
In order to examine the feasibility of our assumption, a representative Reissert compound **2a** was prepared according to the reported method.^{6f} The reaction of **2a** and allyl bromide was carried out in THF in the presence of indium metal at refluxing temperature for 20 min. To our delight, desired compound **3a** was obtained in good yield (70%) along with a trace amount of imidazo[1,5-*a*]quinoline **4a** (6%).⁸ The reaction mechanism can be postulated as shown in Scheme 2. The reaction of **2a** and allylindium reagents produced an imine intermediate (**I**), which produced *N*-benzoylimine intermediate (**II**) by following the path-a. The dihydroquinoline moiety acts as a leaving group to form (**II**). Successive 1,3-H transfers of (**II**) produced a homoallylic amine derivative **3a**. Compound **4a** was formed in a trace amount *via* the dehydrative cyclization (path-b).

Encouraged by the results, various Reissert compounds **2b-e** were prepared in high yields (85 - 89%) and the reactions with allylindium (or methylallylindium) reagents were carried out. As shown in Table 1, the corresponding homoallylic amine derivatives **3b-h** were obtained in reasonable yields (50 - 79%). The reactions with methylallyl bromide (entries 6-8) afforded the corresponding products in higher yields than the cases of allyl bromide (entries 1-5). However, the reactions of **2a** with benzyl bromide or crotyl bromide did not produce the corresponding amines in reasonable yields under the same reaction conditions. The reactions showed very sluggish reactivity and produced many intractable side products.

It is interesting to note that the reaction of a Reissert compound **2f**, derived from isoquinoline, did not produce the corresponding homoallylic amine **5** under the same reaction conditions,



Scheme 2

**Table 1.** In-mediated synthesis of *N*-arylhomoallylic amines

Entry	Substrate (%) ^a	Products (%) ^b
1		
2		
3		
4		
5 ^c		
6 ^d		
7 ^d		
8 ^{c,d}		

^aPrepared by PhCOCl/KCN/CH₂Cl₂/H₂O or PhCOCl/TMSCN/CH₂Cl₂ as reported.⁶ ^bConditions: Substrate (1.0 equiv), allyl bromide (3.0 equiv), In (1.5 equiv), THF, reflux, 20 min. ^cAr is *p*-tolyl. ^dMethylallyl bromide was used.

as shown in Scheme 3. Instead, a imidazo[5,1-*a*]isoquinoline derivative **6** was isolated in low yield (17%). The failure might be due to the less leaving ability of dihydroisoquinoline moiety (Scheme 3) than the dihydroquinoline moiety in the case of qui-

noline (Scheme 2).

The reaction of *N*-hexanoyl derivative **2g** did not produce the corresponding product, quite unexpectedly (Scheme 4). We isolated *N*-hexanoylimine derivative **7** in moderate yield (52%). This compound might be formed *via* an aerobic oxidation of the corresponding *N*-hexanoylimine intermediate (**V**). The reason for this unusual reactivity of **2g** is unclear at this stage.

In summary, an efficient two-step procedure was developed for the introduction of a homoallylic amine moiety at the 2-position of various quinolines.⁹ The reaction involved a synthesis of Reissert compound and a subsequent In-mediated allylation of nitrile.

Experimental Section

Preparation of starting materials. The Reissert compounds were prepared according to the reported procedures.^{6f-i} As a cyanide source KCN was used in most cases; however TMSCN was used for the preparation of **2c**^{6g} and **2g**. The spectroscopic data of unknown compound **2g** are as follows.

Compound 2g: 65%; colorless oil; IR (film) 2234, 1659, 1493, 1335 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 0.84 (t, *J* = 6.9 Hz, 3H), 1.19-1.35 (m, 4H), 1.53-1.73 (m, 2H), 2.32-2.64 (m, 2H), 6.08 (dd, *J* = 9.0 and 6.0 Hz, 1H), 6.51 (d, *J* = 6.0 Hz, 1H), 6.76 (d, *J* = 9.0 Hz, 1H), 7.24-7.38 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) δ 14.11, 22.54, 25.53, 31.49, 33.89, 40.52, 116.16, 122.24, 124.85, 127.07, 127.37, 127.81, 128.94, 129.62, 134.15, 173.02; ESIMS *m/z* 255 (M⁺+H). Anal. Calcd for C₁₆H₁₈N₂O: C, 75.56; H, 7.13; N, 11.01. Found: C, 75.79; H, 7.33; N, 10.89.

Typical procedure for the synthesis of 3a. A stirred mixture of **2a** (130 mg, 0.5 mmol), allyl bromide (181 mg, 1.5 mmol), and indium (86 mg, 0.75 mmol) in THF (2 mL) was heated to reflux for 20 min. After the usual aqueous workup and column chromatographic purification process (hexanes/CH₂Cl₂/EtOAc, 7:2:1) compound **3a** was isolated as a pale yellow solid (106 mg, 70%) along with **4a** (9 mg, 6%). Other compounds **3b-h** were synthesized similarly, and the spectroscopic data of **3a**, **4a**, **3b-h**, **6**, and **7** are as follows.

Compound 3a: 70%; pale yellow solid, mp 116 - 117 °C; IR (KBr) 3325, 1643, 1506, 1484 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.76-2.84 (m, 1H), 2.88-2.97 (m, 1H), 5.00-5.07 (m, 2H), 5.52 (dt, *J* = 6.9 and 6.0 Hz, 1H), 5.67-5.81 (m, 1H), 7.39 (d, *J* = 8.4

Hz, 1H), 7.44-7.56 (m, 4H), 7.72 (ddd, $J = 8.4, 6.9$ and 1.5 Hz, 1H), 7.81 (dd, $J = 8.1$ and 0.9 Hz, 1H), 7.91-7.94 (m, 2H), 8.09 (d, $J = 8.4$ Hz, 1H), 8.11 (d, $J = 6.9$ Hz, NH), 8.14 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 40.40, 53.86, 118.34, 120.15, 126.39, 127.05, 127.35, 127.62, 128.50, 129.01, 129.62, 131.36, 133.35, 134.81, 136.69, 147.19, 159.20, 166.66; ESIMS m/z 303 (M⁺+H). Anal. Calcd for C₂₀H₁₈N₂O: C, 79.44; H, 6.00; N, 9.26. Found: C, 79.51; H, 6.28; N, 9.12.

Compound 4a: 6%; pale yellow oil; IR (film) 1661, 1483, 1454, 1371 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 3.71 (dt, $J = 6.3$ and 1.5 Hz, 2H), 5.09-5.25 (m, 2H), 6.07-6.20 (m, 1H), 6.95 (d, $J = 9.3$ Hz, 1H), 7.14 (ddd, $J = 8.7, 7.2$ and 1.2 Hz, 1H), 7.30 (dd, $J = 7.2$ and 1.2 Hz, 1H), 7.32 (d, $J = 9.3$ Hz, 1H), 7.46-7.53 (m, 4H), 7.59 (dd, $J = 7.8$ and 1.5 Hz, 1H), 7.62-7.65 (m, 2H); ^{13}C NMR (CDCl₃, 75 MHz) δ 32.54, 115.91, 116.99, 117.62, 120.37, 125.26, 125.96, 127.10, 127.42, 128.75, 128.79, 129.02, 129.46, 129.93, 132.65, 134.02, 136.61, 141.37; ESIMS m/z 285 (M⁺+H).

Compound 3b: 61%; pale yellow solid, mp 127 - 128 °C; IR (KBr) 3341, 1645, 1599, 1514, 1483 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 2.58 (s, 3H), 2.79-2.89 (m, 1H), 2.92-3.01 (m, 1H), 5.03-5.11 (m, 2H), 5.55 (dt, $J = 6.9$ and 6.0 Hz, 1H), 5.71-5.85 (m, 1H), 7.39 (d, $J = 8.4$ Hz, 1H), 7.48-7.61 (m, 5H), 7.96-7.99 (m, 2H), 8.03 (d, $J = 8.4$ Hz, 1H), 8.09 (d, $J = 8.4$ Hz, 1H), 8.17 (d, $J = 6.9$ Hz, NH); ^{13}C NMR (CDCl₃, 75 MHz) δ 21.46, 40.40, 53.79, 118.25, 120.10, 126.47, 127.05, 127.37, 128.48, 128.66, 131.31, 131.84, 133.43, 134.87, 136.01, 136.24, 145.75, 158.21, 166.63; ESIMS m/z 317 (M⁺+H).

Compound 3c: 68%; pale yellow solid, mp 125 - 126 °C; IR (KBr) 3323, 1641, 1599, 1514, 1487 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 2.75-2.83 (m, 1H), 2.86-2.93 (m, 1H), 5.00-5.07 (m, 2H), 5.52 (dt, $J = 6.9$ and 6.0 Hz, 1H), 5.68-5.77 (m, 1H), 7.40 (d, $J = 8.7$ Hz, 1H), 7.43-7.51 (m, 3H), 7.62 (dd, $J = 9.0$ and 2.4 Hz, 1H), 7.77 (d, $J = 2.4$ Hz, 1H), 7.90-7.93 (m, 2H), 7.99 (d, $J = 6.9$ Hz, NH), 8.00 (d, $J = 9.0$ Hz, 1H), 8.02 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 40.19, 53.81, 118.48, 120.98, 126.23, 127.01, 127.85, 128.49, 130.44, 130.55, 131.41, 132.03, 133.16, 134.61, 135.72, 145.52, 159.67, 166.66; ESIMS m/z 337 (M⁺+H), 339 (M⁺+2+H). Anal. Calcd for C₂₀H₁₇ClN₂O: C, 71.32; H, 5.09; N, 8.32. Found: C, 71.17; H, 5.34; N, 8.21.

Compound 3d: 50%; pale yellow solid, mp 135 - 136 °C; IR (KBr) 3335, 1645, 1601, 1503, 1483 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 2.74-2.94 (m, 2H), 3.92 (s, 3H), 4.99-5.06 (m, 2H), 5.48 (dt, $J = 7.2$ and 6.0 Hz, 1H), 5.66-5.80 (m, 1H), 7.07 (d, $J = 2.7$ Hz, 1H), 7.34 (d, $J = 8.4$ Hz, 1H), 7.37 (dd, $J = 9.0$ and 2.7 Hz, 1H), 7.43-7.51 (m, 3H), 7.90-7.93 (m, 2H), 7.98 (d, $J = 9.6$ Hz, 1H), 8.02 (d, $J = 8.4$ Hz, 1H), 8.07 (d, $J = 7.2$ Hz, NH); ^{13}C NMR (CDCl₃, 75 MHz) δ 40.45, 53.70, 55.50, 105.12, 118.24, 120.43, 122.26, 127.04, 128.31, 128.49, 130.34, 131.33, 133.49, 134.81, 135.46, 143.24, 156.67, 157.67 166.66; ESIMS m/z 333 (M⁺+H).

Compound 3e: 65%; pale yellow solid, mp 138 - 139 °C; IR (KBr) 3335, 1643, 1612, 1524, 1491 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 2.40 (s, 3H), 2.74-2.84 (m, 1H), 2.86-2.95 (m, 1H), 4.98-5.06 (m, 2H), 5.51 (dt, $J = 7.2$ and 6.0 Hz, 1H), 5.67-5.80 (m, 1H), 7.26 (d, $J = 7.8$ Hz, 2H), 7.38 (d, $J = 8.4$ Hz, 1H), 7.52 (ddd, $J = 8.1, 6.9$ and 1.2 Hz, 1H), 7.71 (ddd, $J = 8.4, 6.9$ and 1.5 Hz, 1H), 7.80 (dd, $J = 8.1$ and 1.2 Hz, 1H), 7.82 (d, $J = 7.8$ Hz, 2H), 8.06 (d, $J = 7.2$ Hz, NH), 8.08 (d, $J = 8.1$ Hz, 1H), 8.12 (d,

$J = 8.7$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 21.74, 40.72, 54.18, 118.61, 120.52, 126.69, 127.39, 127.68, 127.94, 129.32, 129.49, 129.91, 132.27, 133.76, 137.00, 142.05, 147.53, 159.70, 166.98; ESIMS m/z 317 (M⁺+H).

Compound 3f: 79%; pale yellow solid, mp 107 - 108 °C; IR (KBr) 3223, 1639, 1614, 1533, 1501 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 1.78 (s, 3H), 2.76 (dd, $J = 6.9$ and 0.6 Hz, 2H), 4.69-4.70 (m, 1H), 4.79-4.80 (m, 1H), 5.59 (dt, $J = 7.5$ and 6.9 Hz, 1H), 7.39 (d, $J = 8.4$ Hz, 1H), 7.41-7.53 (m, 4H), 7.70 (ddd, $J = 8.4, 6.9$ and 1.5 Hz, 1H), 7.79 (dd, $J = 8.1$ and 1.2 Hz, 1H), 7.85 (d, $J = 7.5$ Hz, NH), 7.88-7.91 (m, 2H), 8.06-8.09 (m, 1H), 8.10 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 22.46, 44.72, 53.02, 114.24, 120.23, 126.24, 126.99, 127.32, 127.56, 128.44, 128.93, 129.49, 131.29, 134.72, 136.47, 141.59, 147.35, 160.04, 166.62; ESIMS m/z 317 (M⁺+H). Anal. Calcd for C₂₁H₂₀N₂O: C, 79.72; H, 6.37; N, 8.85. Found: C, 79.66; H, 6.56; N, 8.71.

Compound 3g: 60%; pale yellow solid, mp 129 - 130 °C; IR (KBr) 3333, 1643, 1601, 1501, 1485 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 1.77 (s, 3H), 2.74 (d, $J = 6.9$ Hz, 2H), 3.92 (s, 3H), 4.68-4.69 (m, 1H), 4.78-4.79 (m, 1H), 5.55 (dt, $J = 7.5$ and 6.9 Hz, 1H), 7.06 (d, $J = 2.7$ Hz, 1H), 7.35 (d, $J = 8.4$ Hz, 1H), 7.36 (dd, $J = 9.3$ and 2.7 Hz, 1H), 7.42-7.51 (m, 3H), 7.80 (d, $J = 7.5$ Hz, NH), 7.87-7.91 (m, 2H), 7.97 (d, $J = 9.3$ Hz, 1H), 8.01 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 22.52, 44.86, 52.89, 55.50, 105.11, 114.21, 120.61, 122.24, 127.04, 128.34, 128.49, 130.33, 131.33, 134.80, 135.30, 141.75, 143.47, 157.53, 157.62, 166.65; ESIMS m/z 347 (M⁺+H).

Compound 3h: 76%; pale yellow solid, mp 114 - 115 °C; IR (KBr) 3325, 1641, 1599, 1504, 1485 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 1.78 (s, 3H), 2.39 (s, 3H), 2.76 (dd, $J = 6.9$ and 0.6 Hz, 2H), 4.68-4.69 (m, 1H), 4.78-4.80 (m, 1H), 5.58 (dt, $J = 7.5$ and 6.9 Hz, 1H), 7.23-7.26 (m, 2H), 7.39 (d, $J = 8.4$ Hz, 1H), 7.51 (ddd, $J = 8.1, 6.9$ and 1.2 Hz, 1H), 7.70 (ddd, $J = 8.4, 6.9$ and 1.5 Hz, 1H), 7.78-7.81 (m, 4H), 8.06-8.09 (m, 1H), 8.10 (d, $J = 8.4$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 21.39, 22.49, 44.76, 53.02, 114.23, 120.34, 126.27, 127.03, 127.36, 127.60, 128.94, 129.13, 129.52, 131.87, 136.51, 141.68, 141.71, 147.40, 160.24, 166.63; ESIMS m/z 331 (M⁺+H).

Compound 6: 17%; pale yellow oil; IR (film) 1638, 1603, 1481, 1462 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 4.00 (dt, $J = 5.7$ and 1.8 Hz, 2H), 5.12-5.22 (m, 2H), 6.13-6.26 (m, 1H), 6.72 (d, $J = 7.2$ Hz, 1H), 7.38 (td, $J = 7.2$ and 0.9 Hz, 1H), 7.41-7.56 (m, 5H), 7.75-7.78 (m, 2H), 7.94 (d, $J = 7.5$ Hz, 1H), 8.07 (d, $J = 8.1$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 34.50, 113.89, 115.93, 120.76, 123.12, 124.33, 126.04, 126.14, 126.93, 127.44, 128.28, 128.60, 128.88, 128.91, 129.98, 133.14, 135.09, 139.33; ESIMS m/z 285 (M⁺+H).

Compound 7: 52%; pale yellow oil; IR (film) 1651, 1626, 1477, 1454, 1385 cm⁻¹; ^1H NMR (CDCl₃, 300 MHz) δ 0.93 (t, $J = 6.9$ Hz, 3H), 1.35-1.52 (m, 4H), 1.76 (s, 3H), 1.87-1.98 (m, 2H), 3.33-3.38 (m, 2H), 3.57 (s, 2H), 4.77-4.78 (m, 1H), 4.81-4.82 (m, 1H), 6.83 (d, $J = 9.6$ Hz, 1H), 7.22 (d, $J = 9.6$ Hz, 1H), 7.34 (ddd, $J = 7.5, 7.2$ and 1.5 Hz, 1H), 7.48 (ddd, $J = 8.7, 7.2$ and 1.5 Hz, 1H), 7.59 (dd, $J = 7.5$ and 1.5 Hz, 1H), 8.06 (d, $J = 8.7$ Hz, 1H); ^{13}C NMR (CDCl₃, 75 MHz) δ 14.02, 22.27, 22.39, 27.10, 31.69, 32.15, 36.05, 111.32, 116.37, 117.02, 119.12, 124.60, 125.91, 126.78, 127.50, 128.48, 130.40, 133.23, 142.97,

144.29; ESIMS m/z 309 ($M^+ + H$). Anal. Calcd for $C_{20}H_{24}N_2O$: C, 77.89; H, 7.84; N, 9.08 Found: C, 77.96; H, 7.56; N, 8.87.

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- Deprotection of *N*-benzoyl group of **3a** was easily conducted in aqueous EtOH in the presence of conc-HCl (reflux, 10 h) in 86%, and the spectroscopic data are as follows: pale yellow oil; IR (film) 3407, 3358, 1618, 1599, 1503, 1379 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 2.46-2.56 (m, 1H), 2.64-2.73 (m, 1H), 3.40 (s, 2H), 4.31 (dd, $J = 7.8$ and 5.4 Hz, 1H), 5.06-5.16 (m, 2H), 5.71-5.85 (m, 1H), 7.42 (d, $J = 8.1$ Hz, 1H), 7.49 (ddd, $J = 8.1$, 6.9 and 1.2 Hz, 1H), 7.67 (ddd, $J = 8.4$, 6.9 and 1.5 Hz, 1H), 7.77 (dd, $J = 8.1$ and 1.2 Hz, 1H), 8.05 (d, $J = 8.4$ Hz, 1H), 8.08 (d, $J = 8.4$ Hz, 1H); ESIMS m/z 199 ($M^+ + H$).