

Facile Synthesis of 1,2,3,4-Tetrasubstituted Pyrroles from Baylis-Hillman Adducts

Seong Jin Kim, Hoo Sook Kim, Taek Hyeon Kim,[†] and Jae Nyoung Kim*

Department of Chemistry and Institute of Basic Science, Chonnam National University, Gwangju 500-757, Korea

*E-mail: kimjn@chonnam.ac.kr

[†]Department of Applied Chemistry, Chonnam National University, Gwangju 500-757, Korea

Received June 4, 2007

Key Words : Pyrroles, Baylis-Hillman adducts, PCC, Decyanomethylation

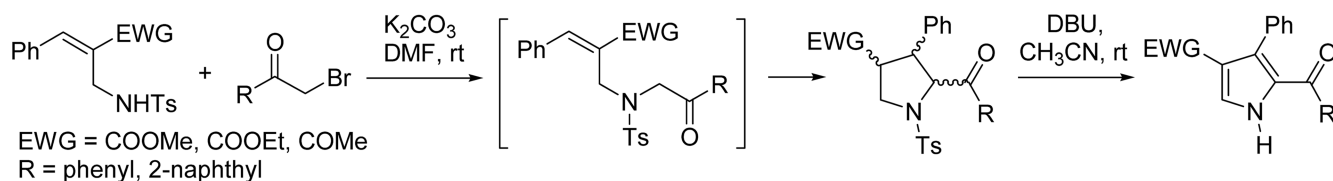
Suitably functionalized pyrroles are the basic skeleton of many biologically important substances and numerous synthetic methods of pyrroles have been investigated extensively.^{1,2} However, the synthesis of pyrrole derivatives from Baylis-Hillman adducts was not developed much.² Recently, we reported the synthesis of 2,3,4-trisubstituted pyrroles starting from the rearranged *aza*-Baylis-Hillman adducts (Scheme 1).³

Meantime we presumed that we could synthesize 1,2,3,4-tetrasubstituted pyrrole derivatives by using the synthetic approach in Scheme 2. As shown in Scheme 2, we imagined that the reaction of Baylis-Hillman acetate **1**, as the representative example, and secondary amine derivatives **2a-d** could give the corresponding S_N2' product **3a-d**, which could be cyclized to **4a-d** under basic conditions. The following acid-catalyzed dehydration and concomitant double bond isomerization of **4a-d** would provide desired pyrroles **5a-d**.

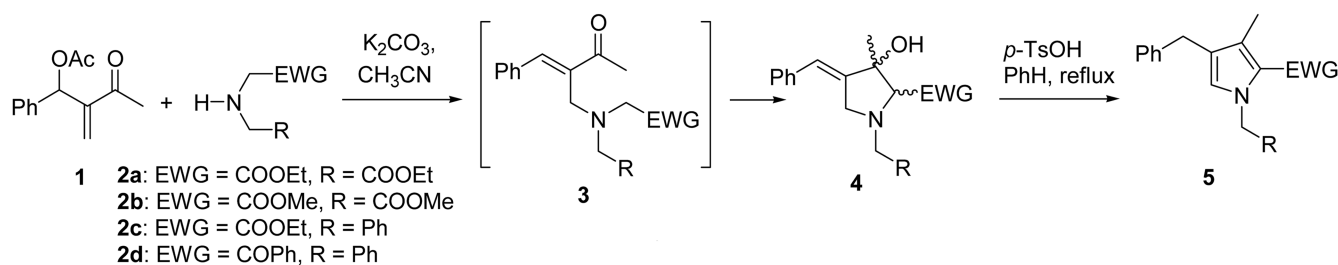
Among the examined conditions the use of K₂CO₃ in CH₃CN gave the best results for the preparation of **4a-d**. As expected we could not observe the formation of **3** (except for **3c**, entry 3 in Table 1),⁴ instead we obtained **4a-d** directly in 50-74% yields as inseparable *syn/anti* mixtures in a one-pot reaction. Based on the ¹H NMR spectra of **4a-d** the ratio of *syn/anti* was 4:1 to 2:1 (footnotes b-d in Table 1), however,

we did not confirm which isomer is the major one. For the reaction of **1** and **2c** we isolated **3c** in 34% yield (entry 3 in Table 1) together with **4c** in 50% yield. For the synthesis of compound **4d** (entry 4) we used **2d**⁵ in slightly excess amount (footnote e in Table 1). The following dehydration step of **4a-d** was carried out under the influence of *p*-TsOH (20-40 mol%) in benzene and we obtained the desired compounds **5a-d** in 41-64% yields. Isomerization of double bond occurred during the dehydration stage simultaneously to afford pyrroles directly. The results are summarized in Table 1.

However, the reaction of **1** and **2e** showed somewhat different reactivity as compared with those of **2a-d** (Scheme 3). When we carried out the reaction of **1** and **2e** in CH₃CN at room temperature the reaction did not show the formation of any new compounds in appreciable amounts presumably due to the limited solubility of **2e** in CH₃CN. Thus we elevated the temperature to refluxing, however, rearranged acetate was the major product in this case. After many trials we could obtain **3e** in 74% yield in aqueous CH₃CN at room temperature. In aqueous CH₃CN the compound **2e** was dissolved completely and the rearrangement of acetate group of **1** to the primary position was minimized at room temperature. With this compound **3e** in our hand we prepared **4e** under the same conditions of Table 1 (CH₃CN,



Scheme 1

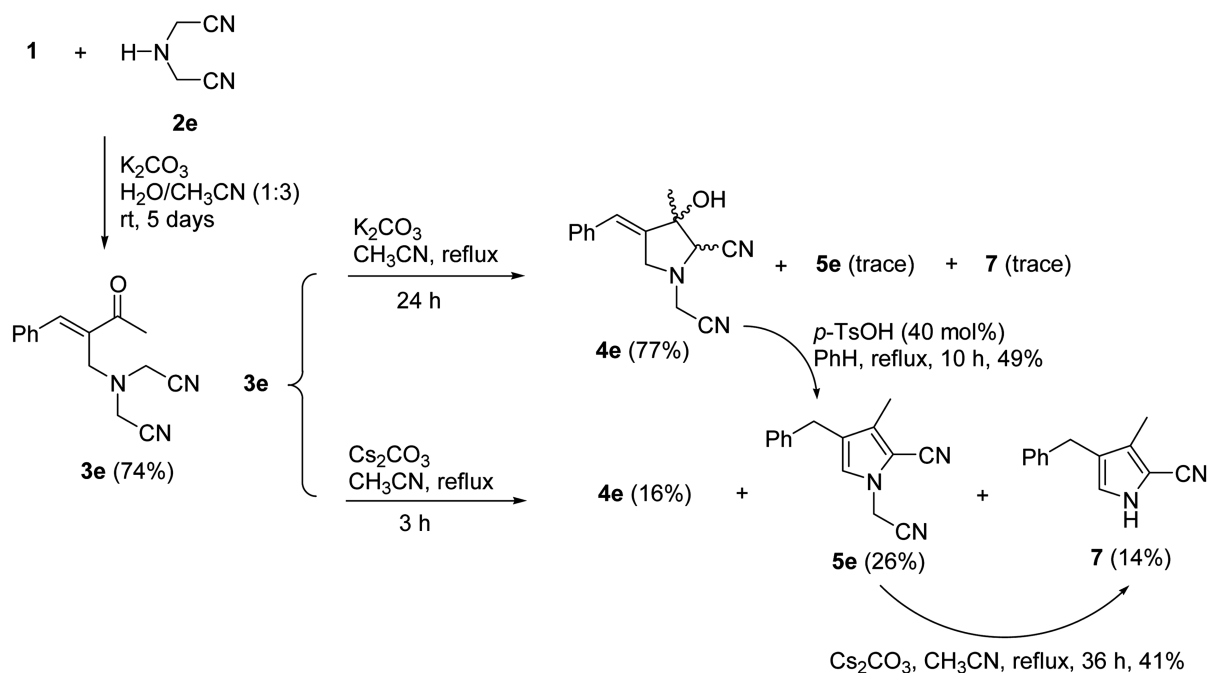


Scheme 2

Table 1. Synthesis of 1,2,3,4-tetrasubstituted pyrroles

Entry	1 + 2	Conditions	3 (%) / 4 (%)	Conditions	5 (%) ^f
1	1 + 2a	K ₂ CO ₃ (1.1 equiv) CH ₃ CN, reflux, 27 h	3a (nd) ^a / 4a (69) ^b	<i>p</i> -TsOH (20 mol%) PhH, reflux, 10 h	5a (64)
2	1 + 2b	K ₂ CO ₃ (1.1 equiv) CH ₃ CN, reflux, 26 h	3b (nd) ^a / 4b (71) ^c	<i>p</i> -TsOH (20 mol%) PhH, reflux, 12 h	5b (47)
3	1 + 2c	K ₂ CO ₃ (2.2 equiv) CH ₃ CN, reflux, 7 days	3c (34) / 4c (50) ^d	<i>p</i> -TsOH (40 mol%) PhH, reflux, 2 days	5c (56)
4	1 + 2d ^e	K ₂ CO ₃ (1.1 equiv) CH ₃ CN, rt, 1 h	3d (nd) ^a / 4d (74) ^d	<i>p</i> -TsOH (20 mol%) PhH, reflux, 12 h	5d (41)

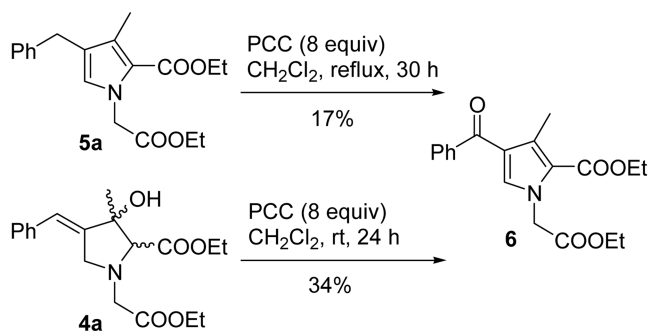
^aNd means not detected. ^bThe ratio is 2:1 (*syn/anti* mixture). ^cThe ratio is 4:1 (*syn/anti* mixture). ^dThe ratio is 3:1 (*syn/anti* mixture). ^eStarting material 2d was prepared by the reaction of benzylamine and phenacyl bromide according to the reference.⁵ The compound 2d was unstable thus we used this compound in a crude state and we used 0.91 equiv of 1. ^fIsolated yield.

**Scheme 3**

K₂CO₃, reflux, 24 h) in 77% yield (*syn/anti*, 3:2). Dehydration of 4e under the same conditions (*p*-TsOH/benzene/reflux) afforded 5e in 49% yield. During the synthesis of 4e we observed the formation of trace amounts of 5e and 7. It is interesting to note that the yields of 5e and 7 were increased with concomitant decrease of 4e when we used Cs₂CO₃ (CH₃CN, reflux, 3 h). The formation of pyrrole derivative 7 can be explained by decyanomethylation of 5e,⁶ and we confirmed the conversion experimentally by transforming 5e into 7 under the same conditions (41% and recovered 5e in 10%).

Finally, we examined the possibility for the oxidation of 5a into 4-benzoylpyrrole derivative 6 as in our previous oxidation involving PCC (pyridinium chlorochromate) in a similar system.⁷ However, the yield of oxidized compound 6 was very low to be useful in a synthetic point of view. It is interesting to note that the oxidation with the precursor 4a instead of 5a showed somewhat improved yield.

In summary, we disclosed the synthesis of poly-substituted

**Scheme 4**

pyrrole derivatives from the reaction of Baylis-Hillman acetate and some secondary amine compounds.⁸

Experimental Section

Typical experimental procedure for the synthesis of compounds 4a and 5a, and the spectroscopic data of 3c,

3e, 4a-e, 5a-e, 6, and 7 are as follows. A stirred mixture of **1** (218 mg, 1.0 mmol), **2a** (189 mg, 1.0 mmol), and K_2CO_3 (152 mg, 1.1 mmol) in CH_3CN (5 mL) was heated to reflux for 27 h. After the usual aqueous workup procedure and column chromatographic purification process (hexanes/EtOAc, 3:1) we obtained **4a** as colorless oil, 240 mg (69%). A solution of **4a** (174 mg, 0.5 mmol) and *p*-TsOH (19 mg, 0.1 mmol) in benzene (4 mL) was heated to reflux for 10 h. After the usual aqueous workup procedure and column chromatographic purification process (hexanes/EtOAc, 6:1) we obtained **5a** as a white solid, 105 mg (64%).

Compound **3c**: 34%; colorless oil; IR (film) 2924, 1737, 1666, 1231, 1189, 1029 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 1.19 (t, $J = 7.2$ Hz, 3H), 2.42 (s, 3H), 3.22 (s, 2H), 3.76 (s, 2H), 3.81 (s, 2H), 4.04 (q, $J = 7.2$ Hz, 2H), 7.19-7.27 (m, 5H), 7.32-7.42 (m, 3H), 7.55-7.58 (m, 3H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 14.14, 26.70, 49.21, 53.37, 57.85, 60.07, 127.10, 128.18, 128.37, 128.83, 129.11, 130.05, 135.11, 138.59, 139.04, 141.62, 171.18, 200.85.

Compound **3e**: 74%; colorless oil; IR (film) 2246, 1664, 1421, 1230, 1132 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 2.51 (s, 3H), 3.55 (s, 4H), 3.64 (s, 2H), 7.42-7.49 (m, 5H), 7.85 (s, 1H).

Compound **4a**: 69% (*syn/anti*, 2:1); colorless oil; IR (film) 3446, 2981, 1738, 1448, 1195, 1097 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz, major isomer) δ 1.27 (t, $J = 7.2$ Hz, 3H), 1.31 (t, $J = 7.2$ Hz, 3H), 1.64 (s, 3H), 2.80 (br s, 1H), 3.51-3.84 (m, 4H), 4.11-4.36 (m, 5H), 6.61 (t, $J = 2.4$ Hz, 1H), 7.20-7.24 (m, 3H), 7.28-7.36 (m, 2H).

Compound **4b**: 71% (*syn/anti*, 4:1); colorless oil; IR (film) 3452, 2954, 1747, 1693, 1442, 1213, 1178 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz, major isomer) δ 1.63 (s, 3H), 3.51-3.90 (m, 6H), 3.70 (s, 3H), 3.77 (s, 3H), 6.61 (t, $J = 2.4$ Hz, 1H), 7.20-7.26 (m, 3H), 7.27-7.36 (m, 2H).

Compound **4c**: 50% (*syn/anti*, 3:1); colorless oil; IR (film) 3454, 2981, 1739, 1448, 1261, 1196 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz, major isomer) δ 1.32 (t, $J = 7.5$ Hz, 3H), 1.60 (s, 3H), 2.75 (br s, 1H), 3.34-3.65 (m, 3H), 3.94-4.05 (m, 2H), 4.21-4.31 (m, 2H), 6.56 (t, $J = 2.4$ Hz, 1H), 7.15-7.21 (m, 3H), 7.24-7.39 (m, 2H).

Compound **4d**: 74% (*syn/anti*, 3:1); colorless oil; IR (film) 3438, 1676, 1448, 1228, 1180, 1092 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz, major isomer) δ 1.55 (s, 3H), 2.68 (br s, 1H), 3.38-4.23 (m, 4H), 4.38 (s, 1H), 6.53 (t, $J = 2.4$ Hz, 1H), 7.17-7.34 (m, 10H), 7.43-7.49 (m, 2H), 7.54-7.60 (m, 1H), 7.93-7.97 (m, 2H).

Compound **4e**: 77% (*syn/anti*, 3:2); colorless oil; IR (film) 3429, 2925, 2222, 1448, 1261, 1101 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz, major isomer) δ 1.66 (s, 3H), 2.60 (br s, 1H), 3.69 (s, 1H), 3.80-3.97 (m, 4H), 6.70 (t, $J = 2.4$ Hz, 1H), 7.21-7.46 (m, 5H) and 1H NMR ($CDCl_3$, 300 MHz, minor isomer) δ 1.71 (s, 3H), 2.54 (br s, 1H), 3.78 (s, 1H), 3.81 (s, 1H), 3.87 (s, 1H), 3.91 (d, $J = 2.4$ Hz, 2H), 6.60 (t, $J = 2.4$ Hz, 1H), 7.21-7.42 (m, 5H).

Compound **5a**: 64%; white solid, mp 42-44 $^{\circ}C$; IR (film) 1755, 1687, 1417, 1298, 1199, 1097 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 1.27 (t, $J = 7.2$ Hz, 3H), 1.32 (t, $J = 7.2$ Hz, 3H), 2.24 (s, 3H), 3.76 (s, 2H), 4.21 (q, $J = 7.2$ Hz, 2H), 4.25 (q, $J = 7.2$ Hz, 2H), 4.87 (s, 2H), 6.42 (s, 1H), 7.12-7.20 (m, 3H), 7.25-7.30 (m, 2H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 11.60, 14.12, 14.34, 31.24, 51.14, 59.69, 61.30, 119.74, 122.66, 125.84, 127.65, 128.32, 128.53, 128.66, 140.81, 162.08, 169.27; LCMS m/z 329 (M^+).

Compound **5b**: 47%; colorless oil; IR (film) 1759, 1693, 1444, 1215, 1124, 1099 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 2.23 (s, 3H), 3.75 (s, 3H), 3.76 (s, 2H), 3.79 (s, 3H), 4.87 (s, 2H), 6.43 (s, 1H), 7.16-7.20 (m, 3H), 7.24-7.30 (m, 2H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 11.54, 31.22, 50.80, 50.97, 52.28, 119.55, 122.77, 125.86, 127.87, 128.33, 128.52, 128.72, 140.68, 162.54, 169.69.

Compound **5c**: 56%; colorless oil; IR (film) 1693, 1452, 1421, 1386, 1297, 1095 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 1.24 (t, $J = 6.9$ Hz, 3H), 2.24 (s, 3H), 3.76 (s, 2H), 4.19 (q, $J = 6.9$ Hz, 2H), 5.43 (s, 2H), 6.55 (s, 1H), 7.01-7.04 (m, 2H), 7.14-7.29 (m, 8H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 11.69, 14.28, 31.21, 52.44, 59.47, 119.60, 122.29, 125.76, 126.41, 127.04, 127.33, 128.28, 128.39, 128.43, 128.59, 138.96, 141.03, 161.86; LCMS m/z 333 (M^+).

Compound **5d**: 41%; colorless oil; IR (film) 1624, 1495, 1446, 1400, 1215, 1173 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 1.63 (s, 3H), 3.73 (s, 2H), 5.37 (s, 2H), 6.68 (s, 1H), 7.05-7.08 (m, 2H), 7.16-7.30 (m, 8H), 7.34-7.40 (m, 2H), 7.44-7.50 (m, 1H), 7.58-7.61 (m, 2H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 12.04, 31.30, 51.99, 122.72, 125.87, 126.80, 127.28, 128.16, 128.23, 128.34, 128.39 (2C), 128.45, 128.47, 129.00, 129.35, 131.59, 138.71, 140.73, 188.34; LCMS m/z 365 (M^+).

Compound **5e**: 49%; colorless oil; IR (film) 2208, 1493, 1425, 1390, 1372 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 2.11 (s, 3H), 3.73 (s, 2H), 4.82 (s, 2H), 6.58 (s, 1H), 7.13-7.16 (m, 2H), 7.19-7.33 (m, 3H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 10.32, 31.25, 35.66, 103.72, 112.38, 113.39, 124.99, 125.64, 126.42, 128.45, 128.63, 132.60, 139.28.

Compound **6**: 34%; colorless oil; IR (film) 2981, 1753, 1693, 1643, 1251, 1203 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 1.29 (t, $J = 7.5$ Hz, 3H), 1.38 (t, $J = 7.5$ Hz, 3H), 2.64 (s, 3H), 4.24 (q, $J = 7.5$ Hz, 2H), 4.32 (q, $J = 7.5$ Hz, 2H), 4.95 (s, 2H), 7.06 (s, 1H), 7.43-7.47 (m, 2H), 7.52-7.55 (m, 1H), 7.76 (m, 2H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 12.55, 14.12, 14.28, 51.79, 60.47, 61.73, 121.91, 122.65, 128.26, 129.04, 131.69, 132.49, 134.92, 140.18, 168.24 (2C), 191.45; LCMS m/z 343 (M^+).

Compound **7**: 41%; pale yellow solid, mp 95-97 $^{\circ}C$; IR (film) 3303, 2212, 1396 cm^{-1} ; 1H NMR ($CDCl_3$, 300 MHz) δ 2.11 (s, 3H), 3.75 (s, 2H), 6.58 (d, $J = 3.0$ Hz, 1H), 7.14-7.22 (m, 3H), 7.26-7.31 (m, 2H), 8.45 (br s, 1H); ^{13}C NMR ($CDCl_3$, 75 MHz) δ 9.96, 31.29, 100.08, 114.45, 121.97, 123.62, 126.12, 128.43, 128.46, 130.64, 140.16; LCMS m/z 196 (M^+).

Acknowledgments. This work was supported by the Korea Research Foundation Grant funded by the Korean Government (MOEHRD, KRF-2006-311-C00384). Spectroscopic data was obtained from the Korea Basic Science Institute, Gwangju branch.

References and Notes

1. For the syntheses and biological activities of pyrrole derivatives, see: (a) Bellina, F.; Rossi, R. *Tetrahedron* **2006**, 62, 7213-7256. (b) Knight, D. W.; Sharland, C. M. *Synlett* **2004**, 119-121. (c) Singh, V.; Kanojiya, S.; Batra, S. *Tetrahedron* **2006**, 62, 10100-10110. (d) Knight, D. W.; Sharland, C. M. *Synlett* **2003**, 2258-2260. (e) Magnus, N. A.; Staszak, M. A.; Udodong, U. E.; Wepsiec, J. P. *Org. Proc. Res. Dev.* **2006**, 10, 899-904. (f) Zen, S.; Harada, K. *Chem. Pharm. Bull.* **1982**, 30, 366-369. (g) Chen, Q.; Wang, T.; Zhang, Y.; Wang, Q.; Ma, J. *Synth. Commun.* **2002**, 32, 1051-1058. (h) Nicolaou, I.; Demopoulos, V. J. *J. Med. Chem.* **2003**, 46, 417-426. (i) Gupton, J. T.; Banner, E. J.; Scharf, A. B.; Norwood, B. K.; Kanters, R. P. F.; Dominey, R. N.; Hempel, J. E.; Kharlamova, A.; Bluhn-Chertudi, I.; Hickenboth, C. R.; Little, B. A.; Sartin, M. D.; Coppock, M. B.; Krumpe, K. E.; Burnham, B. S.; Holt, H.; Du, K. X.; Keertikar, K. M.; Diebes, A.; Ghassemi, S.; Sikorski, J. A. *Tetrahedron* **2006**, 62, 8243-8255. (j) Cadamuro, S.; Degani, I.; Dughera, S.; Fochi, R.; Gatti, A.; Piscopo, L. *J. Chem. Soc., Perkin Trans. 1* **1993**, 273-283. (k) Cohnen, E.; Dewald, R. *Synthesis* **1987**, 566-568. (l) Misra, N. C.; Panda, K.; Ila, H.; Junjappa, H. *J. Org. Chem.* **2007**, 72, 1246-1251.
2. For the examples on the synthesis of pyrroles from Baylis-Hillman adducts, see: (a) Declerck, V.; Ribiere, P.; Martinez, J.; Lamaty, F. *J. Org. Chem.* **2004**, 69, 8372-8381. (b) Shi, M.; Xu, Y.-M. *Eur. J. Org. Chem.* **2002**, 696-701. (c) Roy, A. K.; Pathak, R.; Yadav, G. P.; Maulik, P. R.; Batra, S. *Synthesis* **2006**, 1021-1027.
3. Lee, H. S.; Kim, J. M.; Kim, J. N. *Tetrahedron Lett.* **2007**, 48, 4119-4122.
4. When we carried out the reaction in DMF in the presence of K_2CO_3 at room temperature, the corresponding intermediates **3** could be isolated in moderate yields.
5. For the synthesis of compound **2d**, see: (a) Kawamoto, A.; Wills, M. *Tetrahedron: Asymmetry* **2000**, 11, 3257-3261. (b) Guarna, A.; Bucelli, I.; Machetti, F.; Menchi, G.; Occhiato, E. G.; Scarpi, D.; Trabocchi, A. *Tetrahedron* **2002**, 58, 9865-9870. (c) Deng, B.-L.; Demillequand, M.; Laurent, M.; Touillaux, R.; Belmans, M.; Kemps, L.; Ceresiat, M.; Marchand-Brynaert, J. *Tetrahedron* **2000**, 56, 3209-3217.
6. For the decyanomethylation, see: (a) Katritzky, A. R.; Latif, M.; Urogdi, L. *J. Chem. Soc., Perkin Trans. 1* **1990**, 667-672. (b) Overman, L. E.; Shin, J. *J. Org. Chem.* **1991**, 56, 5005-5007. (c) Yang, T.-K.; Hung, S.-M.; Lee, D.-S.; Hong, A.-W.; Cheng, C.-C. *Tetrahedron Lett.* **1989**, 30, 4973-4976. (d) Padwa, A.; Chen, Y.-Y.; Dent, W.; Nimmesgern, H. *J. Org. Chem.* **1985**, 50, 4006-4014.
7. For the related PCC oxidations, see: (a) Kim, S. J.; Lee, H. S.; Kim, J. N. *Tetrahedron Lett.* **2007**, 48, 1069-1072. (b) Dauben, W. G.; Michno, D. M. *J. Org. Chem.* **1977**, 42, 682-685.
8. For our recent publications on the synthesis of nitrogen-containing five-membered heterocyclic compounds, see: (a) Lee, K. Y.; Lee, Y. J.; Kim, J. N. *Bull. Korean Chem. Soc.* **2007**, 28, 143-146. (b) Kim, S. C.; Lee, K. Y.; Gowrisankar, S.; Kim, J. N. *Bull. Korean Chem. Soc.* **2006**, 27, 1133-1139. (c) Lee, H. S.; Kim, S. J.; Kim, J. N. *Bull. Korean Chem. Soc.* **2006**, 27, 1063-1066.