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### Communications

#### Palladium-catalyzed Acylation of $\beta,\beta$ -Diphenyl- $\alpha$ -(trifluoromethyl)vinylstannane as a Novel Route to 1,3-Disubstituted 2-(Trifluoromethyl)indene Derivatives

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Although a various types of  $\alpha$ -(trifluoromethyl)vinylmetal reagents such as lithium<sup>1,2</sup> and zinc<sup>3-6</sup> have been synthesized and utilized previously, the preparation and synthetic utility of  $\alpha$ -(trifluoromethyl)vinylstannane reagent have been quite limited. Only a couple of papers described about chemistry of  $\alpha$ -(trifluoromethyl)vinylstannane reagent. The  $\alpha$ -(trifluoromethyl)vinylstannane reagent bearing only hydrogens at  $\beta$ -position has been synthesized from the reaction of 2-bromo-trifluoroisopropene with lithium tributylstannate in the presence of CuI and utilized for the cross-coupling reactions with acyl chlorides in the presence of catalytic amount of  $\text{PdCl}(\text{Bn})(\text{PPh}_3)_2$  in HMPA at 65 °C to give  $\alpha$ -(trifluoromethyl)vinyl ketone derivatives.<sup>7</sup> Ichikawa also carried out the reaction of  $\alpha$ -(trifluoromethyl)vinylstannane reagent with  $\alpha,\beta$ -unsaturated acyl chlorides in the presence of catalytic amount of  $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$  and CuCN in toluene at 55-75 °C to give the desired Nazarov substrates.<sup>8</sup> Recently, we reported about the preparation of a novel  $\alpha$ -(trifluoromethyl)vinylstannane reagent<sup>9</sup> bearing two phenyl groups at  $\beta$ -position and the cross coupling reactions of it with aryl iodides to give trifluoromethylated triphenylethene derivatives which are important framework of many nonsteroidal antiestrogens.<sup>10</sup> As a part of our continuing studies on the synthetic utility of  $\beta,\beta$ -diphenyl- $\alpha$ -(trifluoromethyl)vinylstannane reagent, we examined palladium-promoted acylation of this reagent with acyl chlorides to give  $\beta,\beta$ -diphenyl- $\alpha$ -trifluoromethylated enone derivatives which are useful intermediates for the formation of novel 1,3-disubstituted 2-(trifluoromethyl)indene derivatives *via* Friedel-Craft's type of the cyclization. Since nonfluorinated 1,3-disubstituted indene derivative such as Indenestrol A exhibited mammary tumor inhibiting antiestrogen

activity,<sup>11,12</sup> it is expected that 1,3-disubstituted 2-(trifluoromethyl)indene derivatives also have a potential similar activity. Herein, we describe the palladium-promoted acylation of  $\beta,\beta$ -diphenyl- $\alpha$ -(trifluoromethyl)vinylstannane reagent and the formation of 1,3-disubstituted 2-(trifluoromethyl)indene derivatives from the acylated adduct.

A starting material,  $\beta,\beta$ -diphenyl- $\alpha$ -(trifluoromethyl)vinylstannane reagent **1**, was prepared *via* several steps from 2,3,3,3-tetrafluoro-1-phenyl-1-phenylthiopropene.<sup>9</sup> First of all, the acylation reaction of **1** with acetyl chloride was carried out in the presence of several palladium catalyst. The use of Pd catalyst such as  $\text{Pd}(\text{PPh}_3)_4$  or  $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$  in THF, DMF, toluene or HMPA did not provide any acylated product. However, acylation reaction to give acylated product **2a** was successfully accomplished by using a mixture of 10 mol%  $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$  and 10 mol% CuCN in toluene at 50 °C for 6 h. The higher temperature (80 °C) was needed for the completion of acylation of **1** with types of benzoyl chlorides. Therefore, starting material **1** underwent the acylation reaction with a various types of acyl chlorides, such as ethyl chloroformate, furoyl chloride, naphthoyl chloride and benzoyl chlorides bearing a bromo, methoxy, methyl, or nitro on the benzene ring, to give the corresponding trifluoromethylated enone derivatives **2** at 80 °C for 6 h. The experimental results of the acylation reactions are summarized in Table 1.

Reduction of **2a** with  $\text{LiAlH}_4$  (1.5 equiv.) in ether at reflux temperature for 3 h afforded the corresponding allylic alcohols **3a**<sup>13</sup> in 71% yield. The use of  $\text{NaBH}_4$  did not provide the desired product, whereas the starting material was always recovered. The Friedel-Craft's type of cyclization of **3a** was successfully accomplished to give 2-trifluoromethyl-3-methyl-

**Table 1.** The acylation reactions of **1** with acyl chlorides

Compound No.	T (°C)	R	Yield (%) <sup>a</sup>
<b>2a</b>	50	CH <sub>3</sub>	68
<b>2b</b>	80	C <sub>6</sub> H <sub>5</sub>	80
<b>2c</b>	80	( <i>o</i> -CH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	78
<b>2d</b>	80	( <i>p</i> -CH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	88
<b>2e</b>	80	( <i>p</i> -OCH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	85
<b>2f</b>	80	( <i>m</i> -NO <sub>2</sub> )-C <sub>6</sub> H <sub>4</sub>	72
<b>2g</b>	80	( <i>m</i> -CH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	81
<b>2h</b>	80	( <i>m</i> -Br)-C <sub>6</sub> H <sub>4</sub>	77
<b>2i</b>	80	C <sub>2</sub> H <sub>5</sub> O	62
<b>2j</b>	80	2-furanyl	75
<b>2k</b>	80	2-naphthyl	86

<sup>a</sup> Isolated yields.

1-phenylindene (**4a**)<sup>14</sup> by using AlCl<sub>3</sub> (1.2 equiv.) in methylene chloride at -78 °C, followed by the slowly warming to room temperature. The use of dilute H<sub>2</sub>SO<sub>4</sub> instead of AlCl<sub>3</sub> at reflux temperature caused not only to decrease the yield of indene derivatives **4a**, but also to extend the reaction time. The more excess of AlCl<sub>3</sub> (2.0 equiv.) was used to carry out the acylation of **3d** and **3f** because of possible coordination of oxygen with AlCl<sub>3</sub>. Reduction of other types of enone derivatives **2**, followed by treatment with AlCl<sub>3</sub> under the same reaction condition also provided the corresponding 1,3-disubstituted 2-(trifluoromethyl)indene derivatives **4** in good yields. The experimental results of reduction and cyclization reactions are summarized in Table 2. Although 2-(trifluoromethyl)indene has been prepared in a previous literature, this method is a lack of generality and provides low yield preparation.<sup>15</sup>

A typical reaction procedure for the preparation of **2a** is as follows. To a toluene (5 mL) solution of acetyl chloride (0.088 g, 1.12 mmol) and β,β-diphenyl-α-(trifluoromethyl)vinyl-

**Table 2.** The synthesis of 1,3-disubstituted 2-trifluoromethylated indene derivatives **4**

Compound No.	R	Yield of <b>3</b> (%) <sup>a</sup>	Yield of <b>4</b> (%) <sup>a</sup>
<b>3a, 4a</b>	CH <sub>3</sub>	71	76
<b>3b, 4b</b>	C <sub>6</sub> H <sub>5</sub>	74	78
<b>3c, 4c</b>	( <i>p</i> -CH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	69	71
<b>3d, 4d</b>	( <i>p</i> -OCH <sub>3</sub> )-C <sub>6</sub> H <sub>4</sub>	73	68
<b>3e, 4e</b>	( <i>m</i> -Br)-C <sub>6</sub> H <sub>4</sub>	70	73
<b>3f, 4f</b>	( <i>m</i> -NO <sub>2</sub> )-C <sub>6</sub> H <sub>4</sub>	67	66
<b>3g, 4g</b>	2-furanyl	62	63
<b>3h, 4h</b>	2-naphthyl	71	74

<sup>a</sup> Isolated yields.

stannane (0.402 g, 0.75 mmol) was added Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (10 mol%) and CuCN (10 mol%), and the reaction mixture was heated at 50 °C for 6 h under argon atmosphere. After the reaction mixture was quenched with water and then washed with 5% KF solution and brine, solution was extracted with ether twice. The ether solution was dried and chromatographed on SiO<sub>2</sub> column. Elution with a mixture of hexane and ethyl acetate (20 : 1) provided 0.145 g of 2-trifluoromethyl-1,1-diphenyl-1-buten-3-one (**2a**) in 68% yield. **2a**: mp 65–66 °C; <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>) δ 7.41–7.18 (m, 10H), 2.03 (s, 3H); <sup>19</sup>F NMR (100 MHz, CDCl<sub>3</sub>) δ -55.42 (s, 3F); MS, m/z (relative intensity) 290 (M<sup>+</sup>, 81), 289 (100), 275 (12), 255 (22), 227 (29), 213 (30), 207 (17), 178 (29), 176 (19), 151 (13), 127 (19), 105 (33), 77 (13), 51 (10); IR (KBr) 3082, 3059, 3028, 2927, 2855, 1704, 1617, 1608, 1492, 1445, 1418, 1357, 1322, 1261, 1216, 1145, 1111, 1078, 1042, 998, 951, 759, 699, 656 cm<sup>-1</sup>. Anal. Calcd for C<sub>17</sub>F<sub>3</sub>H<sub>13</sub>O: C, 70.32; H, 4.52. Found: C, 70.49; H, 4.60.

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- Spectroscopic data of **3a**: oil; <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>) δ 7.38–7.11 (m, 10H), 4.58 (m, 1H), 1.87 (d, *J* = 8.0 Hz, 1H), 1.52 (d, *J* = 7.4 Hz, 3H); <sup>19</sup>F NMR (100 MHz, CDCl<sub>3</sub>) δ -52.64 (s, 3F); MS, m/z (relative intensity) 292 (M<sup>+</sup>, 44), 277 (15), 274 (19), 259 (15), 237 (13), 223 (16), 209 (26), 178 (18), 171 (21), 167 (100), 165 (33), 151 (29), 127 (59), 77 (22), 51 (17); IR (neat) 3395, 3059, 3028, 2963, 2932, 1623, 1492, 1445, 1318, 1260, 1130, 1077, 1018, 797, 759, 700 cm<sup>-1</sup>. Anal. Calcd for C<sub>17</sub>F<sub>3</sub>H<sub>15</sub>O: C, 69.84; H, 5.18. Found: C, 70.01; H, 5.26.
- Spectroscopic data of **4a**: mp 75–77 °C; <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>) δ 7.41–7.10 (m, 10H), 4.89 (q, *J* = 7.5 Hz, 1H), 1.76 (d, *J* = 7.5 Hz, 3H); <sup>19</sup>F NMR (100 MHz, CDCl<sub>3</sub>) δ -52.50 (s, 3F); MS, m/z (relative intensity) 274 (M<sup>+</sup>, 39), 233 (41), 205 (100), 203 (36), 196 (52), 190 (15), 177 (24), 127 (14), 101 (20), 91 (18), 77 (19), 51 (15); IR (KBr) 3058, 2929, 2855, 2360, 1598, 1492, 1445, 1320, 1262, 1194, 1137, 1034, 763, 732, 700 cm<sup>-1</sup>. Anal. Calcd for C<sub>17</sub>F<sub>3</sub>H<sub>13</sub>: C, 74.43; H, 4.78. Found: C, 74.27; H, 4.89.
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