

Micronutrient cofactor research with extensions to applications

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Following identification of essential micronutrients, there has been a continuum of research aimed at revealing their absorption, transport, utilization as cofactors, and excretion and secretion. Among those cases that have received our attention are vitamin B₆, riboflavin, biotin, lipoate, ascorbate, and certain metal ions. Circulatory transport and cellular uptake of the water-soluble vitamins exhibit relative specificity and facilitated mechanisms at physiological concentrations. Isolation of enzymes and metabolites from micro-organisms and mammals has provided information on pathways involved in cofactor formation and metabolism. Kinases catalysing phosphorylation of B₆ and riboflavin have a preference for Zn²⁺ in stereospecific chelates with adenosine triphosphate. The synthetase for flavin adenine dinucleotide prefers Mg²⁺. The flavin mononucleotide-dependent oxidase that converts the 5'-phosphates of pyridoxine and of pyridoxamine to pyridoxal phosphate is a connection between B₆ and riboflavin and is a primary control point for conversion of B₆ to its coenzyme. Sequencing and cloning of a side-chain oxidase for riboflavin was achieved. Details on binding and function have been delineated for some cofactor systems, especially in several flavoproteins. There is both photochemical oxidation and oxidative catabolism of B₆ and riboflavin. Both biotin and lipoate undergo oxidation of their acid side chains with redox cleavage of the rings. Applications from our findings include the development of affinity absorbents, enhanced drug delivery, delineation of residues in biopolymer modification, pathogen photoinactivation in blood components, and input into human dietary recommendations. Ongoing and future research in the cofactor arena can be expected to add to this panoply. At the molecular level, the way in which the same cofactor can participate in diverse catalytic reactions resides in interactions with surrounding enzyme structures that must be determined case by case. At the level of human intake, more knowledge is desirable for making micronutrient recommendations based on biochemical indicators, especially for the span between infancy and adulthood.

Micronutrient cofactors: Vitamins: Metabolic pathways: Enzymes

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Introduction

The subject of micronutrient cofactors is of major importance because the vitamins and trace elements that constitute this group of dietary ingredients are essential in their function to help man and other organisms utilize such macronutrients as proteins, carbohydrates, fats, and macrominerals that provide energy and the major substances of our bodies.

It should be noted for the reader who has an interest in micronutrient cofactors that there are timely books dealing with nutritional biochemistry of the vitamins (Bender, 1992), the history as well as particulars of conversions of vitamins to coenzymes and other biologically active forms (Combs, 1998), and sources that provide coverage of each vitamin–coenzyme group (McCormick *et al.* 1997b; Rucker *et al.* 2001). There are also periodic symposia resulting in volumes on the updated findings for most of the cofactor groups. This too is the case with ongoing volumes of ‘*Metal ions in biological systems*’, first appearing 28 years ago (Sigel, 1974). Again symposia and conferences on specific metal ion cofactors add to the published detail available to the professional.

Given the invitation to write ‘a reflective personal review’, the author has taken the opportunity to attempt to condense most of his scientific perambulations on micronutrient research into a synoptic treatise. The satisfaction derived from discovery has been enhanced by applications that serve more than to catalogue basic knowledge, so such extensions are also briefly covered in the present review. A more specific acknowledgement of the help provided by mentors, trainees, and other colleagues in a somewhat wider venue of biochemical and nutritional research has been published (McCormick, 2000), so this account will be a more circumscribed focus on findings and applications relating to those micronutrient cofactors that have received our specific attention.

Uptake, transport, and storage of water-soluble vitamins

Although there are differences in the ways that cells take in and route the structurally dissimilar water-soluble vitamins, findings from a number of laboratories indicate commonality of relative specificity and facilitated uptake at physiological concentrations, with passive diffusion becoming predominant at pharmacological levels. Circulatory, less-specific binding proteins help vector the vitamins to organ sites in mammals, which also have pregnancy-induced carriers in some cases. The latter have some parallel in the specific, tight-binding proteins that permit storage of such vitamins as riboflavin and biotin in the eggs of birds. Our studies in this subject area are summarized in the following two sections.

Uptake

The entry of pyridoxine into liver cells is insensitive to Na^+ (and hence Na^+/K^+ ATPase) and dependent on metabolic trapping by pyridoxal kinase (Kozik & McCormick, 1984), whereas uptake by renal proximal tubular cells is similar but may involve Na^+/H^+ exchange and/or pH gradient effects (Bowman & McCormick, 1987, 1989; McCormick, 1989). Disposition of B_6 glucosides was shown to depend upon uptake as well as subsequent metabolic events (Joseph *et al.* 1996; Zhang *et al.* 1993a). Hepatocyte uptake of riboflavin, which is carrier-mediated but not Na^+ -dependent and involves flavokinase-catalysed phosphorylation (Aw *et al.* 1983), has been contrasted with gut (enterocyte) absorption and with uptake by proximal tubular renal cells (Bowman *et al.* 1989). Biotin entry depends on ligandin (glutathione S-transferase) as typical for organic acid anions (Bowers-Komro & McCormick, 1985b).

Transport and storage

Both physical (Pritchard *et al.* 1967) and biological (Lee *et al.* 1973*b*) interactions of biotin with avidin were investigated, as was specificity of avidin (Zempleni *et al.* 1996*e*) and biocytinase (McCormick, 1969). We have delineated properties of the avian carrier and storage proteins for riboflavin (Froehlich *et al.* 1980) including its flavin-binding specificity (Choi & McCormick, 1980), were the first to recognize the pregnancy-induced riboflavin-carrier protein in a mammal (Merrill *et al.* 1979*a*), and elaborated on the occurrence of other cytosolic binding proteins for this vitamin (Merrill *et al.* 1982). Further work has led to the identification of immunoglobulin carriers in man (Merrill *et al.* 1981*a*; Innis *et al.* 1985, 1986).

Vitamin metabolism

Some of the more recent and interesting advances in research on water-soluble vitamins has to do with their biosyntheses. In the past few years, the missing intermediates and enzymes in the pathways for formation of pyridoxine, riboflavin, and biotin have been discovered. Biosynthesis of B₆ in *Escherichia coli* K-12 has been shown to involve condensation of 4-phospho-hydroxy-L-threonine with 1-deoxy-D-xylulose 5-phosphate to form pyridoxine 5'-phosphate (Winkler, 2000). For riboflavin (Bacher *et al.* 2000) the beginning precursor is GTP, which, after four steps, is converted to 5-amino-6-ribitylamino-2,4(1H,3H)-pyrimidinedione. This latter reacts with 3,4-dihydroxybutanone 4-phosphate to form 6,7-dimethyl-8-ribityllumazine, which then dismutates into riboflavin. Enzymes responsible for catalysing the half-dozen steps, including the synthases for lumazine and riboflavin, have been isolated and characterized. In the case of biotin, the terminal metal ion-containing synthases that operate to form dethiobiotin (Schneider & Lindqvist, 1997) and biotin (Flint & Allen, 1997) have been examined. Other studies have centred on the conversions of such vitamins to their coenzymic operating forms and their eventual catabolism. The following four sections outline some of our findings.

Vitamin B₆

Pyridoxal (pyridoxine, pyridoxamine) kinase. Isolation of and comparative studies on both pro- and eucaryotic forms of pyridoxal kinase allowed us to delineate general properties, including the first substantiated role of Zn²⁺ in preference to Mg²⁺ as the cosubstrate ATP complex for the mammalian phosphokinase (McCormick *et al.* 1961), and led to circumscription of inhibitory aspects (McCormick & Snell, 1961), including the potent action of carbonyl reagents (McCormick & Snell, 1959; McCormick *et al.* 1960) and such drugs are known to bind to the kinase (McCormick & Chen, 1999).

Pyridoxine (pyridoxamine) 5'-phosphate oxidase. We succeeded in the first complete purification of pyridoxine (pyridoxamine) 5'-phosphate oxidase, the FMN-dependent enzyme responsible for conversion of the kinase-derived phosphovitamin B₆ to coenzymic pyridoxal 5'-phosphate (Kazarinoff & McCormick, 1975). More facile affinity purification (Bowers-Komro *et al.* 1986) and assays (DePecol & McCormick, 1980) were developed and structural requirements for substrate (Kazarinoff & McCormick, 1973, 1975; DePecol & McCormick, 1980; Merrill *et al.* 1980; Bowers-Komro & McCormick, 1987) and coenzyme specificities (Kazarinoff & McCormick,

1974; Merrill *et al.* 1979b) accomplished. The oxidase requires the 5'-phosphate for substrate but is fairly tolerant of substitutions on the 4-aminomethyl function. Systematic elucidation of the dimeric subunit association (Horiike *et al.* 1979a; Tsuge & McCormick, 1980), active-site amino acid residues (McCormick *et al.* 1976; Horiike *et al.* 1979b; Tsuge & McCormick, 1980; Choi & McCormick, 1981; Bowers-Komro *et al.* 1986), kinetics of which are somewhat different for the two natural substrates (Choi *et al.* 1982, 1983), and ultimately mechanistic delineation of stereochemical aspects (Bowers-Komro & McCormick, 1984b, 1985a; McCormick & Bowers-Komro, 1986) have provided definitive information on the way this essential flavoprotein operates in the ionic abstraction of a substrate hydrogen (Bowers-Komro & McCormick, 1984a), which depends upon the flavin status of an organism (Rasmussen *et al.* 1979, 1980), and participates in the regulation of B₆ metabolism (Merrill *et al.* 1978b; McCormick & Merrill, 1980). The sequences for this essential oxidase from several organisms have been determined (McCormick & Chen, 1999). An important interface between vitamins B₂ and B₆ is now clear.

The scheme given in Fig. 1 outlines the sequential roles of kinase and oxidase in the interconversions of B₆ vitamers toward the coenzyme pyridoxal 5'-phosphate.

Riboflavin

Flavokinase. The enzyme responsible for catalysing phosphorylation of riboflavin to form riboflavin 5'-phosphate (FMN) in mammalian tissue and elsewhere was shown by us to be flavokinase, which is another Zn²⁺-preferring enzyme (McCormick, 1962; Merrill & McCormick, 1980; Nakano & McCormick, 1991b). Its enrichment from liver was accomplished by classic techniques (McCormick, 1962) and then purification done with affinity chromatography (Arsenis & McCormick, 1964a; Merrill & McCormick, 1980; Nakano & McCormick, 1991a). Detailed studies on the specificity of this enzyme (McCormick & Butler, 1962; McCormick *et al.* 1963, 1964; Yang *et al.* 1964; Chassy *et al.* 1965) helped clarify the biological activities of

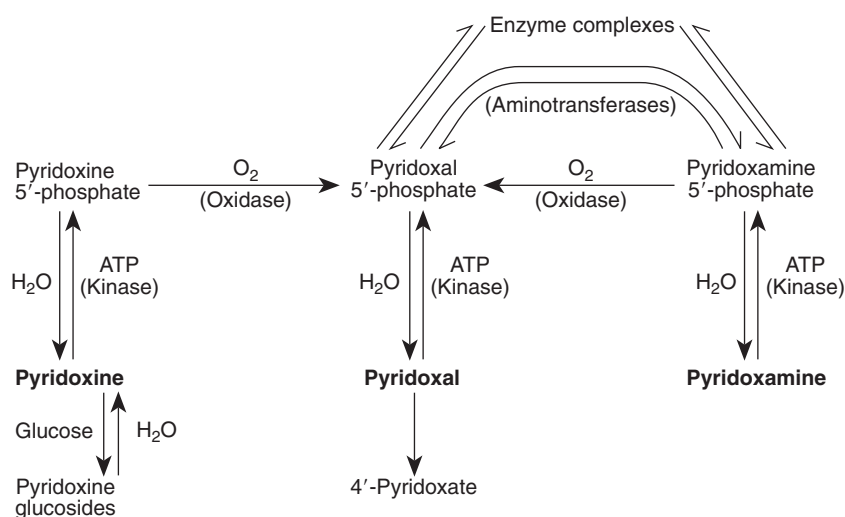


Fig. 1. Vitamin B₆ metabolism. Reactions shown occur in numerous organisms and organs, especially the liver. For the disposition of ingested forms of B₆ with an indication of organ interplay, see McCormick (2001). Both kinase and oxidase are cytosolic.

flavin analogues. Investigations of substrate induction (Merrill *et al.* 1978*a*; Lee & McCormick, 1983) and thyroid hormone stimulation (McCormick *et al.* 1984; Lee & McCormick, 1985) led to recognition of the 'active' and 'inactive' forms, which are poised at the regulation site for the biosynthesis of the flavocoenzymes FMN and FAD.

FAD synthetase. We elaborated the substrate specificity of mammalian FAD synthetase (McCormick, 1964*a,b*; Bowers-Komro *et al.* 1989; McCormick *et al.* 1997*a*) and accomplished its partial (Gomes & McCormick, 1983) and then complete purification using FMN-agarose (Kazarinoff *et al.* 1975; Oka & McCormick, 1987). Further work on this Mg^{2+} -preferring enzyme led to more detailed characterization of the cooperatively interactive kinase-synthetase system and to the kinetic order as regards substrate addition and product removal (Yamada *et al.* 1990).

FMN phosphatase and FAD pyrophosphatase. The interfering, non-specific actions of acid and alkaline phosphatases (McCormick, 1961; McCormick & Russell, 1962) and FAD pyrophosphatase have been separated and generally characterized as degradative hydrolases responsible for breakdown of flavocoenzymes (Lee & McCormick, 1983).

Riboflavin side-chain oxidases. A bacterial ribityl side-chain oxidizing enzyme that had been called a 'hydrolase' was found by us to have relative specificity (Yang & McCormick, 1967*a*), whereas another enzyme narrowly specific for riboflavin (Kekelidze *et al.* 1994, 1995) has been molecularly cloned and sequenced from a fungal organism (Chen & McCormick, 1997*a*) and found to form both aldehyde and acid 'schizoflavin' products at the 5'-terminus (Chen & McCormick, 1997*b*).

Flavin metabolites and analogues. We have helped detail the overall metabolic fate of riboflavin (Foley *et al.* 1967; Yang & McCormick, 1967*b*; McCormick, 1975*b*, 1976*a*; McCormick *et al.* 1984, 1988; Oka & McCormick, 1985; Chastain & McCormick, 1987*a, b*, 1988; Roughead & McCormick, 1991), 8 α -amino acid flavins derived from covalent forms (Addison & McCormick, 1978; Chia *et al.* 1978), and flavin analogues (Ogunmodede & McCormick, 1966; Tu & McCormick, 1969) in the mammal and in milk from cows (Roughead & McCormick, 1990*a*) and in human milk (Roughead & McCormick, 1990*b*). The finding that an 8 α -sulfonyl-riboflavin appears in human urine as a result of catabolic turnover of monoamine oxidase is noteworthy (Chastain & McCormick, 1987*b*). The predominant catabolite of riboflavin to appear in blood plasma following ingestion of the vitamin is the 7 α -hydroxy compound (Zempleni *et al.* 1996*a,b*). The *in vivo* kinetics of riboflavin absorption and disposition have been quantified in normal human subjects (Zempleni *et al.* 1996*a*) and in women with liver cirrhosis (Zempleni *et al.* 1996*c*).

The scheme given in Fig. 2 outlines central aspects of riboflavin transport, metabolism, utilization, and excretion.

Biotin

Biosynthesis. Proof that biotin is formed directly from dethiobiotin was accomplished by our use of the radiolabelled precursor of the vitamin (Tepper *et al.* 1966; Li *et al.* 1968*a*).

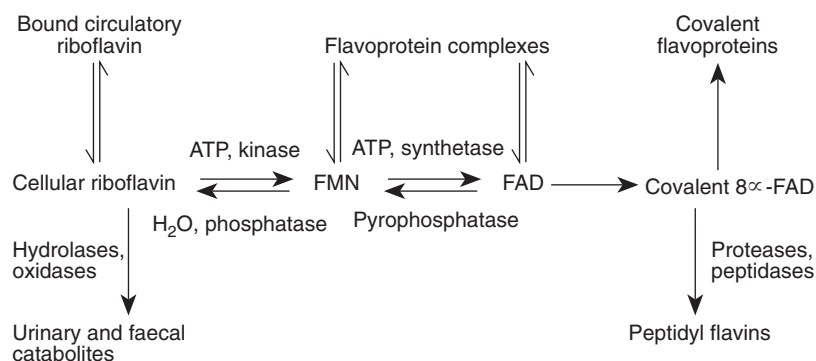


Fig. 2. Riboflavin metabolism. Biosynthesis and turnover of flavocoenzymes occur in most cells, with kinase and synthetase located in the cytosol but degradative enzymes included within organelles.

Catabolism. The fate of biotin and some of its analogues when wholly degraded in a pseudomonad (Brady *et al.* 1965, 1966; Ruis *et al.* 1968; Iwahara *et al.* 1969; Im *et al.* 1970, 1973; Roth *et al.* 1970; Kazarinoff *et al.* 1972; Westendorf & McCormick, 1980) and partly degraded in a fungus (Li *et al.* 1968*b*) and the rat (Lee *et al.* 1972, 1973*a*) was elaborated in our laboratory. Also a discriminating colorimetric reaction for biotin and analogues was developed (McCormick & Roth, 1970). Present knowledge of the metabolism of biotin is based on these detailed studies (McCormick & Wright, 1970; McCormick, 1975*a*, 1976*b*; McCormick & Olson, 1984). Whereas a soil pseudomonad forced to use biotin as the sole source of C, N, S and energy can effect extensive degradation of the vitamin, including the bicyclic ring system, mammals, including man (Zempleni *et al.* 1996*d*), operate more sparingly, mainly on side-chain β -oxidation and oxidation of the ring S.

Based on the numerous catabolites we have isolated and structurally identified and the known function of ϵ -lysyl-linked biotin in mammalian carboxylases, an overview of events is summarized in Fig. 3.

Lipoate

Similar studies have been conducted on the catabolic fate of lipoate, an essential cofactor that is covalently linked to the ϵ -amino functions of specific lysyl residues of transacylases. Lipoate is a vitamin for some microbes but not the mammal, which can biosynthesize it from the level of octanoate. We first detailed total degradation of lipoate in a pseudomonad isolated by enrich-

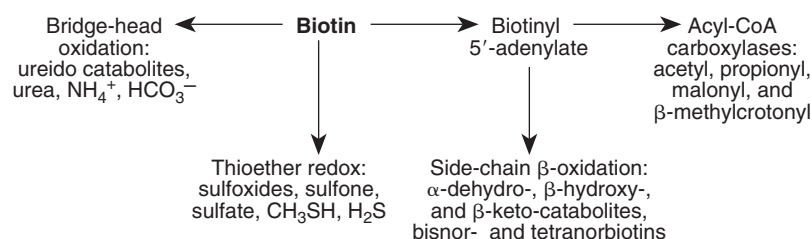


Fig. 3. Biotin metabolism. Though side-chain β -oxidation and S oxidation occur in bacteria and mammals, extensive degradation of the bicyclic ring system is known to occur only in certain bacteria.

ment culture from soil (Shih *et al.* 1972, 1975; Chang *et al.* 1975; Furr *et al.* 1978; Furr & McCormick, 1978), and then in the rat (Harrison & McCormick, 1974; Spence & McCormick, 1976). Syntheses and delineation of the properties of critical side-chain-shortened catabolites, for example bisnor- and tetranorlipoates, were also accomplished (Shih *et al.* 1974) as were HPLC chromatographic separations of metabolites (Howard & McCormick, 1981).

The routes for function and catabolism of lipoate are shown in Fig. 4.

Flavocoenzyme function

To better understand the nature of flavocoenzymes, namely FMN and both non-covalently bound FAD and the less frequent but critical 8α -linked FAD cases, we have examined complexes of free flavins and those that were associated with those specific proteins that serve as binding proteins or flavin enzymes. Some findings are mentioned in the next two sections.

Inter- and intramolecular flavin complexes

Our studies on the nature of inter- and intramolecular complexes of flavins with purines and pyrimidines (Chassy & McCormick, 1965a; Tsibris *et al.* 1965; Roth & McCormick, 1967; McCormick, 1968a) including synthetic analogues of FAD helped elucidate the strength and types of interactions involved, particularly with FAD wherein the adenine moiety quenches the fluorescence of the isoalloxazine ring. Extension of such studies to flavin–aromatic amino acid systems (Föry *et al.* 1968, 1970a,b; MacKenzie *et al.* 1969; McCormick, 1970, 1977b; Wu & McCormick, 1971a,b; Johnson & McCormick, 1973; Johnson *et al.* 1975; McCormick *et al.* 1975; Getoff *et al.* 1978) and ultimately to flavoproteins (McCormick, 1970, 1977a,b; McCormick & Tu, 1970; Wu *et al.* 1970; Tu & McCormick, 1973, 1974; Shiga *et al.* 1975; Merrill *et al.* 1981b) secured the expectation that such interactions are common, particularly with tryptophanyl and tyrosyl residues, and often account for part of the facilitated binding of flavins to proteins.

Flavin-dependent enzymes

The specificity of coenzyme binding and function (Arsenis & McCormick, 1964b; McCormick *et al.* 1964; Chassy & McCormick, 1965b; Roth *et al.* 1966; Tsibris *et al.* 1966; Merrill *et al.* 1979b; Visser *et al.* 1968), nature of active-site residues (McCormick *et al.* 1967; Koster *et al.* 1968; McCormick, 1970; Wu *et al.* 1970; Tu & McCormick, 1973, 1974; Falk *et al.* 1976; Falk

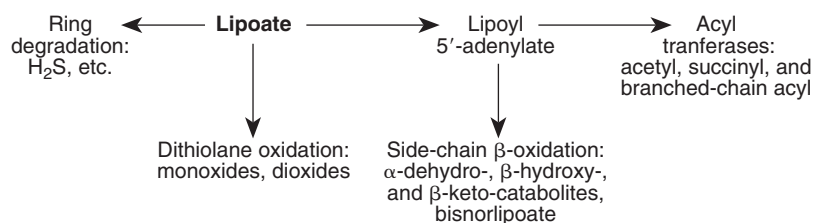


Fig. 4. Lipoate metabolism. More extensive degradation of the dithiolane ring occurs in bacteria utilizing lipoate as a sole source of S and C than in mammals.

& McCormick, 1976; Horiike *et al.* 1979b; Choi & McCormick, 1981) and physical properties of several flavin-dependent enzymes have been elucidated. One common feature is the binding of the pyrimidinoid portion of the isoalloxazine system of FMN within a cleft, which often allows projection of the dimethylbenzenoid edge toward solvent. A prototypic example of using a coenzyme as a photochemical probe for the active site of an enzyme was provided by our work with FAD in D-amino acid oxidase, wherein a tyrosyl as well as lysyl and cysteinyl residues were proven critical (Tu & McCormick, 1973). The stereochemical effects of 8 α -flavin linkage to a cysteinyl residue in monoamine oxidase were confirmed and quantified by our synthesis of the active-site portion of this enzyme (Falk *et al.* 1976; Falk & McCormick, 1976).

Metal ions

We have shown the involvement of K⁺ as an activator and Zn²⁺ chelated with ATP as cosubstrate with pyridoxal kinase (McCormick *et al.* 1961). Also the preference for Zn²⁺ was established for flavokinase (McCormick, 1962; Merrill & McCormick, 1980) and bacterial dihydro-orotase (Sander *et al.* 1965). The metal ion liganding properties of several important functional groups (Griesser *et al.* 1971) including amino acids (Griesser *et al.* 1969; McCormick *et al.* 1969, 1974; Sigel *et al.* 1969a,b, 1970, 1972, 1977; Sigel & McCormick, 1971, 1974; Walker *et al.* 1972), nucleotides (Sigel *et al.* 1967; Sigel & McCormick, 1974), and such vitamins as biotin (Sigel *et al.* 1969c, 1978a; Griesser *et al.* 1970, 1973; Sigel & McCormick, 1974) and lipoate (Sigel *et al.* 1978a,b) have been delineated for important cations of the Irving-Williams series. These latter studies extend our knowledge of the possible interactions within biological metal ion-containing systems.

Derived applications

In biomedical sciences, most investigations are based (and funded) on the expectation that findings will not only shed light on questions of a basic nature but also eventually lead to applications that will directly or indirectly improve the lot of mankind. Over the span of our research, several extensions of the basic findings have led to useful applications. Though few such discoveries nowadays are the result of only one individual or group, at least a significant and literature-documented role has been played by my colleagues and me in the following.

Biochemically specific ('affinity') absorbents

Our numerous studies on the specificity of biopolymers (enzymes, binding-proteins, and polynucleotides) led me to realize nearly a half century ago that one could take advantage of the selective binding of a biopolymer to its substrate, cofactor, or complementary polynucleotide by synthesizing biochemically specific absorbents. A start in this direction was my synthesis in 1959 of the one-methylene-extended homologue of pyridoxal with the intent of coupling this to some matrix for the column purification of pyridoxal kinase. This was elaborated, well before the current term of 'affinity chromatography', with several examples that first used derivatized cellulose, for example liver flavokinase on flavin-cellulose (Arsenis & McCormick, 1964a), avidin on biotinyl-cellulose (McCormick, 1965), some FMN-dependent enzymes on FMN-cellulose (Arsenis & McCormick, 1966), and polyA nucleotides on thymidy-

late-cellulose (Sander *et al.* 1966). As Sephadex gained widespread use in enzymology, we turned to this more manageable material for a matrix, often with linker arms to make an 'affinose'. Examples include riboflavin-binding proteins on flavinyl-affinose (Merrill & McCormick, 1978), pregnancy-specific riboflavin-binding protein on flavinyl-affinose (Merrill *et al.* 1979a), liver flavokinase on flavinyl-affinose (Merrill & McCormick, 1980), pyridoxine (pyridoxamine) 5'-phosphate oxidase on 5'-phosphopyridoxyl-affinose (Bowers-Komro *et al.* 1986), FAD synthetase on FMN-affinose (Oka & McCormick, 1987; McCormick *et al.* 1997a,b), B₆-binding proteins on pyridoxyl-affinose (McCormick *et al.* 1991), and brain flavokinase on flavinyl-affinose (Nakano & McCormick, 1991a). The generalized scheme for formation and use of affinity absorbents is illustrated in Fig. 5.

We also have immobilized enzymes as a means for flow-through catalysis. Examples are with D-amino oxidase (Tu & McCormick, 1972) and flavokinase (Merrill & McCormick, 1979).

Drug delivery

Information from our studies on the specificity of transporters for the vitamins coupled with our knowledge of the enzymic events that occur upon their entry into cells led to the design of vitamin analogue models that exemplify transporter-enhanced delivery of bioactive compounds. A specific case elaborated is the chemical attachment of bioactive amines to vitamin B₆ so as to be 'piggybacked' through the B₆ transporter and metabolically released inside kidney or liver cells as free amine plus the coenzyme pyridoxal 5'-phosphate (Zhang & McCormick, 1991, 1992a,b; Zhang *et al.* 1993b; McCormick, 1994). A means by which some less-transportable compounds of therapeutic use can be imported into cells is exemplified in Fig. 6.

Biopolymer modifications

We established the theoretical bases, including means for calculating number of residues modified and ways to plot correctly such data, in part, to clarify the erroneous manner in which such data had sometimes been published by others. This work (Horiike & McCormick, 1979, 1980) can be extended to experimental protocols for chemical (McCormick, 1970; Horiike *et al.*

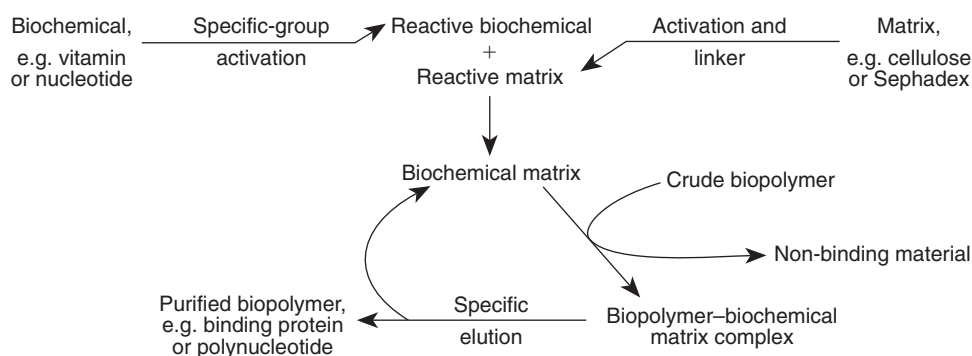


Fig. 5. Biochemically specific ('affinity') absorbents. By attaching a compound that specifically binds a biopolymer to an insoluble matrix in such a manner that allows separation from extraneous materials followed by elution of the biopolymer, purification of the latter can be achieved.

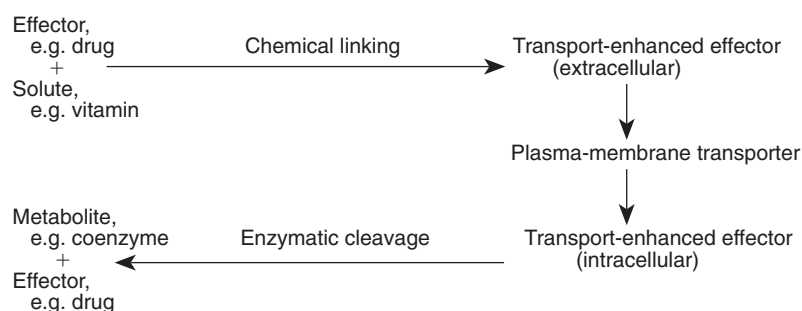


Fig. 6. Transporter-enhanced delivery of bioactive compounds. By attaching a less transportable compound, such as a poorly soluble drug, to a compound that gains facilitated entry and metabolic release within cells, transport of the less transportable compound into such cells can be enhanced.

1979*a,b*; Tsuge & McCormick, 1980; Choi & McCormick, 1981; Nakano *et al.* 1992; Nakano & McCormick, 1992) and photochemical (McCormick *et al.* 1967; Koster *et al.* 1968; McCormick, 1968*b*, 1970; Tu & McCormick, 1973) modifications of enzymes or, for that matter, any biopolymer.

Pathogen photoinactivation

A technique that is being developed for producing pathogen-free blood components is to take advantage of the photochemistry that is obtained with riboflavin and the fact that erythrocytes, platelets and plasma contain no nucleic acid. When this vitamin is irradiated with light at wavelengths that are in the visible range (and hence is not absorbed by simple proteins), the flavin excited state can interact with nucleic acids contained within viral and bacterial pathogens to effect photo-oxidation, especially of guanine residues, to cause killing of such organisms. The side-chain photoproducts of riboflavin that result are mainly those that we have shown are excreted in urine (Chastain & McCormick, 1987*a,b*, 1988) and are secreted in milk from cows (Roughead & McCormick, 1990*a*) and human subjects (Roughead & McCormick, 1990*b*). Fig. 7 summarizes the process of photoinactivation using riboflavin.

Dietary recommendations

Our quantification of vitamins and their metabolites in the urine of human subjects has been helpful in assessment of losses that should be replaced in the diet. This has been the case for riboflavin (Chastain & McCormick, 1987*b*, 1988) and biotin (Zempleni *et al.* 1996*e*). Also knowledge of the secretion of flavins in human milk (Roughead & McCormick, 1990*b*) aids in the estimation of an adequate allowance of riboflavin for infants, just as knowledge of flavin content of cows' milk allows us to estimate the contribution of this important source to our recommended dietary allowance such as periodically published in the US (Food and Nutrition Board and Institute of Medicine, 1998) with counterparts in other countries.

There has been good progress in unravelling the nature of micronutrient cofactors up to the present. Undoubtedly more can and should be learned in future research by others.

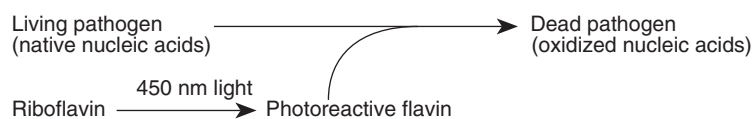


Fig. 7. Pathogen eradication by photoactivation of riboflavin. Because riboflavin can absorb light in the visible region to become a photoreactive oxidant for such bases as guanine within nucleic acids of bacteria and viruses, it can lead to destruction of pathogens within cells and fluids that have no nucleic acids.

Acknowledgements

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