

**An Integrated Hydrogen Production-CO<sub>2</sub> Capture Process from Fossil Fuel**

**Semiannual Progress Report**

**Reporting Period Start Date: March 16, 2004**

**Report Period End Date: August 31, 2004**

**Principal Authors: Z. Wang (PI) and K. B. Bota**  
**Clark Atlanta University, Atlanta, GA 30314**  
**Email: zwang@cau.edu Phone: 404-880-6934**

**D. Day**  
**Scientific Carbons Inc, Blakely, GA**

**Date Report Was Issued: September 2004**

**DOE Award Number: DE-FG26-03NT41914**

**Submitted by: Clark Atlanta University, Atlanta, GA 30314**  
**Eprida Scientific Carbons Inc., Atlanta, GA**

## **DISCLAIMER**

**DISCLAIMER OF LIABILITY:** This report was prepared as accounts of work sponsored by an agency of the U.S. Government. Neither the U.S. Government nor any agency thereof, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights.

**DISCLAIMER OF ENDORSEMENT:** Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the U.S. Government or any agency thereof. The views and opinions of document authors herein do not necessarily state or reflect those of the U.S. Government or any state government or any agency thereof.

## **ABSTRACT**

The major project objective is to determine the feasibility of using the char from coal and/or biomass pyrolysis, ammonia and CO<sub>2</sub> emissions at smokestacks to produce clean hydrogen and a sequestered carbon fertilizer. During this work period, literature review has been completed. The project plan, design and test schedules were made on the basis of discussion with partner in experimental issues. Installation of pilot scale units was finished and major units tests were fully performed. Modification of the pyrolyzer, reformer and gas absorption tank have been done. Integration testing is performing recently. Lab scale tests are in operation phase. The experimental installations are discussed in this paper.

## TABLE OF CONTENTS

DECLAMER .....	2
ABSTRACT .....	3
INTRODUCTION AND OBJECTIVES.....	5
EXPERIMENTAL FACILITIES.....	5
RESULTS AND DISCUSSION.....	10
FUTURE PLANS.....	10
SUMMARY AND CONCLUSIONS.....	10
REFERENCES.....	11

## **INTRODUCTION AND OBJECTIVES**

The increasing anthropogenic CO<sub>2</sub> emissions and possible global warming have challenged the United States and other countries to find new and better ways to meet the world's increasing need for energy while, at the same time, reducing greenhouse gas emissions. The improved technology for integrated hydrogen production/CO<sub>2</sub> capture that we plan to develop through this R&D effort could significantly support President Bush's Global Climate Change Initiative (GCCCI) that commits America to an aggressive strategy to reduce greenhouse gas intensity by 18 percent over the next 10 years. Our new technology concept integrates two significant and complementary hydrogen production and CO<sub>2</sub>-sequestration approaches that have now been developed at Oak Ridge National Laboratory (ORNL) and Clark Atlanta University.

The objective of the proposed study is to determine the feasibility of using the char from a coal and/or biomass pyrolysis-reforming process and CO<sub>2</sub> emissions at a smokestack to form solid NH<sub>4</sub>HCO<sub>3</sub>-char and/or (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub>-char products that may subsequently be used as a fertilizer. Part of the hydrogen from the pyrolysis-reforming process may be converted to ammonia that is used to solidify the CO<sub>2</sub> as NH<sub>4</sub>HCO<sub>3</sub> and/or (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> in the char. The balance of the hydrogen may be purified and sold at market prices or used as a feedstock.

## **EXPERIMENTAL FACILITIES**

This project plans to conduct bench scale experiments and pilot scale tests. The Combustion and Emission Lab at Clark Atlanta University has prepared to conduct the

bench scale experiments. The facility used for pilot scale tests has built in Athens, GA. The following photos show the major experimental equipment and facilities.

Figure 1 is the LINDBERG/BLUE 1.3kw tube furnace and its control consoles, which are used for the production of char material in bench scale tests. Figure 2 is an on-line Fourier Transform Infrared Spectroscopy/Gas Chromatography/Mass Spectrometry (FTIR/GS/MS) system, which are employed to characterize chemicals. Figure 3 shows part of the experimental desk of char analysis and  $\text{NH}_3\text{-CO}_2$  solidification for the bench scale experiments.

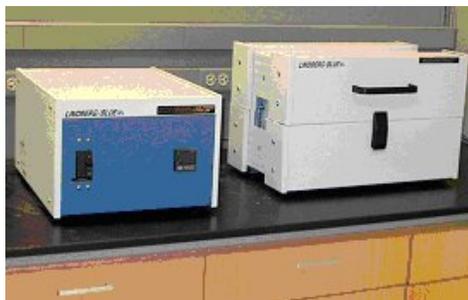


Figure 1 1.3 kW Tube Furnace/control consoles system

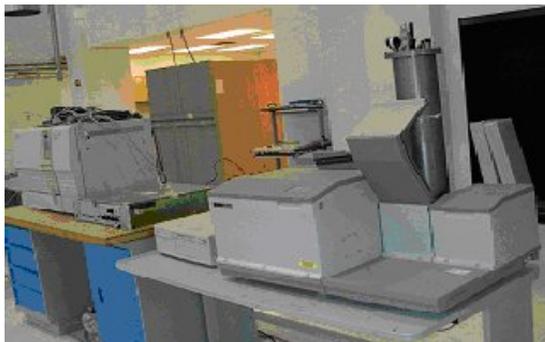


Figure 2 FTIR/GC/MS system



Figure 3 Fume hood and setup for bench scale tests

An overall view of the pilot scale experimental facility was shown in Figure 4. It is the pilot scale test facility. We have finished the installation and passed the tests parts by parts. The overall operation of whole system is running recently. Figure 5 and Figure 6 are the pyrolyzer and reformer, respectively. The pyrolyzer-reformer will be used for producing char and synthesis gas (containing mainly  $H_2$ ,  $CO$  and  $CO_2$ )



Figure 4 the overall view of pilot pyrolyzer-reformer system



Figure 5 Pyrolyzer with Biomass feedstock system



Figure 6 Reformer and Vapor Heater

Figure 7 and Figure 8 are the gas analysis equipment. They are used for both bench scale and pilot scale experiments.



Figure7 Micro-GC and the control computer



Figure 8 Gas (CO, CO<sub>2</sub>, NO<sub>x</sub>, SO<sub>x</sub>, etc.) Analyzers

## RESULTS AND DISCUSSION

Some primary experiments have been done. The pyrolyzer has produced some of char material, which planned to use as the sequestration material. Figure 9 is the photo of sample char material. Figure 10 is the result of primary structure of the formation of Ammonium Bicarbonate.

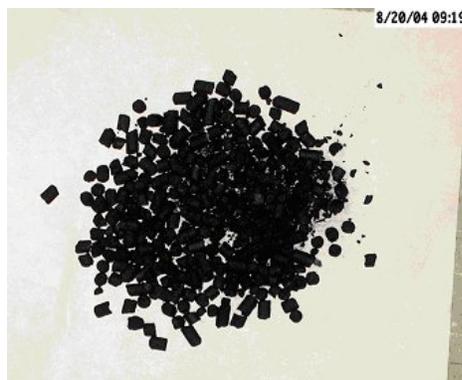


Figure 9 Char material produced by the Pyrolyzer

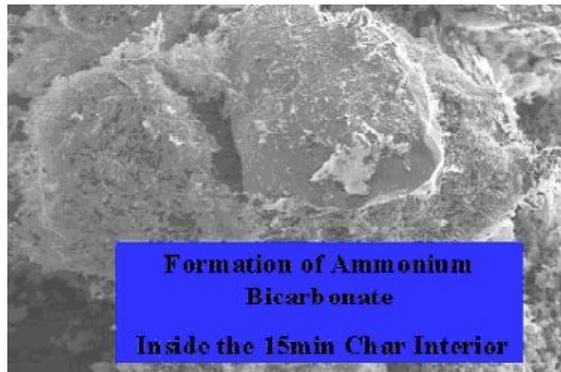


Figure 10 Photo of formation of Ammonium Bicarbonate

## **FUTURE PLANS**

In this report period, we have finished literature reviews, the project planning, installation of the pyrolyzer-reformer, the setup of bench scale experiments, and initiated the characteristic analysis of char products. The future works include the following tasks.

1. Operation of the pyrolyzer-reformer unit
2. Testing, Analysis and characterization of the char and gaseous products
3. Process optimization
4.  $\text{NH}_3\text{-CO}_2$  solidification process
5. Tests to evaluate fertilizer usage

## **CONCLUSIONS**

1. The bench scale equipment is prepared and ready for lab analysis.
2. The pilot scale facilities are installed and finished part of the tests
3. Primary investigation to char characteristics will continue to be conducted.

4. The future works include operation of pyrolyzer-reformer unit,  $\text{NH}_3\text{-CO}_2$  solidification and tests to evaluate fertilizer usage.

## REFERENCES

- Lee, J.W.; Li, R. "A Novel Strategy for  $\text{CO}_2$  Sequestration and Clean Air Protection", Proceedings of Fire National Conference on Carbon Sequestration, Washington, DC, May 14-17, 2001. [http://www.netl.doe.gov/publication/proceedings/01/carbon\\_seq/p12.pdf](http://www.netl.doe.gov/publication/proceedings/01/carbon_seq/p12.pdf)
- Lee, J.W.; Li, R. "Method for Reducing  $\text{CO}_2$ ,  $\text{CO}$ ,  $\text{NO}_x$ , and  $\text{SO}_x$  Emissions", 1998 ORNL Invention Disclosure, ERID 0631; 2002 U.S. Patent No. US 6,447,437 B1.
- Lee, J.W.; Li, R (2002). "Integration of Coal-Fire Energy Systems with  $\text{CO}_2$  Sequestration through  $\text{NH}_4\text{HCO}_3$  Production." *Energy Convers. Manage*, vol. 44, no.9, pg.1451-1464, June 2003.
- Yelverton, F "The use of activated carbon to inactive agriculture chemical spills", North Carolina Cooperative Extension Service, March, 1996. <http://www.bae.ncsu.edu/bae/programs/extension/publicat/wqwm/ag442.html>
- Li, E. Hagaman, C. Tsouris, and J.W. Lee (2002). "Removal of carbon dioxide from flue gas by ammonia carbonation in the gas phase," *Energy & Fuels*, 2003, 17(1), 69-74.
- Miller, D.W., and W.P. Miller, Land application of wastes. P. G217-G241. In M.E. jSummer (ed.) *Handbook of Soil Science*. CRC Press, Boca Raton, FL.
- Miller, D.W., and W.P. Miller, S. Dudka, and M.E. Sumner. 2000, *Characterization of Industrial By-Products*. Soil Sci. Soc. Am., Madison WI.
- Oren, R; Ellsworth, D.; Johnsen, K, et.al. (2001) "Soil fertility limits carbon sequestration by forest ecosystems in a  $\text{CO}_2$ -enriched atmosphere" *Nature* 411, 469-472 (24 May, 2001), Letters to Nature.