



Original Research Article

One-pot and eco-friendly synthesis of silver nanocrystals using *Adiantum raddianum*: Toxicity against mosquito vectors of medical and veterinary importance



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ARTICLE INFO

Article history:

Received 8 March 2016

Received in revised form 29 September 2016

Accepted 13 October 2016

Available online 20 October 2016

Keywords:

Arbovirus

Filariasis

Malaria

Mosquito-borne diseases

Nanotechnology

ABSTRACT

Mosquitoes (Diptera: Culicidae) represent a key threat for millions of humans and animals worldwide, since they act as vectors for devastating parasites and pathogens. Eco-friendly control tools are a priority. Plant-mediated biosynthesis of nanoparticles is rapid and cost-effective. Here we biosynthesized poly-dispersed silver nanocrystals (AgNPs) using a cheap aqueous leaf extract of *Adiantum raddianum*. AgNPs were characterized by UV–vis spectrophotometry, Fourier transform infrared spectroscopy (FTIR), scanning electron microscopy (SEM), transmission electron microscopy (TEM), energy-dispersive spectroscopy (EDX) and X-ray diffraction analysis (XRD). The acute toxicity of *A. raddianum* extract and biosynthesized AgNPs was evaluated against larvae of the malaria vector *Anopheles stephensi*, the dengue vector *Aedes aegypti* and the filariasis vector *Culex quinquefasciatus*. Compared to the leaf aqueous extract, AgNPs showed higher toxicity against *An. stephensi*, *Ae. aegypti* and *Cx. quinquefasciatus* with LC₅₀ values of 10.33, 11.23 and 12.19 µg/ml, respectively. Biosynthesized AgNPs were found safer to non-target organisms *Diplonychus indicus*, *Anisops bouvieri* and *Gambusia affinis*, with respective LC₅₀ values ranging from 517.86 to 635.98 µg/ml. Overall, this study firstly shed light on the potential of *A. raddianum* as a potential bio-resource for rapid, cheap and effective nanosynthesis of novel mosquitocides.

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Introduction

Arthropods represent a key threat for millions of humans and animals worldwide (Benelli et al., 2016a,b; Mehlhorn et al., 2012). *Anopheles* mosquitoes act as malaria vectors in tropical and subtropical areas worldwide, while *Aedes* species are of pivotal importance as dengue, chikungunya and Zika virus vectors (Benelli, 2015a; Benelli and Mehlhorn, 2016). Besides this, large populations of Culicidae can cause irritation and extensive blood loss to livestock. This could lead to reduced productivity and sometimes death. Furthermore, mosquitoes act as vectors of many

different parasites and pathogens that infect livestock and pets, with special reference to horses, dogs and pigs. Good examples are the Eastern Equine Encephalomyelitis virus, the Western Equine Encephalomyelitis virus, the Venezuelan Equine Encephalomyelitis complex and the dog heartworm, *Dirofilaria immitis* (Benelli, 2015a; Mehlhorn et al., 2012). Therefore, Culicidae vector control is a critical requirement in epidemic disease situations, as is an urgent need to develop new and improved mosquito control methods that are economical and effective yet safe for non-target organisms and the environment (Benelli, 2015b; Govindarajan, 2011; Govindarajan et al., 2005).

Nanotechnology is one of the most active areas of research in modern materials science. Nanoparticles exhibit completely new or improved properties based on specific characteristics such as size, distribution and morphology (Benelli, 2016a,b). New

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applications of nanoparticles and nanomaterials are emerging rapidly (Ivanov et al., 2009). Nanocrystalline silver particles have found important applications in the field of high sensitivity biomolecular detection and diagnostics (Kelly et al., 2003), antimicrobials and therapeutics (Savithramma et al., 2011), parasitology and entomology (Benelli, 2016b; Govindarajan et al., 2016a,b,c,d; Veerakumar et al., 2014). However, there is still need for economic, commercially viable as well eco-friendly routes to synthesize effective and stable mosquitocidal silver nanoparticles (Benelli, 2016c).

Eco-friendly nanosynthesis processes include the employ of plant parts (e.g. leaves, bark, rhizomes and roots, see Benelli (2016a) and Muthukumaran et al. (2015) for a review), bacteria and fungi (Saxena et al., 2010). Indeed, chemical synthesis methods lead to presence of some toxic chemical absorbed on the surface that may have adverse effect in the medical applications. Green synthesis provides advancement over chemical and physical method as it is cost effective, environment friendly, easily scaled up for large-scale synthesis and in this method there is no need to use high pressure, energy, temperature and toxic chemicals (Benelli, 2016a,c).

Adiantum raddianum (Adiantaceae) is a plant widely distributed in several Asian countries (Chopra et al., 1956). It is employed in traditional medicine as diuretic, expectorant, emollient, used for coughs, urinary disorders, alopecia and menstrual difficulties (Chopra et al., 1956; Meenakshi et al., 2008). To the best of our

knowledge, the mosquitocidal potential of *A. raddianum* is currently unknown. In this study, we proposed a cheap and rapid method of biosynthesis of AgNPs using the aqueous leaf extract of *A. raddianum*. Bio-reduced AgNPs were characterized by UV–vis spectroscopy, Fourier transform infrared spectroscopy (FTIR), scanning electron microscopy (SEM), transmission electron microscopy (TEM), energy-dispersive spectroscopy (EDX) and X-ray diffraction analysis (XRD). The acute toxicity of *A. raddianum* leaf extract and green-synthesized AgNPs was evaluated against larvae of the malaria vector *Anopheles stephensi*, the dengue, yellow fever and Zika virus vector *Aedes aegypti* and the filariasis vector *Culex quinquefasciatus*. Furthermore, we evaluated the biotoxicity of *A. raddianum* aqueous extract and biosynthesized AgNPs on two non-target aquatic organisms sharing the same ecological niche of mosquito young instars, *Diplonychus indicus* and the biocontrol agent *Gambusia affinis*.

Materials and methods

Materials

Silver nitrate was procured from Merck, India. The glassware was acid-washed thoroughly and then rinsed with Millipore Milli-Q water. Healthy and fresh leaves of *A. raddianum* were collected from Nilgiris, Western Ghats (11° 10' N to 11° 45' N latitude and 76° 14' E to 77° 2' E longitude), Tamil Nadu State, India. The identity was confirmed at the Department of Botany, Annamalai University, Annamalai Nagar, Tamil Nadu. Voucher specimens were numbered (ID: AUDZ-642) and kept in our laboratory and are available upon request.

Preparation of plant leaf extracts

The leaves of *A. raddianum* were dried in the shade and ground to fine powder in an electric grinder. Aqueous extract was prepared by mixing 50 g of dried leaf powder with 500 ml of water (boiled and cooled distilled water) with constant stirring on a magnetic stirrer. The suspension of dried leaf powder in water was left for 3 h and filtered through Whatman n. 1 filter paper and the filtrate were stored in an amber-colored airtight bottle at 10 °C until testing.

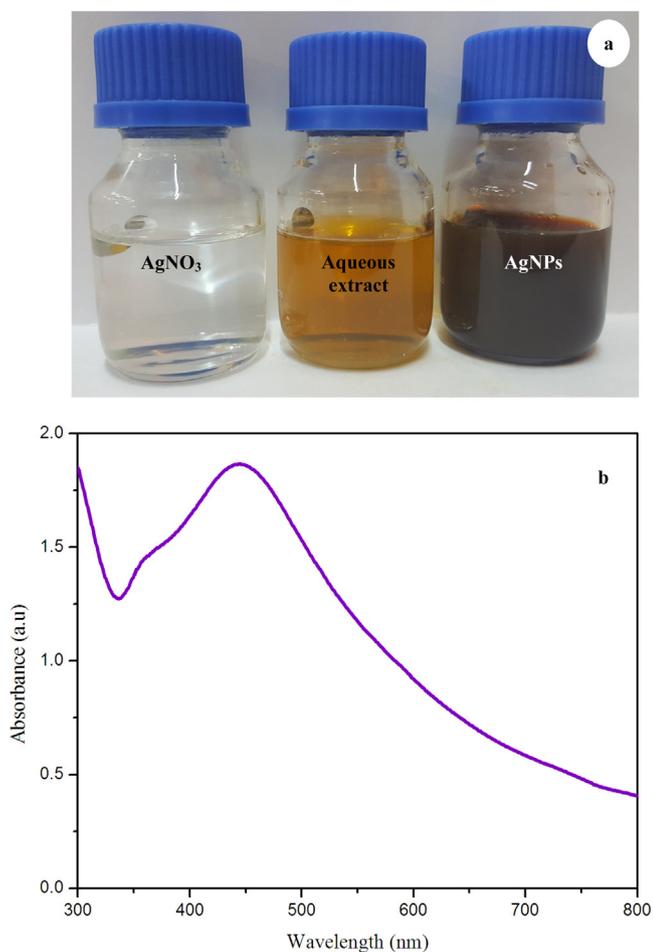


Fig. 1. (a) Color intensity of *Adiantum raddianum* aqueous extract before and after the reduction of silver nitrate (1 mM). The color change indicates Ag^+ reduction to elemental nanosilver. (b) UV–vis spectrum of silver nanoparticles after 180 min from the reaction.

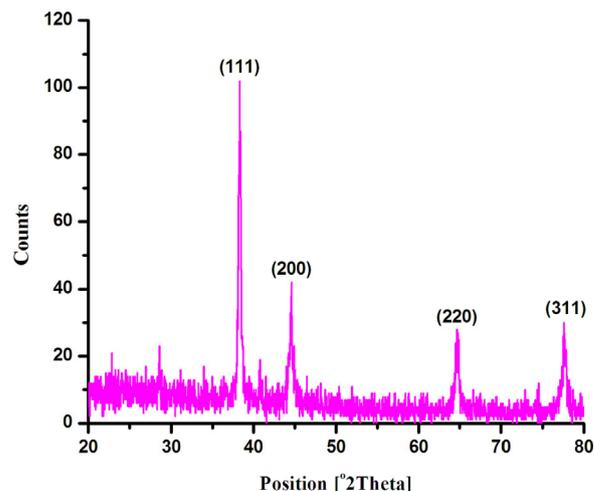


Fig. 2. XRD pattern of silver nanoparticles green synthesized using the *Adiantum raddianum* aqueous extract.

Synthesis of silver nanoparticles

The broth solution of fresh leaves was prepared by taking 10 g of thoroughly washed and finely cut leaves in a 300-ml Erlenmeyer flask along with 100 ml of sterilized double-distilled water and then boiling the mixture for 5 min before finally decanting it. The extract was filtered with Whatman filter paper n. 1, stored at -15°C and tested within a week. The filtrate was treated with aqueous 1 mM AgNO_3 (21.2 mg of AgNO_3 powder in 125 ml Milli-Q water) solution in an Erlenmeyer flask and incubated at room temperature. Eighty-eight milliliters of an aqueous solution of 1 mM silver nitrate was reduced using 12 ml of leaf extract at room temperature for 10 min, resulting in a brown–yellow solution indicating the formation of AgNPs.

Characterization of synthesized silver nanoparticles

The bioreduction of Ag^+ ions was monitored using UV–vis spectrophotometer (UV-160v, Shimadzu, Japan) analysis on size, morphology and composition of AgNPs were performed by scanning electron microscopy (Hitachi S3000 H SEM), transmission electron microscopy (TEM Technite 10 Philips) and energy dispersive X-ray spectrum (EDX). The purified AgNPs were examined for the presence of biomolecules using FT-IR spectrum (Thermo Scientific Nicolet 380 FT-IR Spectrometer) KBr pellets and crystalline AgNPs were determined by XRD analysis.

Mosquito rearing

Laboratory-bred pathogen-free strains of mosquitoes were reared in the vector control laboratory, Department of Zoology, Annamalai University. At the time of adult feeding, these mosquitoes were 3–4 days old after emergences (maintained on raisins and water) and were starved for 12 h before feeding. Each time, 500 mosquitoes per cage were fed on blood using a feeding

unit fitted with Parafilm as membrane for 4 h. *A. aegypti* feeding was done from 12 noon to 4.00 p.m. and *A. stephensi* and *C. quinquefasciatus* were fed during 6.00 p.m. to 10.00 p.m. A membrane feeder with the bottom end fitted with Parafilm was placed with 2 ml of the blood sample (obtained from a slaughterhouse by collecting in a heparinized vial and stored at 4°C) and kept over a netted cage of mosquitoes. The blood was stirred continuously using an automated stirring device, and a constant temperature of 37°C was maintained using a water jacket circulating system. After feeding, the fully engorged females were separated and maintained on raisins. Mosquitoes were held at $28 \pm 2^{\circ}\text{C}$, 70–85% relative humidity, with a photoperiod of 12-h light and 12-h dark.

Acute toxicity against mosquito larvae

Larvicidal activity of the aqueous crude extract and AgNPs from *A. raddianum* was evaluated according to WHO protocol (2005). Based on the wide range and narrow range tests, aqueous crude extract was tested at 70, 140, 210, 280 and $350 \mu\text{m/ml}$ concentrations and Ag NP was tested at 5, 10, 15, 20, and $25 \mu\text{m/ml}$ concentrations. Twenty late III instar larvae were introduced into a 500-ml glass beaker containing 249 ml of dechlorinated water, and 1 ml of desired concentrations of leaf extract or AgNPs was added. For each concentration, five replicates were performed. Larval mortality was recorded at 24 h after exposure, during which no food was given to the larvae. Each test included a set control groups (silver nitrate and distilled water) with five replicates for each individual concentration.

Biotoxicity on non-target organisms

The AgNP toxicity on non-target organisms was assessed following the method by Sivagnaname and Kalyanasundaram (2004). The effect of aqueous extract and AgNPs of the potential

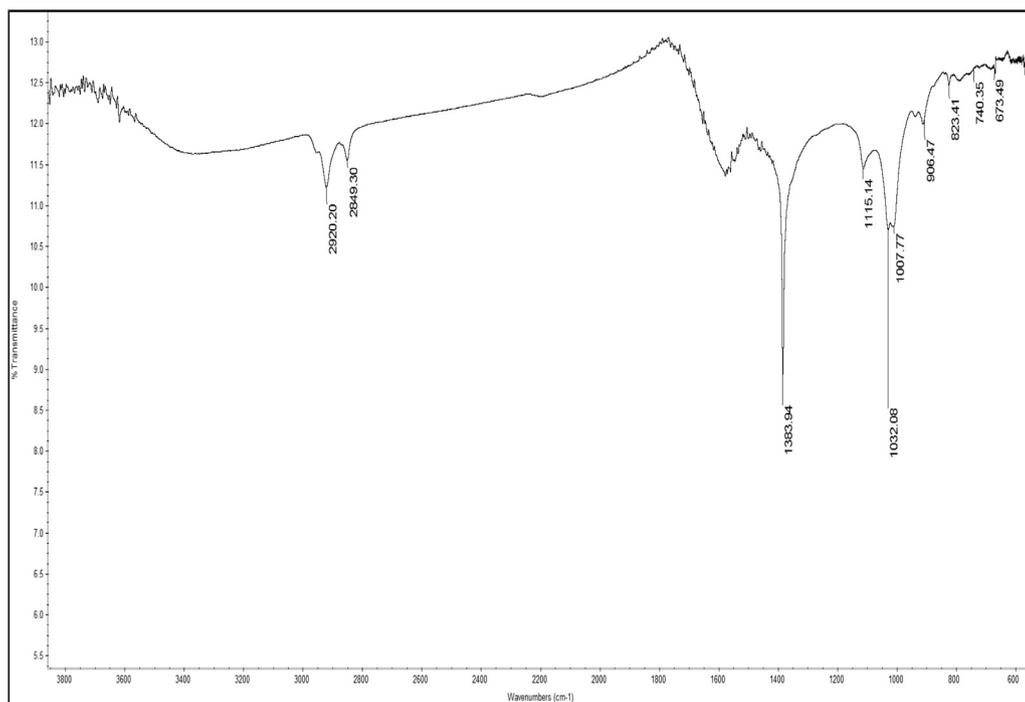


Fig. 3. FTIR spectrum of silver nanoparticles green synthesized using the *Adiantum raddianum* aqueous leaf extract.

plant was tested against non-target organisms *D. indicus* and *G. affinis*. The species were field collected and separately maintained in cement tanks (85 cm diameter and 30 cm depth) containing water at 27 ± 3 °C.

The *A. raddianum* aqueous extract and AgNPs were tested at concentrations of even 50 times higher the doses tested on mosquito larvae. Ten replicates will be performed for each concentration along with four replicates of untreated controls. The non-target organisms were observed for mortality and other abnormalities such as sluggishness and reduced swimming activity after 48 h exposure. The exposed non-target organisms were also observed continuously for ten days to understand the post treatment effect of this extract on survival and swimming activity.

Data analysis

Mortality data were subjected to probit analysis. LC_{50} and LC_{90} were calculated using the method by Finney (1971). In experiments evaluating biotoxicity on non-target organisms, the Suitability Index (SI) was calculated for each non-target species

using the following formula (Deo et al., 1988).

$$SI = \frac{LC_{50} \text{ of non-target organisms}}{LC_{50} \text{ of target vector species}}$$

All data were analyzed using the SPSS Statistical Software Package version 16.0. A probability level of $P < 0.05$ was used for the significance of differences between values.

Results

Biosynthesis and characterization of silver nanoparticles

In our experiments, when the plant extract was added to silver nitrate solution, the formation of AgNPs occurred and color changed rapidly to dark brown (Fig. 1a). AgNPs were characterized using UV–vis spectroscopy, where an intense, broad absorption peak was observed at 447 nm (Fig. 1b). In XRD analysis, four diffraction peaks were observed at 37.67, 43.84, 64.03 and 76.96 represent the (111), (200), (220) and (311) (Fig. 2), while main peaks linked to the presence of reducing and stabilizing biomolecules are shown in the FTIR spectrum (Fig. 3). SEM of AgNPs showed that nanoparticles were mostly spherical or with

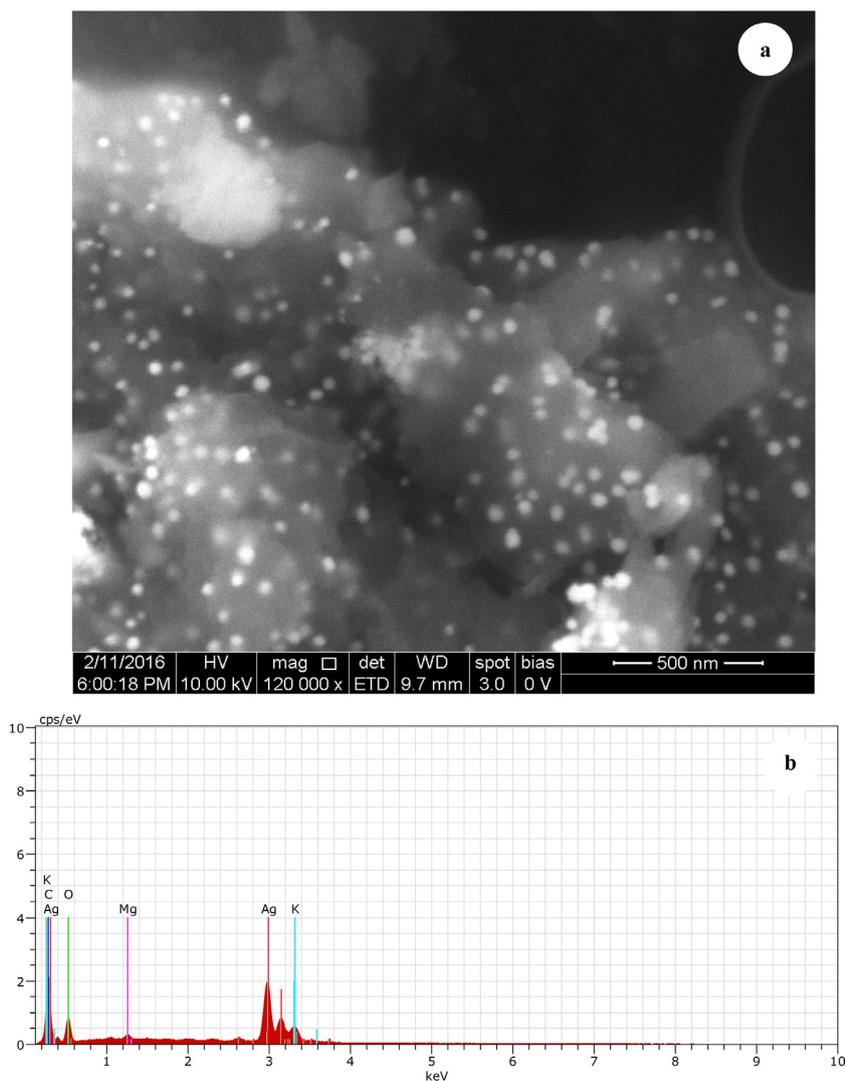


Fig. 4. Silver nanoparticles biofabricated using the *Adiantum raddianum* leaf extract: (a) SEM (1,20,000 \times); (b) EDX spectrum showing the chemical composition.

cubic structures (Fig. 4a). Fig. 4b shows spot EDX analysis, confirming the presence of silver in the sample. Fig. 5 shows the TEM of AgNPs synthesized using *A. raddianum* leaf extract. Among shapes, spheres, truncated triangles, and decahedral morphologies dominate, ranging from 9.69 to 13.9 nm with an average size of 10.9 nm.

Acute toxicity against mosquito larvae

In laboratory conditions, both the *A. raddianum* leaf extract and AgNPs showed dose-dependent larvicidal effect against all tested mosquito species (Tables 1 and 2). Compared to the leaf aqueous extract, green-synthesized AgNPs showed higher toxicity against *A. stephensi*, *A. aegypti*, and *C. quinquefasciatus* with LC₅₀ values of 10.33, 11.23 and 12.19 µg/mL, respectively (Table 2).

Biotoxicity on non-target organisms

Both toxicity treatments achieved negligible toxicity against *D. indicus* and *G. affinis*, with LC₅₀ values ranging from 517.86 to 8430.41 µg/mL (Tables 3 and 4). Focal observations highlighted that longevity and swimming activity of the study species were not altered for a week after testing. SI indicated that *A. raddianum*-fabricated AgNPs were less toxic to the non-target organism tested if compared to the targeted mosquitoes (Table 5).

Discussion

In UV–vis spectroscopy, an intense broad absorption peak was observed at 447 nm after 180 min from the reaction, and it was probably due to surface plasmon resonance (SPR). The SPR peak is very sensitive to the size and shape of the nanoparticles, amount of extract, silver nitrate concentration and the type of biomolecules present in the leaf extract (Govindarajan and Benelli, 2016a,b). Our UV–vis results are in agreement with previous research (Singh et al., 2010; Zargar et al., 2011), where the AgNPs were observed as stable in solution and also showed little aggregation. Besides, the plasmon bands were broadened with an absorption tail in longer wavelengths, and this may be related to the size distribution of nanoparticles (Ahmad et al., 2003).

The XRD pattern confirmed the crystalline nature of green-synthesized silver nanoparticles (Benelli, 2016a). Four diffraction peaks were observed at 38.22, 44.37, 64.54 and 77.47 represent the (111), (200), (220) and (311), reflections and the face-centered cubic structure of metallic silver, respectively. FTIR spectroscopy shed light on the identity of the biomolecules responsible for reduction and efficient stabilization of the AgNPs. The peak at 2920 cm⁻¹ indicates carboxylic acid (Li et al., 2007). 2849 cm⁻¹ may indicate the presence of methylene group (CH₂) next to OH or NH₂ functional group. The band at 1384 cm⁻¹ may corresponds to C–C stretching of aromatic amines. The bands at 1115 cm⁻¹, 1032 cm⁻¹ and 1007 cm⁻¹ might indicate the presence of C–O stretching alcohols, carboxylic acids, esters and ethers. The peak obtained at 823 cm⁻¹ could be due to the bending vibrations of C–H groups of phenyl rings. 673 cm⁻¹ probably corresponds to C–H stretching strong vinyl di-substituted alkenes (Prakash et al., 2013).

SEM and TEM showed that the green-synthesized nanoparticles were mostly spherical, but also sub-triangular and decahedral morphologies were detected, with an average size of 14.96 nm. As a general trend, the shape of plant-synthesized Ag NP was spherical, with exception of some neem-synthesized AgNPs. They are poly-disperse, with spherical or flat, plate-like, morphology, and mean size range of 5–35 nm in size (Shankar et al., 2004). For example, AgNPs fabricated using *Emblica officinalis* were also predominantly spherical with an average size of 16.8 nm ranging from 7.5 to 25 nm

(Ankamwar et al., 2005). Most of the AgNPs was roughly circular in shape with smooth edges. In agreement with these findings, AgNPs from *Annona squamosa* leaf extract were spherical in shape with an average size ranging from 20 to 100 nm (Vivek et al., 2012) while Thirunavokkarasu et al. (2013) reported spherical nanoparticles with size ranging from 8 to 90 nm in *Desmodium gangeticum*. The present TEM analysis also showed that the surfaces of the AgNPs were surrounded by a black thin layer of some material, which might be due to the capping organic constituents of the plant broth, as also highlighted by Rafiuddin (2013).

In laboratory conditions, the *A. raddianum* aqueous leaf extract showed larvicidal activity against *A. stephensi*, *A. aegypti*, and *C. quinquefasciatus*; with LC₅₀ values ranging from 143.22 to 169.94 µg/mL. In latest years, a growing number of evidences have been provided about the larvicidal efficacy of plant-borne larvicides (Veerekumar et al., 2013; Benelli, 2015a,b). Furthermore, the *A. raddianum*-synthesized AgNPs were highly toxic against *A. stephensi*, *A. aegypti*, and *C. quinquefasciatus* larvae, with LC₅₀ ranging from 10.33 to 12.19 µg/mL. In latest years, a growing number of plant-synthesized AgNPs have been studied for their excellent larvicidal activity against important mosquito vectors (Benelli, 2016a,c). Combination of metal nanoparticles with bioactive principles bestows improved efficiency. The present study implicated that the percentage of mosquito larvicidal mortality increased by many folds with the addition of bio-stabilized AgNPs. For instance, Muthukumar et al. (2015) studied the potential of *Gmelina asiatica*-mediated synthesis of AgNPs against *A. stephensi*, *A. aegypti*, and *C. quinquefasciatus*, obtaining higher LC₅₀ values (i.e., 22.44, 25.77, and 27.83 µg/mL, respectively). A further example is the larvicidal activity of *Leucas aspera*-synthesized AgNPs against *A. aegypti* and *A. stephensi*, with LC₅₀ ranging from 13.06 to 25.54 ppm for *A. aegypti*, and from 12.45 ppm to 22.26 ppm for *A. stephensi* (Sivapriyajothi et al., 2014). Similarly, low doses of AgNPs biosynthesized using *Euphorbia hirta* leaf extract have been reported as toxic against *A. stephensi*, with LC₅₀ values ranging from 10.14 ppm (1 instar larvae), to 34.52 ppm (pupae) (Priyadarshini et al., 2012).

The observed larvicidal activity may be attributed to the interaction between silver nanoparticles and the extracellular

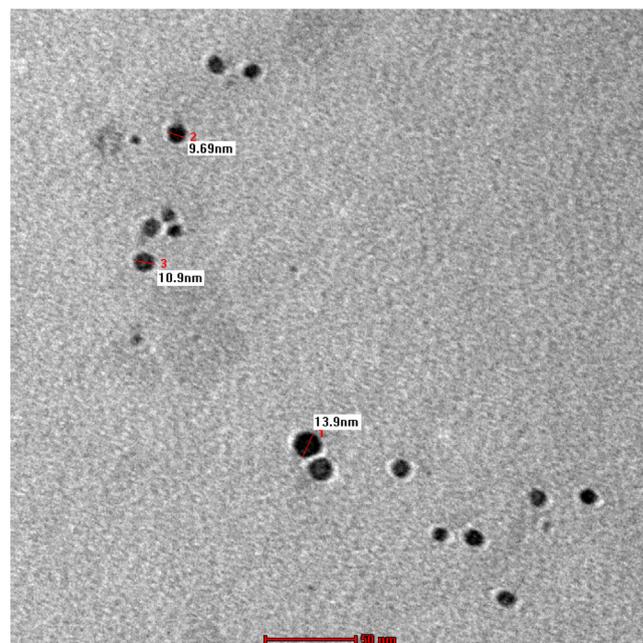


Fig. 5. TEM of silver nanoparticles green synthesized using the *Adiantum raddianum* aqueous extract.

Table 1Larvicidal activity of *Adiantum raddianum* aqueous leaf extract against the mosquito vectors *Anopheles stephensi*, *Aedes aegypti* and *Culex quinquefasciatus*.

Mosquito species	Concentration ($\mu\text{g/ml}$)	Mortality (%) \pm SD ^a	LC ₅₀ ($\mu\text{g/ml}$) (LCL-UCL)	LC ₉₀ ($\mu\text{g/ml}$) (LCL-UCL)	Slope	Regression equation	χ^2 (d.f.)	
<i>An. stephensi</i>	70	28.0 \pm 1.2	143.22	285.74	3.46	$y = 12.03 + 0.259x$	3.351 (4)	
	140	49.2 \pm 0.8	(126.35–158.04)	(264.80–313.31)				n.s.
	210	67.5 \pm 0.8						
	280	88.3 \pm 0.6						
	350	99.1 \pm 0.4						
<i>Ae. aegypti</i>	70	25.6 \pm 0.8	155.96	306.19	3.14	$y = 7.97 + 0.262x$	2.170 (4)	
	140	44.2 \pm 0.6	(139.17–171.02)	(283.71–335.98)				n.s.
	210	63.8 \pm 0.6						
	280	84.5 \pm 0.4						
	350	97.3 \pm 0.6						
<i>Cx. quinquefasciatus</i>	70	21.9 \pm 0.6	169.94	321.63	2.65	$y = 3.1 + 0.269x$	2.573 (4)	
	140	40.6 \pm 0.4	(153.84–184.80)	(298.31–352.54)				n.s.
	210	59.3 \pm 0.4						
	280	80.2 \pm 0.6						
	350	95.4 \pm 0.8						

No mortality was observed in the control.

SD = standard deviation.

LC₅₀ = lethal concentration that kills 50% of the exposed organisms.LC₉₀ = lethal concentration that kills 90% of the exposed organisms.

UCL = 95% upper confidence limit.

LCL = 95% lower confidence limit.

 χ^2 = chi square.

d.f. = degrees of freedom.

n.s. = not significant ($\alpha = 0.05$).^a Values are mean \pm SD of five replicates.**Table 2**Larvicidal activity of silver nanocrystals synthesized using the *Adiantum raddianum* leaf extract against the mosquito vectors *Anopheles stephensi*, *Aedes aegypti* and *Culex quinquefasciatus*.

Mosquito species	Concentration ($\mu\text{g/ml}$)	Mortality (%) \pm SD ^a	LC ₅₀ ($\mu\text{g/ml}$) (LCL-UCL)	LC ₉₀ ($\mu\text{g/ml}$) (LCL-UCL)	Slope	Regression equation	χ^2 (d.f.)	
<i>An. stephensi</i>	5	27.5 \pm 0.8	10.33	20.51	3.35	$y = 11.42 + 3.644x$	3.667 (4)	
	10	49.2 \pm 0.8	(9.14–11.39)	(19.01–22.48)				n.s.
	15	66.3 \pm 0.6						
	20	88.4 \pm 0.4						
	25	99.0 \pm 0.6						
<i>Ae. aegypti</i>	5	24.2 \pm 0.6	11.23	21.81	2.99	$y = 7.17 + 3.714x$	2.496 (4)	
	10	45.6 \pm 0.4	(10.05–12.29)	(20.23–23.91)				n.s.
	15	62.1 \pm 0.4						
	20	85.3 \pm 0.6						
	25	97.2 \pm 0.8						
<i>Cx. quinquefasciatus</i>	5	20.5 \pm 0.4	12.19	22.86	2.56	$y = 2.39 + 3.814x$	2.082 (4)	
	10	41.3 \pm 0.4	(11.05–13.24)	(21.23–25.03)				n.s.
	15	59.4 \pm 0.6						
	20	80.6 \pm 0.6						
	25	96.2 \pm 0.4						

No mortality was observed in the control.

SD = standard deviation.

LC₅₀ = lethal concentration that kills 50% of the exposed organisms.LC₉₀ = lethal concentration that kills 90% of the exposed organisms.

UCL = 95% upper confidence limit.

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n.s. = not significant ($\alpha = 0.05$).^a Values are mean \pm SD of five replicates.

lipoprotein matrix. This interaction increases the permeability of the plasma membrane of cells. Additionally, an interaction between AgNPs and the sulfur-containing proteins or phosphorous containing compound leads to denaturation of organelles and enzymes and reduces ATP synthesis, which finally causes the loss of the cellular function and cell death (Sap-lam et al., 2010; Sondi and Salopek-Sondi, 2004).

In our experiments, the biotoxicity of *A. raddianum* aqueous extract and green-synthesized AgNPs on non-target organisms *D. indicus* and *G. affinis*, with LC₅₀ ranging from 517.86 to 8430.41 $\mu\text{g/ml}$. In addition, the SI pointed out that *A. raddianum*-fabricated AgNPs were less toxic to the non-target organism tested if

compared to the targeted mosquito larvae. Currently, limited knowledge is available about the acute toxicity of mosquitocidal nanoparticles towards non-target aquatic species (Benelli, 2016a). *Pergularia rubra*- and *Pergularia daemia*-synthesized AgNPs did not exhibit any evident toxicity effect against *Poecilia reticulata* fishes, after 48 h of exposure to LC₅₀ and LC₉₀ values calculated on IV instar larvae of *A. aegypti* and *A. stephensi* (Patil et al., 2012). Subarani et al. (2013) did not reported toxicity effects of *Vinca rosea*-synthesized AgNPs against *P. reticulata*, after 72 h of exposure to dosages toxic against *A. stephensi* and *C. quinquefasciatus*. Similarly, Haldar et al. (2013) did not detected toxicity of AgNPs produced using dried green fruits of *D. roxburghii* against

Table 3Biototoxicity of *Adiantum raddianum* aqueous leaf extract against two non-target organisms sharing the same ecological niche of *Anopheles*, *Aedes* and *Culex* mosquito vectors.

Non-target organism	Concentration ($\mu\text{g/ml}$)	Mortality (%) \pm SD ^a	LC ₅₀ ($\mu\text{g/ml}$) (LCL-UCL)	LC ₉₀ ($\mu\text{g/ml}$) (LCL-UCL)	Slope	Regression equation	χ^2 (d.f.)
<i>D. indicus</i>	3000	26.3 \pm 1.2	6329.54	12263.35	2.94	$y = 9.58 + 0.006x$	5.691 (4) n.s.
	6000	48.2 \pm 0.8	(5641.24–6943.39)	(11383.45–13417.63)			
	9000	65.8 \pm 0.6					
	12000	87.2 \pm 1.2					
	15000	100.0 \pm 0.0					
<i>G. affinis</i>	4000	27.8 \pm 0.8	8430.41	17154.12	3.98	$y = 11.6 + 0.004x$	1.096 (4) n.s.
	8000	46.2 \pm 0.6	(7417.96–9318.29)	(15868.05–18862.34)			
	12000	68.3 \pm 0.6					
	16000	85.6 \pm 0.4					
	20000	97.1 \pm 0.6					

No mortality was observed in the control.

SD = standard deviation.

LC₅₀ = lethal concentration that kills 50% of the exposed organisms.LC₉₀ = lethal concentration that kills 90% of the exposed organisms.

UCL = 95% upper confidence limit.

LCL = 95% lower confidence limit.

 χ^2 = chi square.

d.f. = degrees of freedom.

n.s. = not significant ($\alpha = 0.05$).^a Values are mean \pm SD of five replicates.**Table 4**Acute toxicity of silver nanocrystals synthesized using the *Adiantum raddianum* leaf extract against two non-target organisms sharing the same ecological niche of *Anopheles*, *Aedes* and *Culex* mosquito vectors.

Non-target organism	Concentration ($\mu\text{g/ml}$)	Mortality (%) \pm SD ^a	LC ₅₀ ($\mu\text{g/ml}$) (LCL-UCL)	LC ₉₀ ($\mu\text{g/ml}$) (LCL-UCL)	Slope	Regression equation	χ^2 (d.f.)
<i>D. indicus</i>	250	28.7 \pm 1.2	517.86	1006.83	2.99	$y = 10.42 + 0.074x$	6.167 (4) n.s.
	500	46.2 \pm 1.2	(460.60–568.80)	(934.37–1101.83)			
	750	66.5 \pm 0.4					
	1000	89.4 \pm 0.4					
	1250	100.0 \pm 0.0					
<i>G. affinis</i>	300	26.4 \pm 0.8	635.98	1261.72	3.34	$y = 10.34 + 0.061x$	2.656 (4) n.s.
	600	48.3 \pm 0.8	(563.63–700.02)	(1169.29–1383.67)			
	900	65.2 \pm 1.2					
	1200	87.3 \pm 1.2					
	1500	98.1 \pm 0.4					

No mortality was observed in the control.

SD = standard deviation.

LC₅₀ = lethal concentration that kills 50% of the exposed organisms.LC₉₀ = lethal concentration that kills 90% of the exposed organisms.

UCL = 95% upper confidence limit.

LCL = 95% lower confidence limit.

 χ^2 = chi square.

d.f. = degrees of freedom.

n.s. = not significant ($\alpha = 0.05$).^a Values are mean \pm SD of five replicates.**Table 5**Suitability index of two non-target organisms over young instars of *Anopheles stephensi*, *Aedes aegypti* and *Culex quinquefasciatus* exposed to *Adiantum raddianum* aqueous leaf extract and green-synthesized silver nanocrystals.

Treatment	Non-target organism	<i>An. stephensi</i>	<i>Ae. aegypti</i>	<i>Cx. quinquefasciatus</i>
Aqueous leaf extract	<i>Diplonychus indicus</i>	44.19	40.58	37.24
	<i>Gambusia affinis</i>	58.86	54.05	49.60
Silver nanoparticles	<i>Diplonychus indicus</i>	50.13	46.11	42.48
	<i>Gambusia affinis</i>	61.56	56.63	52.17

P. reticulata, after 48 h-exposure to LC₅₀ of IV instar larvae of *A. stephensi* and *C. quinquefasciatus*. Rawani et al. (2013) showed that mosquitocidal AgNPs synthesized using *Solanum nigrum* berry extracts were not toxic against two mosquito predators, *Toxorhynchites* larvae and *Diplonychus annulatum*, and *Chironomus circumdatus* larvae, exposed to lethal concentrations of dry nanoparticles calculated on *A. stephensi* and *C. quinquefasciatus* larvae. AgNPs synthesized using the 2,7-bis [2-(diethylamino)-ethoxy] fluorene isolate from the *Melia azedarach* leaves did not show

acute toxicity against *Mesocyclops pehpeiensis* copepods (Ramanibai and Velayutham, 2015).

Later on, Govindarajan et al. (2016e) assessed the biotoxicity of *C. spinarum*-synthesized silver nanoparticles on the non-target aquatic organisms *Anisops bouvieri*, *D. indicus* and *G. affinis*. Toxicity testing revealed minimal toxicity, obtaining LC₅₀ values in the range of 424 to 6402 $\mu\text{m/ml}$. Similarly, Govindarajan et al. (2016f) reported that the *Malva sylvestris*-synthesized silver nanoparticles exhibited minimal biotoxicity against nontarget organisms

D. indicus and *G. affinis*, as with LC₅₀ values ranging from 813 to 10,459 µg/ml. *B. cristata*-fabricated silver nanoparticles tested on the non-target organisms *A. bouvieri*, *D. indicus* and *G. affinis*, showed LC₅₀ values ranging from 633 to 8595 µg/ml (Govindarajan and Benelli, 2016a). Moreover, the aqueous extract and biosynthesized AgNPs of *Quisqualis indica* had a moderate biotoxic effect on two mosquito predators *A. bouvieri* (LC₅₀ 653 µg/ml) and *G. affinis* (LC₅₀ 2183 µg/ml) (Govindarajan et al., 2016g). In addition, the exposure to extremely low doses (e.g. 1 ppm) of green-synthesized metal nanoparticles did not negatively impact the predation rates of different mosquito predators (Murugan et al., 2015). It has been formulated that very low doses of plant-synthesized metal nanoparticles may reduce the motility of mosquito larvae, enhancing predation of mosquito fishes, tadpoles, odonate nymphs and other mosquito natural enemies (Benelli, 2016a,b).

Conclusions

Overall, we synthesized silver nanoparticles using a cheap aqueous extract of *A. raddianum* leaves as reducing and stabilizing agent. Our AgNPs were mostly spherical in shape, crystalline in nature, with face-centered cubic geometry, and the mean size was 9–13 nm. This research highlighted that *A. raddianum*-synthesized AgNPs are easy to produce, stable over time, and can be employed at low dosages to strongly reduce populations of vectors mosquitoes of medical and veterinary importance without detrimental effects on predation rates of non-target aquatic organisms, such as *D. indicus* and *G. affinis*.

Conflicts of interest

The authors declare no conflicts of interest.

Compliance with ethical standards

All applicable international and national guidelines for the care and use of animals were followed. All procedures performed in studies involving animals were in accordance with the ethical standards of the institution or practice at which the studies were conducted.

Acknowledgments

Three anonymous reviewers improved an earlier version of our study. The authors would like to thank Professor and Head, Department of Zoology, Annamalai University for the laboratory facilities provided. The authors would also like to acknowledge the cooperation of staff members of the VCRC (ICMR), Pondicherry.

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