



Crystal structure of azimsulfuron

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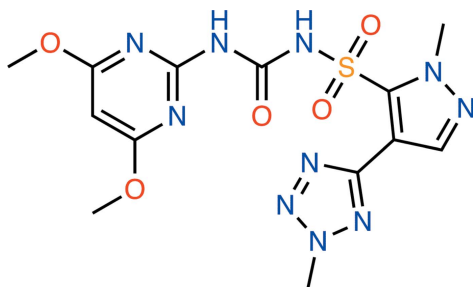
The title compound {systematic name: 1-(4,6-dimethoxy-pyrimidin-2-yl)-3-[1-methyl-4-(2-methyl-2*H*-tetrazol-5-yl)pyrazol-5-ylsulfonfyl]urea}, C₁₃H₁₆N₁₀O₅S, is a sulfonylurea herbicide. In this compound, the dihedral angles between the planes of the central pyrazole and the terminal dimethoxy-pyrimidine and tetrazole rings are 79.10 (8) and 17.21 (16)°, respectively. In the crystal, N—H···O hydrogen bonds link adjacent molecules, forming R₂²(8) inversion dimers. In addition, weak C—H···O and C—H···N hydrogen bonds and weak π – π interactions [ring centroid separation = 3.8255 (12) Å] are present, resulting in a three-dimensional architecture.

Keywords: crystal structure; azimsulfuron; hydrogen bonding; herbicide.

CCDC reference: 1405357

1. Related literature

For information on the herbicidal properties of the title compound, see: Valle *et al.* (2006); Boschini *et al.* (2007). For a related crystal structure, see: Chopra *et al.* (2004).



2. Experimental

2.1. Crystal data

C ₁₃ H ₁₆ N ₁₀ O ₅ S	$\gamma = 66.374 (3)^\circ$
$M_r = 424.42$	$V = 907.16 (14) \text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 8.5884 (7) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.9165 (7) \text{ \AA}$	$\mu = 0.23 \text{ mm}^{-1}$
$c = 12.2788 (13) \text{ \AA}$	$T = 173 \text{ K}$
$\alpha = 73.190 (5)^\circ$	$0.42 \times 0.10 \times 0.09 \text{ mm}$
$\beta = 75.819 (4)^\circ$	

2.2. Data collection

Bruker APEXII CCD diffractometer	16122 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2013)	4123 independent reflections
$T_{\min} = 0.909$, $T_{\max} = 0.979$	3357 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.036$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$	266 parameters
$wR(F^2) = 0.118$	H-atom parameters constrained
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.56 \text{ e \AA}^{-3}$
4123 reflections	$\Delta\rho_{\text{min}} = -0.49 \text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N3—H3N···O3 ⁱ	0.88	2.09	2.877 (2)	149
C1—H1A···O3 ⁱⁱ	0.98	2.55	3.381 (3)	143
C1—H1C···O4 ⁱⁱⁱ	0.98	2.44	3.225 (3)	137
C11—H11A···N9 ^{iv}	0.98	2.57	3.357 (3)	137
C13—H13B···N6 ^v	0.98	2.62	3.533 (3)	155

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x+1, y-1, z$; (iii) $-x+2, -y+1, -z+2$; (iv) $x-1, y+1, z$; (v) $-x+1, -y+1, -z+1$.

Data collection: APEX2 (Bruker, 2013); cell refinement: SAINT (Bruker, 2013); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL2013 (Sheldrick, 2015); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HG5444).

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supporting information

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S1. Comment

Azimsulfuron [systematic name: 1-(4,6-dimethoxypyrimidin-2-yl)-3-[1-methyl-4-(2-methyl-2*H*-tetrazol-5-yl)pyrazol-5-ylsulfonyl]urea] is a sulfonylurea herbicide, a group of pesticides widely used all over the world for controlling weeds in several crops, rice, wheat, maize, barley, sugar beet, and tomato (Valle *et al.*, 2006). However, until now its crystal structure has not been reported. In the title compound (Fig. 1), the dihedral angles between the planes of the central pyrazol and the terminal dimethoxypyrimidinyl and tetrazol rings are 79.10 (8) and 17.21 (16)°, respectively. All bond lengths and bond angles are normal and comparable to those observed in similar crystal structures (Chopra *et al.*, 2004).

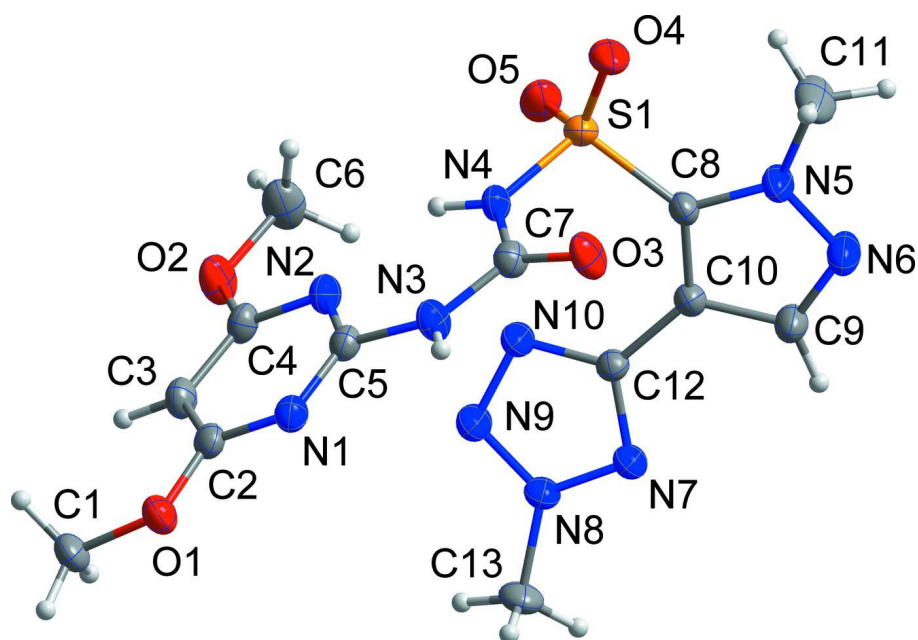
In the crystal structure (Fig. 2), molecules are linked by a pairs of N—H···O hydrogen bonds (Table 1), forming inversion dimmers with an $R_2^2(8)$ ring motif. In addition, weak C—H···O and C—H···N hydrogen bonding and weak intermolecular π – π interactions between the terminal tetrazol ring systems [$\text{Cg2} \cdots \text{Cg2}^i = 3.8255(12) \text{ \AA}$] are present (Cg2 is the centroid of the N7–N8–N9–N10–C12 ring) [for symmetry codes: (i), $-x, -y + 1, -z + 1$].

S2. Experimental

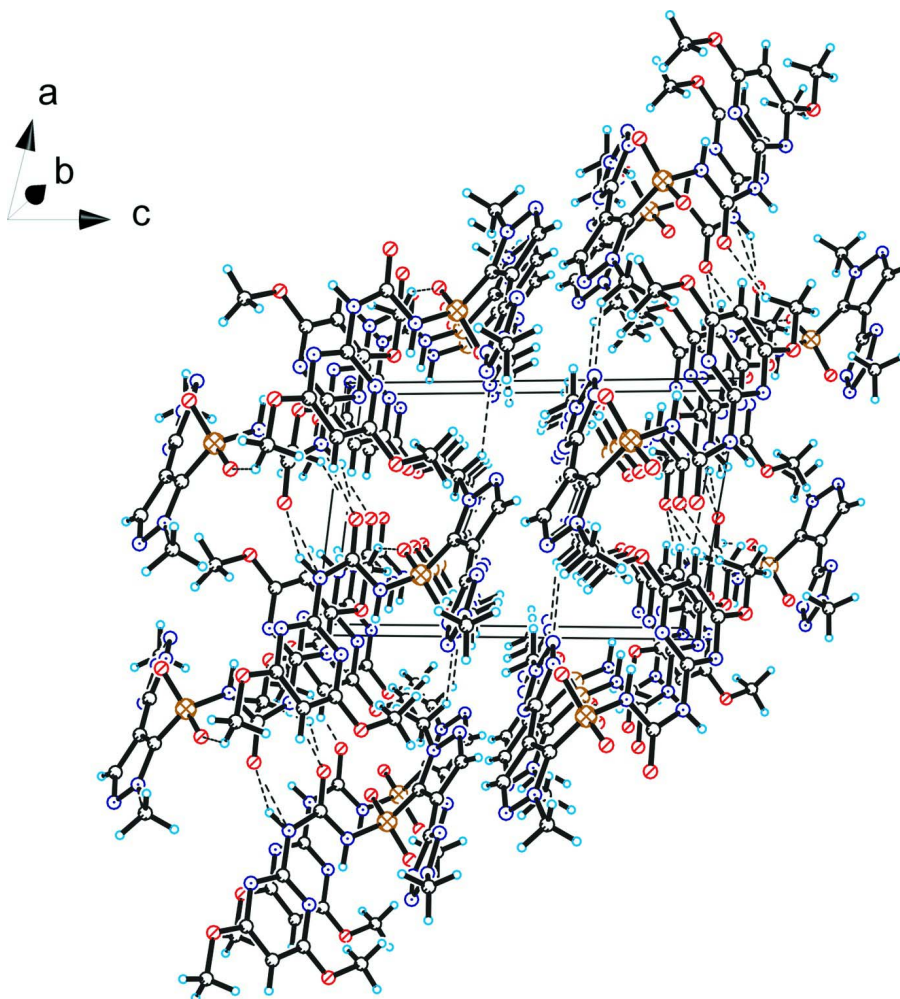
The title compound was purchased from the Dr. Ehrenstorfer GmbH Company. Slow evaporation of a solution in CH_2Cl_2 gave single crystals suitable for X-ray analysis.

S3. Refinement

All H-atoms were positioned geometrically and refined using a riding model with $d(\text{N—H}) = 0.88 \text{ \AA}$, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for urea N—H, $d(\text{C—H}) = 0.98 \text{ \AA}$, $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$ for methyl group, $d(\text{C—H}) = 0.95 \text{ \AA}$, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for aromatic C—H.

**Figure 1**

The asymmetric unit of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radius.

**Figure 2**

Crystal packing viewed along the *b* axis. The hydrogen bonds are shown as dashed lines.

1-(4,6-Dimethoxypyrimidin-2-yl)-3-[1-methyl-4-(2-methyl-2*H*-tetrazol-5-yl)pyrazol-5-ylsulfonyl]urea

Crystal data

$C_{13}H_{16}N_{10}O_5S$

$M_r = 424.42$

Triclinic, $P\bar{1}$

$a = 8.5884(7) \text{ \AA}$

$b = 9.9165(7) \text{ \AA}$

$c = 12.2788(13) \text{ \AA}$

$\alpha = 73.190(5)^\circ$

$\beta = 75.819(4)^\circ$

$\gamma = 66.374(3)^\circ$

$V = 907.16(14) \text{ \AA}^3$

$Z = 2$

$F(000) = 440$

$D_x = 1.554 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4687 reflections

$\theta = 2.3\text{--}26.9^\circ$

$\mu = 0.23 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Block, colourless

$0.42 \times 0.10 \times 0.09 \text{ mm}$

Data collection

Bruker APEXII CCD

diffractometer

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2013)

$T_{\min} = 0.909$, $T_{\max} = 0.979$

16122 measured reflections
 4123 independent reflections
 3357 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$

$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 1.8^\circ$
 $h = -11 \rightarrow 11$
 $k = -12 \rightarrow 12$
 $l = -15 \rightarrow 15$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.118$
 $S = 1.04$
 4123 reflections
 266 parameters
 0 restraints

Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0543P)^2 + 0.5222P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.56 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.49 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.74797 (6)	0.79478 (5)	0.73248 (4)	0.02370 (14)
O1	1.10371 (18)	−0.02315 (15)	1.11503 (12)	0.0280 (3)
O2	1.32890 (18)	0.30498 (17)	0.81805 (13)	0.0356 (4)
O3	0.53148 (17)	0.65412 (16)	0.91451 (13)	0.0326 (4)
O4	0.6527 (2)	0.91670 (16)	0.78851 (14)	0.0379 (4)
O5	0.90400 (19)	0.79382 (17)	0.65785 (14)	0.0361 (4)
N1	0.9372 (2)	0.21202 (18)	1.04786 (14)	0.0236 (4)
N2	1.0436 (2)	0.38255 (18)	0.89699 (14)	0.0238 (4)
N3	0.7525 (2)	0.44579 (18)	0.97760 (14)	0.0254 (4)
H3N	0.6779	0.4182	1.0336	0.031*
N4	0.8032 (2)	0.63736 (17)	0.82771 (13)	0.0214 (3)
H4N	0.9133	0.5838	0.8284	0.026*
N5	0.4566 (2)	0.88624 (18)	0.63630 (16)	0.0303 (4)
N6	0.3789 (2)	0.8477 (2)	0.57414 (18)	0.0394 (5)
N7	0.7377 (2)	0.40229 (17)	0.58472 (14)	0.0247 (4)
N8	0.8855 (2)	0.28937 (17)	0.59920 (14)	0.0243 (4)
N9	0.9977 (2)	0.32420 (18)	0.63093 (16)	0.0298 (4)
N10	0.9232 (2)	0.46781 (18)	0.63767 (15)	0.0280 (4)
C1	1.2673 (3)	−0.1457 (2)	1.11176 (19)	0.0308 (5)
H1A	1.3074	−0.1629	1.0333	0.046*
H1B	1.2544	−0.2373	1.1643	0.046*
H1C	1.3512	−0.1201	1.1355	0.046*
C2	1.0957 (2)	0.1075 (2)	1.04081 (16)	0.0228 (4)
C3	1.2338 (2)	0.1330 (2)	0.96351 (17)	0.0262 (4)
H3	1.3452	0.0575	0.9589	0.031*
C4	1.1988 (2)	0.2757 (2)	0.89353 (17)	0.0244 (4)

C5	0.9207 (2)	0.3420 (2)	0.97398 (16)	0.0215 (4)
C6	1.2928 (3)	0.4514 (3)	0.7412 (2)	0.0393 (6)
H6A	1.2067	0.4667	0.6948	0.059*
H6B	1.3985	0.4573	0.6904	0.059*
H6C	1.2486	0.5297	0.7861	0.059*
C7	0.6855 (2)	0.5842 (2)	0.90723 (16)	0.0231 (4)
C8	0.6097 (2)	0.7766 (2)	0.65845 (16)	0.0215 (4)
C9	0.4833 (3)	0.7125 (2)	0.5569 (2)	0.0316 (5)
H9	0.4601	0.6571	0.5153	0.038*
C10	0.6314 (2)	0.6612 (2)	0.60716 (16)	0.0212 (4)
C11	0.3620 (3)	1.0278 (3)	0.6757 (3)	0.0496 (7)
H11A	0.2815	1.0973	0.6224	0.074*
H11B	0.4431	1.0732	0.6783	0.074*
H11C	0.2981	1.0074	0.7527	0.074*
C12	0.7652 (2)	0.5127 (2)	0.60905 (16)	0.0204 (4)
C13	0.9157 (3)	0.1370 (2)	0.5897 (2)	0.0326 (5)
H13A	1.0396	0.0807	0.5778	0.049*
H13B	0.8653	0.1424	0.5244	0.049*
H13C	0.8625	0.0857	0.6606	0.049*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0261 (3)	0.0177 (2)	0.0299 (3)	−0.00848 (19)	−0.0094 (2)	−0.00349 (18)
O1	0.0263 (7)	0.0221 (7)	0.0311 (8)	−0.0045 (6)	−0.0073 (6)	−0.0024 (6)
O2	0.0173 (7)	0.0365 (8)	0.0414 (9)	−0.0065 (6)	−0.0006 (6)	0.0009 (7)
O3	0.0177 (7)	0.0287 (7)	0.0380 (8)	−0.0007 (6)	0.0005 (6)	−0.0025 (6)
O4	0.0478 (10)	0.0220 (7)	0.0494 (9)	−0.0038 (7)	−0.0234 (8)	−0.0146 (7)
O5	0.0301 (8)	0.0364 (8)	0.0425 (9)	−0.0205 (7)	−0.0075 (7)	0.0047 (7)
N1	0.0213 (8)	0.0249 (8)	0.0240 (8)	−0.0068 (7)	−0.0050 (7)	−0.0050 (6)
N2	0.0176 (8)	0.0255 (8)	0.0253 (8)	−0.0046 (7)	−0.0039 (7)	−0.0049 (7)
N3	0.0174 (8)	0.0254 (8)	0.0260 (8)	−0.0041 (7)	0.0002 (7)	−0.0024 (7)
N4	0.0162 (7)	0.0200 (7)	0.0251 (8)	−0.0028 (6)	−0.0050 (6)	−0.0042 (6)
N5	0.0265 (9)	0.0225 (8)	0.0423 (10)	0.0018 (7)	−0.0164 (8)	−0.0132 (7)
N6	0.0336 (10)	0.0331 (10)	0.0538 (12)	0.0015 (8)	−0.0243 (9)	−0.0174 (9)
N7	0.0234 (8)	0.0194 (7)	0.0312 (9)	−0.0041 (7)	−0.0068 (7)	−0.0078 (7)
N8	0.0260 (8)	0.0182 (7)	0.0293 (9)	−0.0047 (7)	−0.0063 (7)	−0.0087 (6)
N9	0.0252 (9)	0.0236 (8)	0.0411 (10)	−0.0021 (7)	−0.0102 (8)	−0.0131 (7)
N10	0.0233 (8)	0.0212 (8)	0.0409 (10)	−0.0024 (7)	−0.0103 (8)	−0.0121 (7)
C1	0.0283 (11)	0.0193 (9)	0.0403 (12)	−0.0001 (8)	−0.0127 (9)	−0.0055 (8)
C2	0.0234 (10)	0.0227 (9)	0.0236 (9)	−0.0058 (8)	−0.0094 (8)	−0.0056 (7)
C3	0.0180 (9)	0.0271 (10)	0.0298 (10)	−0.0027 (8)	−0.0066 (8)	−0.0060 (8)
C4	0.0165 (9)	0.0297 (10)	0.0262 (10)	−0.0071 (8)	−0.0042 (8)	−0.0056 (8)
C5	0.0173 (9)	0.0246 (9)	0.0228 (9)	−0.0049 (8)	−0.0044 (7)	−0.0076 (7)
C6	0.0283 (11)	0.0411 (13)	0.0386 (13)	−0.0145 (10)	0.0007 (10)	0.0043 (10)
C7	0.0200 (9)	0.0242 (9)	0.0230 (9)	−0.0051 (8)	−0.0010 (8)	−0.0075 (7)
C8	0.0213 (9)	0.0172 (8)	0.0249 (9)	−0.0028 (7)	−0.0079 (8)	−0.0050 (7)
C9	0.0297 (11)	0.0264 (10)	0.0419 (12)	−0.0028 (9)	−0.0170 (10)	−0.0124 (9)

C10	0.0209 (9)	0.0187 (9)	0.0237 (9)	−0.0057 (7)	−0.0056 (8)	−0.0040 (7)
C11	0.0405 (14)	0.0317 (12)	0.0737 (18)	0.0128 (11)	−0.0255 (13)	−0.0302 (12)
C12	0.0206 (9)	0.0198 (8)	0.0219 (9)	−0.0056 (7)	−0.0053 (7)	−0.0066 (7)
C13	0.0393 (12)	0.0181 (9)	0.0415 (12)	−0.0063 (9)	−0.0074 (10)	−0.0125 (8)

Geometric parameters (Å, °)

S1—O4	1.4201 (15)	N8—N9	1.314 (2)
S1—O5	1.4243 (16)	N8—C13	1.462 (2)
S1—N4	1.6284 (16)	N9—N10	1.325 (2)
S1—C8	1.751 (2)	N10—C12	1.350 (2)
O1—C2	1.338 (2)	C1—H1A	0.9800
O1—C1	1.444 (2)	C1—H1B	0.9800
O2—C4	1.337 (2)	C1—H1C	0.9800
O2—C6	1.444 (3)	C2—C3	1.387 (3)
O3—C7	1.217 (2)	C3—C4	1.383 (3)
N1—C5	1.321 (2)	C3—H3	0.9500
N1—C2	1.341 (2)	C6—H6A	0.9800
N2—C4	1.330 (2)	C6—H6B	0.9800
N2—C5	1.342 (2)	C6—H6C	0.9800
N3—C7	1.373 (2)	C8—C10	1.389 (3)
N3—C5	1.397 (2)	C9—C10	1.393 (3)
N3—H3N	0.8800	C9—H9	0.9500
N4—C7	1.378 (2)	C10—C12	1.459 (2)
N4—H4N	0.8800	C11—H11A	0.9800
N5—N6	1.340 (3)	C11—H11B	0.9800
N5—C8	1.358 (2)	C11—H11C	0.9800
N5—C11	1.466 (3)	C13—H13A	0.9800
N6—C9	1.325 (3)	C13—H13B	0.9800
N7—N8	1.324 (2)	C13—H13C	0.9800
N7—C12	1.332 (2)		
O4—S1—O5	119.58 (10)	N2—C4—C3	123.37 (18)
O4—S1—N4	109.86 (9)	O2—C4—C3	117.43 (17)
O5—S1—N4	104.30 (9)	N1—C5—N2	127.89 (17)
O4—S1—C8	107.79 (9)	N1—C5—N3	114.04 (17)
O5—S1—C8	109.46 (10)	N2—C5—N3	118.05 (16)
N4—S1—C8	104.90 (9)	O2—C6—H6A	109.5
C2—O1—C1	117.09 (16)	O2—C6—H6B	109.5
C4—O2—C6	118.04 (16)	H6A—C6—H6B	109.5
C5—N1—C2	114.99 (17)	O2—C6—H6C	109.5
C4—N2—C5	114.96 (16)	H6A—C6—H6C	109.5
C7—N3—C5	129.98 (17)	H6B—C6—H6C	109.5
C7—N3—H3N	115.0	O3—C7—N3	121.66 (18)
C5—N3—H3N	115.0	O3—C7—N4	122.49 (17)
C7—N4—S1	123.00 (13)	N3—C7—N4	115.85 (16)
C7—N4—H4N	118.5	N5—C8—C10	107.00 (17)
S1—N4—H4N	118.5	N5—C8—S1	122.90 (14)

N6—N5—C8	111.29 (16)	C10—C8—S1	130.02 (14)
N6—N5—C11	117.40 (17)	N6—C9—C10	111.85 (19)
C8—N5—C11	131.12 (18)	N6—C9—H9	124.1
C9—N6—N5	105.78 (17)	C10—C9—H9	124.1
N8—N7—C12	101.76 (15)	C8—C10—C9	104.07 (16)
N9—N8—N7	113.84 (15)	C8—C10—C12	131.15 (18)
N9—N8—C13	123.11 (16)	C9—C10—C12	124.59 (18)
N7—N8—C13	122.87 (17)	N5—C11—H11A	109.5
N8—N9—N10	106.55 (15)	N5—C11—H11B	109.5
N9—N10—C12	105.48 (16)	H11A—C11—H11B	109.5
O1—C1—H1A	109.5	N5—C11—H11C	109.5
O1—C1—H1B	109.5	H11A—C11—H11C	109.5
H1A—C1—H1B	109.5	H11B—C11—H11C	109.5
O1—C1—H1C	109.5	N7—C12—N10	112.36 (16)
H1A—C1—H1C	109.5	N7—C12—C10	121.13 (17)
H1B—C1—H1C	109.5	N10—C12—C10	126.48 (17)
O1—C2—N1	112.43 (17)	N8—C13—H13A	109.5
O1—C2—C3	124.43 (17)	N8—C13—H13B	109.5
N1—C2—C3	123.14 (18)	H13A—C13—H13B	109.5
C4—C3—C2	115.60 (17)	N8—C13—H13C	109.5
C4—C3—H3	122.2	H13A—C13—H13C	109.5
C2—C3—H3	122.2	H13B—C13—H13C	109.5
N2—C4—O2	119.19 (17)		
O4—S1—N4—C7	59.80 (18)	C5—N3—C7—N4	−7.5 (3)
O5—S1—N4—C7	−170.87 (15)	S1—N4—C7—O3	−2.7 (3)
C8—S1—N4—C7	−55.81 (17)	S1—N4—C7—N3	176.66 (13)
C8—N5—N6—C9	0.1 (3)	N6—N5—C8—C10	0.1 (2)
C11—N5—N6—C9	−175.4 (2)	C11—N5—C8—C10	174.9 (2)
C12—N7—N8—N9	0.4 (2)	N6—N5—C8—S1	177.18 (16)
C12—N7—N8—C13	175.78 (18)	C11—N5—C8—S1	−8.1 (3)
N7—N8—N9—N10	−0.4 (2)	O4—S1—C8—N5	15.8 (2)
C13—N8—N9—N10	−175.73 (18)	O5—S1—C8—N5	−115.78 (18)
N8—N9—N10—C12	0.2 (2)	N4—S1—C8—N5	132.80 (17)
C1—O1—C2—N1	177.72 (17)	O4—S1—C8—C10	−167.94 (18)
C1—O1—C2—C3	−1.3 (3)	O5—S1—C8—C10	60.5 (2)
C5—N1—C2—O1	−178.07 (16)	N4—S1—C8—C10	−50.9 (2)
C5—N1—C2—C3	1.0 (3)	N5—N6—C9—C10	−0.4 (3)
O1—C2—C3—C4	179.64 (18)	N5—C8—C10—C9	−0.3 (2)
N1—C2—C3—C4	0.7 (3)	S1—C8—C10—C9	−177.09 (17)
C5—N2—C4—O2	178.90 (17)	N5—C8—C10—C12	−175.5 (2)
C5—N2—C4—C3	−0.2 (3)	S1—C8—C10—C12	7.8 (3)
C6—O2—C4—N2	−1.7 (3)	N6—C9—C10—C8	0.4 (3)
C6—O2—C4—C3	177.50 (19)	N6—C9—C10—C12	176.0 (2)
C2—C3—C4—N2	−1.1 (3)	N8—N7—C12—N10	−0.3 (2)
C2—C3—C4—O2	179.76 (18)	N8—N7—C12—C10	−178.32 (17)
C2—N1—C5—N2	−2.7 (3)	N9—N10—C12—N7	0.1 (2)
C2—N1—C5—N3	175.99 (17)	N9—N10—C12—C10	177.97 (18)

C4—N2—C5—N1	2.3 (3)	C8—C10—C12—N7	159.2 (2)
C4—N2—C5—N3	−176.29 (17)	C9—C10—C12—N7	−15.0 (3)
C7—N3—C5—N1	−173.08 (19)	C8—C10—C12—N10	−18.5 (3)
C7—N3—C5—N2	5.7 (3)	C9—C10—C12—N10	167.3 (2)
C5—N3—C7—O3	171.9 (2)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N3—H3N \cdots O3 ⁱ	0.88	2.09	2.877 (2)	149
C1—H1A \cdots O3 ⁱⁱ	0.98	2.55	3.381 (3)	143
C1—H1C \cdots O4 ⁱⁱⁱ	0.98	2.44	3.225 (3)	137
C11—H11A \cdots N9 ^{iv}	0.98	2.57	3.357 (3)	137
C13—H13B \cdots N6 ^v	0.98	2.62	3.533 (3)	155

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x+1, y-1, z$; (iii) $-x+2, -y+1, -z+2$; (iv) $x-1, y+1, z$; (v) $-x+1, -y+1, -z+1$.