



Crystal structure of 3-(morpholin-4-yl)-1-phenyl-3-(pyridin-2-yl)propan-1-one

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Received 26 November 2014; accepted 30 November 2014

Edited by W. T. A. Harrison, University of Aberdeen, Scotland

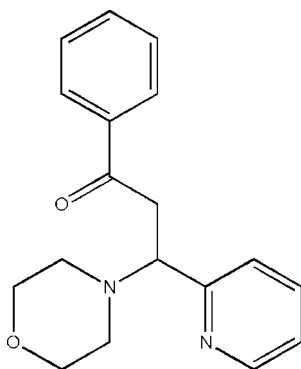
In the title compound $C_{18}H_{20}N_2O_2$, the morpholine ring adopts a chair conformation with the exocyclic N—C bond in an equatorial orientation. The N atom of the morpholine ring and the C atom of the carbonyl group are in an *anti* conformation about the central C—C bond [torsion angle = $-162.92(11)^\circ$] and the dihedral angle between the planes of the benzene ring and the pyridine ring is $83.30(5)^\circ$. In the crystal, pairs of very weak C—H $\cdots\pi$ interactions link the molecules into inversion dimers.

Keywords: crystal structure; morpholin-4-yl; pyridin-2-yl; propan-1-one; biological activity.

CCDC reference: 1036843

1. Related literature

For background to the biological activity of morpholine derivatives, see: Panneerselvam *et al.* (2009); Subhashini *et al.* (2013); Sawant *et al.* (2013); Dave & Sasaki (2006); For related structures, see: Chen *et al.* (2011); Meti *et al.* (2013);



2. Experimental

2.1. Crystal data

$C_{18}H_{20}N_2O_2$	$V = 3140.1(3) \text{ \AA}^3$
$M_r = 296.36$	$Z = 8$
Orthorhombic, $Pbca$	Mo $K\alpha$ radiation
$a = 12.4554(6) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 8.2204(4) \text{ \AA}$	$T = 295 \text{ K}$
$c = 30.6681(17) \text{ \AA}$	$0.20 \times 0.15 \times 0.10 \text{ mm}$

2.2. Data collection

Bruker APEXII CCD diffractometer	16093 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1996)	3812 independent reflections
$T_{\min} = 0.954$, $T_{\max} = 0.975$	2547 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.028$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$	199 parameters
$wR(F^2) = 0.111$	H-atom parameters constrained
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
3812 reflections	$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg2$ is the centroid of the $C10$ – $C14$ / $N1$ ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C2-H2\cdots Cg2^i$	0.93	2.90	3.780 (6)	159

Symmetry code: (i) $-x, -y + 1, -z$.

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

Acknowledgements

FMMA acknowledges the PG and Research Department of Chemistry and the Management of Jamal Mohamed College (Autonomous) for their kind support.

Supporting information for this paper is available from the IUCr electronic archives (Reference: HB7328).

References

- Bruker (2008). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chen, X.-Y., Zhao, M.-M., Qian, X. & Hou, S.-G. (2011). *Acta Cryst.* **E67**, o3484.
- Dave, R. & Sasaki, N. A. (2006). *Tetrahedron Asymmetry*, **17**, 388–401.
- Meti, G. Y., Kamble, R. R., Ravi, A. J., Arunkashi, H. K. & Devarajegowda, H. C. (2013). *Acta Cryst.* **E69**, o129.
- Panneerselvam, P., Priya, M., Gnanarupa, , Kumar, N., Ramesh, & Saravanan, G. (2009). *Indian J. Pharm. Sci.* **71**, 428–432.
- Sawant, R. T., Stevenson, J., Odell, L. R. & Arvidsson, P. I. (2013). *Tetrahedron Asymmetry*, **24**, 134–141.

Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

Subhashini, N. J. P., Amanaganti, J., Boddu, L. & Acharya Nagarjuna, P.
(2013). *J. Chem. Pharm. Res.* **5**, 140–147.

supporting information

Acta Cryst. (2015). E71, o24–o25 [https://doi.org/10.1107/S2056989014026292]

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S1. Comment

Morpholines are six-membered heterocycles featuring both cyclic amine and ether functional group. These compounds possess important applications in pharmaceuticals and in industries (Panneerselvam *et al.*, 2009; Subhashini *et al.*, 2013). Chiral morpholine derivatives have found numerous applications in asymmetric synthesis as chiral auxiliaries as well as chiral ligands (Sawant *et al.*, 2013; Dave & Sasaki 2006).

The geometric parameters of the title molecule (Fig. 1) agree well with reported similar structure (Chen *et al.*, 2011; Meti *et al.*, 2013). The morpholine (N2/O2/C15—C18) ring adopts a chair conformation [$Q = 0.5756(3) \text{ \AA}$, $\Theta = 179.09(3)^\circ$, $\varphi = 332.57(5)^\circ$]. The phenyl ring makes a dihedral angles of $83.30(5)^\circ$ with the pyridine ring. In the crystal, a weak C—H $\cdots\pi$ interaction is observed.

S2. Experimental

To an ethanolic solution of acetophenone (3.0 ml, 0.025 mol) taken in a round bottom flask, morpholine (2.1 ml, 0.025 mol) and pyridine-2-carboldehyde (2.6 ml, 0.025 mol) were added. The reaction mixture was kept over a magnetic stirrer and stirred well in an ice cold condition for 3 hr. The colourless solid formed was filtered and washed several times with petroleum ether (40–60%). The crude solid obtained was dried and recrystallized using absolute alcohol. The recrystallized product was dried over vacuum. The yield is 78% and MP is 445 K.

S3. Refinement

H atoms were positioned geometrically and refined using riding model with C—H = 0.93 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic C—H, C—H = 0.98 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for C—H, C—H = 0.97 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for C—H₂,

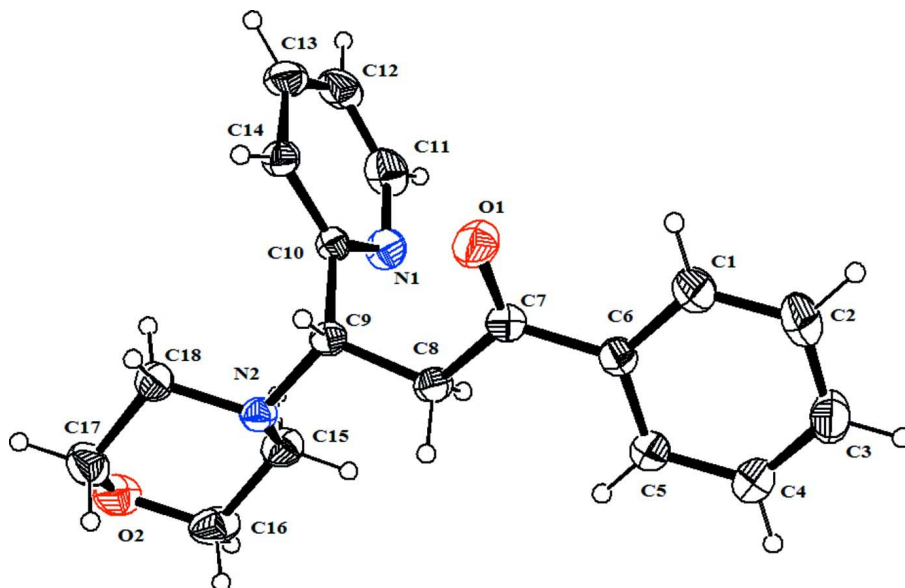


Figure 1

The molecular structure of (I), with 30% probability displacement ellipsoids for non-H atoms.

3-(Morpholin-4-yl)-1-phenyl-3-(pyridin-2-yl)propan-1-one

Crystal data

$C_{18}H_{20}N_2O_2$

$M_r = 296.36$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 12.4554$ (6) Å

$b = 8.2204$ (4) Å

$c = 30.6681$ (17) Å

$V = 3140.1$ (3) Å³

$Z = 8$

$F(000) = 1264$

$D_x = 1.254$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3812 reflections

$\theta = 1.3$ – 28.4°

$\mu = 0.08$ mm⁻¹

$T = 295$ K

Block, colourless

$0.20 \times 0.15 \times 0.10$ mm

Data collection

Bruker APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 0 pixels mm⁻¹

ω and ϕ scans

Absorption correction: multi-scan

(*SADABS*; Sheldrick, 1996)

$T_{\min} = 0.954$, $T_{\max} = 0.975$

16093 measured reflections

3812 independent reflections

2547 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\max} = 28.4^\circ$, $\theta_{\min} = 1.3^\circ$

$h = -15 \rightarrow 15$

$k = -10 \rightarrow 10$

$l = -35 \rightarrow 40$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.111$

$S = 1.03$

3812 reflections

199 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from

neighbouring sites

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0416P)^2 + 0.6298P]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.14 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.16 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.42802 (13)	−0.1732 (2)	0.02138 (5)	0.0604 (4)
H1	0.5020	−0.1642	0.0182	0.073*
C2	0.36782 (16)	−0.2413 (2)	−0.01152 (6)	0.0746 (5)
H2	0.4013	−0.2768	−0.0369	0.089*
C3	0.25900 (16)	−0.2571 (2)	−0.00707 (5)	0.0680 (5)
H3	0.2186	−0.3050	−0.0291	0.082*
C4	0.21022 (13)	−0.2020 (2)	0.02998 (6)	0.0653 (5)
H4	0.1362	−0.2121	0.0330	0.078*
C5	0.26935 (12)	−0.13170 (19)	0.06292 (5)	0.0529 (4)
H5	0.2350	−0.0932	0.0878	0.063*
C6	0.37980 (11)	−0.11808 (15)	0.05909 (4)	0.0412 (3)
C7	0.44770 (10)	−0.04735 (16)	0.09421 (4)	0.0417 (3)
C8	0.39381 (10)	0.03707 (17)	0.13181 (4)	0.0429 (3)
H8A	0.3510	−0.0418	0.1477	0.051*
H8B	0.3453	0.1190	0.1204	0.051*
C9	0.47105 (9)	0.11758 (15)	0.16326 (4)	0.0364 (3)
H9	0.5269	0.0372	0.1697	0.044*
C10	0.52724 (10)	0.26260 (15)	0.14299 (4)	0.0368 (3)
C11	0.51464 (16)	0.4990 (2)	0.10454 (6)	0.0666 (5)
H11	0.4726	0.5746	0.0898	0.080*
C12	0.62216 (16)	0.5287 (2)	0.10752 (6)	0.0678 (5)
H12	0.6520	0.6215	0.0951	0.081*
C13	0.68473 (13)	0.4190 (2)	0.12904 (5)	0.0610 (5)
H13	0.7585	0.4344	0.1313	0.073*
C14	0.63651 (11)	0.28526 (18)	0.14735 (5)	0.0452 (3)
H14	0.6775	0.2099	0.1627	0.054*
C15	0.33073 (11)	0.2736 (2)	0.20287 (5)	0.0532 (4)
H15A	0.2787	0.2424	0.1809	0.064*
H15B	0.3602	0.3787	0.1949	0.064*
C16	0.27665 (12)	0.2851 (2)	0.24674 (6)	0.0662 (5)
H16A	0.2195	0.3651	0.2453	0.079*
H16B	0.2447	0.1808	0.2539	0.079*
C17	0.43388 (14)	0.2128 (2)	0.28202 (5)	0.0632 (4)
H17A	0.4040	0.1078	0.2898	0.076*
H17B	0.4846	0.2440	0.3045	0.076*
C18	0.49150 (11)	0.19882 (18)	0.23921 (4)	0.0462 (3)
H18A	0.5247	0.3021	0.2320	0.055*
H18B	0.5477	0.1176	0.2415	0.055*
N1	0.46563 (10)	0.36867 (15)	0.12143 (4)	0.0526 (3)
N2	0.41661 (8)	0.15290 (13)	0.20488 (4)	0.0394 (3)
O1	0.54495 (8)	−0.05833 (15)	0.09246 (4)	0.0655 (3)

O2	0.35010 (10)	0.32957 (14)	0.27995 (4)	0.0678 (3)
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Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0564 (9)	0.0742 (11)	0.0506 (9)	−0.0007 (8)	0.0083 (8)	−0.0126 (8)
C2	0.0842 (13)	0.0925 (14)	0.0470 (10)	−0.0019 (11)	0.0067 (9)	−0.0220 (9)
C3	0.0815 (12)	0.0728 (11)	0.0497 (9)	−0.0123 (10)	−0.0134 (9)	−0.0073 (9)
C4	0.0567 (9)	0.0784 (12)	0.0607 (10)	−0.0166 (9)	−0.0049 (8)	−0.0085 (9)
C5	0.0495 (9)	0.0615 (9)	0.0477 (8)	−0.0098 (7)	0.0030 (7)	−0.0074 (7)
C6	0.0473 (7)	0.0366 (7)	0.0399 (7)	−0.0009 (6)	0.0012 (6)	0.0008 (6)
C7	0.0410 (7)	0.0392 (7)	0.0449 (8)	0.0005 (6)	0.0027 (6)	0.0003 (6)
C8	0.0379 (7)	0.0446 (7)	0.0461 (8)	−0.0049 (6)	0.0040 (6)	−0.0039 (6)
C9	0.0328 (6)	0.0363 (7)	0.0402 (7)	0.0024 (5)	0.0011 (5)	0.0002 (6)
C10	0.0376 (7)	0.0382 (7)	0.0346 (6)	−0.0004 (5)	0.0020 (5)	−0.0025 (5)
C11	0.0910 (13)	0.0498 (9)	0.0589 (10)	−0.0004 (9)	−0.0032 (9)	0.0166 (8)
C12	0.0926 (13)	0.0549 (10)	0.0559 (10)	−0.0279 (10)	0.0183 (9)	0.0017 (8)
C13	0.0555 (9)	0.0663 (10)	0.0613 (10)	−0.0229 (8)	0.0155 (8)	−0.0162 (9)
C14	0.0389 (7)	0.0512 (8)	0.0456 (8)	−0.0027 (6)	0.0030 (6)	−0.0067 (7)
C15	0.0404 (7)	0.0569 (9)	0.0624 (10)	0.0062 (7)	0.0048 (7)	−0.0090 (8)
C16	0.0500 (9)	0.0641 (10)	0.0844 (12)	−0.0057 (8)	0.0237 (9)	−0.0198 (9)
C17	0.0810 (11)	0.0612 (10)	0.0473 (9)	−0.0109 (9)	0.0080 (8)	−0.0065 (8)
C18	0.0483 (8)	0.0451 (8)	0.0453 (8)	−0.0029 (7)	−0.0002 (6)	−0.0014 (6)
N1	0.0530 (7)	0.0496 (7)	0.0553 (7)	0.0018 (6)	−0.0067 (6)	0.0125 (6)
N2	0.0355 (5)	0.0409 (6)	0.0419 (6)	−0.0012 (5)	0.0040 (5)	−0.0024 (5)
O1	0.0412 (6)	0.0847 (9)	0.0707 (8)	0.0064 (5)	0.0016 (5)	−0.0247 (6)
O2	0.0733 (7)	0.0648 (7)	0.0654 (7)	−0.0114 (6)	0.0215 (6)	−0.0235 (6)

Geometric parameters (\AA , $^\circ$)

C1—C2	1.376 (2)	C11—N1	1.338 (2)
C1—C6	1.380 (2)	C11—C12	1.364 (3)
C1—H1	0.9300	C11—H11	0.9300
C2—C3	1.368 (3)	C12—C13	1.362 (2)
C2—H2	0.9300	C12—H12	0.9300
C3—C4	1.366 (2)	C13—C14	1.373 (2)
C3—H3	0.9300	C13—H13	0.9300
C4—C5	1.377 (2)	C14—H14	0.9300
C4—H4	0.9300	C15—N2	1.4604 (17)
C5—C6	1.3852 (19)	C15—C16	1.508 (2)
C5—H5	0.9300	C15—H15A	0.9700
C6—C7	1.4878 (19)	C15—H15B	0.9700
C7—O1	1.2159 (15)	C16—O2	1.417 (2)
C7—C8	1.5039 (19)	C16—H16A	0.9700
C8—C9	1.5146 (18)	C16—H16B	0.9700
C8—H8A	0.9700	C17—O2	1.419 (2)
C8—H8B	0.9700	C17—C18	1.500 (2)
C9—N2	1.4741 (16)	C17—H17A	0.9700

C9—C10	1.5157 (18)	C17—H17B	0.9700
C9—H9	0.9800	C18—N2	1.4565 (17)
C10—N1	1.3365 (17)	C18—H18A	0.9700
C10—C14	1.3803 (17)	C18—H18B	0.9700
C2—C1—C6	120.73 (15)	C13—C12—C11	118.42 (15)
C2—C1—H1	119.6	C13—C12—H12	120.8
C6—C1—H1	119.6	C11—C12—H12	120.8
C3—C2—C1	120.36 (16)	C12—C13—C14	118.55 (15)
C3—C2—H2	119.8	C12—C13—H13	120.7
C1—C2—H2	119.8	C14—C13—H13	120.7
C4—C3—C2	119.47 (16)	C13—C14—C10	119.99 (15)
C4—C3—H3	120.3	C13—C14—H14	120.0
C2—C3—H3	120.3	C10—C14—H14	120.0
C3—C4—C5	120.76 (15)	N2—C15—C16	109.39 (13)
C3—C4—H4	119.6	N2—C15—H15A	109.8
C5—C4—H4	119.6	C16—C15—H15A	109.8
C4—C5—C6	120.18 (14)	N2—C15—H15B	109.8
C4—C5—H5	119.9	C16—C15—H15B	109.8
C6—C5—H5	119.9	H15A—C15—H15B	108.2
C1—C6—C5	118.48 (13)	O2—C16—C15	111.65 (12)
C1—C6—C7	119.19 (12)	O2—C16—H16A	109.3
C5—C6—C7	122.33 (12)	C15—C16—H16A	109.3
O1—C7—C6	120.35 (13)	O2—C16—H16B	109.3
O1—C7—C8	120.85 (12)	C15—C16—H16B	109.3
C6—C7—C8	118.80 (11)	H16A—C16—H16B	108.0
C7—C8—C9	113.98 (10)	O2—C17—C18	111.37 (13)
C7—C8—H8A	108.8	O2—C17—H17A	109.4
C9—C8—H8A	108.8	C18—C17—H17A	109.4
C7—C8—H8B	108.8	O2—C17—H17B	109.4
C9—C8—H8B	108.8	C18—C17—H17B	109.4
H8A—C8—H8B	107.7	H17A—C17—H17B	108.0
N2—C9—C8	110.20 (10)	N2—C18—C17	110.24 (12)
N2—C9—C10	114.37 (10)	N2—C18—H18A	109.6
C8—C9—C10	112.07 (11)	C17—C18—H18A	109.6
N2—C9—H9	106.6	N2—C18—H18B	109.6
C8—C9—H9	106.6	C17—C18—H18B	109.6
C10—C9—H9	106.6	H18A—C18—H18B	108.1
N1—C10—C14	121.74 (13)	C10—N1—C11	116.89 (13)
N1—C10—C9	116.80 (11)	C18—N2—C15	108.87 (11)
C14—C10—C9	121.45 (12)	C18—N2—C9	112.48 (10)
N1—C11—C12	124.40 (17)	C15—N2—C9	115.74 (11)
N1—C11—H11	117.8	C16—O2—C17	109.40 (12)
C12—C11—H11	117.8		
C6—C1—C2—C3	−0.8 (3)	N1—C11—C12—C13	0.0 (3)
C1—C2—C3—C4	1.2 (3)	C11—C12—C13—C14	−1.1 (2)
C2—C3—C4—C5	−0.3 (3)	C12—C13—C14—C10	1.3 (2)

C3—C4—C5—C6	−0.9 (3)	N1—C10—C14—C13	−0.4 (2)
C2—C1—C6—C5	−0.5 (2)	C9—C10—C14—C13	−179.19 (13)
C2—C1—C6—C7	179.11 (16)	N2—C15—C16—O2	59.10 (17)
C4—C5—C6—C1	1.3 (2)	O2—C17—C18—N2	−58.71 (16)
C4—C5—C6—C7	−178.26 (14)	C14—C10—N1—C11	−0.6 (2)
C1—C6—C7—O1	−10.1 (2)	C9—C10—N1—C11	178.23 (13)
C5—C6—C7—O1	169.48 (15)	C12—C11—N1—C10	0.8 (3)
C1—C6—C7—C8	170.08 (13)	C17—C18—N2—C15	57.48 (15)
C5—C6—C7—C8	−10.3 (2)	C17—C18—N2—C9	−172.87 (12)
O1—C7—C8—C9	5.6 (2)	C16—C15—N2—C18	−57.27 (15)
C6—C7—C8—C9	−174.59 (11)	C16—C15—N2—C9	174.89 (11)
C7—C8—C9—N2	−162.92 (11)	C8—C9—N2—C18	168.06 (11)
C7—C8—C9—C10	68.50 (15)	C10—C9—N2—C18	−64.62 (14)
N2—C9—C10—N1	−79.35 (15)	C8—C9—N2—C15	−65.92 (14)
C8—C9—C10—N1	47.00 (15)	C10—C9—N2—C15	61.40 (14)
N2—C9—C10—C14	99.47 (14)	C15—C16—O2—C17	−58.86 (18)
C8—C9—C10—C14	−134.18 (13)	C18—C17—O2—C16	58.37 (16)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg2 is the centroid of the C10—C14/N1 ring.

$D\cdots H\cdots A$	$D\cdots H$	$H\cdots A$	$D\cdots A$	$D\cdots H\cdots A$
C2—H2 \cdots Cg2 ⁱ	0.93	2.90	3.780 (6)	159

Symmetry code: (i) $-x, -y+1, -z$.