



# Crystal structure of (2*E*,4*E*)-5-[bis(2-hydroxyethyl)amino]-1-(4-chlorophenyl)-5-phenylpenta-2,4-dien-1-one

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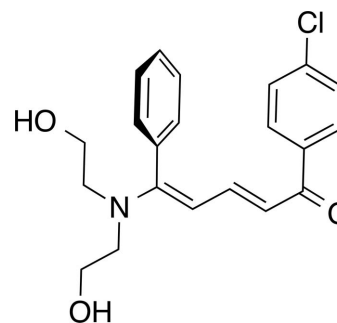
In the title compound, C<sub>21</sub>H<sub>22</sub>ClNO<sub>3</sub>, the pentadiene unit is nearly planar [maximum deviation = 0.023 (1) Å], but the carbonyl O atom deviates significantly [by 0.304 (1) Å] from its mean plane, which is twisted with respect to the phenyl and chlorobenzene rings by 71.34 (13) and 46.40 (13)°, respectively. In the crystal, inversion-related molecules are linked by two pairs of O—H...O hydrogen bonds, forming chains propagating along [01 $\bar{1}$ ], enclosing R<sup>2</sup><sub>2</sub>(16) and R<sup>2</sup><sub>2</sub>(22) ring motifs. The chains are linked *via* C—H...O hydrogen bonds and C—H... $\pi$  interactions into a three-dimensional supramolecular architecture.

**Keywords:** crystal structure; dienes; enamines; hydrogen bonding; C—H... $\pi$  interactions.

**CCDC reference:** 1431635

## 1. Related literature

For crystal structures of 1-aryl-5-phenylpenta-2,4-dien-1-ones, see: Kashino & Haisa (1980); Fischer *et al.* (2007*a,b*); Patil *et al.* (2007); Zhao *et al.* (2007); Silva *et al.* (2011); Vologzhanina *et al.* (2013); Golovanov *et al.* (2014). For non-linear optical properties of 1,5-diarylpent-2,4-dien-1-ones, see: Singh & Miyata (1996). For the biological activity of related chalcones, see: Karaman *et al.* (2012); Nielsen *et al.* (2005); Wu *et al.* (2011).



## 2. Experimental

### 2.1. Crystal data

C<sub>21</sub>H<sub>22</sub>ClNO<sub>3</sub>  
 $M_r$  = 371.85  
 Triclinic,  $P\bar{1}$   
 $a$  = 6.6258 (1) Å  
 $b$  = 11.0019 (2) Å  
 $c$  = 13.8592 (3) Å  
 $\alpha$  = 110.980 (1)°  
 $\beta$  = 99.401 (2)°  
 $\gamma$  = 93.338 (1)°  
 $V$  = 923.14 (3) Å<sup>3</sup>  
 $Z$  = 2  
 Cu  $K\alpha$  radiation  
 $\mu$  = 2.00 mm<sup>-1</sup>  
 $T$  = 120 K  
 0.18 × 0.06 × 0.06 mm

### 2.2. Data collection

Bruker APEXII CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 2005)  
 $T_{\min}$  = 0.715,  $T_{\max}$  = 0.890  
 8297 measured reflections  
 3028 independent reflections  
 2686 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}}$  = 0.030

### 2.3. Refinement

$R[F^2 > 2\sigma(F^2)]$  = 0.031  
 $wR(F^2)$  = 0.092  
 $S$  = 0.99  
 3028 reflections  
 235 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max}$  = 0.19 e Å<sup>-3</sup>  
 $\Delta\rho_{\min}$  = -0.20 e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C12–C17 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O2—H2B...O3 <sup>i</sup>	0.84	1.92	2.7475 (15)	168
O3—H3B...O1 <sup>ii</sup>	0.84	1.87	2.6983 (16)	169
C7—H7A...O1 <sup>iii</sup>	0.95	2.59	3.497 (2)	159
C20—H20B...O2 <sup>iv</sup>	0.99	2.49	3.3396 (18)	143
C21—H21A...Cg1 <sup>iii</sup>	0.99	2.73	3.5791 (17)	144

Symmetry codes: (i)  $-x+1, -y, -z$ ; (ii)  $-x+2, -y+1, -z+1$ ; (iii)  $-x+1, -y+1, -z+1$ ; (iv)  $x+1, y, z$ .

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and publCIF (Westrip, 2010).

## Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: XU5877).

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## supporting information

*Acta Cryst.* (2015). E71, o870–o871 [https://doi.org/10.1107/S2056989015019568]

## Crystal structure of (2*E*,4*E*)-5-[bis(2-hydroxyethyl)amino]-1-(4-chlorophenyl)-5-phenylpenta-2,4-dien-1-one

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### S1. Structural commentary

By the reaction between diethanolamine and (*E*)-1-(4-chlorophenyl)-5-phenylpent-2-en-4-yn-1-one the title compound was synthesized.

All bond lengths and valence angles are characteristic of single, double and aromatic bonds (Allen *et al.*, 1987), although the length of the C3—C4 bond (1.416 (2) Å) indicates slight delocalization of electron density along polyenone chain. In contrast with previously characterized 1-aryl-5-phenylpent-2,4-dien-1-ones (Kashino & Haisa, 1980; Fischer *et al.*, 2007a,b) Patil *et al.*, 2007; Zhao *et al.*, 2007; Silva *et al.*, 2011) and the (*E*,*Z*)-1-(4-chlorophenyl)-5-phenyl-5-(phenylsulfanyl)penta-2,4-dien-1-one (Vologzhanina *et al.*, 2013), the title compound adopts the *cis*-orientation of C(3) and C(6) atoms in respect to the C(4)=C(5) double bond (Figure S1) which was previously observed only for (*E*,*E*)-1-(4-chlorophenyl)-5-phenyl-5-(piperidin-1-yl)penta-2,4-dien-1-one (Golovanov *et al.*, 2014). Besides, it is the first representative of 1-aryl-5-phenylpent-2,4-dien-1-ones with the *s-trans* conformation of the enone fragment. As the result coplanarity between pentdienone and phenyl rings is absent, whilst the other 1,5-diarylpentdienones are *quasi*-planar. The angles between the meanplane of the pent-2,4-dien-1-one chain (RMSD = 0.11 (8) Å) and those of chlorophen-4-yl and phenyl rings are equal to, respectively, 137.65 (6) and 72.48 (4) °.

Due to the presence of two donor H(O) atoms, hydrogen bonding realizes in the crystal of the title compound. Despite the presence of chlorine and nitrogen atoms, only O—H...O bifurcate bonding was found with the oxygen atom of keto-group (Figure S2). The resulting H-bonded chain motif is characterized by O...O distances as short as 2.748 (2) and 2.698 (2) Å and OHO angles equal to 168 and 169 °.

### S2. Synthesis and crystallization

A solution of (499 mg, 1.87 mmol) (*E*)-1-(4-chlorophenyl)-5-phenylpent-2-en-4-yn-1-one and (236 mg, 2.24 mmol) diethanolamine in 95% EtOH (7 ml) was heated 10 h under reflux. The mixture was cooled, and the precipitate of adduct was filtered off, washed on a filter with 2 ml of cold 50% EtOH, and dried. Yield is 87 %. The single crystals of the product were obtained by slow crystallization from 95% EtOH. M.p. 370–371 K.

### S3. Refinement

H atoms were placed in the calculated positions with O—H = 0.84 and C—H = 0.95–0.99 Å, and refined in ride mode with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$  and  $1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $1.2U_{\text{iso}}(\text{C})$  for the others.

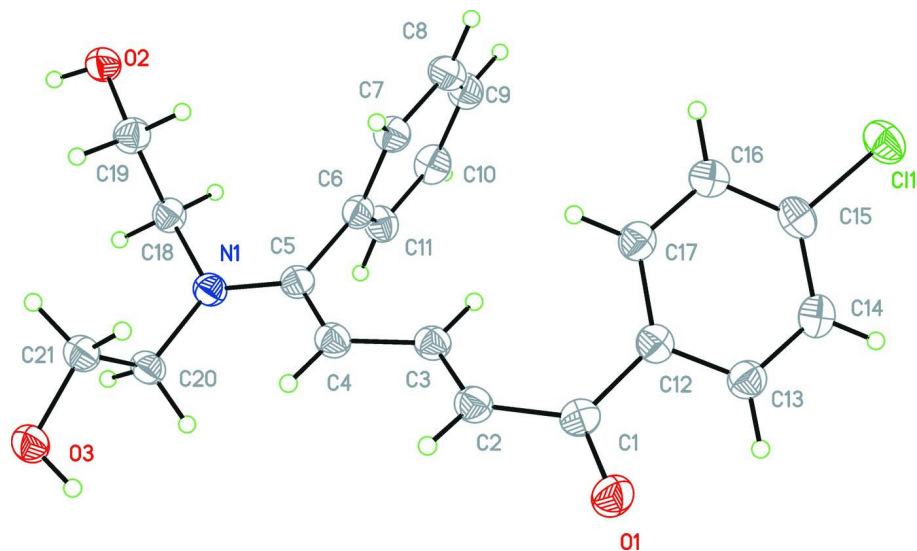


Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

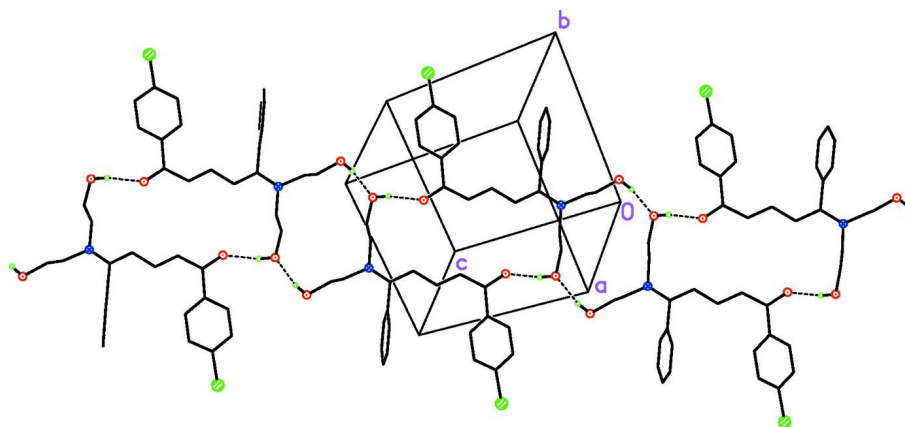


Figure 2

Fragment of a classic H-bonded chain (shown with dashed lines). The H(C) atoms are omitted for clarity.

**(2*E*,4*E*)-5-[Bis(2-hydroxyethyl)amino]-1-(4-chlorophenyl)-5-phenylpenta-2,4-dien-1-one**

*Crystal data*

$C_{21}H_{22}ClNO_3$

$M_r = 371.85$

Triclinic,  $P\bar{1}$

Hall symbol:  $-P\ 1$

$a = 6.6258\ (1)\ \text{\AA}$

$b = 11.0019\ (2)\ \text{\AA}$

$c = 13.8592\ (3)\ \text{\AA}$

$\alpha = 110.980\ (1)^\circ$

$\beta = 99.401\ (2)^\circ$

$\gamma = 93.338\ (1)^\circ$

$V = 923.14\ (3)\ \text{\AA}^3$

$Z = 2$

$F(000) = 392$

$D_x = 1.338\ \text{Mg m}^{-3}$

Melting point: 370 K

Cu  $K\alpha$  radiation,  $\lambda = 1.54178\ \text{\AA}$

Cell parameters from 2512 reflections

$\theta = 3.5\text{--}67.5^\circ$

$\mu = 2.00\ \text{mm}^{-1}$

$T = 120\ \text{K}$

Needle, yellow

$0.18 \times 0.06 \times 0.06\ \text{mm}$

*Data collection*

Bruker APEXII CCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan  
 (SADABS; Bruker, 2005)  
 $T_{\min} = 0.715$ ,  $T_{\max} = 0.890$

8297 measured reflections  
 3028 independent reflections  
 2686 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.030$   
 $\theta_{\max} = 64.9^\circ$ ,  $\theta_{\min} = 3.5^\circ$   
 $h = -7 \rightarrow 7$   
 $k = -12 \rightarrow 12$   
 $l = -14 \rightarrow 16$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.031$   
 $wR(F^2) = 0.092$   
 $S = 0.99$   
 3028 reflections  
 235 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.060P)^2 + 0.130P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.005$   
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.20 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.20029 (6)	1.12487 (4)	0.63194 (3)	0.03690 (14)
O1	0.91844 (17)	0.73532 (11)	0.65905 (8)	0.0324 (3)
O2	0.01218 (16)	0.13121 (10)	−0.03794 (8)	0.0299 (3)
H2B	0.0429	0.0713	−0.0890	0.045*
O3	0.83459 (16)	0.06098 (10)	0.18632 (8)	0.0307 (3)
H3B	0.9239	0.1182	0.2330	0.046*
N1	0.52198 (18)	0.30611 (12)	0.12564 (9)	0.0246 (3)
C1	0.7721 (2)	0.71163 (14)	0.58302 (11)	0.0256 (3)
C2	0.7492 (2)	0.59076 (14)	0.49358 (12)	0.0266 (3)
H2A	0.8258	0.5243	0.5034	0.032*
C3	0.6301 (2)	0.56068 (14)	0.39630 (12)	0.0249 (3)
H3A	0.5444	0.6229	0.3857	0.030*
C4	0.6263 (2)	0.44205 (14)	0.30992 (12)	0.0254 (3)
H4A	0.7134	0.3803	0.3201	0.030*
C5	0.5039 (2)	0.41119 (14)	0.21228 (12)	0.0241 (3)
C6	0.3479 (2)	0.49831 (14)	0.19510 (11)	0.0246 (3)

C7	0.1775 (2)	0.51077 (15)	0.24391 (12)	0.0274 (3)
H7A	0.1595	0.4631	0.2877	0.033*
C8	0.0346 (2)	0.59274 (16)	0.22843 (13)	0.0318 (3)
H8A	−0.0810	0.6011	0.2618	0.038*
C9	0.0598 (2)	0.66242 (15)	0.16460 (13)	0.0322 (4)
H9A	−0.0384	0.7182	0.1540	0.039*
C10	0.2289 (3)	0.65052 (15)	0.11611 (13)	0.0325 (4)
H10A	0.2460	0.6980	0.0720	0.039*
C11	0.3729 (2)	0.56964 (15)	0.13177 (12)	0.0295 (3)
H11A	0.4894	0.5628	0.0991	0.035*
C12	0.6260 (2)	0.81178 (15)	0.58804 (11)	0.0254 (3)
C13	0.7033 (2)	0.94446 (15)	0.64078 (12)	0.0290 (3)
H13A	0.8468	0.9689	0.6691	0.035*
C14	0.5731 (3)	1.04078 (15)	0.65230 (12)	0.0311 (3)
H14A	0.6268	1.1309	0.6867	0.037*
C15	0.3643 (2)	1.00379 (15)	0.61308 (12)	0.0296 (3)
C16	0.2823 (2)	0.87320 (15)	0.56061 (12)	0.0285 (3)
H16A	0.1383	0.8495	0.5337	0.034*
C17	0.4143 (2)	0.77749 (15)	0.54807 (11)	0.0267 (3)
H17A	0.3600	0.6877	0.5119	0.032*
C18	0.3543 (2)	0.25212 (14)	0.03273 (11)	0.0262 (3)
H18A	0.4102	0.1966	−0.0281	0.031*
H18B	0.2993	0.3252	0.0155	0.031*
C19	0.1793 (2)	0.17064 (15)	0.04944 (12)	0.0287 (3)
H19A	0.2299	0.0920	0.0590	0.034*
H19B	0.1316	0.2230	0.1142	0.034*
C20	0.6923 (2)	0.22827 (14)	0.13012 (12)	0.0263 (3)
H20A	0.8160	0.2875	0.1771	0.032*
H20B	0.7240	0.1899	0.0587	0.032*
C21	0.6511 (2)	0.11742 (15)	0.16928 (12)	0.0277 (3)
H21A	0.5997	0.1523	0.2359	0.033*
H21B	0.5436	0.0492	0.1165	0.033*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0381 (2)	0.0326 (2)	0.0451 (3)	0.01395 (16)	0.01386 (17)	0.01676 (17)
O1	0.0319 (6)	0.0311 (6)	0.0294 (6)	0.0036 (4)	−0.0017 (4)	0.0091 (4)
O2	0.0251 (5)	0.0301 (6)	0.0299 (6)	0.0051 (4)	0.0018 (4)	0.0072 (4)
O3	0.0286 (6)	0.0287 (5)	0.0301 (6)	0.0081 (4)	0.0001 (4)	0.0073 (4)
N1	0.0233 (6)	0.0258 (6)	0.0246 (7)	0.0044 (5)	0.0055 (5)	0.0087 (5)
C1	0.0249 (7)	0.0279 (7)	0.0256 (8)	0.0010 (6)	0.0049 (6)	0.0122 (6)
C2	0.0246 (7)	0.0262 (7)	0.0300 (8)	0.0047 (6)	0.0052 (6)	0.0116 (6)
C3	0.0227 (7)	0.0255 (7)	0.0286 (8)	0.0029 (5)	0.0069 (6)	0.0117 (6)
C4	0.0248 (7)	0.0251 (7)	0.0283 (8)	0.0049 (6)	0.0064 (6)	0.0116 (6)
C5	0.0230 (7)	0.0236 (7)	0.0277 (8)	0.0019 (5)	0.0081 (5)	0.0106 (6)
C6	0.0248 (7)	0.0227 (7)	0.0236 (8)	0.0023 (6)	0.0026 (5)	0.0066 (5)
C7	0.0277 (8)	0.0278 (7)	0.0290 (8)	0.0031 (6)	0.0068 (6)	0.0128 (6)

C8	0.0267 (8)	0.0323 (8)	0.0377 (9)	0.0067 (6)	0.0090 (6)	0.0129 (6)
C9	0.0314 (8)	0.0294 (8)	0.0348 (9)	0.0083 (6)	0.0008 (6)	0.0125 (6)
C10	0.0392 (9)	0.0305 (8)	0.0310 (9)	0.0042 (7)	0.0044 (6)	0.0165 (6)
C11	0.0325 (8)	0.0290 (8)	0.0287 (8)	0.0055 (6)	0.0092 (6)	0.0113 (6)
C12	0.0283 (8)	0.0286 (7)	0.0211 (8)	0.0042 (6)	0.0066 (5)	0.0105 (6)
C13	0.0278 (8)	0.0305 (8)	0.0274 (8)	0.0024 (6)	0.0046 (6)	0.0099 (6)
C14	0.0364 (9)	0.0263 (7)	0.0290 (8)	0.0033 (6)	0.0061 (6)	0.0088 (6)
C15	0.0343 (8)	0.0310 (8)	0.0288 (8)	0.0109 (6)	0.0116 (6)	0.0142 (6)
C16	0.0272 (8)	0.0324 (8)	0.0275 (8)	0.0043 (6)	0.0057 (6)	0.0130 (6)
C17	0.0289 (8)	0.0279 (7)	0.0231 (8)	0.0021 (6)	0.0055 (6)	0.0092 (6)
C18	0.0276 (8)	0.0270 (7)	0.0227 (8)	0.0040 (6)	0.0044 (6)	0.0077 (6)
C19	0.0277 (8)	0.0301 (8)	0.0273 (8)	0.0031 (6)	0.0039 (6)	0.0105 (6)
C20	0.0230 (7)	0.0284 (7)	0.0271 (8)	0.0069 (6)	0.0072 (6)	0.0084 (6)
C21	0.0258 (8)	0.0277 (7)	0.0283 (8)	0.0062 (6)	0.0041 (6)	0.0089 (6)

*Geometric parameters (Å, °)*

Cl1—C15	1.7423 (15)	C9—H9A	0.9500
O1—C1	1.2483 (19)	C10—C11	1.385 (2)
O2—C19	1.4179 (18)	C10—H10A	0.9500
O2—H2B	0.8400	C11—H11A	0.9500
O3—C21	1.4209 (17)	C12—C17	1.397 (2)
O3—H3B	0.8400	C12—C13	1.398 (2)
N1—C5	1.3644 (19)	C13—C14	1.387 (2)
N1—C20	1.4624 (18)	C13—H13A	0.9500
N1—C18	1.4672 (19)	C14—C15	1.380 (2)
C1—C2	1.436 (2)	C14—H14A	0.9500
C1—C12	1.499 (2)	C15—C16	1.386 (2)
C2—C3	1.361 (2)	C16—C17	1.389 (2)
C2—H2A	0.9500	C16—H16A	0.9500
C3—C4	1.416 (2)	C17—H17A	0.9500
C3—H3A	0.9500	C18—C19	1.522 (2)
C4—C5	1.373 (2)	C18—H18A	0.9900
C4—H4A	0.9500	C18—H18B	0.9900
C5—C6	1.4978 (19)	C19—H19A	0.9900
C6—C11	1.393 (2)	C19—H19B	0.9900
C6—C7	1.397 (2)	C20—C21	1.530 (2)
C7—C8	1.388 (2)	C20—H20A	0.9900
C7—H7A	0.9500	C20—H20B	0.9900
C8—C9	1.385 (2)	C21—H21A	0.9900
C8—H8A	0.9500	C21—H21B	0.9900
C9—C10	1.387 (2)		
C19—O2—H2B	109.5	C13—C12—C1	118.42 (13)
C21—O3—H3B	109.5	C14—C13—C12	120.79 (15)
C5—N1—C20	120.41 (12)	C14—C13—H13A	119.6
C5—N1—C18	121.58 (12)	C12—C13—H13A	119.6
C20—N1—C18	117.00 (11)	C15—C14—C13	119.04 (15)

O1—C1—C2	119.50 (13)	C15—C14—H14A	120.5
O1—C1—C12	118.48 (13)	C13—C14—H14A	120.5
C2—C1—C12	122.02 (13)	C14—C15—C16	121.69 (14)
C3—C2—C1	127.31 (13)	C14—C15—C11	118.85 (12)
C3—C2—H2A	116.3	C16—C15—C11	119.45 (12)
C1—C2—H2A	116.3	C17—C16—C15	118.86 (14)
C2—C3—C4	123.85 (13)	C17—C16—H16A	120.6
C2—C3—H3A	118.1	C15—C16—H16A	120.6
C4—C3—H3A	118.1	C16—C17—C12	120.77 (14)
C5—C4—C3	123.74 (13)	C16—C17—H17A	119.6
C5—C4—H4A	118.1	C12—C17—H17A	119.6
C3—C4—H4A	118.1	N1—C18—C19	112.61 (12)
N1—C5—C4	123.35 (13)	N1—C18—H18A	109.1
N1—C5—C6	116.34 (13)	C19—C18—H18A	109.1
C4—C5—C6	120.23 (13)	N1—C18—H18B	109.1
C11—C6—C7	119.24 (13)	C19—C18—H18B	109.1
C11—C6—C5	120.35 (13)	H18A—C18—H18B	107.8
C7—C6—C5	120.40 (13)	O2—C19—C18	110.76 (12)
C8—C7—C6	120.05 (14)	O2—C19—H19A	109.5
C8—C7—H7A	120.0	C18—C19—H19A	109.5
C6—C7—H7A	120.0	O2—C19—H19B	109.5
C9—C8—C7	120.30 (15)	C18—C19—H19B	109.5
C9—C8—H8A	119.9	H19A—C19—H19B	108.1
C7—C8—H8A	119.9	N1—C20—C21	114.63 (12)
C10—C9—C8	119.85 (14)	N1—C20—H20A	108.6
C10—C9—H9A	120.1	C21—C20—H20A	108.6
C8—C9—H9A	120.1	N1—C20—H20B	108.6
C11—C10—C9	120.18 (15)	C21—C20—H20B	108.6
C11—C10—H10A	119.9	H20A—C20—H20B	107.6
C9—C10—H10A	119.9	O3—C21—C20	110.36 (12)
C10—C11—C6	120.38 (14)	O3—C21—H21A	109.6
C10—C11—H11A	119.8	C20—C21—H21A	109.6
C6—C11—H11A	119.8	O3—C21—H21B	109.6
C17—C12—C13	118.83 (14)	C20—C21—H21B	109.6
C17—C12—C1	122.62 (13)	H21A—C21—H21B	108.1
O1—C1—C2—C3	−164.68 (15)	C5—C6—C11—C10	−179.81 (14)
C12—C1—C2—C3	14.9 (2)	O1—C1—C12—C17	−142.76 (15)
C1—C2—C3—C4	175.65 (14)	C2—C1—C12—C17	37.7 (2)
C2—C3—C4—C5	179.10 (15)	O1—C1—C12—C13	33.1 (2)
C20—N1—C5—C4	−7.3 (2)	C2—C1—C12—C13	−146.49 (15)
C18—N1—C5—C4	160.89 (14)	C17—C12—C13—C14	−0.8 (2)
C20—N1—C5—C6	169.64 (12)	C1—C12—C13—C14	−176.78 (13)
C18—N1—C5—C6	−22.20 (19)	C12—C13—C14—C15	1.6 (2)
C3—C4—C5—N1	170.29 (14)	C13—C14—C15—C16	−1.4 (2)
C3—C4—C5—C6	−6.5 (2)	C13—C14—C15—C11	177.71 (11)
N1—C5—C6—C11	−64.96 (19)	C14—C15—C16—C17	0.5 (2)
C4—C5—C6—C11	112.05 (17)	C11—C15—C16—C17	−178.65 (11)



N1—C5—C6—C7	116.19 (15)	C15—C16—C17—C12	0.4 (2)
C4—C5—C6—C7	−66.80 (19)	C13—C12—C17—C16	−0.2 (2)
C11—C6—C7—C8	0.5 (2)	C1—C12—C17—C16	175.64 (13)
C5—C6—C7—C8	179.36 (14)	C5—N1—C18—C19	−76.12 (17)
C6—C7—C8—C9	0.1 (2)	C20—N1—C18—C19	92.42 (15)
C7—C8—C9—C10	−0.2 (2)	N1—C18—C19—O2	174.26 (11)
C8—C9—C10—C11	−0.3 (2)	C5—N1—C20—C21	86.90 (16)
C9—C10—C11—C6	0.9 (2)	C18—N1—C20—C21	−81.78 (16)
C7—C6—C11—C10	−0.9 (2)	N1—C20—C21—O3	−171.02 (11)

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

Cg1 is the centroid of the C12–C17 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O2—H2B $\cdots$ O3 <sup>i</sup>	0.84	1.92	2.7475 (15)	168
O3—H3B $\cdots$ O1 <sup>ii</sup>	0.84	1.87	2.6983 (16)	169
C7—H7A $\cdots$ O1 <sup>iii</sup>	0.95	2.59	3.497 (2)	159
C20—H20B $\cdots$ O2 <sup>iv</sup>	0.99	2.49	3.3396 (18)	143
C21—H21A $\cdots$ Cg1 <sup>iii</sup>	0.99	2.73	3.5791 (17)	144

Symmetry codes: (i)  $-x+1, -y, -z$ ; (ii)  $-x+2, -y+1, -z+1$ ; (iii)  $-x+1, -y+1, -z+1$ ; (iv)  $x+1, y, z$ .