



## Analytical Methods

## Preliminary results of mercury levels in raw and cooked seafood and their public health impact

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## ABSTRACT

Mercury is toxic for human health and one of the main routes of exposure is through consumption of contaminated fish and shellfish. The objective of this work was to assess the possible mercury contamination of bivalves (*Anomalocardia brasiliensis*, *Lucina pectinata*, *Callinectes sapidus*), crustacean (*C. sapidus*) and fish (*Bagre marinus* and *Diapterus rhombeus*) collected on Salinas da Margarida, BA (Brazil), a region which carciniculture, fishing and shellfish extraction are the most important economic activities. The effect of cooking on Hg concentration in the samples was also studied. The results showed that Hg concentration was generally higher in the cooked samples than in raw samples. This increase can be related to the effect of Hg pre-concentration, formation of complexes involving mercury species and sulfhydryl groups present in tissues and/or loss of water and fat. The highest concentrations were found in *B. marinus* samples ranging 837.0–1585.3  $\mu\text{g kg}^{-1}$ , which exceeded those recommended by Brazilian Health Surveillance Agency (ANVISA). In addition, Hg values found in the other samples also suggest the monitoring of the Hg concentrations in seafood consumed from the region.

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## 1. Introduction

Mercury (Hg) is an environmental pollutant due to its high toxicity, even at low concentrations, and its capacity to bioaccumulate in organisms and biomagnify in the food chain (US EPA, 1997, 2003; Porto, Araujo, & Feldberga, 2005; UNEP, 2013). In the environment, the inorganic forms ( $\text{Hg}^0$  and  $\text{Hg(II)}$ ) are easily transformed in organic species (Bisinoti & Jardim, 2003; Poulin & Gibb, 2008; Voegborlo & Akagi, 2007). Among the main transformations is the formation of methylmercury (MeHg) (Blum, Popp, Drazen, Choy, & Johnson, 2013). This chemical species enters into the aquatic food chain and reaches its maximum concentration in the tissues of fish at the top of this chain, and MeHg accounts for 80% to 95% of the total mercury (Afonso et al., 2008; Blum et al., 2013; EFSA, 2012). The its most serious effects in humans are damages in the nervous system (FAO/WHO, 2004; Hightower & Moore, 2003).

Due to the capable of Hg accumulating along the trophic chain, especially in aquatic environments, human exposure to mercury occurs mainly through the consumption of fish and seafood in which a significant concentration of mercury is present (Hightower & Moore, 2003; US EPA, 2009). Additionally, an increase of mercury concentrations has been observed around

the world in several fish species along the last decades, some with levels exceeding human toxicological thresholds (US FDA, 2013; US EPA, 1997, 2003, 2004, 2009). The World Health Organization (WHO) assumes that foods with mercury concentrations of 0.5  $\text{mg kg}^{-1}$  (wet weight) or more are inadequate for human consumption (WHO, 1996). In Brazil, the maximum limits of mercury are 0.5  $\text{mg kg}^{-1}$  and 1.0  $\text{mg kg}^{-1}$  for non-carnivorous and predatory fish, respectively (ANVISA, 1998, 2013).

Recent studies evaluated the Hg concentrations and their species after different culinary treatments in fish, and according to the authors no significant conversion of Hg species or variation in the total Hg concentrations were observed by this treatments (Schmidt et al., 2015). However, there are still many controversies about the influence of cooking in the Hg concentrations (Amyot & Ouédraogo, 2011; Kalogeropoulos et al., 2012; Perello, Martí-Cid, Llobet, & Domingo, 2008; Schmidt et al., 2015).

Salinas da Margarida is a Brazilian municipality, which belongs to the state of Bahia and it is known as one of the major Brazilian producers and distributors of shrimp. In addition to carciniculture, the main economic activities are fishing and shellfish extraction (IBGE – Instituto Brasileiro de Geografia e Estatística., 2013). Around the year of 2009, an oil rig was implanted in São Roque do Paraguaçu, a locality near Salinas da Margarida, and it is known that Hg, as well as some other trace elements, is present in all geological hydrocarbons (Sofowote et al., 2011).

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The present work is the first one to examine the Hg concentrations, in cooked and raw samples of *Anomalocardia brasiliana*, *Lucina pectinata*, *Callinectes sapidus*, *Bagre marinus* and *Diapterus rhombeus* from Salinas da Margarida, Todos os Santos Bay (TSB), in order to evaluate the possible Hg contamination in the commercialized and consumed seafood after the implantation of the oil rig.

## 2. Materials and methods

### 2.1. Sampling

Samples of two bivalve mollusks (*L. pectinata*, *A. brasiliana*), one crustacean (*C. sapidus*) and two fish (*D. rhombeus* and *B. marinus*) were collected during two periods, in May and June of 2013, on Salinas da Margarida. Two samples of each organism were collected in both sampling periods, totalizing 10 samples. Approximately 1.0 kg of *A. brasiliana*, *L. pectinata* and *C. sapidus* were collected in surface sediment (up to 20 cm deep) during low tide. Ten *D. rhombeus*, up to 150 g each, were caught with cast nets at depths up to 10 m. Three *B. marinus*, approximately 1.0 kg each, were caught by hooks and fishing line at depths between 20 and 50 m. Immediately after being caught, fish were slaughtered by severing the spinal cord, in accordance with Brazilian laws (BRASIL, 2000). The fish samples were identified, labeled in separate polyethylene bags, transferred in ice boxes, and transported to the laboratory. Then muscle of samples was removed and homogenized.

### 2.2. Reagents

Analytical grade reagents were utilized (Merck, Darmstadt, Germany), a standard reference solution of  $1.000 \text{ mg L}^{-1}$  (Merck, Darmstadt, Germany) was used to build the Hg analytical curve, and purified water obtained from a Milli-Q system (Millipore, Billerica, MA, USA) was used to prepare all of the aqueous solutions.

### 2.3. Sample preparation

All samples were separated in two groups for the evaluation of the mercury levels before and after cooking to observe if there are some variations in the Hg concentrations in the cooked samples. Therefore, one part of the sample was only dried in an oven with air circulation at  $60 \text{ }^\circ\text{C}$  for 48 h, and it was then ground in an impact mill. The other part was placed in a beaker with approximately 1 L of ultrapure water and heated on a hot plate for 30 min at a temperature of  $80 \text{ }^\circ\text{C}$ . After cooking, the samples were dried and ground under the same conditions as the raw samples. It is noteworthy that the analysis was performed on entire crustacean, the shells were removed from the bivalve mollusks before analysis, and only the muscles of the fish were analyzed.

### 2.4. Apparatus

The analyses were carried out with a Direct Mercury Analyzer<sup>®</sup> (DMA-80, Milestone Srl, Italy), according to adapted method US EPA (2007). This equipment typically contains an automatic sampler, a quartz furnace, a cobalt-manganese oxide catalyst, a gold-coated sand amalgamator and an atomic absorption detection cell with three different path lengths (165, 120 and 4 mm). The sample is introduced into the quartz furnace, where it is heated up to  $200 \text{ }^\circ\text{C}$  (drying temperature) for 60 s and  $650 \text{ }^\circ\text{C}$  (ashing temperature, maximum temperature allowed by the software of the equipment) for 105 s, which allows Hg volatilization and reduction. Air (99.99%) was used as the combustion and carrier gases (Melendez-Perez & Fostier, 2013).

### 2.5. Optimization tests

It was first assessed whether the measurement of the mercury concentration could be significantly affected by the sample amount analyzed. Some collected samples (bivalves, fish and crustaceans) were used to assess the effect of mass variation between 5 and 100 mg.

### 2.6. Validation of the analytical method

The limits of detection (LOD) and quantitation (LOQ) were determined as 3 and 10 times the standard deviation of the residuals from the linear regression, respectively, divided by the angular coefficient value of the linear equation (Miller & Miller, 2000). For each type of sample (fish, crustacean and bivalves), the precision was assessed by the relative standard deviation from ten analytical replicates of one sample. The accuracy was checked by the addition and recovery of two concentrations of Hg(II), 1 ng and 10 ng, and by the analysis of the following three certified reference materials: DOLT-4 (Dogfisher liver), NIST 1566th (Oyster tissue) and MURST ISS-A2 (Antarctic krill). For the evaluation of matrix effects, calibration curves were constructed using samples spiked with known amounts of Hg(II) standard solution in each matrix. The slope of each curve was compared with the slope of the curve constructed in acidic solution.

### 2.7. Statistical analysis

Mercury contents were calculated on a fresh weight (fw) basis. Data were expressed as the mean  $\pm$  standard deviation, and the differences between the mercury contents of the raw and cooked samples in both sampling periods were assessed using a two-tailed Student's *t*-test for  $n = 3$  with a 95% confidence level.

## 3. Results and discussion

### 3.1. Effect of the sample mass

Fig. 1 shows the effect of the sample mass on the determined Hg concentrations for bivalves, fish and crustaceans. No significant difference was observed at the confidence level of 95% between masses of 10 and 50 mg for the crustacean samples, 5 and 30 mg for the bivalves and 20 and 60 mg for the fish. Thus, it was weighed 10 mg of bivalves and 30 mg of fish and crustacean.

### 3.2. Method validation

The limits of detection and quantification were 0.021 ng Hg and 0.072 ng Hg, respectively. The linear range was from 0.1 ng to 25 ng Hg. Accuracy was determined by the analysis of reference materials and addition and recovery tests. The results obtained were in good agreement with the certified values, with recoveries for the reference materials ranging from 91% to 107% and from 85% to 89% for the addition and recovery tests (Table 1). The precision (RSD) was generally less than 4% (Table 1).

### 3.3. Mercury concentrations in fish, bivalves and crustacean

Mercury concentrations were expressed in micrograms per kilogram wet weight taking into account the humidity which was of 77% for the raw fish and 80% for the raw bivalves and crustacean. The concentrations in the raw samples ranged 837.0–1585.3  $\mu\text{g kg}^{-1}$  for *B. marinus*, 53.0–212.0  $\mu\text{g kg}^{-1}$  for *D. rhombeus*, 365.0–725.0  $\mu\text{g kg}^{-1}$  for *L. pectinata*, 124.0–203.0  $\mu\text{g kg}^{-1}$  for *A. brasiliana* and 83.0–149.0  $\mu\text{g kg}^{-1}$  for *C. sapidus* (Fig. 2).

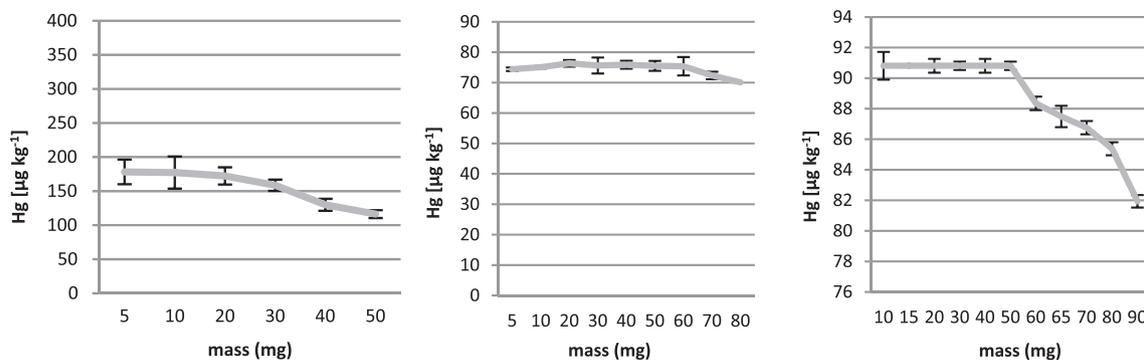


Fig. 1. Study of the effects of different bivalves, fish and crustacean sample masses (mean ± standard deviation) to n = 10.

Table 1

Recoveries from the Hg addition and of certified reference materials with precision (expressed as relative standard deviation).

Matrix	Sample value without addition (ng)	After addition of 1 ng Hg		After addition of 10 ng Hg		Precision (%)	Accuracy (%)
		Obtained value (ng)	Recovery (%)	Obtained value (ng)	Recovery (%)		
Fish	0.9	1.8	90	9.4	85	2	91 (DOLT-4)
Bivalve	6.5	7.4	90	15.3	88	4	107 (NIST 1566a)
Crab	3.2	4.2	100	12.1	89	3	103 (MURST-ISS A2)

The lowest and highest concentrations of mercury were obtained for the samples of *B. marinus* and *C. sapidus*, respectively. This can be related to the way each organism metabolizes the mercury furthermore factors such as each organism’s mass, size, age, and eating habits (Horvat, Lucotte, & Malm, 2007).

Regarding to the organisms studied in this work, fish take the highest position in the food chain; therefore, the highest concentrations of mercury in fish may be related to the processes of bioaccumulation and biomagnification (Blum et al., 2013; EFSA, 2012). On the other hand, the lower levels of Hg in crustaceans may be a consequence of the sample choices, as the samples were collected during a period of change, in which the crustacean loses its carapace and part of its contamination historic. It is worth mentioning that the sampling in the period of change was motivated by the high consumption of *C. sapidus* in this phase.

Many studies have investigated the concentrations of total mercury in fish, bivalves and crustaceans from different parts of the world (Bisi et al., 2012; Carbonel, Bravo, Fernández, & Tarazona,

2009; Hajeb et al., 2009; Ikem & Egiebor, 2005; McClain, Chumchal, Drenner, & Newland, 2006; Ordiano-Flores, Rosiles-Martínez, & Galván-Magaña, 2012; Schimidt et al., 2015; Zhang, Campbell, & Johnson, 2012). However, this present paper shows the latest determinations of the mercury in fish and shellfish of the TSB, unheard to Salinas da Margarida, and the results demonstrate the need for continued monitoring of element in this region, mainly in predatory fish that exceeded the limits of Brazilian recommendations and were considered inadequate for human consumption.

3.4. Influence of cooking on the mercury concentrations

During cooking, water content of fish tissues decreased on an average by 10–20% for boiled samples, therefore, the humidity values considered for calculation of wet weight were 67% for fish and 60% for bivalves and crustacean.

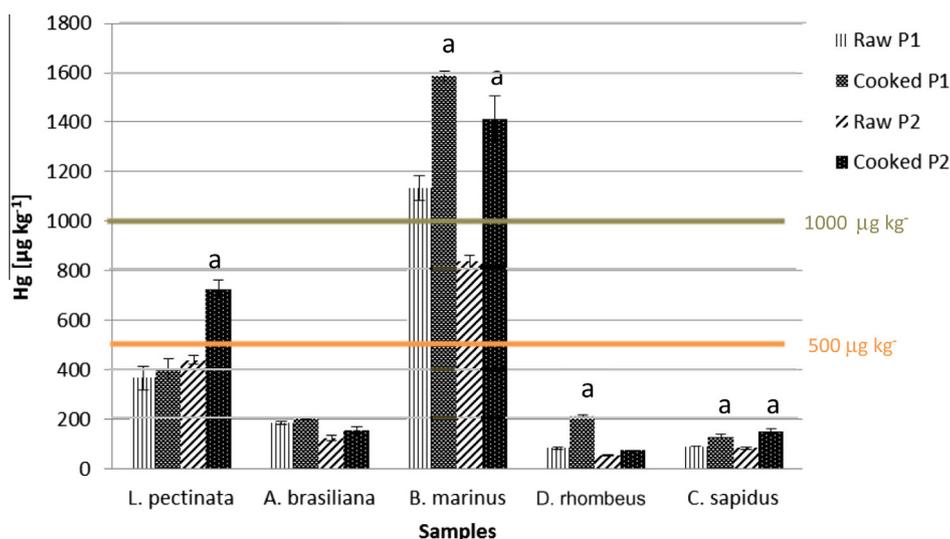


Fig. 2. Total mercury concentrations, in wet weight, with respective deviations shown by error bars for the raw and cooked samples from the 1st and 2nd sampling periods. The small letter “a” represents the results with significant difference (p < 0.05), according Student-t test, between raw and cooked samples.

In general, cooked samples presented higher Hg concentration than raw samples (Fig. 2) with significant increase ( $p < 0.05$ ) after cooking. A possible explanation for this result is that the heating can provide the formation of complexes between mercury species and sulfhydryl groups present in tissues, such as methylmercury-cysteine (Clarkson & Magos, 2006). This process can result in a reduction of the possible loss of mercury during the drying step. Another possibility is that the loss of some minerals in the form of salts during cooking may produce the effect of pre-concentrating of the mercury because of an increase in the mercury/mass ratio related to the mass decrease and the maintenance of the initial concentration of mercury. Lastly, the variation of the concentration can be in function of water loss during the cooking, due to the humidity in the cooked samples is lowest than in raw samples, thus, the Hg concentrations are higher in the most of cooked samples (Perelló et al., 2008).

The raw and cooked *B. marinus* samples from P2 and the cooked *B. marinus* samples from P2, besides the cooked *L. pectinata* samples from P2, presented mercury concentrations higher than the values that Brazilian recommendations considers acceptable for human consumption ( $0.5 \text{ mg kg}^{-1}$  for non-carnivorous fish and  $1.0 \text{ mg kg}^{-1}$  for predator fish). The remaining samples, despite having lower levels than those permitted by current legislation, warn of the necessity of regular monitoring of the food consumed in the region of Salinas da Margarida.

Kalogeropoulos et al. (2012) determined the concentrations of mercury and other elements in raw and cooked samples from 6 species of fish, squid, shrimp and mussels. The authors also showed that the mercury concentrations increased in all samples after cooking. Similar results were found for sardines and tuna by Perelló et al. (2008) and Amyot and Ouédraogo (2011), and according to Perelló et al. (2008) the increase of Hg levels can be due to loss of water and fat.

Although many authors have investigated the mercury concentrations in aquatic organisms around the world, there are still divergences if cooking modifies the results of Hg concentrations. Therefore, the results of previous studies that showed mercury concentrations above of specified limits may be significantly changed by the cooking process, thus, the new concentrations should be considered inadequate for human consumption. This approach can provide more representative data for calculations of the mercury exposure of the population of Salinas da Margarida through the ingestion of marine organisms.

#### 4. Conclusions

This paper presents the latest determinations of mercury in fish and shellfish sampled in the TSB, also taking account the consumption form of seafood by the population. According to our results, the intake of *B. marinus* from Salinas da Margarida could constitute a risk for people, therefore, it is necessary to constantly monitor the mercury in fish and seafood consumed in Salinas da Margarida. It was show that the Hg concentrations can vary significantly after cooking, thus, our results denote that it would be important to evaluate the levels of mercury in food based on the most common way of public consumption.

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#### References

- Afonso, C., Lourenço, H. M., Pereira, C., Martins, M. L., Carvalho, M. L., Castro, M., et al. (2008). Total and organic mercury, selenium and a-tocopherol in some deep-water species. *Journal of the Science of Food and Agriculture*, 88, 2543–2550.
- Amyot, M., & Ouédraogo (2011). Effects of various cooking methods and food components on bioaccessibility of mercury from fish. *Environmental Research*, 111, 1064–1069.
- Agência Nacional de Vigilância Sanitária – ANVISA. Portaria N° 685, de 27/08/1998. Available from <<http://e-legis.anvisa.gov.br/leisref/public/showAct.php?id=90>>.
- Agência Nacional de Vigilância Sanitária – ANVISA. (Brasil, 2013). RESOLUÇÃO – RDC No – 42, DE 29 DE AGOSTO DE 2013. Dispõe sobre o Regulamento Técnico MERCOSUL sobre Limites Máximos de Contaminantes Inorgânicos em Alimentos. In: <[http://portal.anvisa.gov.br/wps/wcm/connect/8100bb8040eac2e8b590b79cca79f4cf/RDC+n%C2%BA+42\\_2013\\_final.pdf?MOD=AJPERES](http://portal.anvisa.gov.br/wps/wcm/connect/8100bb8040eac2e8b590b79cca79f4cf/RDC+n%C2%BA+42_2013_final.pdf?MOD=AJPERES)> Accessed 13.03.14.
- Bisi, T. L., Lepoint, G., Azevedo, A. F., Dorneles, P. R., Flache, L., Dasd, K., et al. (2012). Trophic relationships and mercury biomagnification in Brazilian tropical coastal food webs. *Ecological Indicators*, 18, 291–302.
- Bisinoti, M. C., & Jardim, W. F. (2003). Production of organic mercury from Hg<sup>0</sup>: experiments using microcosms. *Journal of the Brazilian Chemical Society*, 14, 244–248.
- Blum, J. D., Popp, B. N., Drazen, J. C., Choy, A., & Johnson, M. W. (2013). Methylmercury production below the mixed layer in the North Pacific Ocean. *Nature Geoscience*, 6, 879–884.
- BRASIL. Ministério da Agricultura. Instrução Normativa n° 3, de 7 de Janeiro de 2000. Regulamento técnico de métodos de insensibilização para o abate humanitário de animais de açougue. S.D.A./M.M.A. Diário oficial da União, Brasília, p 14–16, de 24 de Janeiro de 2000, Seção I. [online] Accessed in <[www.agricultura.gov.br/das/dispoa/Anexo%20Abate.htm](http://www.agricultura.gov.br/das/dispoa/Anexo%20Abate.htm)>, in 2012.
- Carbonel, G., Bravo, J. C., Fernández, C., & Tarazona, J. V. (2009). A new method for total mercury and methyl mercury analysis in muscle of seawater fish. *Bulletin Environmental Contamination Toxicology*, 83, 210–213.
- Clarkson, T. W., & Magos, L. (2006). The toxicology of mercury and its chemical compounds. *Critical Reviews in Toxicology*, 36, 609–662.
- EFSA. (2012). EFSA panel on contaminants in the food chain (CONTAM); scientific opinion on the risk for public health related to the presence of mercury and methylmercury in food. URL: Accessed 17.06.14.
- Food and Agricultural Organization/World Health Organization. (2004). Safety evaluation of certain food additives and contaminants, sixty-first meeting of the Joint FAO/WHO Expert Committee on Food Additives, WHO Food Additives Series, No. 52. Geneva: World Health Organization, International Programme on Chemical Safety. <[http://whqlibdoc.who.int/trs/WHO\\_TRS\\_922.pdf](http://whqlibdoc.who.int/trs/WHO_TRS_922.pdf)>. Accessed 22.06.2014.
- Hajeb, P., Jinap, S., Ismail, A., Fatimah, A. B., Jamilah, B., & Abdul Rahim, M. (2009). Assessment of mercury level in commonly consumed marine fishes in Malaysia. *Food Control*, 20, 79–84.
- Hightower, J. M., & Moore, D. (2003). Mercury levels in high-end consumers of fish. *Environmental Health Perspectives*, 111, 604–608.
- Horvat, M., Lucotte, M., & Malm, O. (2007). Recovery of mercury-contaminated fisheries. *Ambio*, 36, 33–44.
- IBGE – Instituto Brasileiro de Geografia e Estatística. (2013). On-line: <<http://cidades.ibge.gov.br/xtras/perfil.php?lang=&codmun=292730&search=bahia%7Csalinas-da-margarida>> Accessed 15.06.13.
- Ikem, A., & Egiebor, N. O. (2005). Assessment of trace elements in canned fishes (mackerel, tuna, salmon, sardines and herrings) marketed in Georgia and Alabama (United States of America). *Journal of Food Composition and Analysis*, 18, 771–787.
- Kalogeropoulos, N., Karavoltos, S., Sakellari, A., Avramidou, S., Dassenakis, M., & Scoullou, M. (2012). Heavy metals in raw, fried and grilled Mediterranean finfish and shellfish. *Food and Chemical Toxicology*, 50, 3702–3708.
- McClain, W. C., Chumchal, M. M., Drenner, R. W., & Newland, L. W. (2006). Mercury concentrations in fish from lake Meredith, Texas: Implications for the issuance of fish consumption advisories. *Environmental Monitoring and Assessment*, 123, 249–258.
- Melendez-Perez, J. J., & Fostier, A. H. (2013). Assessment of direct mercury Analyzer® to quantify mercury in soils and leaf samples. *Journal of the Brazilian Chemical Society*, 24, 1880–1886.
- Miller, J. N., & Miller, J. C. (2000). *Statistics and chemometrics for analytical chemistry* (4th ed.). Edinburgh: Pearson, p. 271.
- Ordiano-Flores, A., Rosiles-Martinez, R., & Galván-Magaña, F. (2012). Biomagnification of mercury and its antagonistic interaction with selenium in yellow fin tuna *Thunnus albacares* in the trophic web of Baja California Sur, Mexico. *Ecotoxicology and Environmental Safety*, 86, 182–187.
- Perello, G., Martí-Cid, R., Llobet, J. M., & Domingo, J. E. (2008). Effects of various cooking processes on the concentrations of arsenic, cadmium, mercury, and lead in foods. *Journal of Agricultural and Food Chemistry*, 56, 11262–11269.
- Porto, J. I. R., Araujo, C. S. O., & Feldberga, E. (2005). Mutagenic effects of mercury pollution as revealed by micronucleus test on three Amazonian fish species. *Environmental Research*, 97, 287–292.

- Poulin, J., Gibb, H. (2008). Mercury: Assessing the environmental burden of disease at national and local levels. WHO Environmental Burden of Disease Series No. 16. <[http://whqlibdoc.who.int/publications/2008/9789241596572\\_eng.pdf](http://whqlibdoc.who.int/publications/2008/9789241596572_eng.pdf)> Accessed 19.06.14.
- Schmidt, L., Bizzi, C. A., Duarte, F. A., Muller, E. I., Krupp, E., Feldmann, J., et al. (2015). Evaluation of Hg species after culinary treatments of fish. *Food Control*, 47, 413–419.
- Sofowote, U. M., Hung, H., Rastogi, A. K., Westgate, J. N., Deluca, P. F., Su, Y. S., et al. *Atmospheric Environment*, 45, 967–976.
- United Nations Environment Programme – UNEP. (2013). Global Mercury Assessment 2013: Sources, Emissions, Releases, and Environmental Transport. Geneva, Switzerland.
- United States Environmental Protection Agency (US EPA). (1997). Mercury study report to Congress. Volume III: Fate and transport of mercury in the environment. EPA/452/R-97/005. <<http://www.epa.gov/mercury/report.htm>>. Accessed 18.06.14.
- United States Environmental Protection Agency (US EPA). (2003). Mercury in Marine Life Database. Office of Wetlands, Oceans, and Watersheds Oceans and Coastal Protection Division, Washington.
- United States Environmental Protection Agency (US EPA). (2004). National listing of fish advisories (EPA-823-F-04-016). Office of Water, Washington United States Environmental Protection Agency.
- United States Environmental Protection Agency (US EPA). (2007). Method 7473, Mercury in Solids and Solutions by Thermal Decomposition, Amalgamation, and Atomic Absorption Spectrophotometry.
- United States Environmental Protection Agency (US EPA). (2009). EPA releases report on fish contamination in US lakes and reservoirs (EPA-823-F-09-008). Office of Water, Washington United States Environmental Protection Agency.
- US FDA. (2013). Mercury concentrations in fish: FDA Monitoring Program (1990–2010). Food and Drug Administration, Washington, DC. <<http://www.fda.gov/Food/FoodborneIllnessContaminants/Metals/ucm191007.htm>> Accessed 16.06.14.
- Voegborlo, R. B., & Akagi, H. (2007). Determination of mercury in fish by cold vapour atomic absorption spectrometry using an automatic mercury analyzer. *Food Chemistry*, 100, 853–858.
- WHO (World Health Organization). (1996). Health criteria other supporting information. In Guidelines for drinking water Quality (Vol. 2, pp. 31–388), 3rd ed. Geneva.
- Zhang, L., Campbell, L. M., & Johnson, T. B. (2012). Seasonal variation in mercury and food web biomagnification in Lake Ontario, Canada. *Environmental Pollution*, 16, 178 e 184.