

Crystal structure of fenbuconazole

Gihaeng Kang, Jineun Kim,* Hyunjin Park and Tae Ho Kim*

Department of Chemistry and Research Institute of Natural Sciences, Gyeongsang National University, Jinju 52828, Republic of Korea. *Correspondence e-mail: tkim@gnu.ac.kr, jekim@gnu.ac.kr

Received 12 August 2015; accepted 18 August 2015

Edited by W. T. A. Harrison, University of Aberdeen, Scotland

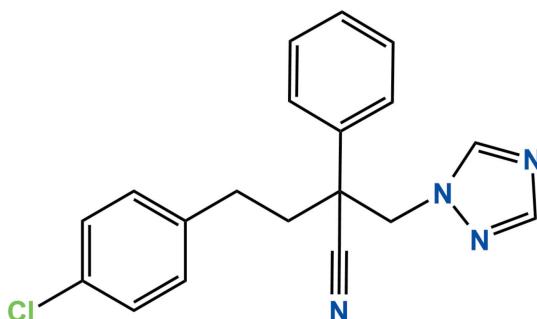
In the title compound, $C_{19}H_{17}ClN_4$ [systematic name: (*RS*)-4-(4-chlorophenyl)-2-phenyl-2-(1*H*-1,2,4-triazol-1-ylmethyl)-butyronitrile], which is the conazole fungicide fenbuconazole, the dihedral angles between the planes of the central benzene and the terminal chlorophenyl and triazole rings are 32.77 (5) and 32.97 (5) $^\circ$, respectively. The C—C—C—C linkage between the tertiary C atom and the benzene ring has an *anti* orientation [torsion angle = 174.47 (12) $^\circ$]. In the crystal, C—H \cdots N hydrogen bonds and very weak C—Cl \cdots π interactions [$\text{Cl}\cdots\pi$ = 3.7892 (9) Å] link adjacent molecules, forming two-dimensional networks lying parallel to the (101) plane. The planes are linked by weak π — π interactions [centroid–centroid separation = 3.8597 (9) Å], resulting in a three-dimensional architecture.

Keywords: crystal structure; fungicide; fenbuconazole; C—Cl \cdots π interactions; π — π interactions.

CCDC reference: 1419334

1. Related literature

For information on the fungicidal properties of the title compound, see: Li *et al.* (2012). For related crystal structures, see: Rizzoli *et al.* (2009); Yin *et al.* (2014).



2. Experimental

2.1. Crystal data

$C_{19}H_{17}ClN_4$
 $M_r = 336.82$
Monoclinic, $P2_1/n$
 $a = 12.4606$ (3) Å
 $b = 6.7404$ (2) Å
 $c = 20.5394$ (5) Å
 $\beta = 95.455$ (2) $^\circ$

$V = 1717.28$ (8) Å 3
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.23$ mm $^{-1}$
 $T = 173$ K
 $0.18 \times 0.07 \times 0.03$ mm

2.2. Data collection

Bruker APEXII CCD
diffractometer
Absorption correction: multi-scan
(SADABS; Bruker, 2013)
 $T_{\min} = 0.959$, $T_{\max} = 0.993$

15784 measured reflections
3936 independent reflections
3044 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.108$
 $S = 1.04$
3936 reflections

217 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.25$ e Å $^{-3}$
 $\Delta\rho_{\min} = -0.30$ e Å $^{-3}$

Table 1
Hydrogen-bond geometry (Å, $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C8—H8A \cdots N1 ⁱ	0.99	2.53	3.522 (2)	178
C11—H11 \cdots N1 ⁱ	0.95	2.60	3.533 (2)	166
C17—H17A \cdots N1 ⁱⁱ	0.99	2.58	3.5101 (18)	156
C18—H18 \cdots N4 ⁱⁱⁱ	0.95	2.46	3.277 (2)	144

Symmetry codes: (i) $x, y - 1, z$; (ii) $-x + 1, -y + 2, -z + 2$; (iii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *APEX2* (Bruker, 2013); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2015); molecular graphics: *DIAMOND* (Brandenburg, 2010); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

Acknowledgements

This research was supported by the Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science and Technology (No. 2015R1D1A4A01020317).

Supporting information for this paper is available from the IUCr electronic archives (Reference: HB7481).

References

Brandenburg, K. (2010). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.

- Bruker (2013). *APEX2, SAINT and SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Li, Y., Dong, F., Liu, X., Xu, J., Li, J., Kong, Z., Chen, X. & Zheng, Y. (2012). *Environ. Sci. Technol.* **46**, 2675–2683.
- Rizzoli, C., Marku, E. & Greci, L. (2009). *Acta Cryst. E* **65**, o663.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Sheldrick, G. M. (2015). *Acta Cryst. C* **71**, 3–8.
- Yin, B.-T., Yan, C.-Y., Peng, X.-M., Zhang, S.-L., Rasheed, S., Geng, R.-X. & Zhou, C.-H. (2014). *Eur. J. Med. Chem.* **71**, 148–159.

supporting information

Acta Cryst. (2015). E71, o680–o681 [https://doi.org/10.1107/S205698901501542X]

Crystal structure of fenbuconazole

Gihaeng Kang, Jineun Kim, Hyunjin Park and Tae Ho Kim

S1. Comment

Fenbuconazole, [systematic name: (*RS*)-4-(4-chlorophenyl)-2-phenyl-2-(1*H*-1,2,4-triazol-1-ylmethyl)butyronitrile], is a conazole fungicide and it has been used for the control of leaf spot, yellow and brown rust, powdery mildew, and net blotch on various agricultural and horticultural crops (Li *et al.*, 2012). However, until now its crystal structure has not been reported. The dihedral angles between the planes of the central benzene and the terminal chlorophenyl and triazole rings are 32.77 (5) and 32.97 (5) $^{\circ}$, respectively. All bond lengths and bond angles are normal and comparable to those observed in similar crystal structures (Rizzoli *et al.*, 2009; Yin *et al.*, 2014).

In the crystal structure (Fig. 2), C—H \cdots N hydrogen bonds and weak C3—Cl1 \cdots Cg1^{iv} (Cg1 is the centroid of the N2—N3—C18—N4—C19 ring) interaction [3.7892 (9) Å] with a chlorophenyl ring are observed (Table 1), forming two-dimensional networks parallel to (101) plane. In addition, the planes are linked by weak intermolecular $\pi\cdots\pi$ interaction between the terminal chlorophenyl ring systems [Cg2 \cdots Cg2^v, 3.8597 (9) Å], resulting in a three-dimensional architecture. (Cg2 is the centroid of the C1—C6 ring) [for symmetry codes: (iv), $-x + 1, -y + 1, -z + 2$, (v), $-x, -y + 1, -z + 2$].

S2. Experimental

The title compound was purchased from the Dr Ehrenstorfer GmbH Company. Slow evaporation of a solution in CH₃CN gave brown plates suitable for X-ray analysis.

S3. Refinement

All H-atoms were positioned geometrically and refined using a riding model with d(C—H) = 0.99 Å, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for CH₂ group and d(C—H) = 0.95 Å, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for aromatic C—H.

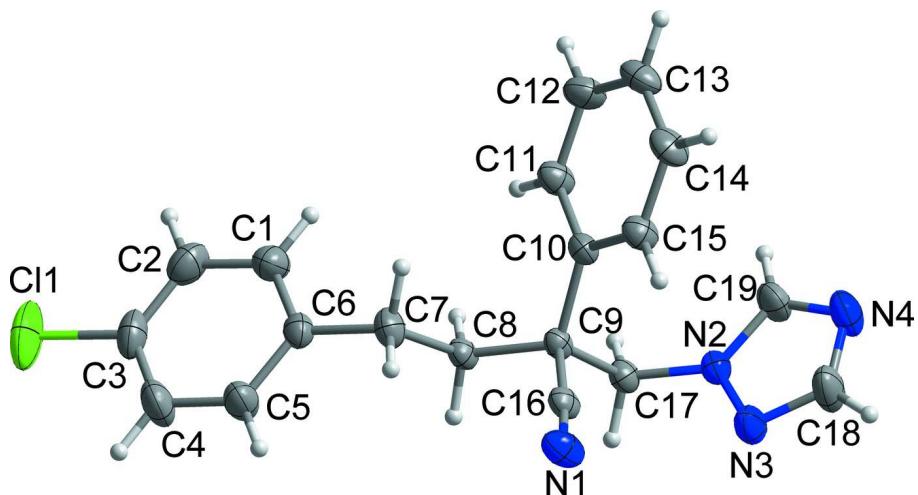
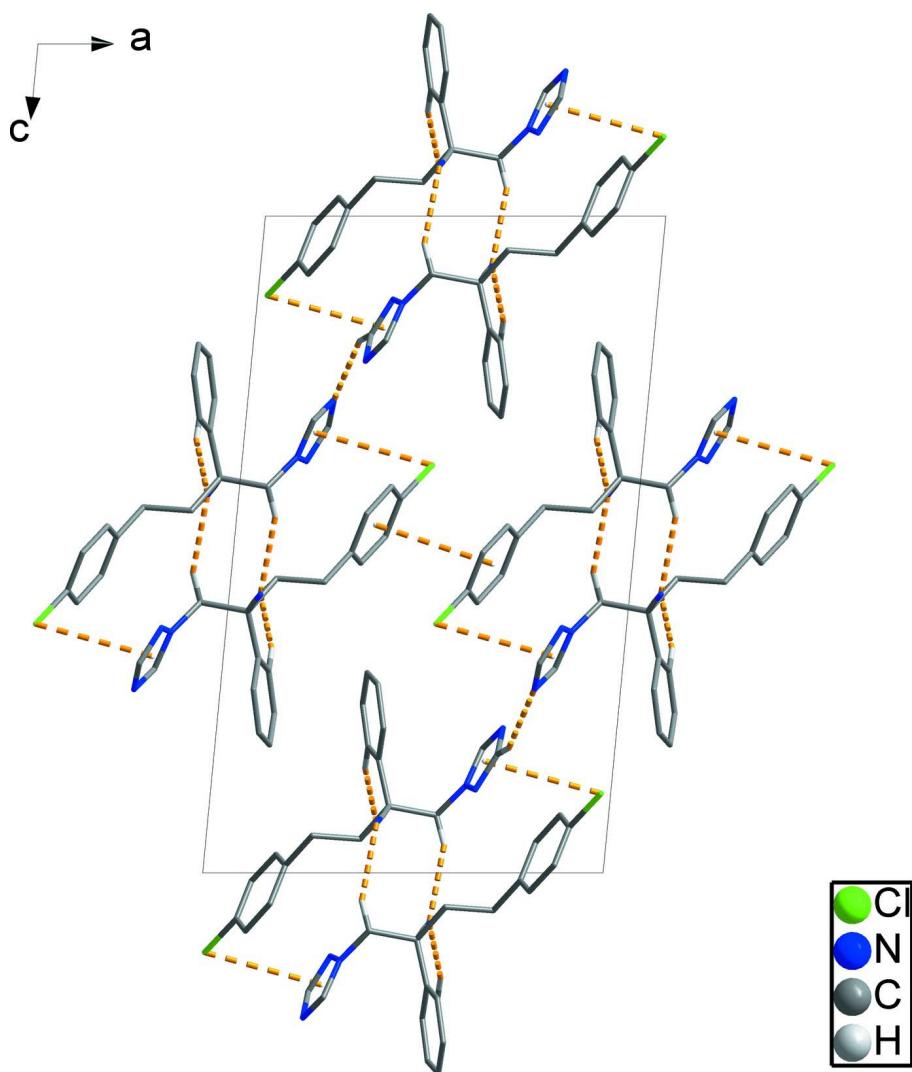


Figure 1

The asymmetric unit of the title compound with displacement ellipsoids drawn at the 50% probability level.

**Figure 2**

Crystal packing viewed along the b axis. The intermolecular interactions are shown as dashed lines.

(*RS*)-4-(4-Chlorophenyl)-2-phenyl-2-(1*H*-1,2,4-triazol-1-ylmethyl)butyronitrile

Crystal data

$C_{19}H_{17}ClN_4$
 $M_r = 336.82$
 Monoclinic, $P2_1/n$
 $a = 12.4606 (3) \text{ \AA}$
 $b = 6.7404 (2) \text{ \AA}$
 $c = 20.5394 (5) \text{ \AA}$
 $\beta = 95.455 (2)^\circ$
 $V = 1717.28 (8) \text{ \AA}^3$
 $Z = 4$

$F(000) = 704$
 $D_x = 1.303 \text{ Mg m}^{-3}$
 $\text{Mo } K\alpha \text{ radiation, } \lambda = 0.71073 \text{ \AA}$
 Cell parameters from 3787 reflections
 $\theta = 3.2\text{--}27.4^\circ$
 $\mu = 0.23 \text{ mm}^{-1}$
 $T = 173 \text{ K}$
 Plate, brown
 $0.18 \times 0.07 \times 0.03 \text{ mm}$

Data collection

Bruker APEXII CCD
diffractometer
 φ and ω scans
Absorption correction: multi-scan
(*SADABS*; Bruker, 2013)
 $T_{\min} = 0.959$, $T_{\max} = 0.993$
15784 measured reflections

3936 independent reflections
3044 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$
 $\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 1.8^\circ$
 $h = -15 \rightarrow 16$
 $k = -8 \rightarrow 8$
 $l = -26 \rightarrow 26$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.108$
 $S = 1.04$
3936 reflections
217 parameters
0 restraints

Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.043P)^2 + 0.4967P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.30 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.02411 (4)	0.19334 (10)	1.12090 (3)	0.0689 (2)
N1	0.42210 (11)	1.1422 (2)	0.92410 (6)	0.0367 (3)
N2	0.63664 (9)	0.7989 (2)	0.86689 (5)	0.0277 (3)
N3	0.66105 (11)	0.9948 (2)	0.87314 (6)	0.0398 (3)
N4	0.71532 (12)	0.8721 (2)	0.77946 (7)	0.0445 (4)
C1	0.17939 (13)	0.3822 (3)	0.97206 (8)	0.0384 (4)
H1	0.2005	0.3286	0.9325	0.046*
C2	0.12369 (14)	0.2639 (3)	1.01229 (9)	0.0447 (4)
H2	0.1056	0.1311	1.0003	0.054*
C3	0.09501 (12)	0.3415 (3)	1.06994 (8)	0.0410 (4)
C4	0.12079 (14)	0.5320 (3)	1.08810 (8)	0.0485 (5)
H4	0.1014	0.5833	1.1284	0.058*
C5	0.17565 (13)	0.6493 (3)	1.04672 (8)	0.0398 (4)
H5	0.1933	0.7822	1.0589	0.048*
C6	0.20501 (10)	0.5763 (2)	0.98807 (7)	0.0278 (3)
C7	0.26520 (11)	0.7035 (3)	0.94298 (7)	0.0316 (3)
H7A	0.2559	0.8451	0.9541	0.038*
H7B	0.2341	0.6831	0.8973	0.038*
C8	0.38533 (11)	0.6541 (2)	0.94830 (6)	0.0237 (3)
H8A	0.3937	0.5096	0.9418	0.028*
H8B	0.4171	0.6864	0.9931	0.028*
C9	0.44939 (10)	0.7655 (2)	0.89855 (6)	0.0218 (3)

C10	0.41229 (10)	0.7205 (2)	0.82676 (6)	0.0229 (3)
C11	0.38360 (12)	0.5284 (2)	0.80795 (7)	0.0312 (3)
H11	0.3807	0.4280	0.8402	0.037*
C12	0.35906 (13)	0.4822 (3)	0.74246 (7)	0.0364 (4)
H12	0.3390	0.3506	0.7301	0.044*
C13	0.36370 (13)	0.6264 (3)	0.69533 (7)	0.0369 (4)
H13	0.3467	0.5943	0.6505	0.044*
C14	0.39293 (13)	0.8170 (3)	0.71325 (7)	0.0361 (4)
H14	0.3968	0.9162	0.6807	0.043*
C15	0.41687 (12)	0.8646 (2)	0.77900 (6)	0.0297 (3)
H15	0.4364	0.9967	0.7912	0.036*
C16	0.43710 (11)	0.9796 (2)	0.91178 (6)	0.0257 (3)
C17	0.56995 (11)	0.7062 (2)	0.91202 (6)	0.0255 (3)
H17A	0.5963	0.7450	0.9572	0.031*
H17B	0.5765	0.5603	0.9086	0.031*
C18	0.70813 (15)	1.0295 (3)	0.81948 (8)	0.0472 (5)
H18	0.7352	1.1567	0.8096	0.057*
C19	0.66923 (13)	0.7306 (3)	0.81087 (7)	0.0357 (4)
H19	0.6603	0.5979	0.7958	0.043*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0454 (3)	0.0820 (4)	0.0822 (4)	-0.0076 (3)	0.0220 (2)	0.0428 (3)
N1	0.0555 (8)	0.0248 (8)	0.0306 (7)	0.0036 (7)	0.0081 (6)	0.0001 (6)
N2	0.0283 (6)	0.0304 (7)	0.0253 (6)	-0.0013 (5)	0.0069 (5)	0.0004 (5)
N3	0.0489 (8)	0.0351 (8)	0.0384 (7)	-0.0131 (7)	0.0195 (6)	-0.0055 (6)
N4	0.0514 (8)	0.0463 (10)	0.0397 (7)	-0.0003 (7)	0.0240 (6)	0.0004 (7)
C1	0.0448 (9)	0.0389 (10)	0.0317 (8)	-0.0065 (8)	0.0057 (6)	-0.0035 (7)
C2	0.0441 (9)	0.0366 (10)	0.0529 (10)	-0.0123 (8)	0.0020 (8)	0.0032 (8)
C3	0.0275 (7)	0.0508 (12)	0.0455 (9)	-0.0038 (8)	0.0086 (6)	0.0185 (8)
C4	0.0521 (10)	0.0585 (13)	0.0388 (9)	-0.0014 (10)	0.0241 (8)	0.0008 (9)
C5	0.0474 (9)	0.0360 (10)	0.0382 (8)	-0.0064 (8)	0.0164 (7)	-0.0052 (7)
C6	0.0231 (6)	0.0334 (9)	0.0271 (7)	0.0002 (6)	0.0036 (5)	0.0036 (6)
C7	0.0317 (7)	0.0344 (9)	0.0297 (7)	0.0026 (7)	0.0084 (6)	0.0068 (7)
C8	0.0299 (7)	0.0226 (8)	0.0193 (6)	-0.0001 (6)	0.0054 (5)	0.0023 (6)
C9	0.0274 (6)	0.0185 (7)	0.0200 (6)	0.0006 (6)	0.0053 (5)	0.0021 (5)
C10	0.0243 (6)	0.0246 (8)	0.0202 (6)	0.0026 (6)	0.0046 (5)	0.0004 (6)
C11	0.0427 (8)	0.0269 (9)	0.0245 (7)	0.0000 (7)	0.0055 (6)	0.0024 (6)
C12	0.0482 (9)	0.0306 (9)	0.0302 (8)	0.0004 (8)	0.0024 (6)	-0.0069 (7)
C13	0.0427 (8)	0.0468 (11)	0.0208 (7)	0.0091 (8)	0.0014 (6)	-0.0042 (7)
C14	0.0465 (9)	0.0397 (10)	0.0223 (7)	0.0063 (8)	0.0045 (6)	0.0075 (7)
C15	0.0385 (8)	0.0267 (9)	0.0242 (7)	0.0014 (7)	0.0036 (6)	0.0038 (6)
C16	0.0334 (7)	0.0257 (9)	0.0186 (6)	-0.0012 (7)	0.0053 (5)	0.0032 (6)
C17	0.0288 (7)	0.0262 (8)	0.0220 (6)	0.0010 (6)	0.0048 (5)	0.0036 (6)
C18	0.0574 (11)	0.0420 (11)	0.0467 (9)	-0.0119 (9)	0.0284 (8)	-0.0007 (8)
C19	0.0392 (8)	0.0379 (10)	0.0322 (8)	0.0044 (7)	0.0142 (6)	-0.0043 (7)

Geometric parameters (\AA , $\text{^{\circ}}$)

C11—C3	1.7461 (16)	C8—C9	1.5492 (18)
N1—C16	1.1445 (19)	C8—H8A	0.9900
N2—C19	1.3376 (18)	C8—H8B	0.9900
N2—N3	1.3585 (18)	C9—C16	1.479 (2)
N2—C17	1.4444 (17)	C9—C10	1.5332 (17)
N3—C18	1.3178 (19)	C9—C17	1.5537 (18)
N4—C19	1.314 (2)	C10—C15	1.386 (2)
N4—C18	1.351 (2)	C10—C11	1.388 (2)
C1—C6	1.379 (2)	C11—C12	1.386 (2)
C1—C2	1.382 (2)	C11—H11	0.9500
C1—H1	0.9500	C12—C13	1.377 (2)
C2—C3	1.372 (2)	C12—H12	0.9500
C2—H2	0.9500	C13—C14	1.376 (2)
C3—C4	1.367 (3)	C13—H13	0.9500
C4—C5	1.388 (2)	C14—C15	1.392 (2)
C4—H4	0.9500	C14—H14	0.9500
C5—C6	1.382 (2)	C15—H15	0.9500
C5—H5	0.9500	C17—H17A	0.9900
C6—C7	1.5124 (19)	C17—H17B	0.9900
C7—C8	1.5273 (19)	C18—H18	0.9500
C7—H7A	0.9900	C19—H19	0.9500
C7—H7B	0.9900		
C19—N2—N3	109.39 (12)	C16—C9—C8	106.43 (11)
C19—N2—C17	130.06 (14)	C10—C9—C8	114.28 (11)
N3—N2—C17	119.81 (12)	C16—C9—C17	109.49 (12)
C18—N3—N2	101.96 (13)	C10—C9—C17	108.56 (10)
C19—N4—C18	102.36 (13)	C8—C9—C17	107.94 (10)
C6—C1—C2	121.52 (15)	C15—C10—C11	118.93 (13)
C6—C1—H1	119.2	C15—C10—C9	120.87 (13)
C2—C1—H1	119.2	C11—C10—C9	119.90 (12)
C3—C2—C1	118.87 (17)	C12—C11—C10	120.47 (14)
C3—C2—H2	120.6	C12—C11—H11	119.8
C1—C2—H2	120.6	C10—C11—H11	119.8
C4—C3—C2	121.31 (15)	C13—C12—C11	120.22 (16)
C4—C3—C11	119.53 (14)	C13—C12—H12	119.9
C2—C3—C11	119.16 (15)	C11—C12—H12	119.9
C3—C4—C5	119.02 (16)	C14—C13—C12	119.92 (14)
C3—C4—H4	120.5	C14—C13—H13	120.0
C5—C4—H4	120.5	C12—C13—H13	120.0
C6—C5—C4	121.15 (17)	C13—C14—C15	120.11 (14)
C6—C5—H5	119.4	C13—C14—H14	119.9
C4—C5—H5	119.4	C15—C14—H14	119.9
C1—C6—C5	118.10 (14)	C10—C15—C14	120.35 (15)
C1—C6—C7	120.59 (13)	C10—C15—H15	119.8
C5—C6—C7	121.30 (15)	C14—C15—H15	119.8

C6—C7—C8	111.85 (12)	N1—C16—C9	175.67 (15)
C6—C7—H7A	109.2	N2—C17—C9	112.33 (11)
C8—C7—H7A	109.2	N2—C17—H17A	109.1
C6—C7—H7B	109.2	C9—C17—H17A	109.1
C8—C7—H7B	109.2	N2—C17—H17B	109.1
H7A—C7—H7B	107.9	C9—C17—H17B	109.1
C7—C8—C9	114.21 (11)	H17A—C17—H17B	107.9
C7—C8—H8A	108.7	N3—C18—N4	115.46 (16)
C9—C8—H8A	108.7	N3—C18—H18	122.3
C7—C8—H8B	108.7	N4—C18—H18	122.3
C9—C8—H8B	108.7	N4—C19—N2	110.83 (15)
H8A—C8—H8B	107.6	N4—C19—H19	124.6
C16—C9—C10	110.05 (11)	N2—C19—H19	124.6
C19—N2—N3—C18	-0.44 (17)	C16—C9—C10—C11	-159.29 (13)
C17—N2—N3—C18	-171.49 (14)	C8—C9—C10—C11	-39.62 (17)
C6—C1—C2—C3	-1.1 (3)	C17—C9—C10—C11	80.90 (15)
C1—C2—C3—C4	-0.2 (3)	C15—C10—C11—C12	-0.5 (2)
C1—C2—C3—C11	179.73 (13)	C9—C10—C11—C12	-174.29 (13)
C2—C3—C4—C5	1.0 (3)	C10—C11—C12—C13	0.5 (2)
C11—C3—C4—C5	-178.99 (14)	C11—C12—C13—C14	0.1 (2)
C3—C4—C5—C6	-0.4 (3)	C12—C13—C14—C15	-0.6 (2)
C2—C1—C6—C5	1.5 (2)	C11—C10—C15—C14	0.0 (2)
C2—C1—C6—C7	-179.47 (15)	C9—C10—C15—C14	173.75 (13)
C4—C5—C6—C1	-0.8 (2)	C13—C14—C15—C10	0.5 (2)
C4—C5—C6—C7	-179.76 (15)	C19—N2—C17—C9	-94.71 (18)
C1—C6—C7—C8	-77.25 (17)	N3—N2—C17—C9	74.23 (16)
C5—C6—C7—C8	101.70 (17)	C16—C9—C17—N2	-65.98 (14)
C6—C7—C8—C9	174.47 (12)	C10—C9—C17—N2	54.17 (16)
C7—C8—C9—C16	60.28 (15)	C8—C9—C17—N2	178.56 (11)
C7—C8—C9—C10	-61.39 (16)	N2—N3—C18—N4	0.2 (2)
C7—C8—C9—C17	177.73 (12)	C19—N4—C18—N3	0.0 (2)
C16—C9—C10—C15	27.07 (17)	C18—N4—C19—N2	-0.34 (19)
C8—C9—C10—C15	146.73 (13)	N3—N2—C19—N4	0.52 (18)
C17—C9—C10—C15	-92.74 (15)	C17—N2—C19—N4	170.36 (14)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
C8—H8A···N1 ⁱ	0.99	2.53	3.522 (2)	178
C11—H11···N1 ⁱ	0.95	2.60	3.533 (2)	166
C17—H17A···N1 ⁱⁱ	0.99	2.58	3.5101 (18)	156
C18—H18···N4 ⁱⁱⁱ	0.95	2.46	3.277 (2)	144

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1, -y+2, -z+2$; (iii) $-x+3/2, y+1/2, -z+3/2$.