

# Pr<sup>3+</sup>- and Pr<sup>3+</sup>/Er<sup>3+</sup>-Doped Selenide Glasses for Potential 1.6 μm Optical Amplifier Materials

Yong Gyu Choi, Bong Je Park, Kyong Hon Kim, and Jong Heo

**1.6 μm emission originated from Pr<sup>3+</sup>: (<sup>3</sup>F<sub>3</sub>, <sup>3</sup>F<sub>4</sub>) → <sup>3</sup>H<sub>4</sub> transition in Pr<sup>3+</sup>- and Pr<sup>3+</sup>/Er<sup>3+</sup>-doped selenide glasses was investigated under an optical pump of a conventional 1480 nm laser diode. The measured peak wavelength and full-width at half-maximum of the fluorescent emission are ~1650 nm and ~120 nm, respectively. A moderate lifetime of the thermally coupled upper manifolds of  $212 \pm 10$  μs together with a high stimulated emission cross-section of  $(3 \pm 1) \times 10^{-20}$  cm<sup>2</sup> promises to be useful for 1.6 μm band fiber-optic amplifiers that can be pumped with an existing high-power 1480 nm laser diode. Codoping Er<sup>3+</sup> enhances the emission intensity by way of a nonradiative Er<sup>3+</sup>: <sup>4</sup>I<sub>13/2</sub> → Pr<sup>3+</sup>: (<sup>3</sup>F<sub>3</sub>, <sup>3</sup>F<sub>4</sub>) energy transfer. The Dexter model based on the spectral overlap between donor emission and acceptor absorption describes well the energy transfer from Er<sup>3+</sup> to Pr<sup>3+</sup> in these glasses. Also discussed in this paper are major transmission loss mechanisms of a selenide glass optical fiber.**

## I. INTRODUCTION

Optical communication bandwidths are limited by the clarity of the transmission window in the SiO<sub>2</sub>-based optical fiber. By virtue of recent development of elaborated processing technologies, mass production of OH-free silica optical fibers which have transparent characteristics in the entire spectrum from ~1200 to ~1700 nm is currently available [1]. Thus, in order to fully utilize the whole transparent region of the dry fibers for large capacity optical communication systems of several tens Tb/s, optical amplifiers covering the wavelengths from shorter than 1300 nm to about 1700 nm are necessary.

Recently, a significant technical improvement has been made in Raman fiber amplifiers and semiconductor optical amplifiers as well as rare-earth (RE) doped fiber amplifiers. However, the RE doped fiber amplifiers have several superior characteristics over the Raman and semiconductor amplifiers. Some of them are possibilities to have a relatively high signal-to-noise ratio, good gain stability, and low signal crosstalk [2]. This paper deals only with fiber materials related to new RE doped fiber amplifiers. Erbium ions doped in either germano-silicate or tellurite glasses have been well demonstrated to provide signal amplification over the wavelength region of 1530 to 1600 nm [3]-[5]. Praseodymium and dysprosium ions are proven to be useful for amplifiers in the 1.3 μm band [6], [7]. Thulium ions in fluoride glasses amplify optical signals in the wavelengths of 1450 to 1520 nm [8] and of longer than 1650 nm [9]. So far, there has been no RE-host combination optimized for optical amplifiers in wavelength regions around 1350 to 1450 nm where the hydroxyl ions cause an additional absorption in the conventional silica fibers, and around 1610 to 1650 nm where the gain-shifted erbium-doped fiber amplifiers cannot provide a practical optical gain. In this paper, we report measured results of spec-

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troscopic properties of a  $\text{Pr}^{3+}$ -doped low phonon energy glass, which is considered as a possible candidate material for 1.6  $\mu\text{m}$  band fiber amplifiers. Sensitizing effect of  $\text{Er}^{3+}$  on the 1.6  $\mu\text{m}$  emission from the  $\text{Pr}^{3+}$ -doped glass for optical pump at around 1480 nm is also discussed.

Magnitude of multiphonon relaxation is one of the parameters that govern emission intensity of some radiative intra-4f-configurational transitions of RE ions in a dielectric medium. One representative example of such transitions is the 1.3  $\mu\text{m}$  emission originated from the  $\text{Pr}^{3+}$ :  $^1\text{G}_4 \rightarrow ^3\text{H}_5$  transition [6]. The  $\text{Pr}^{3+}$  ion actually has a luminescence centered at about 1.6  $\mu\text{m}$  from  $(^3\text{F}_3, ^3\text{F}_4) \rightarrow ^3\text{H}_4$  transition, but due to the tightly spaced energy levels of an energy gap of  $\sim 1350 \text{ cm}^{-1}$ , the radiative transition is virtually forbidden in most glasses, even in sulfide glasses [10]. A lasing action at  $\sim 1.6 \mu\text{m}$  has been demonstrated by using this transition in  $\text{LaCl}_3$  crystal [11], but no detailed spectroscopic data on this transition from a glass host have been reported yet [12].

The multiphonon relaxation process from the  $(^3\text{F}_3, ^3\text{F}_4)$  level can be reduced in glasses having lower vibrational phonon energy than the sulfide glasses ( $\sim 350 \text{ cm}^{-1}$ ) [13], [14]. Some germano-selenide glass systems have thermal stability good enough to be drawn into a fiber waveguide without any crystallization [15], and possess the most intense phonon energy at about  $200 \text{ cm}^{-1}$  [16]. In this study, we used Ge-As-Ga-Se glasses, and regarded them as a representative of selenide glasses.

## II. EXPERIMENTAL

$\text{Ge}_{30}\text{-As}_{80}\text{-Ga}_2\text{-Se}_{60}$  (mol%) composition was used as a host glass. Glasses were prepared from Ge, Ga, As, and Se powders with a purity better than 99.999%. Metals of praseodymium and erbium with purity of 99.999% were used as a starting material of rare-earths. Up to 1 mol% praseodymium could be solved in the host. Introduction of small amount of gallium to the Ge-As-Se system enhanced the RE solubility as did it in Ge-As-S glasses [17]. Approximately 15-g batches were weighed in an Ar-purged glove box with  $\text{O}_2$  and  $\text{OH}^-$  concentrations less than 2 ppm and 5 ppm, respectively. Silica ampoules containing the starting materials were sealed, and melted in a rocking furnace. Temperature was first increased to  $500^\circ\text{C}$  with a heating rate of  $2^\circ\text{C}/\text{min}$  and then subsequently elevated to  $1000^\circ\text{C}$  by  $1^\circ\text{C}/\text{min}$ . After the melting at  $1000^\circ\text{C}$  for 12 h, ampoules with the melt inside were removed from the furnace and quenched into water. Samples were annealed at around their glass transition temperatures ( $\sim 360^\circ\text{C}$ ) for two hours. Approximately 2 mm-thick discs were sliced and optically polished for the measurements.

A UV/VIS/NIR spectrophotometer and an FTIR spectrom-

eter were used to measure absorption and transmission spectra in the regions of 500 to 2500 nm and  $400$  to  $4000 \text{ cm}^{-1}$ , respectively. A conventional 1480 nm laser diode was used as an excitation source. The emission signals were detected using an arrangement equipped with a 0.5-m monochromator, InGaAs PIN detector, lock-in amplifier and a computer. Re-absorption of the fluorescences was minimized by focusing the excitation beam immediately below the surface of the samples and letting the emitted lights travel the shortest distance inside the sample. Decay traces were monitored with a 500 MHz digitizing oscilloscope at a proper mechanical chopper frequency. A decay lifetime (or duration time) was determined as the first  $e^{-1}$ -folding time of the fluorescence emission intensity.

## III. RESULTS

### 1. Absorption Properties

UV-side absorption spectrum of a 1 mol%  $\text{Pr}^{3+}$ -doped selenide glass is shown in Fig. 1. Host absorption starts at around 800 nm. Shown in Fig. 2 is IR-side transmission spectrum of a  $\text{Pr}^{3+}$ -doped selenide glass. Transmission cutoff of the host glass appears at  $\sim 600 \text{ cm}^{-1}$ , which suggests that this glass consists of lower phonon vibration modes than the conventional sulfide glasses. Based on the measured absorption and transmission spectra, one can draw a schematic energy level diagram of  $\text{Pr}^{3+}$  in these glasses as shown in Fig. 3. The  $^3\text{F}_4$  and  $^3\text{F}_2$  multiplets are thermally coupled with the  $^3\text{F}_3$  and  $^3\text{H}_6$  levels, respectively [18]. The Boltzmann distribution predicts that approximately

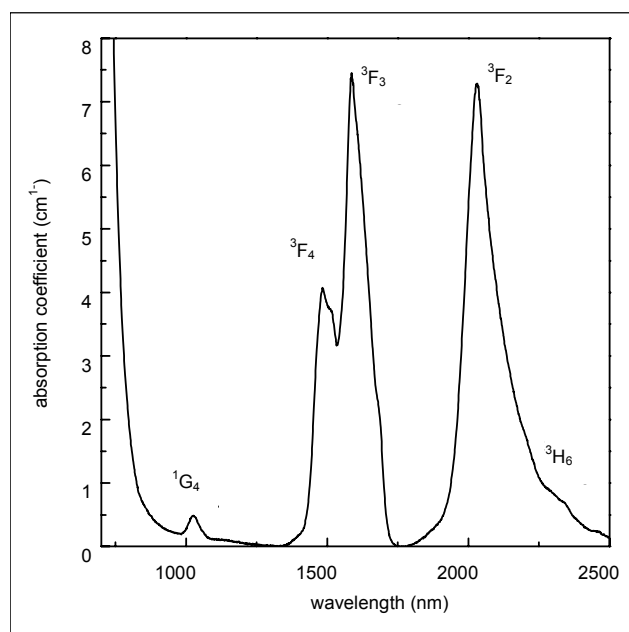


Fig. 1. Absorption spectrum of a 1.0 mol%  $\text{Pr}^{3+}$ -doped selenide glass.

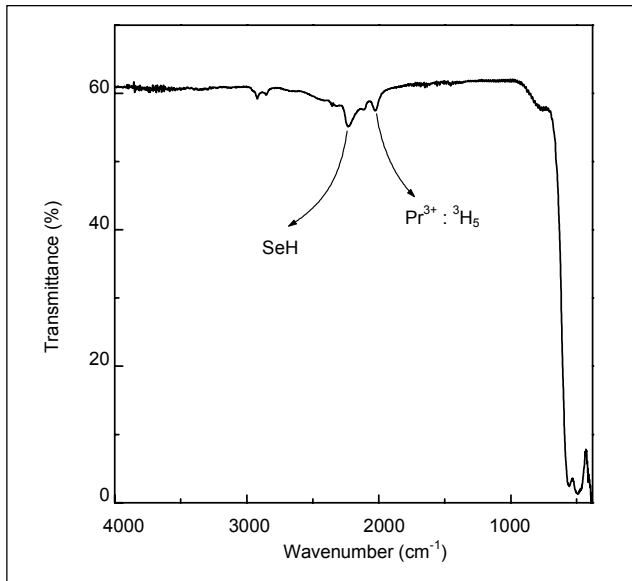


Fig. 2. Infrared transmission spectrum of a  $\text{Pr}^{3+}$ -doped selenide glass.

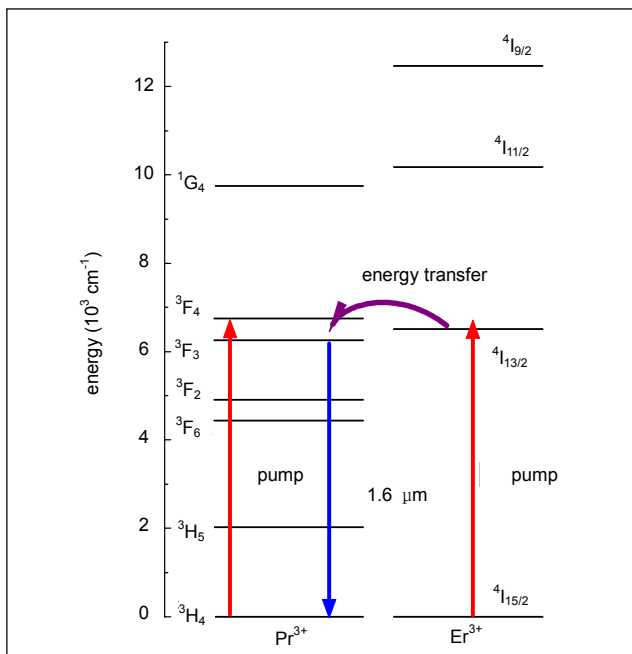


Fig. 3. Schematic energy level diagram of  $\text{Pr}^{3+}$  and  $\text{Er}^{3+}$  in selenide glass.

90% and 91% of populations reside in the corresponding lower-lying levels, *i.e.*,  $^3\text{F}_3$  and  $^3\text{H}_6$ , in this selenide glass, respectively. Thus, we consider the first two thermally coupled manifolds as a ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) level, and the other two manifolds as a ( $^3\text{F}_2$ ,  $^3\text{H}_6$ ) level. Absorption cross-section spectra of the  $\text{Pr}^{3+}$ :  $^3\text{H}_4 \rightarrow (^3\text{F}_3, ^3\text{F}_4)$  and  $\text{Er}^{3+}$ :  $^4\text{I}_{15/2} \rightarrow ^4\text{I}_{13/2}$  transitions are presented in Fig. 4. The absorption cross-section of  $\text{Pr}^{3+}$  peaks at  $\sim 1585$  nm with a value of  $\sim 2.2 \times 10^{-20} \text{ cm}^2$ , and at 1480 nm it is  $\sim 1.2 \times 10^{-20} \text{ cm}^2$ .

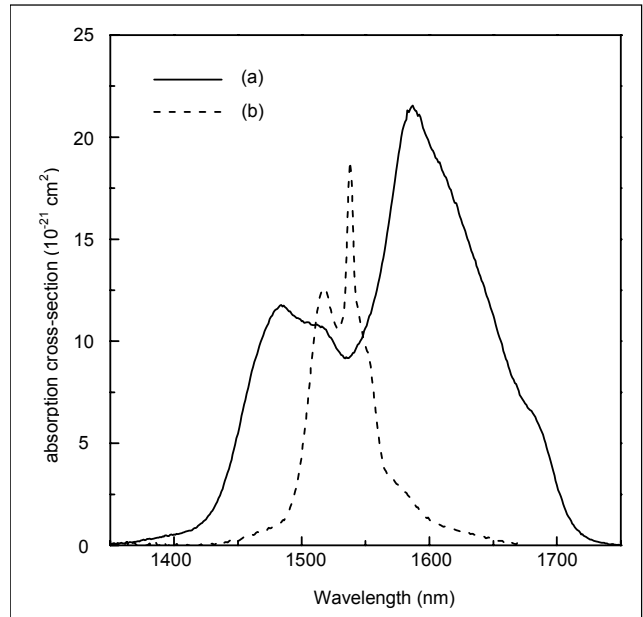


Fig. 4. Absorption cross-section spectra of (a)  $\text{Pr}^{3+}$ :  $^3\text{H}_4 \rightarrow (^3\text{F}_3, ^3\text{F}_4)$  and (b)  $\text{Er}^{3+}$ :  $^4\text{I}_{15/2} \rightarrow ^4\text{I}_{13/2}$  in selenide glass.

Meanwhile, the maximum absorption cross-section of  $\text{Er}^{3+}$  is  $\sim 1.9 \times 10^{-20} \text{ cm}^2$  at 1538 nm.

Judd-Ofelt analysis was done using five absorption transitions. Lineshape spectra of the ( $^3\text{F}_4$ ,  $^3\text{F}_3$ ) and ( $^3\text{F}_2$ ,  $^3\text{H}_6$ ) manifolds were divided into four corresponding individual transitions with the assumption that each transition appeared to have a Lorentzian lineshape [12]. Three Judd-Ofelt intensity parameters, *i.e.*,  $\Omega_2$ ,  $\Omega_4$  and  $\Omega_6$  [19] were calculated to be 1.01, 10.54 and 2.53 in  $10^{-20} \text{ cm}^2$ , respectively. Uncertainty involved in the analysis was estimated to be  $\sim 15\%$ . Arranged in Table 1 are the radiative transition probabilities, branching ratio, and radiative lifetimes of some interested transitions. Branching ratio of the 1.6  $\mu\text{m}$  emission is at  $\sim 77\%$ .

## 2. Emission from Single Doped Glasses

Figure 5(a) shows the emission spectrum originated from the ( $^3\text{F}_3$ ,  $^3\text{F}_4$ )  $\rightarrow ^3\text{H}_4$  transition. The peak wavelength and full-width at half-maximum are  $\sim 1650$  nm and  $\sim 120$  nm, respectively. The peak emission cross-section is about  $(3 \pm 1) \times 10^{-20} \text{ cm}^2$ , which is mostly attributed to high refractive index of the host ( $\sim 2.5$ ) at near-infrared wavelengths. The Lorentzian local field correction factor represented in terms of refractive index should be incorporated when an electric dipole induced light emission of a rare-earth ion embedded in a dielectric medium is considered. Thus, based on Judd-Ofelt formalism, emission cross-section of a certain manifold is proportional to refractive index of the dielectric medium. Measured lifetimes of the fluorescing level are shown in Fig. 6. Up to 0.05 mol% ( $\sim 1.7 \times 10^{19} \text{ ions/cm}^3$ ) of  $\text{Pr}^{3+}$  the

Table 1. Results of Judd-Ofelt analysis performed on a  $\text{Pr}^{3+}$ -doped  $\text{Ge}_{30}\text{-As}_8\text{-Ga}_2\text{-Se}_{60}$  glass.

transition	wavelength ( $\mu\text{m}$ )	radiative transition probability* ( $\text{s}^{-1}$ )	branching ratio (%)	calculated lifetime (ms)
$^3\text{H}_5 \rightarrow ^3\text{H}_4$	4.9	43	100	23.0
$^3\text{H}_6 \rightarrow ^3\text{H}_5$	4.5	52	48	9.1
$^3\text{H}_6 \rightarrow ^3\text{H}_4$	2.4	57	52	
$^3\text{F}_2 \rightarrow ^3\text{H}_6$	1.4	1	$\sim 0$	0.447
$^3\text{F}_2 \rightarrow ^3\text{H}_5$	3.4	360	16	
$^3\text{F}_2 \rightarrow ^3\text{H}_4$	2.0	1873	84	
$^3\text{F}_3 \rightarrow ^3\text{F}_2$	7.2	3	$\sim 0$	0.272
$^3\text{F}_3 \rightarrow ^3\text{H}_6$	4.8	104	3	
$^3\text{F}_3 \rightarrow ^3\text{H}_5$	2.3	743	20	
$^3\text{F}_3 \rightarrow ^3\text{H}_4$	1.6	2836	77	
$^3\text{F}_4 \rightarrow ^3\text{F}_3$	2.2	$\sim 0$	$\sim 0$	0.500
$^3\text{F}_4 \rightarrow ^3\text{F}_2$	5.4	3	$\sim 0$	
$^3\text{F}_4 \rightarrow ^3\text{H}_6$	3.9	251	13	
$^3\text{F}_4 \rightarrow ^3\text{H}_5$	2.1	871	43	
$^3\text{F}_4 \rightarrow ^3\text{H}_4$	1.5	880	44	
$^1\text{G}_4 \rightarrow ^3\text{F}_4$	3.4	85	4	0.450
$^1\text{G}_4 \rightarrow ^3\text{F}_3$	3.0	9	$\sim 0$	
$^1\text{G}_4 \rightarrow ^3\text{F}_2$	2.0	31	1	
$^1\text{G}_4 \rightarrow ^3\text{H}_6$	1.8	751	34	
$^1\text{G}_4 \rightarrow ^3\text{H}_5$	1.3	1216	55	
$^1\text{G}_4 \rightarrow ^3\text{H}_4$	1.0	127	6	

\* Magnetic dipole transition is not considered, and error involved in the calculations is  $\sim 15\%$ .

upper level lifetime is nearly constant to be  $\sim 212 \pm 10 \mu\text{s}$ , whereas the lifetime decreases as the concentration further increases. The concentration quenching mechanisms acting on the  $(^3\text{F}_3, ^3\text{F}_4)$  manifolds are probably cross-relaxation processes corresponding to either  $(^3\text{F}_3, ^3\text{F}_4) : ^3\text{H}_5 \rightarrow (^3\text{F}_2, ^3\text{H}_6) : (^3\text{F}_2, ^3\text{H}_6)$  or  $(^3\text{F}_3, ^3\text{F}_4) : ^3\text{H}_4 \rightarrow (^3\text{F}_2, ^3\text{H}_6) : ^3\text{H}_5$  [20]. Therefore, the  $\text{Pr}^{3+}$  concentration should be kept lower than 0.05 mol% to minimize the concentration quenching.

### 3. Emission from Codoped Glasses

We added  $\text{Er}^{3+}$  into the  $\text{Pr}^{3+}$ -doped glasses as a sensitizer to enhance the excitation efficiency of the 1480 nm pumping. Figure 5(b) shows a typical emission spectrum of  $\text{Pr}^{3+}/\text{Er}^{3+}$ -codoped glasses. The emission intensity increases with the introduction of  $\text{Er}^{3+}$ . To confirm energy transfer from  $\text{Er}^{3+}$  to  $\text{Pr}^{3+}$ , duration times of the codoped samples were measured at 1650

nm for the  $(^3\text{F}_3, ^3\text{F}_4)$  level and at 1535 nm for the  $\text{Er}^{3+} : ^4\text{I}_{13/2}$  level. Figure 7 shows the measured duration times of the two manifolds under pumping at 1480 nm. Lifetime of the  $\text{Er}^{3+} : ^4\text{I}_{13/2}$  level decreases as the  $\text{Er}^{3+}$  concentration increases in the codoped samples, while the apparent duration time of the  $\text{Pr}^{3+} : (^3\text{F}_3, ^3\text{F}_4)$  level increases. Lifetimes of the  $^4\text{I}_{13/2}$  level in  $\text{Er}^{3+}$ -single-doped and  $\text{Pr}^{3+}/\text{Er}^{3+}$ -codoped glasses where  $\text{Er}^{3+}$  concentration is 0.1 mol% are  $\sim 3.26$  ms and  $\sim 0.75$  ms, respectively. This confirms that a fast energy transfer between the  $\text{Er}^{3+} : ^4\text{I}_{13/2}$  and  $\text{Pr}^{3+} : (^3\text{F}_3, ^3\text{F}_4)$  levels takes place as illustrated in Fig. 3. On the other hand, the lifetime increase of the  $(^3\text{F}_3, ^3\text{F}_4)$  level is dedicated to the prolonged population feeding from the  $\text{Er}^{3+} : ^4\text{I}_{13/2}$  even after stop of the excitation. The maximum  $\text{Er}^{3+}$  concentration can be determined in consideration of magnitude of the  $^4\text{I}_{15/2} \rightarrow ^4\text{I}_{13/2}$  ground state absorption at the 1.6  $\mu\text{m}$  band and RE solubility of the host.

A direct excitation to the upper Stark levels of the  $(^3\text{F}_3, ^3\text{F}_4)$

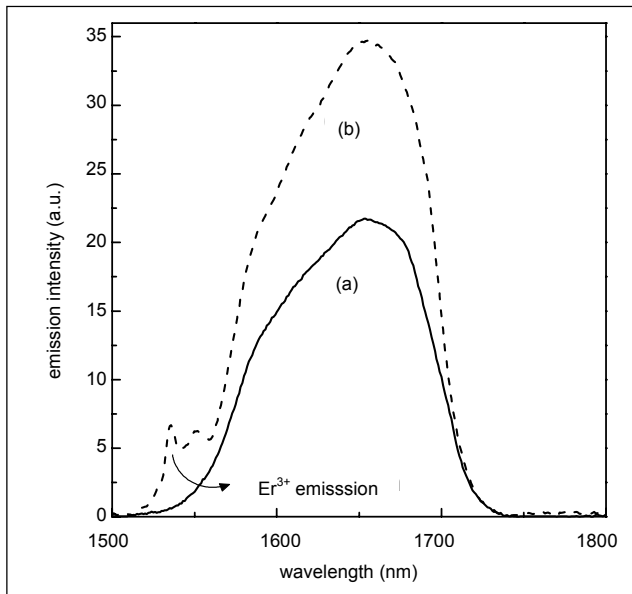


Fig. 5. Emission spectra of 0.05 mol%  $\text{Pr}^{3+}$ -doped glasses, where codoped  $\text{Er}^{3+}$  concentration is (a) 0 and (b) 0.2 mol%.

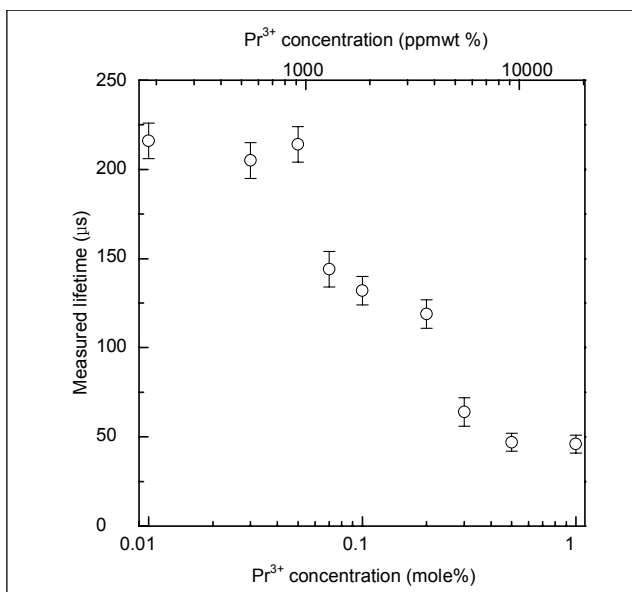


Fig. 6. Measured lifetimes of the ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) manifolds against  $\text{Pr}^{3+}$  concentrations.

multiplet is preferred to make an efficient emission at 1.6  $\mu\text{m}$  because the UV-side absorption of the host starts at around 800nm as shown in Fig. 1. If excited with a 1.5  $\mu\text{m}$  pump source instead of the 1480 nm one,  $\text{Pr}^{3+}/\text{Er}^{3+}$ -codoped glasses may emit the 1.6  $\mu\text{m}$  luminescence with a greater intensity.

## IV. DISCUSSION

### 1. Energy Transfer from $\text{Er}^{3+}$ to $\text{Pr}^{3+}$

According to the Dexter's classical theory on the nonradia-

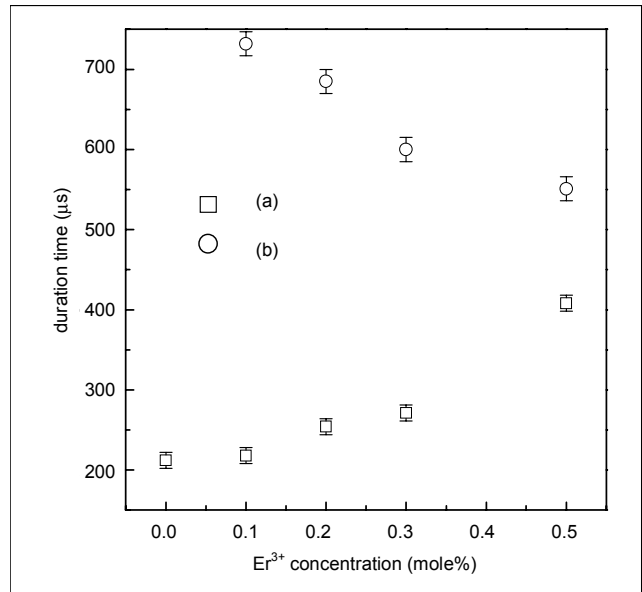


Fig. 7. Duration times of (a)  $\text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) and (b)  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2}$  levels in  $\text{Pr}^{3+}/\text{Er}^{3+}$ -codoped glasses.  $\text{Pr}^{3+}$  concentration is fixed at 0.05 mol%. Note that the first  $e^{-1}$ -folding times in codoped samples are influenced by population feeding from  $\text{Er}^{3+}$  to  $\text{Pr}^{3+}$  occurring even after stop of excitation. Thus, the first  $e^{-1}$ -folding times indicate duration times in the case of codoped samples.

tive energy transfer, overlap of spectral lineshapes between the energy-donor emission and acceptor absorption plays a crucial role in determining the magnitude of electric multipole-induced nonradiative energy transfers [21]. A quantitative analysis involving other parameters such as integrated absorption cross-section of the energy-accepting level and energy migration among donors can be done [19], but it is not considered in this paper. Normalized lineshape spectra of the emission and absorption transitions of  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2} \leftrightarrow ^4\text{I}_{15/2}$  and  $\text{Pr}^{3+}$ :  $^3\text{H}_4 \leftrightarrow (^3\text{F}_3, ^3\text{F}_4)$  are shown in Fig. 8. The overlapped area is much larger in the forward  $\text{Er}^{3+} \rightarrow \text{Pr}^{3+}$  transfer than in the backward case. Thus, it is evident that the forward energy transfer, i.e.,  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2} \rightarrow \text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ), would be much faster than the backward  $\text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ )  $\rightarrow \text{Er}^{3+}$ :  $^4\text{I}_{13/2}$  transfer. The agile energy transfer of  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2} \rightarrow \text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) evidenced by this static spectroscopic consideration is further confirmed by the data related to the variations of measured lifetimes of the  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2}$  and  $\text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) levels. Lifetime of the  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2}$  level abruptly decreases as shown in Fig. 9 when  $\text{Pr}^{3+}$  ions are introduced. The decay trace of the  $^4\text{I}_{13/2}$  level in a 0.1 mol%  $\text{Er}^{3+}$ -single-doped glass shows a nearly exponential behavior with a  $e^{-1}$  folding lifetime of  $\sim 3.26$  ms, while a nonexponential decay curve appears when 0.05 mol% of  $\text{Pr}^{3+}$  codopes the  $\text{Er}^{3+}$ -single-doped glass, and at the same time the lifetime becomes  $\sim 0.75$  ms. The nonexponentiality in a decay trace usually results from a nonradiative energy transfer [22].

Table 2. Representative phonon energy ( $\hbar\omega$ ), calculated lifetime ( $\tau_0$ ), measured lifetime ( $\tau_m$ ), branching ratio ( $\beta$ ), quantum efficiency ( $\eta$ ), and product of measured lifetime and emission cross-section ( $\sigma_{se}\tau_m$ ) associated with the 1.3  $\mu\text{m}$  and 1.6  $\mu\text{m}$  emissions in some  $\text{Pr}^{3+}$ -doped low phonon energy glasses.

glasses	transition	$\hbar\omega$ ( $\text{cm}^{-1}$ )	$\tau_0$ ( $\mu\text{s}$ )	$\tau_m$ ( $\mu\text{s}$ )	$\beta$ (%)	$\eta$ (%)	$\sigma_{se}\tau_m$ ( $10^{-26} \text{ cm}^2 \text{ s}$ )
fluoride*	$^1\text{G}_4 \rightarrow ^3\text{H}_5$	$\sim 500$	2484	110	60	4	36
sulfide**	$^1\text{G}_4 \rightarrow ^3\text{H}_5$	$\sim 350$	510	300	52	60	250
selenide***	$(^3\text{F}_3, ^3\text{F}_4) \rightarrow ^3\text{H}_4$	$\sim 200$	$\sim 320 \pm 50$	$212 \pm 10$	$\sim 77$	$65 \pm 20$	$636 \pm 200$

\*ZBLAN [24], \*\*Ga-La-S [25], \*\*\*Ge-As-Ga-Se, this work

Also, a rough estimation on the energy transfer rate ( $W_{et}$ ) can be calculated using the following [23]:

$$W_{et} = 1 - \frac{\tau_c}{\tau_s},$$

where  $\tau_s$  and  $\tau_c$  denote the lifetimes without and with an energy acceptor, respectively. Therefore, the energy transfer rate of  $\text{Er}^{3+}: ^4\text{I}_{13/2} \rightarrow \text{Pr}^{3+}: (^3\text{F}_3, ^3\text{F}_4)$  becomes  $\sim 77\%$ .

## 2. Characteristics of the 1.6 $\mu\text{m}$ Emission

As a suitable host of  $\text{Pr}^{3+}$  ion for the 1.3  $\mu\text{m}$  amplifiers, glasses with a low phonon energy are requested to enhance the fluorescing level lifetime. So far, fluoride and sulfide glasses have been considered as a host of the 1.3  $\mu\text{m}$  amplifiers. As stated in Introduction section, the 1.6  $\mu\text{m}$  emission is virtually forbidden even in a sulfide glass, so that some characteristic parameters associated with the  $\text{Pr}^{3+}: ^1\text{G}_4 \rightarrow ^3\text{H}_5$  transition in fluoride and sulfide glasses are compared with those of the  $(^3\text{F}_3, ^3\text{F}_4) \rightarrow ^3\text{H}_4$  transition in the current selenide glass in Table 2. Intrinsic radiative lifetime of the  $(^3\text{F}_3, ^3\text{F}_4)$  level was estimated in consideration of the population distribution. Product of measured lifetime ( $\tau_m$ ) and emission cross-section ( $\sigma_{se}$ ) is a useful figure of merit from which one can evaluate the efficiency of an optical fiber amplifier [26]. As shown in Table 2, the value of  $\sigma_{se}\tau_m$  for the selenide glass is higher than that for the sulfide glass. It is because, in the selenide glasses, the emission cross-section is much higher than that of the sulfide glasses, even though both of the glasses feature similar fluorescing level lifetimes.

Oscillator strength of the  $\text{Pr}^{3+}: ^3\text{H}_4 \rightarrow ^1\text{G}_4$  transition is much weaker than that of the  $^3\text{H}_4 \rightarrow (^3\text{F}_3, ^3\text{F}_4)$ . Moreover, the 1.3  $\mu\text{m}$ -fluorescing  $^1\text{G}_4$  manifold is located at about  $\sim 1020 \text{ nm}$ , where no commercial high-power laser diodes are currently available. However, strong and stable laser diodes operating at  $1480 \text{ nm}$  have been well developed and can be effectively used as pump sources for the 1.6  $\mu\text{m}$  emission.

## 3. Possibility of a Low Loss Selenide Optical Fiber

The minimum transmission loss wavelength in chalcogenide glasses is usually determined by the intersection between the weak absorption tail and the multiphonon absorption tail in case the extrinsic vibrational impurity absorption does not exist [15]. Due to the reduced multiphonon absorption in such heavy glass fibers, the minimum loss wavelength is shifted to longer wavelength-side than that of the conventional oxide glass fibers. As a result, the magnitude of the weak absorption tail is almost responsible for the clarity of a selenide optical fiber at the 1.6  $\mu\text{m}$  band. The weak absorption tail is caused by the additional band gap states, which are formed from electronic impurities such as transition metal elements or defects such as valence alternation pairs and dangling bonds [27]. In a highly pure and homogeneous glass, it is assumed that the band gap states caused by the defects is formed at the glass transition temperature during the melt quenching process [15]. Chalcogenide glasses are vulnerable to the formation of the defects in their electronic band gap [28]. Mainly because of the weak absorption, one can hardly anticipate a selenide optical fiber with a transmission loss comparable to that of the oxide glass fibers. However, the propagation loss of  $\sim 2 \text{ dB/m}$  at the 1.6  $\mu\text{m}$  band can be ascertained in the selenide glass fibers [29]. Fortunately, the additional extrinsic absorptions associated with the molecular vibrational species such as  $\text{OH}^-$  and  $\text{SeH}^-$  are not located at the 1.6  $\mu\text{m}$  band if their concentration in the glasses is kept below a readily achievable limit [30].

The current host glass exhibits a superior thermal stability. Its glass transition and onset of crystallization temperatures are  $\sim 355^\circ\text{C}$  and  $\sim 510^\circ\text{C}$ , respectively. Fiber drawing from the glass rods was successful [31]. About 5 m-long fiber with a diameter of  $\sim 120 \mu\text{m}$  was coated with a UV-curable polymer. Optical loss of  $\sim 9 \text{ dB/m}$  was measured from the drawn fiber at  $1313 \text{ nm}$  with an  $\text{Nd}^{3+}$ : YLF laser by the conventional cut back method. Thus, elaboration on the materials purification is further needed. Drawing of a core/clad structured fiber is being tried. We anticipate that a mono-mode selenide optical fiber

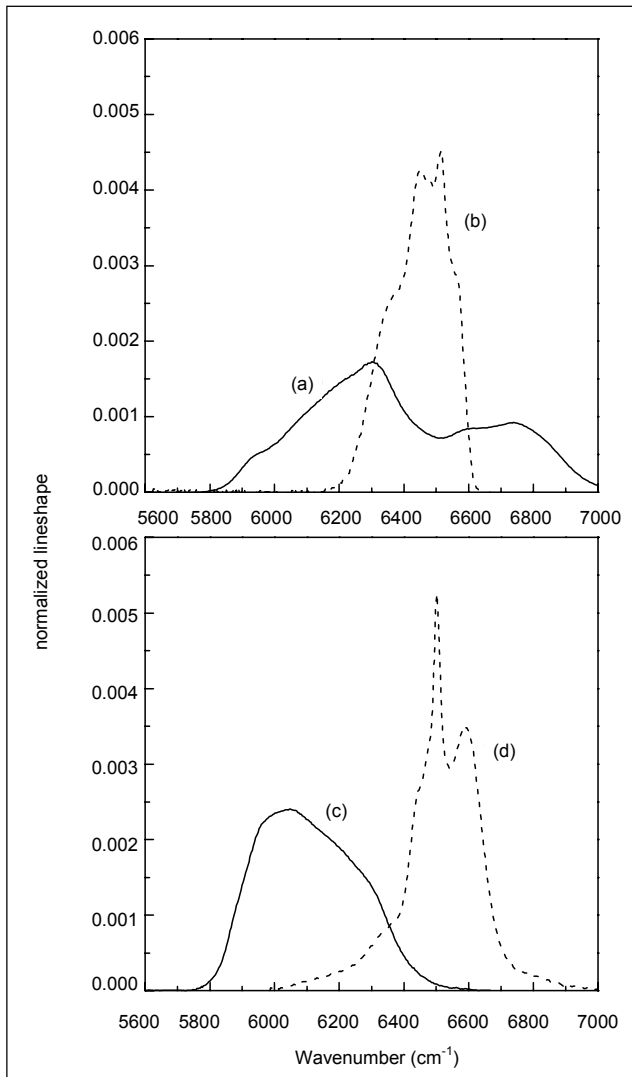


Fig. 8. Lineshape spectra of (a) absorption of  $\text{Pr}^{3+}$ , (b) emission of  $\text{Er}^{3+}$ , (c) emission of  $\text{Pr}^{3+}$  and (d) absorption of  $\text{Er}^{3+}$  in selenide glass. Note that the area under each spectrum is identical to each other.

with optical loss less than  $\sim 5$  dB/m at the  $1.6 \mu\text{m}$  band is quite promising mainly because of the good thermal stability of the host glasses used in this study. A numerical simulation regarding the  $\text{Pr}^{3+}$ -doped selenide fiber amplifiers indicates that a favorable optical gain may be achievable even in a 50 cm-long fiber, though the background transmission loss of the fiber is assumed to be 5 dB/m [31].

## V. CONCLUSION

The measured peak wavelength and full-width at half-maximum of the  $1.6 \mu\text{m}$  emission from  $\text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ )  $\rightarrow$   $^3\text{H}_4$  transition in  $\text{Pr}^{3+}$ - and  $\text{Pr}^{3+}/\text{Er}^{3+}$ -doped selenide glasses are  $\sim 1650$  nm and  $\sim 120$  nm, respectively. A moderate lifetime of

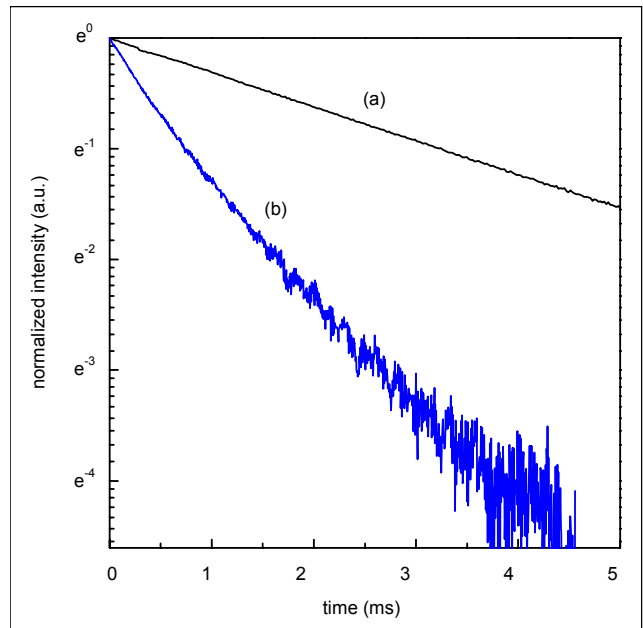


Fig. 9. Decay traces of 0.1 mol%  $\text{Er}^{3+}$ -doped selenide glasses of which  $\text{Pr}^{3+}$  concentration is (a) 0 and (b) 0.05 mol%.

the thermally coupled upper manifolds ( $\sim 212 \pm 10 \mu\text{s}$ ) together with a high stimulated emission cross-section of  $\sim (3 \pm 1) \times 10^{-20} \text{ cm}^2$  promises to be useful for  $1.6 \mu\text{m}$  band fiber-optic amplifiers that can be pumped with existing 1480 nm high-power laser diodes. Codoping of  $\text{Er}^{3+}$  into a  $\text{Pr}^{3+}$ -doped low phonon energy glass significantly enhanced the emission intensity by way of a nonradiative  $\text{Er}^{3+}$ :  $^4\text{I}_{13/2} \rightarrow \text{Pr}^{3+}$ : ( $^3\text{F}_3$ ,  $^3\text{F}_4$ ) energy transfer.

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